

IMPROVED SEPARATION OF SEMI-VOLATILES

Dan DiFeo — SGE, Incorporated. 2007 Kramer Lane, Austin, Texas 78758, USA.
Peter Dawes, Angus Hibberd — SGE International Pty. Ltd. 7 Argent Place, Ringwood 3134, Australia.

INTRODUCTION

Semi-volatile contaminants in water and soil are some of the most targeted compounds for analysis in environmental laboratories today. Semi-volatile organic compounds are usually found in the soil and ground water as a result of industrial waste processes. Their presence has serious health and environmental implications, which requires monitoring. This class of contaminants constitutes a wide range of compounds with vastly different physical properties making this one of the most challenging analyses to perform and as such, is extremely difficult to analyze in a cost effective turnaround time. The requirement to meet low detection limits for even the highest boiling point contaminants has meant the use of low-bleed or mass-spectrum grade columns are required. The US environmental protection agency recommends that the analysis be carried out on a 5% phenyl column.

ADVANTAGES OF USING SOLGEL-1ms™ & BPX5

This poster shows improved separation of the US EPA 8270 mix on a BPX5-5% phenyl column and a SOLGEL-1ms — 100% polydimethylsiloxane column. The SOLGEL-1ms capillary column used for this analysis is an exceptionally low bleed, high temperature column. The sol-gel material used encapsulates the 100% polydimethylsiloxane into a sol-gel matrix. The sol-gel matrix, which is essentially a synthetic glass, is then chemically bonded to the fused silica surface. This type of bonding surface has certain advantages over conventional 100% polydimethylsiloxane. These advantages include lower bleed at higher temperatures, excellent inertness and increased resistance to degradation.

The BPX5 column is also an extremely inert and low bleed capillary column with a maximum temperature

of 360/370°C. The low bleed characteristics of these two columns is especially beneficial for the late eluting semi-volatiles. The lower bleed levels at elevated temperatures leads to better signal to noise ratios allowing for lower detection limits to be achieved. The high degree of inertness exhibited by these two capillary columns gives excellent peak shape allowing for reproducible quantitation.

The upper temperature limits of the BPX5 and SOLGEL-1ms™ columns allows the user to bake out any high boiling contaminants out of the column after the analysis without damaging the stationary phase. Baking out high boiling contaminants ensures that they do not interfere with retention times or elevate baseline in future analyses and can be incorporated into the GC method resulting in less instrument downtime.

SUMMARY

The SOLGEL-1ms™ and the BPX5 capillary columns are the ideal choice for analysis of the US EPA 8270 semi-volatiles mixture. Excellent separation of the various components in less than 40 minutes is easily achieved. The low bleed levels at high temperatures allow for very low detection limits of even the most difficult analytes.

ACKNOWLEDGMENTS

We would like to thank Mark Ferry from ECS/MDL, USA for supplying the chromatograms and data for this poster.

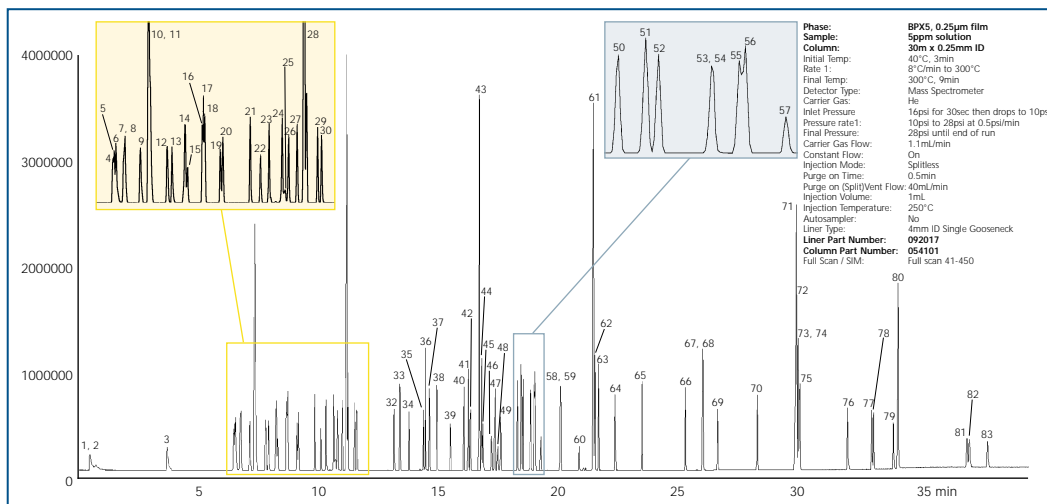


Figure 1. Analysis of US EPA 8270 mix on BPX5.

