



Webinar 5. Teaching strategies for AO2/3

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Carrotgate



A student investigated the effect of sugar solutions on pieces of carrot. ..

Concentration	% change in mass
0.0	+24
0.2	+12
0.4 etc	+1 etc

Suggest why the student calculated the % change in mass

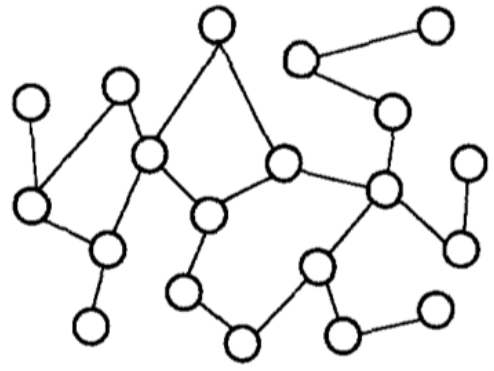
Why we created Blueprint

The examiners reported that, overall, A01 (knowledge) questions were answered well, including answers to simple factual questions and questions which required basic calculations such as percentage change. The problems were with in-depth knowledge: answering A02 questions (apply knowledge) and A03 questions (analyse, interpret, evaluate, draw conclusions). [2018]

Students found the concept of speciation difficult to apply when describing natural selection in the context of orchid colour.

Students struggled to apply their knowledge of what happens during the digestive process to the visking tubing experiment.

- Candidates who did less well on this paper generally did the following:
- Found it difficult to apply what they had learnt to unfamiliar situations.



AO2/3 means expert knowledge

Force predicts motion

Newton's 1st law

Newton's 2nd law

Resistive forces

Inertia

Vectors

Balanced & unbalanced

Acceleration

3 takeaways

1. Scaffold: teach cognitive strategies
2. Hug: use increasingly complex problems
3. Bridge: learn the deep structure

Scaffold

Teach cognitive strategies

How would you solve these?

- 1 The radio waves have a frequency of 4.8×10^9 Hz

Wave speed of electromagnetic waves = 3.0×10^8 m/s

Calculate the wavelength of the radio waves.

Give your answer to 2 significant figures.
- 2 Plants need nitrate ions in order to make proteins.

A plant is growing in soil flooded with water.

Explain why the plant cannot absorb enough nitrate ions.

Domain-specific strategy

1

The radio waves have a frequency of 4.8×10^9 Hz

Wave speed of electromagnetic waves = 3.0×10^8 m/s

Calculate the wavelength of the radio waves.

Give your answer to 2 significant figures.

1. Identify need for procedure: wave equation
2. Use maths skill to rearrange equation
3. Substitute values
4. Calculate answer

Domain-general strategy

2

Plants need nitrate ions in order to make proteins.

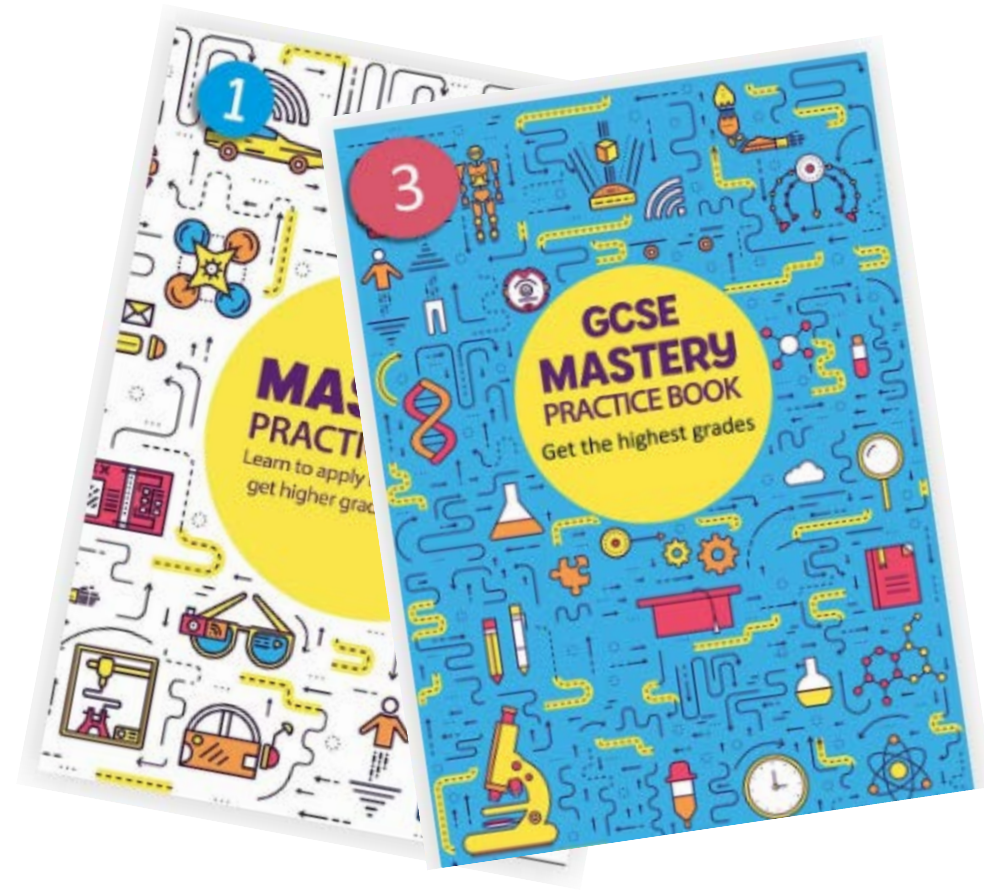
A plant is growing in soil flooded with water.

Explain why the plant cannot absorb enough nitrate ions.

- Detect that the question is about active transport
- Recall everything you know about this concept
- Solve the problem using this information

Why teach cognitive strategies?

- Many students don't learn them spontaneously
- Scaffolding complex task
- Support metacognition
- Research says they work



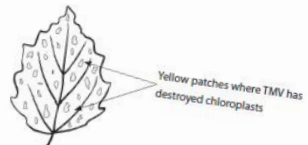
The worked example

- Domain-specific strategy
- Model expert thinking

Example

7.1 Functions of cell parts

1 The diagram shows a leaf infected with the tobacco mosaic virus (TMV).



Yellow patches where TMV has destroyed chloroplasts

Explain how destroying chloroplasts will affect the growth of the plant.

Detect
I need to think about the function of chloroplasts.

Recall
chloroplasts

1. Chloroplasts are green structures inside plant cells.
2. Their function is to carry out photosynthesis.
3. Photosynthesis is a process plants use to make food for themselves.
4. Growth is a life process. Organisms need food to grow.

Solve

IF TMV destroys some of the chloroplasts,
THEN there will be fewer chloroplasts.

IF chloroplasts carry out photosynthesis
THEN fewer chloroplasts means less photosynthesis.

SO the plant will produce less food.
SO the plant will grow less.

OVERALL destroying chloroplasts will reduce the growth of the plant.

How is this cell part useful to the organism?

How does this cell part affect growth?

The worked example

Example

2.3 Bulb Brightness

1 In the circuit 1, the ammeter A1 reads 0.3A.
In circuit 2 a second bulb is added.

What do the ammeters in circuit 2 read?
A $> 0.3\text{ A}$ B 0.3 A C $< 0.3\text{ A}$

Detect
I need to understand how an extra bulb changes the current.

Recall
1. Current is a flow of electrons. The flow and the current are the same everywhere in a loop.
2. Each bulb has resistance and reduces the current.
3. The wire in a bulb gets hot and lights up. So the bulb transfers energy from the circuit, as heat and light.

Solve
As current is the same all around a series circuit, each ammeter reads the same. Adding a bulb increases the resistance. Increasing the resistance reduces the current. So the current with two bulbs will be less than 0.3A. The answer is C - less than 0.3A.

What is the current in A2 in the first circuit?

