

Safety Data Sheet

FC27-12

2-(HEPTAFLUOROPROPOXY)TETRAFLUOROPROPIONYL FLUORIDE

Revised 16-November-2016

1. IDENTIFICATION OF SUBSTANCE

Product Identifier

Product name 2-(Heptafluoropropoxy)tetrafluoropropionyl fluoride

Catalog numberFC27-12BrandFluoryx LabsCAS No.2062-98-8

Relevant identified uses of the Substance or mixture and uses advised against

Identified uses: Laboratory chemicals; manufacture of substances

Details of the supplier of the safety data sheet

Fluoryx Labs

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USA

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2. HAZARDS IDENTIFICATION

Emergency Overview

OSHA Classification Corrosive, irritant, moisture sensitive

GHS Classification Corrosive, moisture sensitive, respiratory irritant

H314; Skin Corr. 1B, R34/34/35

H335; STOT SE 3a

Label Elements

Signal Word:

Pictogram:

Danger



Hazard Statements

H314 H335

Precautionary Phrases

P280

P301 + P330 + P331

Causes severe skin burns and eye damage.

May cause respiratory irritation.

Wear protective gloves/protective clothing/eye

protection/face protection.

IF SWALLOWED: rinse mouth. Do NOT induce

vomiting.

P303 + P361 + P353 IF ON SKIN (or hair): Remove/Take off

immediately all contaminated clothing. Rinse skin

with water/shower.

P304 + P340 IF INHALED: Remove victim to fresh air and keep

at rest in a position comfortable for breathing.

P305 + P351 + P338 IF IN EYES: Rinse cautiously with water for several

minutes. Remove contact lenses, if present and

easy to do. Continue rinsing.

P501 Dispose of contents/container to local regulations.

Potential Health Effects

Eye Contact: Corrosive, contact causes severe eye burns which may result in permanent tissue and

corneal damage.

Skin Contact: Corrosive causes skin burns and severe irritation.

Inhalation: Corrosive. Vapor may be strongly irritating to the respiratory tract.

Ingestion: Risk of internal burns if ingested.

Target Organs: Eyes. Skin. Respiratory system. G.I. Tract.

3. COMPOSITION AND INFORMATION ON INGREDIENTS

Synonyms 2,3,3,3-tetrafluoro-2-(heptafluoropropoxy)propionyl fluoride;

Propanoyl fluoride, 2,3,3,3-tetrafluoro-2-(1,1,2,2,3,3,3-heptafluoropropoxy)-;

Perfluoro(2-methyl-3-oxahexanoyl) fluoride;

HFPO dimer:

Hexafluoropropene oxide dimer;

Heptafluoropropoxytetrafluoropropionylfluoride; Undecafluoro-(2-methyl-3-oxahexanoyl) fluoride; 2-(Perfluoropropoxy)perfluoropropanoyl fluoride; Tetrafluoro-2-(heptafluoro-1-propoxy)propanoyl fluoride

Chemical formula

 $C_6F_{12}O_2$

CF₃CF₂CF₂OCF(CF₃)C(O)F

Components

Components			
Material	CAS#	EINECS #	TSCA Listed
2-(Heptafluoropropoxy)tetrafluoropropionyl	2062-98-8	218-173-8	Yes
fluoride			

4. FIRST AID MEASURES

Description of First Aid Measures

Ingestion: IF SWALLOWED: rinse mouth. Do NOT induce

vomiting.

Inhalation: IF INHALED: Remove victim to fresh air and keep at

rest in a position comfortable for breathing.

Skin Contact: IF ON SKIN (or hair): Remove/Take off immediately

all contaminated clothing. Rinse skin with

water/shower.

Eye Contact: IF IN EYES: Rinse cautiously with water for several

minutes. Remove contact lenses, if present and

easy to do. Continue rinsing.

Most Important Symptoms and Effects Symptoms may be delayed. Severe burns may

occur.

Indications of Any Immediate Medical Attention

Large exposures can lead to hypocalcemia due to release of hydrogen fluoride upon exposure to moisture. Apply a 2.5% calcium gluconate gel to

exposed skin.

5. FIRE FIGHTING MEASURES

Flammable Properties:

Flash point: None below the boil

Method: Closed cup

Extinguishing Media:

Suitable: Carbon dioxide, dry chemical powder, sand.

Unsuitable: Do not use water.

Fire fighting instructions: Wear self-contained breathing apparatus and

protective clothing to prevent contact with skin,

lungs, and eyes.

Special Hazards: Corrosive. Emits toxic fumes of hydrogen fluoride

(HF) under fire conditions or exposure to water.

6. ACCIDENTAL RELEASE MEASURES

Personal Precautions: Wear respirator, chemical safety goggles, rubber

boots, and heavy rubber gloves. If outside keep bystanders upwind and away from danger point. Mark out the contaminated area with signs and prevent access to unauthorized personnel. Do not attempt to take action without suitable protective clothing - see section 8 of SDS. Turn leaking containers leak-side up to prevent the escape of

liquid.

Environmental PrecautionsDo not discharge to drains or rivers. Do not release

into environment.

Clean-Up Procedures: Clean-up should be dealt with only by qualified

personnel familiar with the specific substance. Absorb into dry earth, sand, or vermiculite. Transfer to a closable, labeled salvage container for disposal

by an appropriate method.

