

Red Quiz 1

Lessons 1, 2, 3, 4, Labs 1, 2

Please supply the requested information.

1. Name the English gentleman who was responsible for the three laws of motion that serve as the foundation of classical physics. _____
2. If you were able to let go of a ball in outer space, far away from the influence of gravity, without pushing it in any particular direction, would the ball remain in that exact spot or would it move in one direction or another?
3. If a book were placed in your hand and you could measure the force applied to your hand by the book, could you determine how much force your hand was applying to the book?
4. How many decimeters are in one meter?
5. How many centimeters are in one meter?

Match the three laws of motion with the word or phrase that best describes it:

- | | |
|---------------------|-----------------------|
| 6. _____ First Law | a. equal and opposite |
| 7. _____ Second Law | b. $F = ma$ |
| 8. _____ Third Law | c. inertia |

Match each word or phrase with the definition or description that best fits it:

- | | |
|---------------------------------|---|
| 9. _____ gravity | a. the entire range of gravitational influence |
| 10. _____ force | b. a rate of increase or decrease in the rate of travel |
| 11. _____ velocity | c. a ball, thrown upward, continues upward after it is released |
| 12. _____ acceleration | d. a rate of travel |
| 13. _____ gravitational field | e. the natural force of attraction between any two masses |
| 14. _____ inertia | f. gravitational acceleration constant |
| 15. _____ 9.8 m/sec^2 | g. a natural push or pull |

Circle the best response(s).

16. The farther an object falls under gravity the (greater, lesser) will be its velocity by the end of its travel.
17. Objects falling under the influence of gravity will fall with a constant (acceleration, velocity)
18. Galileo, the Italian student of nature, lived in which century? (1400's, 1600's, 1800's)
19. In its third second of fall, an object falling under gravity will fall at a velocity of (9.8 m/sec, 29.4 m/sec, 941.2 m/sec)
20. A ball thrown upward approaches its (lowest, highest) velocity when it reaches its highest point from the ground.
21. Two cannon balls of different masses, when dropped from the Leaning Tower of *Pizza* will hit the ground (at different times, at the same time).
22. If an object having a mass of 1 gram were pushed so that it accelerated at a rate of 1 meter per second for each second of the push, what was the force of the push?
 - a. $F = 1 \text{ gram} \times 1 \text{ meter per second for every second of the fall}$
 - b. $F = 1 \text{ gram} + 1 \text{ meter per second for every second of the fall}$
 - c. $F = 0.5 \text{ gram} \times 1 \text{ meter per second}$
 - d. $F = 1 \text{ gram}$

Red Quiz 2

Lessons 5, 6, 7, 8, Labs 3, 4

Circle the letter(s) corresponding to the correct response(s).

1. Two lines running side by side and that are the same distance from one another at all adjacent points are:
 - a. orbital
 - b. purple
 - c. perpendicular
 - d. parallel

2. Substances that flow are called:
 - a. solids
 - b. flowers
 - c. fluids
 - d. oozers

3. Fluids have a tendency to travel in complex motions that result from being compressed (squished together) in one place while spreading out in another. This pattern of compression and decompression is described as:
 - a. wave motion
 - b. orbital motion
 - c. linear motion
 - d. slinky motion

4. A branch of mathematics that deals with waves and curves is called:
 - a. fluidometry
 - b. ergonomometry
 - c. trigonometry
 - d. waveometry

5. Work is often defined as the application of:
 - a. force over mass
 - b. force over a distance
 - c. energy over a force
 - d. distance over time

6. Simple machines have the ability to:
 - a. increase the output of work
 - b. increase the output of energy
 - c. increase the output of force
 - d. increase the output of people

7. The price you pay for this increase is paid by an increase in:
- a. distance
 - b. energy
 - c. work
 - d. people

Choose the best answers from among the words at the end. You may use a word more than once.

8. A satellite travels around the earth in a circular pattern. This is because of two influences in different directions. First, its own _____ makes it tend to continue in a straight line, but _____ continually pulls it toward the center of the earth. The resulting motion is called an _____. All the planets circle the sun in this type of motion as does the moon around the earth.
9. In science, a word that is used to describe a particular part of the universe we wish to consider is the word _____.
10. Of the two kinds of energy, stored energy is called _____ energy.
11. Of the two kinds of energy, energy of motion is called _____ energy.
12. The amount of potential energy in a system has the tendency to _____. This represents a(n) _____ in the level of organization of the system.
13. A slinky is a good example of a solid material which displays _____ motion, which is typical of fluids.
14. A slinky or a jump rope, when twirled from the ends has a tendency to form a number of waves at evenly-spaced distances. Those waves are called _____ waves, and they are separated by evenly spaced points between waves that are called _____.
15. A force pulling an object toward the center of its orbit is called a _____ force.
16. The price paid for doing work is called _____. The amount you pay is directly related to the amount of work that is done, although you may lose some to heat if your system is inefficient.

jumping, centripetal, wiggly, central, system, bumps, kinetic, stored, frenetic, linear, wave, orbital, inertia, sitting, standing, potential, nodes, knots, decrease, increase, not-movey thingies, centripetal, toward-the-middle force, gravity, orbit, energy, dollars, friction, dereliction

Red Quiz 3

Lessons 9, 10, 11, 12, Labs 5, 6

True or false?