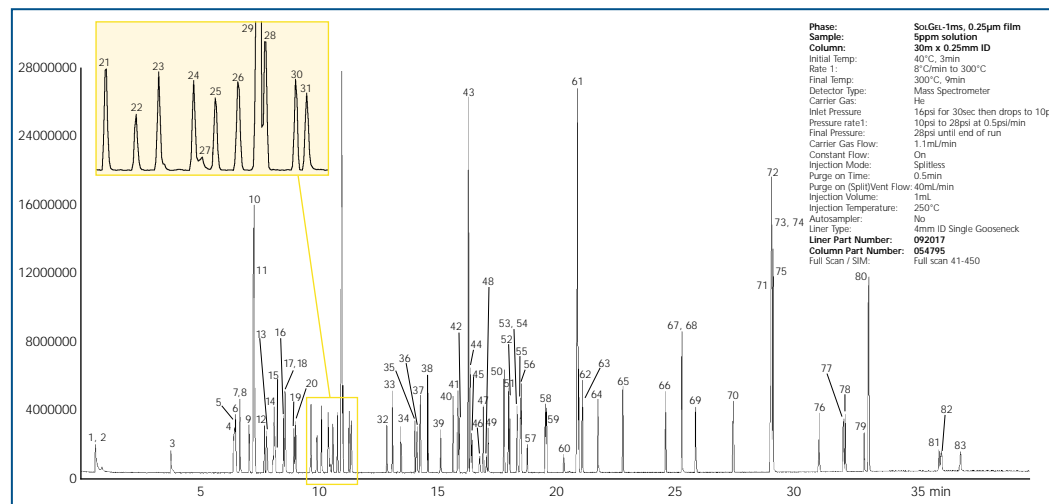


Figure 2. Analysis of US EPA 8270 mix on SOLGEL-1ms.

Components	12. 1,2-Dichlorobenzene	24. bis-(2-Chloroethoxy) methane	36. 2,4,5-Trichlorophenol	48. 4-Nitrophenol	60. Pentachlorophenol	72. Chrysene-d12
1. Pyridine	13. Benzyl alcohol	25. Benzoic acid	37. 2-Fluorobiphenyl	49. 2,4-Dinitrotoluene	61. Phenanthrene-d10	73. Chrysene
2. n-Nitrosodimethylamine	14. 2-Methyl phenol	26. 2,4-Dichlorophenol	38. 2-Chloronaphthalene	50. Diethylphthalate	62. Phenanthrene	74. 3,3-Dichlorobenzidine
3. 2-Fluorophenol	15. bis-(2-chloroisopropyl)ether	27. 1,2,4-Trichlorobenzene	39. 2-Nitroaniline	51. Fluorene	63. Anthracene	75. bis(2-Ethylhexyl)phthalate
4. Phenol-d5	16. n-Nitroso-di-n-propylamine	28. Naphthalene-d8	40. Dimethyl phthalate	52. 4-Chlorophenyl phenyl ether	64. Carbazole	76. Di-n-octyl phthalate
5. Phenol	17. Hexachloroethane	29. Naphthalene	41. Acenaphthylene	53. 2-Methyl-4,6-dinitrophenol	65. Di-n-butyl phthalate	77. Benzofluoranthene
6. Aniline	18. 4-Methylphenol	30. Hexachlorobutadiene	42. 2,6-Dinitrotoluene	54. 4-Nitroaniline	66. Fluoranthene	78. Benzofluoranthene
7. 2-Chlorophenol	19. Nitrobenzene-d5	31. 4-Chloroaniline	43. Acenaphthene-d10	55. n-Nitrosophenylamine	67. Benzo[a]pyrene	79. Benzofluoranthene
8. bis-(2-chloroethyl) ether	20. Nitrobenzene	32. 4-Chloro-3-methylphenol	44. Acenaphthene	56. Azobenzene	68. Pyrene	80. Perylene-d12
9. 1,3-Dichlorobenzene	21. Isophorene	33. 2-Methylnaphthalene	45. 3-Nitroaniline	57. 2,4,6-Tribromophenol	69. n-Terphenyl-d14	81. indeno[1,2,3-cd]perylene
10. 1,4-Dichlorobenzene-d4	22. 2-Nitrophenol	34. Hexachlorocyclopentadiene	46. 2,4-Dinitrophenol	58. 4-Bromophenyl phenyl ether	70. Butyl bromide phthalate	82. Dibenz[a,h]anthracene
11. 1,4-Dichlorobenzene	23. 2,4-Xylenol	35. 2,4,6-Trichlorophenol	47. Dibenzofuran	59. Hexachlorobenzene	71. Benzo[a]anthracene	83. Benzo[g,h,i]perylene



SGE, Incorporated
2007 Kramer Lane, Austin, Texas 78758, USA
Toll Free: (800) 945 6154 Tel: (512) 837 7190 Fax: (512) 836 9159
Email: usa@sge.com Web: www.sge.com