Why doesn't it matter where the ammeter and bulbs are?

- Proven to benefit novices
- Reduces cognitive load

Teaching cognitive strategies

7.3 Types of cell transport

Plants need minerals A,B and C for healthy growth. The table shows the concentration of these minerals in plant root cells and in the surrounding soil in normal conditions.

Mineral	Concentration in plant root cells in mol/dm ³	Concentration in surrounding soil in mol/dm ³
A	0.7	0.5
B	0.2	0.4
C	0.8	1.0

Heavy rain can leave the soil waterlogged, which decreases oxygen reaching the root cells.

For which mineral will the movement into the cells be reduced?
Explain your choice.

**Detect**

2. Find the key information

Plants need minerals A,B and C for healthy growth. The table shows the concentration of these minerals in plant root cells and in the surrounding soil in normal conditions.

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Normal conditions



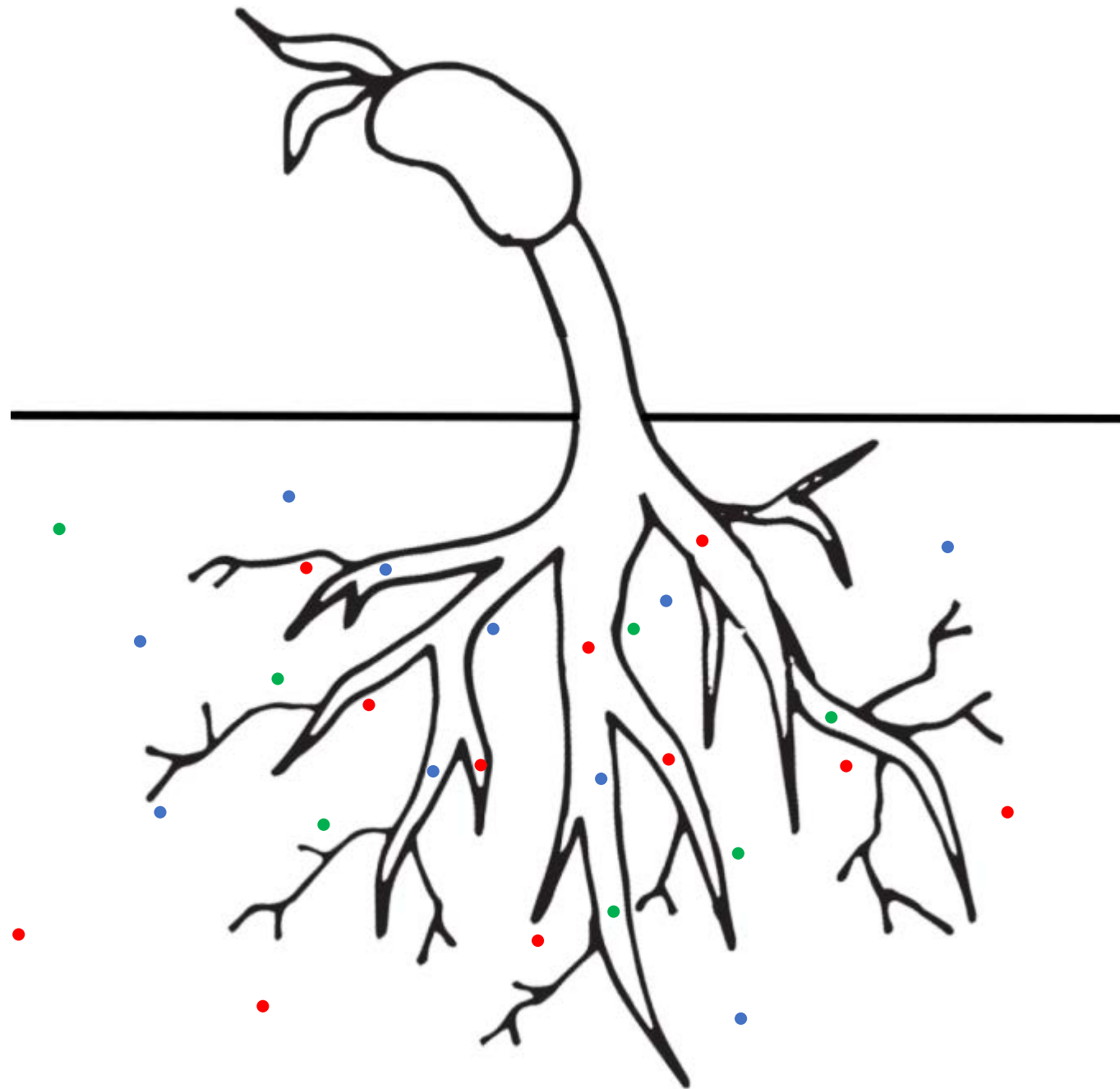
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Heavy rain can leave the soil waterlogged, which decreases oxygen reaching the root cells.

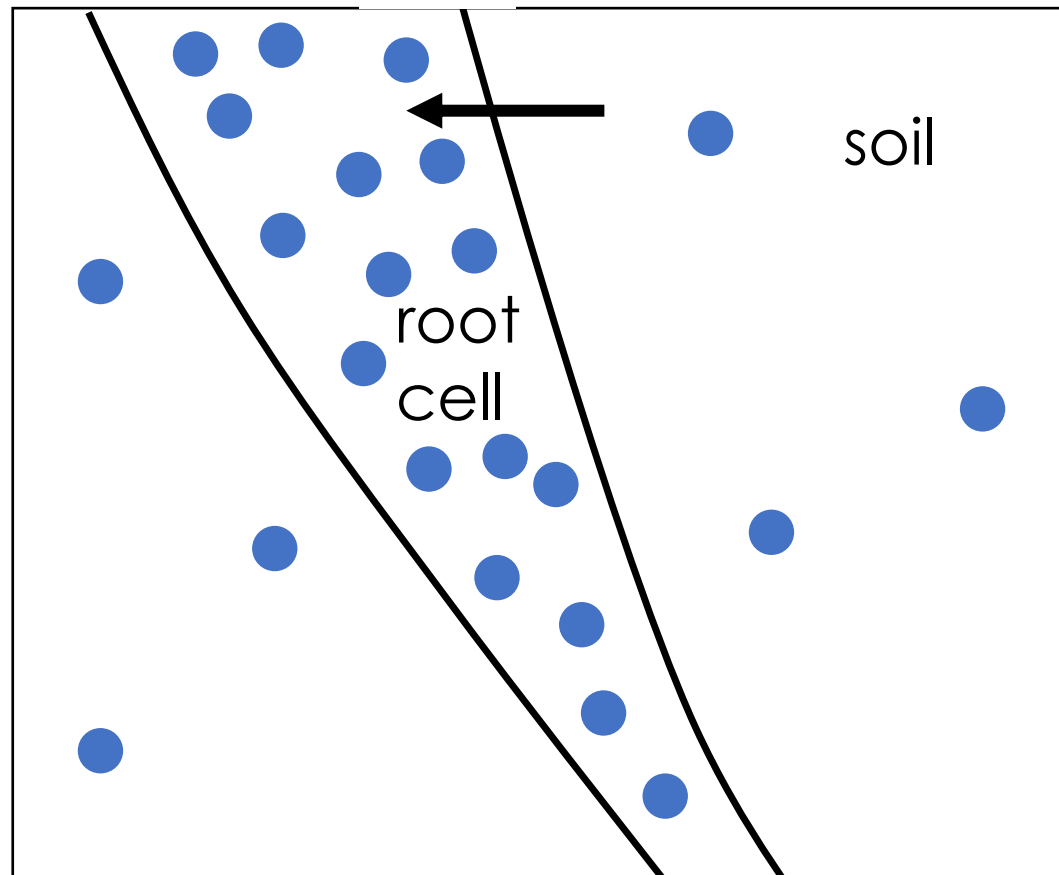
For which mineral will the movement into the cells be reduced?