7. HANDLING AND STORAGE

Personal Precautions: Avoid direct contact with the substance. Ensure

there is sufficient ventilation of the area. Do not handle in a confined space. Avoid the formation or spread of mists in the air. Only use in a fume hood.

Storage Conditions: Keep tightly closed. Store under inert atmosphere:

product reacts with water to produce hydrofluoric acid. Protect from atmospheric moisture. Store at

2 to 8 °C.

Suitable Packaging: Must only be kept in original packaging.

8. EXPOSURE CONTROLS AND PERSONAL PROTECTION

Engineering Controls: Safety shower, eye bath, and self-contained

breathing apparatus on hand in case of emergency.

Mechanical exhaust required.

Personal Protective Equipment

Eye/face protection: Chemical safety goggles.

Respirators: Use organic vapor and acid gas respirators with

components tested and approved under appropriate government standards such as NIOSH (US) or CEN

(EU).

Protective clothing: Where there is potential for skin contact, have

available and wear as appropriate impervious

gloves, apron, pants, and jacket.

Hand protection: Compatible chemical-resistant gloves.

General hygiene measures: Wash thoroughly after handling. Wash

contaminated clothing before reuse.

9. PHYSICAL AND CHEMICAL PROPERTIES

Molecular Weight 332.04 g/mol

Appearance and Color Clear colorless liquid

OdorAcrid odor of hydrogen fluorideBoiling Point54 - 56 °C @ 760 mm Hg

Melting PointNot availableVapor PressureNot available

Vapor Density (Air = 1) > 1

Saturated Vapor Conc. Not available

Density 1.61 g/mL @ 25 °C

Flash Point None below the boil (closed cup)

Explosion Limits Not available **Refractive Index** $n_D^{25} = 1.3$

Solubility in Water Reacts with water to produce HF

10. STABILITY AND REACTIVITY

Stability: Stable in sealed containers under dry, inert

atmospheres.

Conditions to Avoid: Heat, flames, sparks, and moisture.

Materials to Avoid: Avoid aluminum, strong oxidizing agents, water,

alcohols, and strong acids.

Hazardous Decomposition Products: In combustion, emits toxic fumes including carbon

monoxide, carbon dioxide, and hydrogen fluoride.

Hazardous Polymerization Reactions: Will not polymerize under normal conditions.

11. TOXICOLOGICAL INFORMATION

Acute Toxicity No data.

Skin corrosion/irritation Strong corrosive effect on skin and mucous

membranes.

Serious eye damage/irritation Strong corrosive effect.

Respiratory or skin sensitization No sensitizing effect known.

Germ Cell mutagenicity
Not known.
Carcinogenicity
Not known.
Reproductivity toxicity
Aspiration hazard
STOT-repeated exposure
STOT-single exposure
Not known.
Not known.
Not known.

To the best of our knowledge, the acute and chronic toxicity of this substance is not fully known. No classification data on carcinogenic properties of this material is available from the EPA, IARC, NTP, OSHA, or ACGIH.

12. ECOLOGICAL INFORMATION

General: Take care to prevent chemicals from entering the

ground, water courses or drainage systems.

Toxicity: Toxic to fish.

Persistance and Degradability: Not known

Bio-Accumulation Potential: Not known

Mobility in Soil: Not known

Other Adverse Effects: Toxic to aquatic organisms including fish,

invertebrates, and algae.

Not known

13. DISPOSAL CONSIDERATIONS

Results of PBT and vPvB Assessment:

Disposal Operations: Carefully hydrolyze material by mixing with basic

water and/or alcohol in a hood. This releases hydrofluoric acid which should be neutralized. Alternately, absorb onto clay or vermiculite and dispose of absorbent material as solid waste.

Offer unused product to Fluoryx.

Follow all chemical pollution control regulations.

14. TRANSPORTATION INFORMATION

Mode DOT/IMDG/IATA/ICAO

UN Number 3265 Class (Subsidiary) 8

Proper Shipping Name Corrosive liquid, acidic, organic, n.o.s.

Hazard Label (Subsidiary) Corrosive liquid

8

Packing Group

Shipping Hazard Label 8. Corrosive liquid

15. SAFETY, HEALTH, AND ENVIRONMENTAL REGULATIONS

Safety, Health and Environmental Regulations: Product is not subject to any additional regulations

or provisions.

Safety Assessment No chemical safety assessment.

Note: The regulatory information given above only indicates the principal regulations specifically applicable to the product described in the safety data sheet. The user's attention is drawn to the possible existence of additional provisions which complete these regulations. Refer to all applicable national, international and local regulations or provisions.

16. OTHER INFORMATION

FC27-12 SDS

HMIS codes:

HTN:

Legal Disclaimer:

Health: 3

Flammability: 1 Reactivity: 2

2918.99.5000

For R&D use only. Not for drug, household, or other uses. The previous information is based upon our current knowledge and experience of our product and is not exhaustive. It applies to the product as defined by the specifications. In case of combinations or mixtures, one must confirm that no new hazards are likely to exist. In any case, the user is not exempt from observing all legal, administrative and regulatory procedures relating to the product, personal hygiene, and integrity of the work environment. Unless noted to the contrary, the technical information applies only to pure product.

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End of SDS