

Determination of Trace Elements in Lead for Battery Applications using Atomic Absorption Analysis

Sample and Standard Preparation procedures for Trace Elements in Lead

[1] *Preparation of Lead (Pb) Samples:*
Drill Pb metal to make small pieces. Weigh 2.0 grams of these pieces on a balance. Place the sample in a 400ml Pyrex breaker and add 10ml H₂O, 2.5 grams of tartaric acid and 7ml HNO₃. Warm on a hot-plate until the sample dissolves. Dilute to 100ml in a volumetric flask with water (H₂O). This prepares a 2% sample solution, with a dilution factor of 50.

[2] *Preparation of Sulfuric Acid (H₂SO₄) Samples:*
Using a graduated cylinder, place 75ml of water (H₂O) into a 100ml volumetric flask. Slowly add sulfuric acid (H₂SO₄) to the water, swirling to mix. **Solution will bubble up and get hot, do NOT spill it on yourself – wear gloves!** Add H₂SO₄ to bring the final volume to 100ml. This prepares a 25% sample solution, with a dilution factor of 4.

[3] *Preparation of Pure / Waste Water (H₂O) Samples:*
Using a digital pipettor or glass pipet, add 1ml nitric acid (HNO₃) to 100ml volumetric flask. Add water sample to 100ml mark. There is basically no dilution and the energy relates directly to the sample.

[4] *Preparation of Multi-element Standards for Calibration:*

Using a pipet or pipettor, add 100ml of each of the 1000ppm stock concentrate standards (Buck standards) to a 1 Liter volumetric flask containing 250ml water (H₂O) and 50ml nitric acid (HNO₃). The following groupings will prepare stable 100ppm Stock Standards . (Add water to bring final volume to 1 Liter):

- A) Bi, Ni, Ag, Zn, Cu, Cd
- B) Fe, Sb, Sn, As, Al, Ca

Dilute the 100ppm Stock Standards into the following Working Standards:

25ml	Stock + 1ml HNO ₃ to 100ml in Volumetric Flask =	25ppm
10ml	“ “	10ppm
5ml	“ “	5ppm
2ml	“ “	2ppm
1ml	“ “	1ppm
0.5ml	“ “	0.5ppm

Use these standards for Cu, Ag, Fe, Ca:
0, 0.5, 2, 5ppm.

Use these standards for Ni, Zn, Cd:
0, 0.5, 1, 2ppm.

Use these standards for Bi, Sb, Sn, As, Al, Pb:
0, 5, 10, 25ppm.

Determination of Trace Element in Lead

Samples:	Lead Sample #19, Lead Sample #21, Sulfuric Acid (~98%)
Sample Prep:	2% solutions of lead in 5% HNO ₃ / 2.5% tartaric acid; 10% solution of sulfuric acid (1:10 dilution)
Calibration:	0.5 and 2.5 µg / ml (ppm) analyte metal standard in 2% high-purity lead matrix, 2% high-purity lead matrix blank; for lead samples. 1.0 µg / ml (ppm) analyte metal standard in pure (distilled / deionized) water, pure water blank; for sulfuric acid sample.
Instrument:	Buck 210VGP Atomic Absorption Spectrophotometer, Giant Pulse and In-Line D ₂ Correction, and Model 420 Hydride Generation system.
Conditions:	Standard operating conditions for 210 unit; analytical parameters and correction modes as listed per element; air / acetylene flame for Ni, Ag, Zn, Cu, Fe, Cd; nitrous oxide / acetylene for Al, Ca; argon hydrogen for As, Sb, Sn, Bi.
Results:	Values are weight percent (% w/w) in original sample: Data based on 1:50 Pb dilution and 1:10 H ₂ SO ₄ dilution: D.L. [detectability] based on 2-sigma statistics for Pb samples.

Element	Wavelength	D.L.	Lead # 19	Lead # 21	H ₂ SO ₄
Ni	232nm	0.0015%	<0.0015%	0.0018%	<0.0015%
Ag	328	0.0003	0.0017	0.0018	0.0005
Zn	213	0.0003	0.0009	0.0004	0.0027
Cu	324	0.0005	0.0150	0.0093	0.0021
Bi	223	0.0008	0.0167	0.0184	<0.0008
Fe	248	0.001	<0.001	0.002	0.005
Sb	217	0.0005	<0.0005	0.0012	<0.0005
Sn	224	0.0007	<0.0007	0.0009	<0.0007
As	193	0.0001	<0.0001	0.0011	<0.0001
Cd	228	0.0005	0.0006	0.0009	0.0008
Al	309	0.005	<0.005	<0.005	0.006
Ca	422	0.0005	0.0006	0.0022	0.0472

These data show the powerful flexibility and stability of the Buck 210VGP system for the wide-ranging requirements of the manufacturing industry. The overall high sensitivity of the various trace metals supports the interference-free quality of the data. The combination of unique components provides an unmatched system in performance and economy.



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