Explain your choice.

- mineral A
- mineral B
- mineral C



Mineral	Concentration in plant root cells in mol/dm ³	Concentration in surrounding soil in mol/dm ³
A	0.7	0.5



**Detect**

3. Identify the Unknown

Plants need minerals A,B and C for healthy growth. The table shows the concentration of these minerals in plant root cells and in the surrounding soil in normal conditions.

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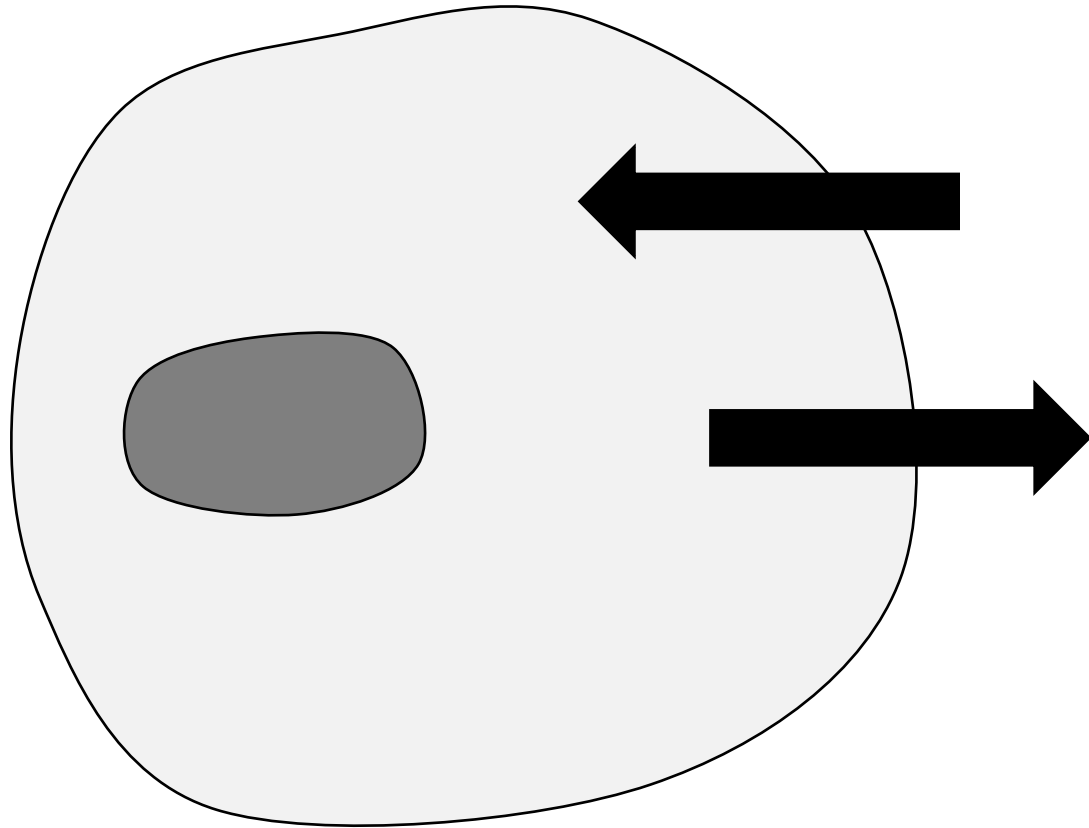
Different
conditions

Heavy rain can leave the soil waterlogged, which decreases oxygen reaching the root cells.

For which mineral will the movement into the cells be reduced?
Explain your choice.

**Detect**

4. Decide the key concept

**Cell transport:**

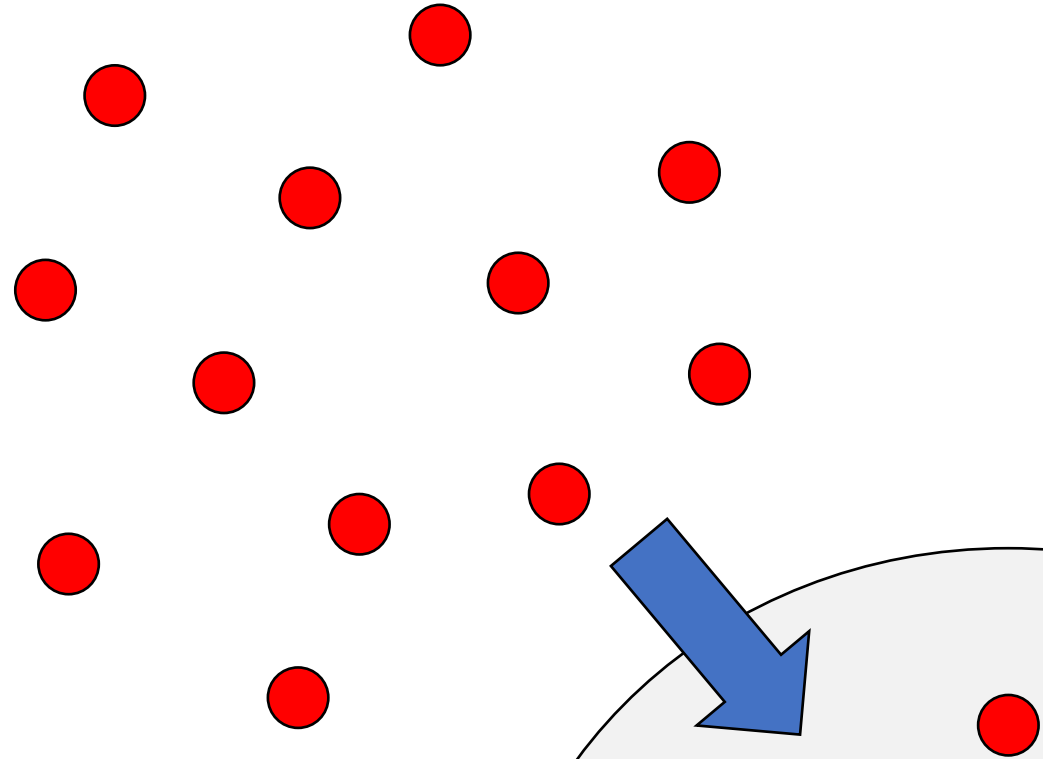
Diffusion

Active transport



Recall

DOWN gradient

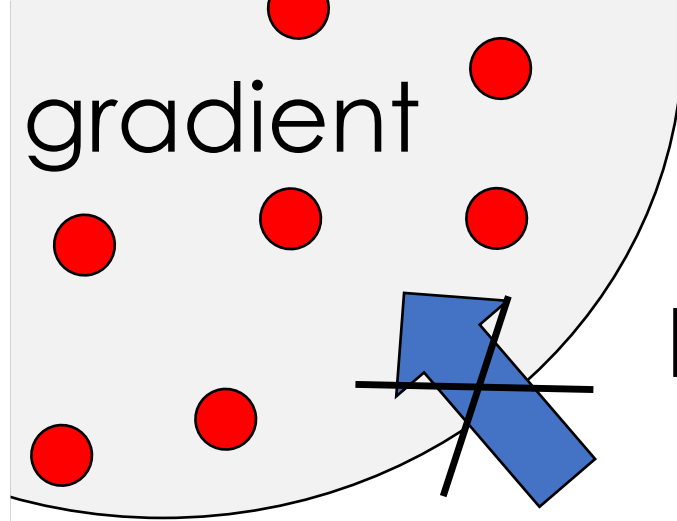


Diffusion



Recall

UP gradient

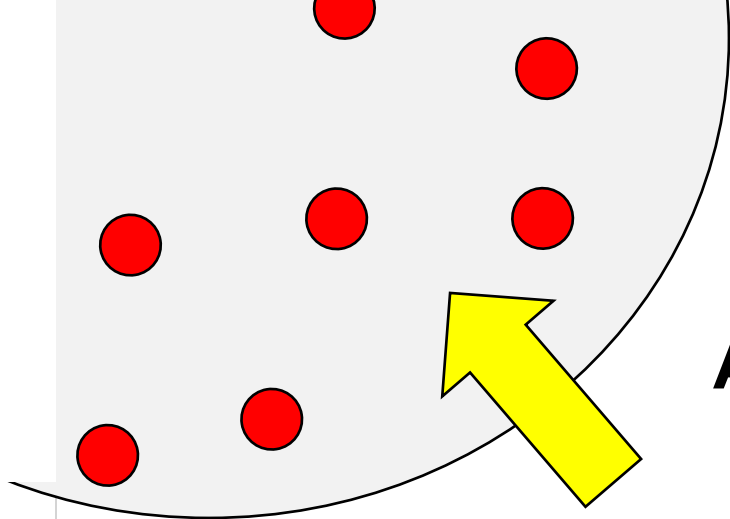


No diffusion

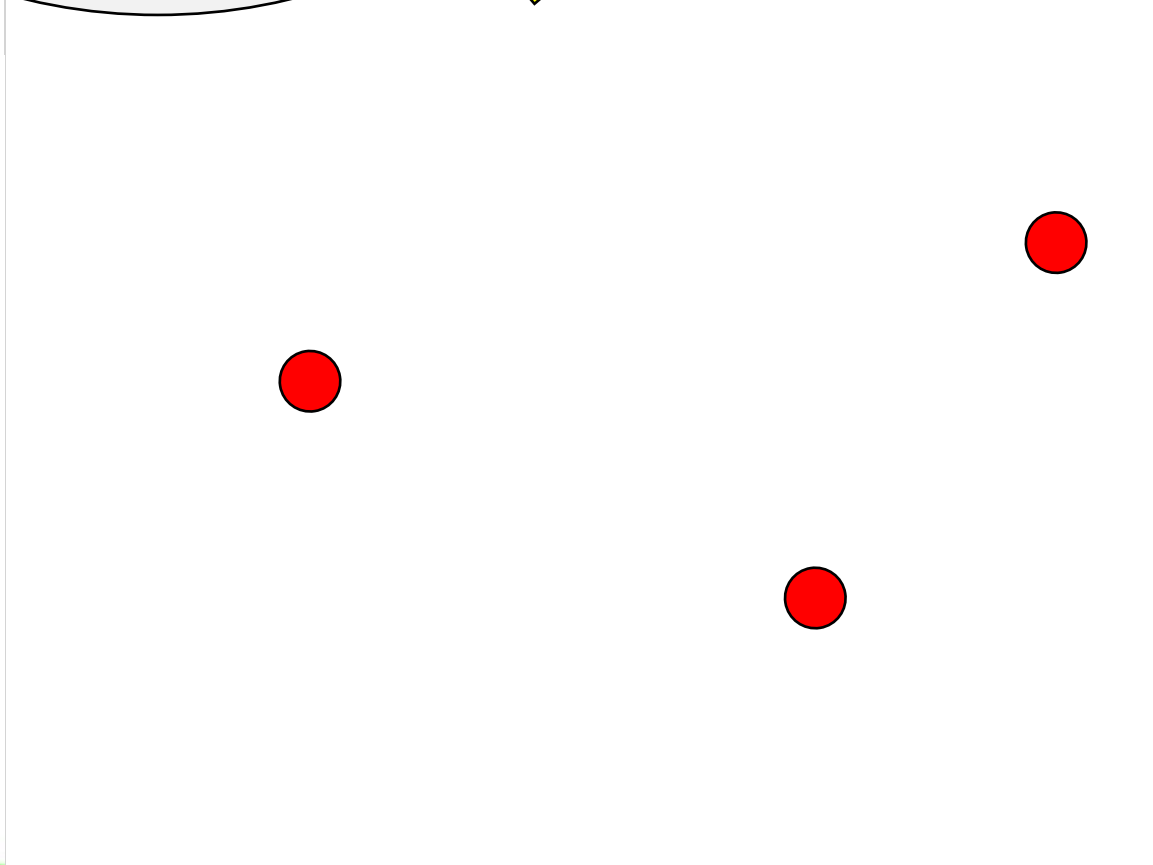
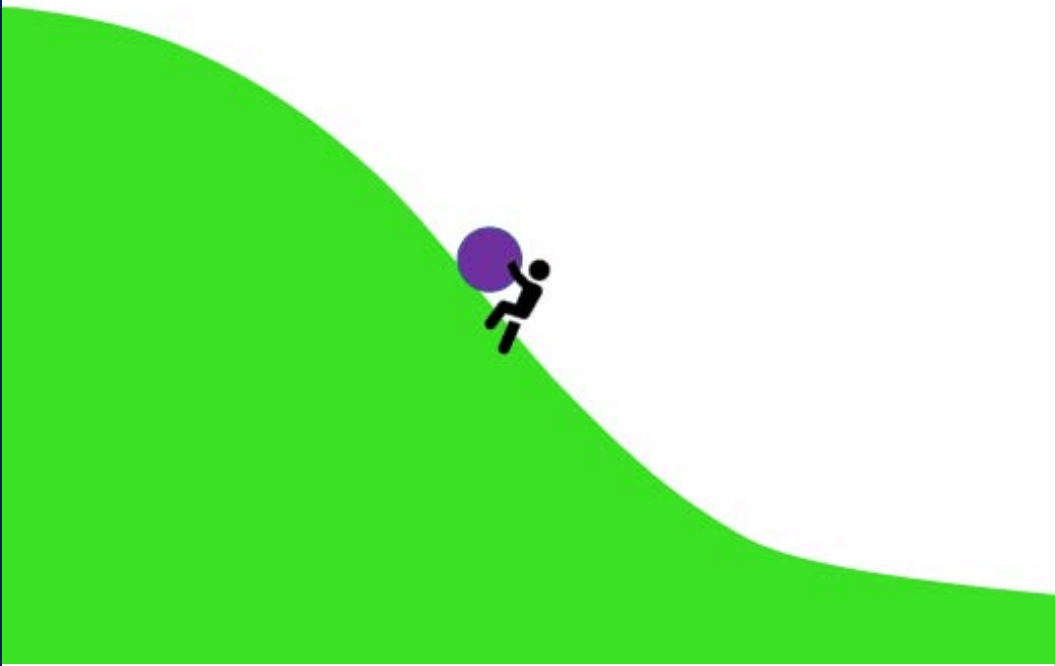




Recall



Active transport



**Solve**

How to structure an explanation

- ✔ **Claim** A simple statement that answers the question
- ✔ **Evidence** Data or observations that supports the claim.
- ✔ **Reasoning** Scientific ideas that show how the evidence supports the claim

active transport →

Mineral	Concentration in plant root cells in mol/dm ³	Concentration in surrounding soil in mol/dm ³
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Solve

- ✓ Claim
- ✓ Evidence
- ✓ Reasoning

Mineral A: its movement will be affected by a reduction in oxygen.

active transport →

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Solve

- ✓ Claim
- ✓ Evidence
- ✓ Reasoning

Mineral A: its movement will be affected by a reduction in oxygen.

The concentration of A inside the root cells is higher than the soil.



Solve

Active transport →

Mineral	Concentration in plant root cells in mol/dm ³	Concentration in surrounding soil in mol/dm ³
A	0.7	0.5
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Solve

- ✓ Claim
- ✓ Evidence
- ✓ Reasoning

Mineral A: its movement will be affected by a reduction in oxygen.

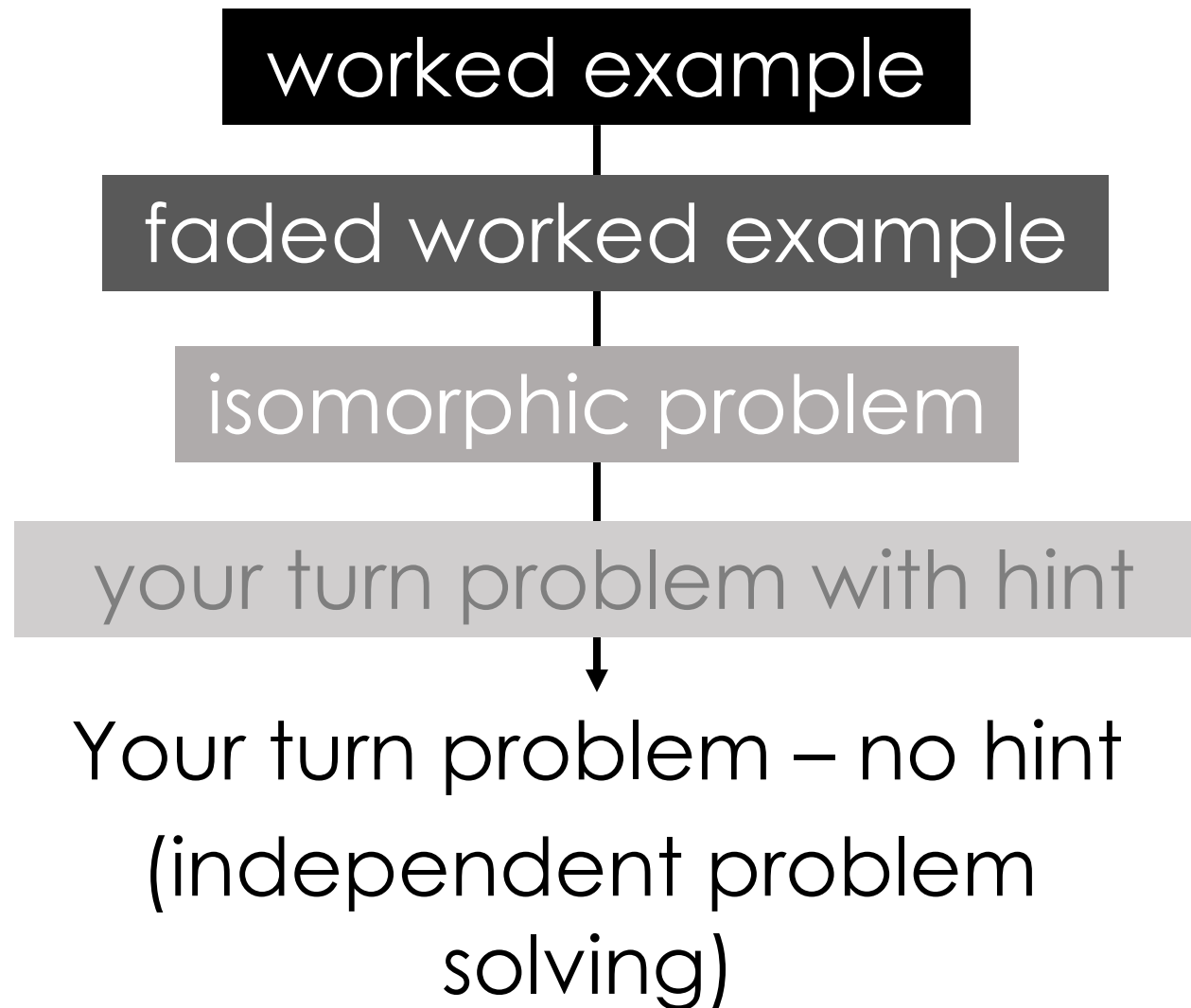
The concentration of A inside the root cells is higher than the soil.

So it cannot move in by diffusion, but must move by active transport.

Active transport require energy which comes from respiration. Aerobic respiration provides the most energy but requires oxygen.

If the cell receives less oxygen, it will start switch to anaerobic respiration, which produces less energy. So the active transport of A is reduced.

Scaffold from example to independence



worked example

faded worked example

isomorphic problem

your turn problem with hint

Your turn problem – no hint

Mineral	Concentration in mmol/dm^3	
	Outside cells	Inside cells
Calcium	120	5
Chloride	3	28
Potassium	6	135
Sodium	139	11

Scientists studied the movement of minerals that normally flow into human cells.

They measured the initial concentrations of ions outside and inside the cells.

Then they added cyanide, a poison that stops respiration.

Explain the effect of stopping respiration on the movement of each mineral.

worked example

faded worked example

isomorphic problem

your turn problem with hint

Your turn problem – no hint



Detect

✓ Values

The table shows the concentration of four different minerals outside and inside cells. The minerals normally flow into cells. The concentration is higher outside cells for calcium and sodium. It is higher inside cells for chloride and potassium.

✓ Unknown

I need to explain the effect of stopping respiration on the movement of each mineral.

✓ Concept

The concept I need to use is cell transport.

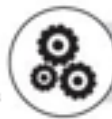
worked example

faded worked example

isomorphic problem

your turn problem

Your turn problem



Recall

1. How substances move in/out of cells by diffusion.

2. How substances move in/out of cells by active transport.

3. How substances get their energy from respiration.

worked example

faded worked example

isomorphic problem

your turn problem with hint

Your turn



Solve

- ✓ Claim The movement of _____ and _____ into cells will be affected by stopping respiration. The movement of _____ and _____ will not be affected.
- ✓ Evidence Before respiration was stopped, the concentration of chloride and potassium were _____ in the inside of the cells compared to the outside. The concentration of calcium and sodium were _____ inside the cells compared to the outside.
- ✓ Reasoning Chloride and potassium move into cells by _____. This requires _____, which is released by respiration. Calcium and sodium move into cells by _____, which does not require _____.

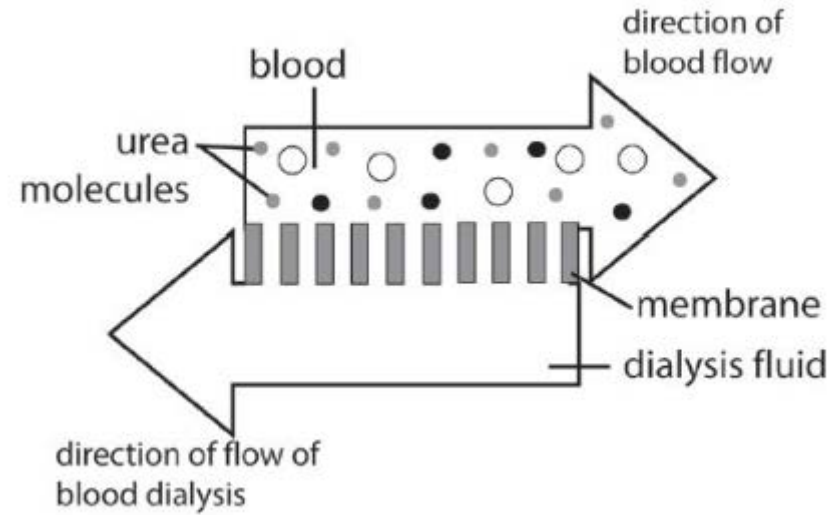
worked example

faded worked example

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Your turn problem – no hint



In kidney disease urea builds up in the blood. The diagram shows how a treatment works.

Explain how a moving stream of dialysis fluid can help the patient.

Hint: Urea only diffuses if there is a concentration difference.

Hug

Use increasingly
complex problems

Goal of education : transfer



From osmosis in potatoes



To a variety of contexts



When skills and knowledge learned in one situation help you learn or perform better on something else.

Low-road transfer



- Similar conditions e.g. right -> left hand drive
- Fairly automatic and unconscious
- Transfer happens easily

‘Hugging’ the eventual goal



Small-side games



Are the best preparation for big games

Stay as close as possible to the eventual performance: start with small, simple version and add complexity.

Put hugging into practice

7KC-Heat & temperature

Temperature difference

Temperature difference: When two objects are at different temperatures, energy moves from the warmer to the cooler until they are at the same temperature. The bigger the difference, the faster energy moves.

Blueprint performances

Simpler

Acquire

Goal To determine what variables affects the rate at which a substance cools, using a graph.

Apply

Goal To determine how a variable affects the rate of temperature change from a graph.

Analyse

Goal To draw conclusions from graphs about the affect of several variables on rate of temperature change.

More
complex



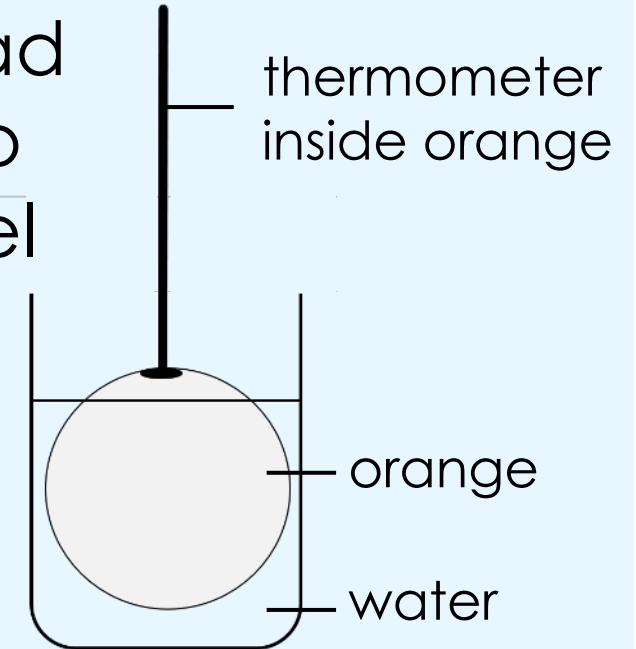
“Now for the real puzzle. To find the time of death, you need to work out how long the dead body took to cool to 29.5°C . Let's investigate!”



What variables could affect how quickly an object cools down?



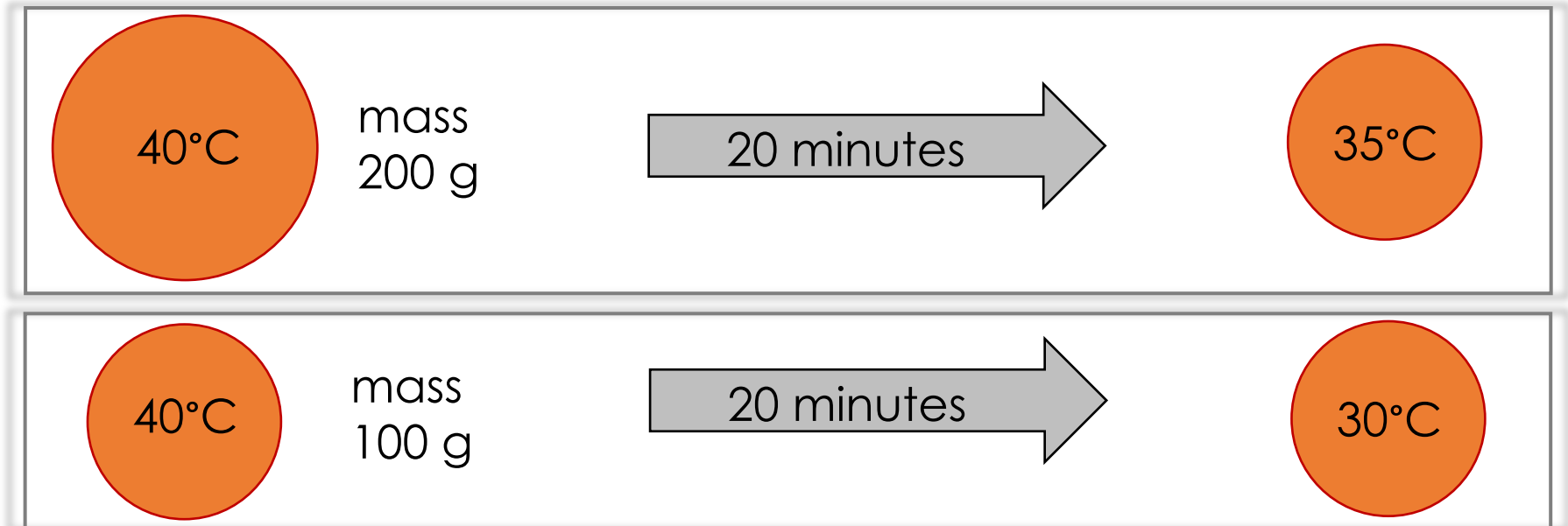
“ I don't have dead bodies for you to test! Use a model – an orange,”



Activity: What affects the rate of cooling?

- Work in pairs to investigate one independent variable.
- Put the orange in the water and measure the temperature of the orange every 2 mins for 20 mins.
- Record your results
- Draw a line graph and write a conclusion.

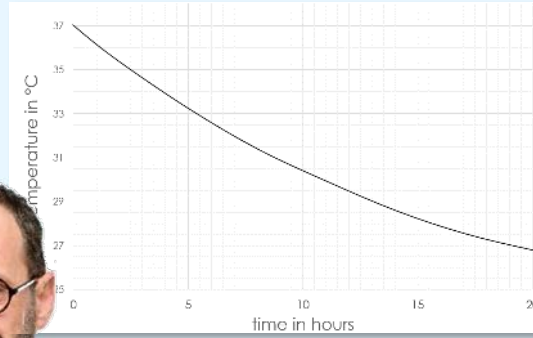
Mass affects rate of cooling



The smaller the mass, the greater the temperature drop.



How does mass affect how much heat is transferred?



“Lee's lawyer used a cooling curve to estimate the time of death as 7pm. That means Lee wasn't home to kill her...but is this estimate correct?”

Activity: Study the evidence

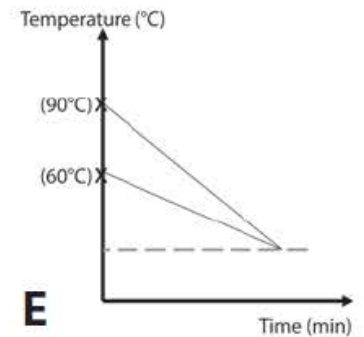
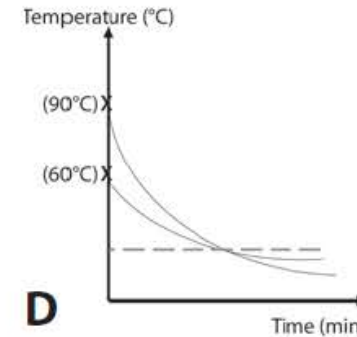
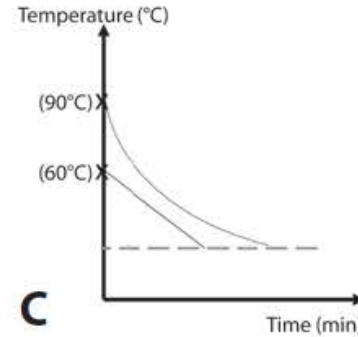
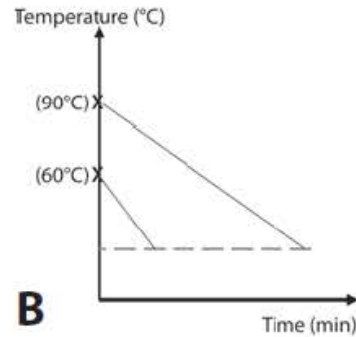
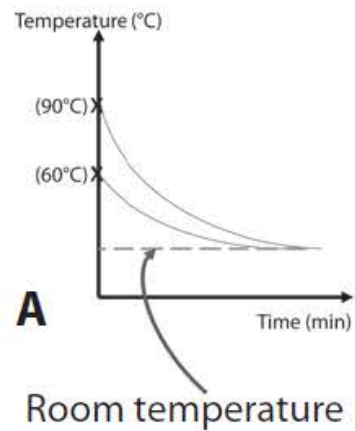
SS4

- Study the graph Lee's lawyer used. How did he use it to conclude that Clare was killed at 7pm?
- Use what you have learnt about cooling to decide if she could have been killed after 10pm.

3.5 Temperature Graphs

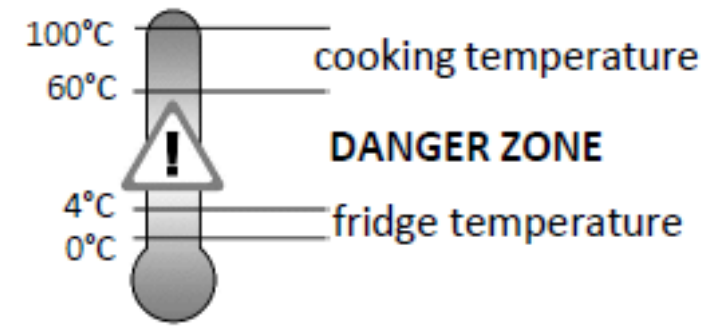


- 1** Two beakers of hot water are left to cool.
Which set of graphs shows how the temperatures change with time?



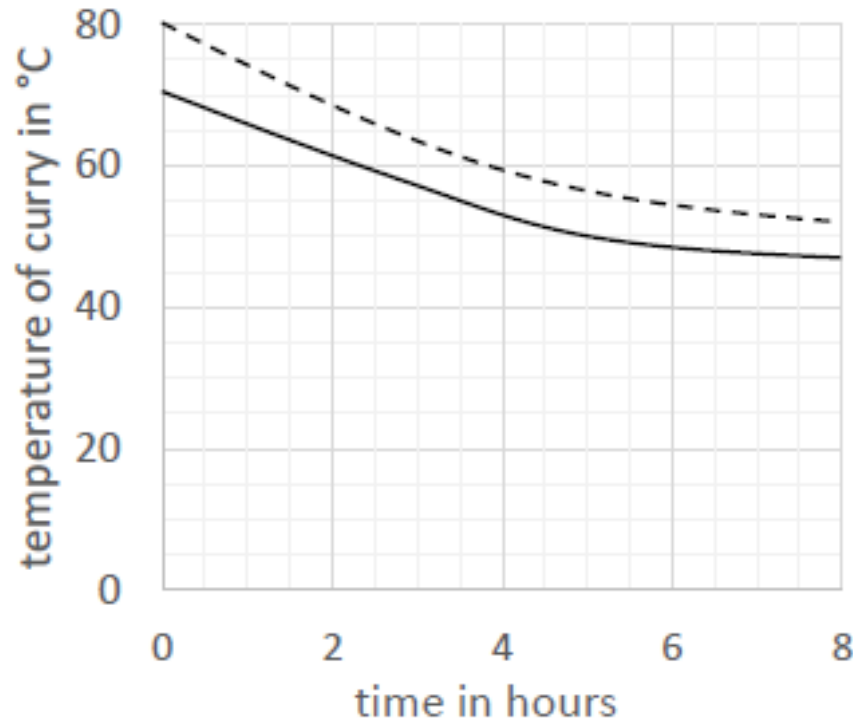
Analyse

The danger zone! When food is between the temperature of 4 and 60°C bacteria start to multiply rapidly. Leave the food for too long in this danger zone and it will not be safe to eat.

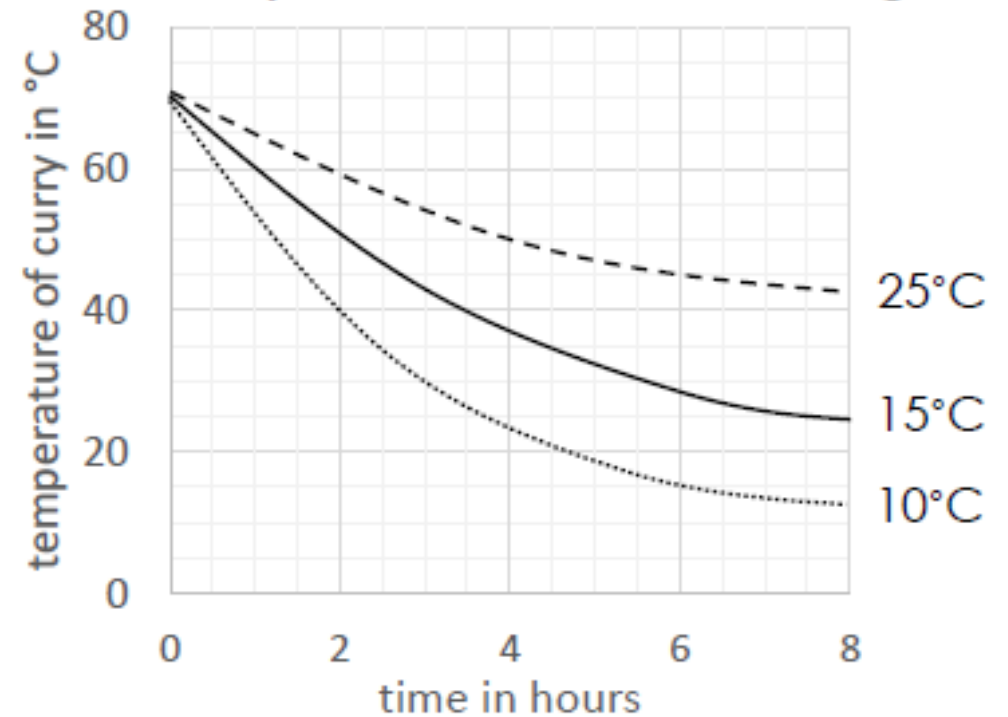


Luca wants to work out how to avoid his curry being too long in the danger zone. He investigated 4 different factors and plotted the results below

1. Initial temperature of the curry



2. Temperature of the surroundings



Bridge

Learn the deep
structure

High-road transfer



- Different conditions e.g. chess to politics
- Transfer not easy - takes effort
- Requires search for patterns and principles

'Bridging' between situations

- Build an abstract idea from concrete experience
- Use bodily and visual analogies – how people think
- Help students 'bridge' (generalise) across contexts

Beyond tell and practice 1

Practicing Versus Inventing With Contrasting Cases: The Effects of Telling First on Learning and Transfer

Daniel L. Schwartz, Catherine C. Chase, Marily A. Oppezzo, and Doris B. Chin
Stanford University

Being told procedures and concepts before problem solving can inadvertently undermine the learning of deep structures in physics. If students do not learn the underlying structure of physical phenomena, they will exhibit poor transfer. Two studies on teaching physics to adolescents compared the effects of “telling” students before and after problem solving. In Experiment 1 ($N = 128$), students in a tell-and-practice condition were told the relevant concepts and formulas (e.g., density) before practicing on a set of contrasting cases for each lesson. Students in an invent-with-contrasting-cases (ICC) condition had to invent formulas using the same cases and were told only afterward. Both groups exhibited equal proficiency at using the formulas on word problems. However, ICC students better learned the ratio structure of the physical phenomena and transferred more frequently to semantically unrelated topics that also had a ratio structure (e.g., spring constant). Experiment 2 ($N = 120$) clarified the sources of the effects while showing that ICC benefited both low- and high-achieving students.

Example: density, a difficult concept



$$\text{Density} = \frac{\text{mass (g)}}{\text{volume (cm}^3\text{)}}$$

- It's a derived quantity (mass / volume)
- It involves proportional thinking
- Students equate density with weight

Concrete analogy

**EXPLORE**

Why are some materials heavier than others?

Let's use an analogy: crowded clowns



Companies send clowns to parties in buses.

Unfortunately some clowns are unhappy because their companies make the buses too crowded. Your job is to calculate the crowdedness of each company's bus.

Activity: Calculate crowdedness

SS2

- Invent a way of comparing each company's crowdedness.

Contrasting cases

Complete mastery

activate 1

acquire

apply

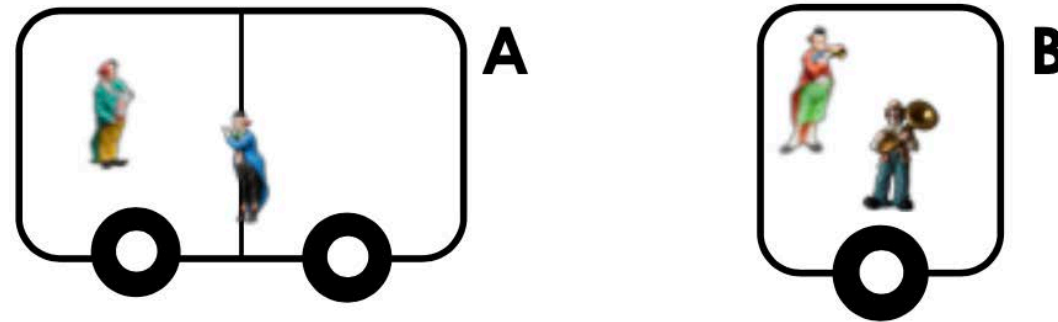
assess

analyse

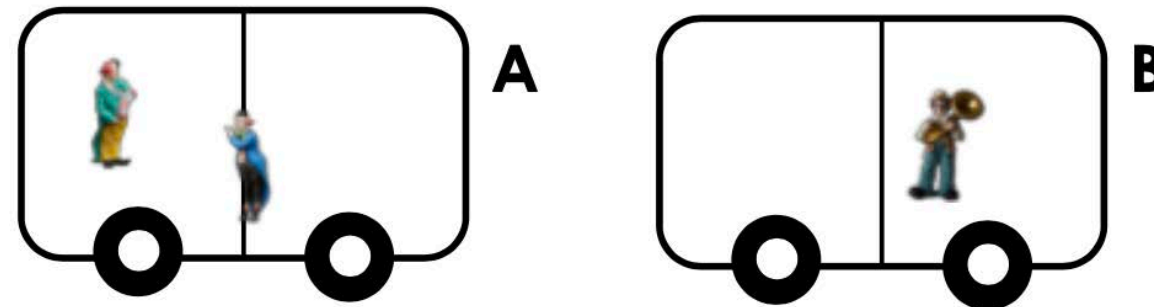
test

teaching materials

1. Which bus is more crowded: A, B or the same?
Why?



2. Which bus is more crowded: A, B or the same?
Why?



Invention

5. Now use your rule. Calculate a number for the crowdedness of each bus.

SS2

Complete mastery

activate 1

acquire

apply

assess

analyse

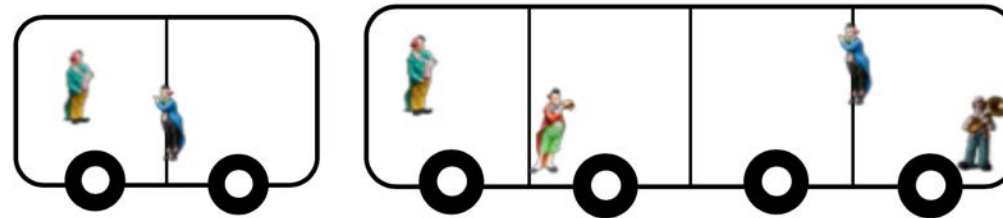
test

teaching materials

Clowns for U bus Crowdedness = _____



Rent-a-clown bus Crowdedness = _____



Fun clowns bus Crowdedness = _____



Beyond tell and practice 2

Standard T&P instruction is important because it delivers the explanations and solutions invented by experts, and students need opportunities to hear and practice these ideas. For students to gain this benefit without undermining transfer, the current studies suggest that expositions should happen after students have explored novel deep structures. The inventing activity can serve as “preparation for future learning” by readying students to appreciate more fully the expert solutions and deep structures when they are explained (Bransford & Schwartz, 1999). Giving students the end-product of expertise too soon short-cuts the need to find the deep structure that the expertise describes. Students in the T&P condition focused on what they had been told, and they applied the formulas sequentially to the problems, which reduced their chances of finding the deep structure. Without an appreciation of deep structure, students are less likely to see the structure in new situations that differ on the surface, and they will fail to transfer.

Do you need 'Question Sets'?

DENSITY

... what you learned in maths about ratios

transfer

Example

3 1 What is the ratio of grey to white squares?

There are 3 grey squares to 1 white square.
Or, 3:1

$$\frac{3}{1}$$

Your turn

1. What is the ratio of grey to white squares?

DENSITY

... how mass affects density (if volume is fixed)

notice

Example

2 4 What is the ratio of squares to white squares?

The ratio is 2:4. You can make the numbers smaller by dividing both sides by 2.
The simplified ratio is 1:2.

DENSITY

... the same idea in different situations

Example

Here are the ingredients for 1 mug of hot chocolate. What do you need for 4 mugs?

4 mugs needs 4 x the amount of ingredients

The ratio milk: chocolate = 200 ml : 4 squares

So the amounts are 800 ml 16 squares

DENSITY

... how you can identify a substance from its mass: volume ratio

Example

volume: 2 cm³ mass: 10 g (Cube A)
volume: 2 cm³ mass: 20 g (Cube B)

1. Which is heavier for its size? Give a reason.
2. Which cube has a higher density?

Your turn

1. The density of silver is 10.5 g/cm³.
Volume = 2 cm³ Mass = 21 g (Cube A)
Volume = 4 cm³ Mass = 44 g (Cube B)

a) Which cube (A or B) made of silver?
b) Give a reason.

Your turn

1. The density of aluminium is 2.7 g/cm³.
Volume = 3.5 cm³ Mass = 10.7 g (Cube A)
Volume = 1.2 cm³ Mass = 3.24 g (Cube B)

a) Which cube (A or B) made of pure aluminium?
b) Give a reason.

For the following questions use the list

Substance	Density in g/cm ³
Copper	8.9
Gold	19.3
Zinc	7.1
Water	1.0
Stainless steel	7.5 - 8.0

2. A block has a mass of 58.5 g and a volume of 2 m³.
3. A metal bar has a volume of 20 000 cm³.
4. Oona's ring has a volume of 0.5 cm³.
5. A beaker has a mass of 200 g. 55 g of water is added. The total mass increases to 258 g. Is the liquid pure water?

1. Cube A has a volume of 10 cm³ and a mass of 1 kg. Cube B has a volume of 20 cm³ and a mass of 1 kg. Which cube is more dense? Give a reason.

2. Cube A has a volume of 5 cm³ and a mass of 10 g. Cube B has a volume of 4 cm³ and a mass of 10 g. Which cube has a lower density?

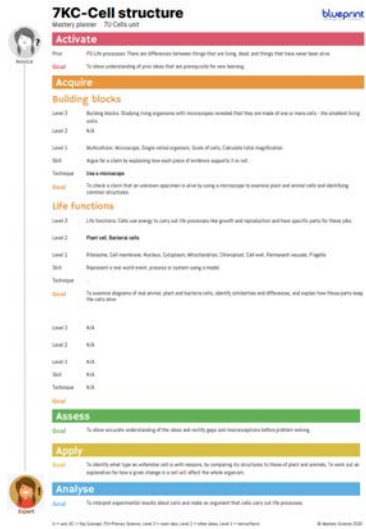
3. Whitney fills balloons with 10 g of two different gases. She then measures the volume of the balloons.

Balloon A has a volume of 15 cm³.
Balloon B has a volume of 9 cm³.
Which gas has the lowest density?

4. Here are two different liquids. Each has a mass of 50 g.

Liquid A has volume of 48 cm³
Liquid B has a volume of 30 cm³
Which liquid has the higher density?

Next steps



Blueprint planners

Year 7/8 now

Year 9-11 coming soon



Complete Mastery

Year 7/8 course on offer until 31st July

Further webinars?