1. The law of conservation of energy states that energy cannot change forms.
2. Anytime we do work against a force (such as gravity), we pay for that work with energy.
3. Falling water can be used to turn generators. This takes the potential energy from the generator and passes it on to the falling water.
4. A resistance converts heat energy into electricity.
5. A light bulb uses a filament made from tungsten to convert electrical energy to light.
6. The explosion of an atomic bomb is an example of a conversion of mass-energy to heat, light and kinetic energy.
7. The hot dogs we eat provide us with energy in the form of chemical bonds.
8. In the real world, it is possible to transfer energy perfectly from one marble to another such that no energy is lost to the surroundings in the form of sound or heat.

Match the following energy changes with the devices that cause them:

- | | |
|---|----------------------|
| 9. _____ electrical energy to light energy | a. rocket engine |
| 10. _____ chemical energy to light energy | b. hydroelectric dam |
| 11. _____ chemical energy to gravitational energy | c. glowing embers |
| 12. _____ thermal energy to light energy | d. kerosene lamp |
| 13. _____ gravitational energy to electric energy | e. steam generator |
| 14. _____ thermal energy to electric energy | f. light bulb |

Red Quiz 4

Lessons 13, 14, 15, 16, Labs 7, 8

Select the best response(s) from the list below.

1. Name the four natural forces.
2. Particles having like charges _____ one another.
3. Particles having unlike charges _____ one another.
4. The force that is seen in the presence of all large masses is _____.
5. Electricity that flows through a conductive medium is called electric _____, while electrical charges that are simply accumulated in one location are called _____ (“staying”) charges.
6. Name the three types of particles that make up atoms.

static, gravitation, strong nuclear, weak nuclear, medium nuclear, energy, work, distance, repel, repulse, attract, adore, proton, boson, lepton, electron, neutron, neutron, acceleron, intron, exon, current, electromagnetism

True or false?

7. Gravity is a natural pull with no push.
8. The electromagnetic force pushes but does not pull.
9. Electric current flows from a location of high potential energy to an area of low potential energy.
10. Work is done when nearby, opposite-charged particles are separated.
11. Protons carry a negative charge.

Red Quiz 5

Lessons 17, 18, 19, 20, Labs 9, 10

Select the best response(s) from the list below.

1. A electromagnetic device that is used to lock and unlock car doors automatically is called a _____.
2. The range of influence of a magnet is called its magnetic _____.
3. Atoms of a particular element are identified by their number of _____.
4. The _____ force is responsible for holding several positive-charged protons in a single nucleus.
5. The number of protons in an atom is also called its _____.

atoms, protons, hologram, solenoid, neutrons, selenium, molecules, pole, electrons, yard, pasture, field, elements, nuclear, electromagnetic, gravitational, atomic number, atomic mass, mass number

Match the following common elements with their descriptions:

- | | |
|--------------------|---|
| 6. _____ iodine | a. main ingredient in a penny |
| 7. _____ oxygen | b. tincture for cuts |
| 8. _____ aluminum | c. shiny bumpers |
| 9. _____ neon | d. gas for party balloons |
| 10. _____ copper | e. colorful lighted signs |
| 11. _____ helium | f. radioactive gas found in basements |
| 12. _____ hydrogen | g. light, explosive gas |
| 13. _____ radon | h. a gas in the atmosphere we must breathe in to live |
| 14. _____ chromium | i. common kitchen foil |

True or false?

15. Every electric current is accompanied by a magnetic field.
16. Atoms are an unstable form of matter.
17. Protons are about 2,000 times the size of electrons.
18. A compass points toward the geographic north and south poles.
19. The north pole of one magnet attracts the north pole of another.

20. Skin is made up of atoms.
21. The electromagnetic force is primarily responsible for the activity of an atomic bomb.
22. An electromagnet is a magnetic device made by passing electrical current through a conductor.

Red Quiz 6

Lessons 21, 22, 23, 24, Labs 11, 12

Circle the best response(s) to complete the sentence(s).

1. Anytime work is done against a force, (potential, kinetic) energy is stored up.
2. (Deeper, Shallower) liquid is associated with greater pressure.
3. Molecules of a hot substance move (faster, slower) than molecules of the substance when it is cold.

True or false?

4. Matter is a highly stable form of energy, but can sometimes disassemble to release huge amounts of energy.
5. Heat is a form of matter.
6. In general when matter is heated, it expands.
7. *Locomotion* is movement from place to place.
8. Heat has no effect on the way matter interacts.

Circle the letter(s) corresponding to the correct response(s).

9. Which of the following require an input of work?
 - a. separating opposite-charged particles
 - b. pushing together opposite-charged particles
 - c. separating like-charged particles
 - d. pushing together like-charged particles
 - e. separating magnetic poles having opposite charges
 - f. pushing together magnetic poles having opposite charges
 - g. separating magnetic poles having like charges
 - h. pushing together magnetic poles having like charges
 - i. separating two earth-sized masses
 - j. pushing together two earth-sized masses
10. Which of the following are best described as fluids? (Circle all the correct responses.)
 - a. solids
 - b. liquids
 - c. gases
11. Which of the following are properties of fluids? (Circle all the correct responses.)
 - a. increasing pressure with increasing depth
 - b. can be suctioned and siphoned
 - c. flowing
 - d. surface tension

12. Which is the correct name of the device pictured in the textbook for detecting and measuring radioactivity?
- Google counter
 - Geister counter
 - Geiger counter
 - Gamma counter
13. A generator is a device that is used to take mechanical energy and convert it to what kind of energy?
- light energy
 - heat energy
 - chemical bond energy
 - electric energy
14. The temperature of your freezer is more likely to be set at:
- 40°C
 - 20°C
 - 0°C
 - 10°C
15. The temperature of your house is comfortable at:
- 0°C
 - 20°C
 - 40°C
 - 72°C
16. Mark each of the following forces as either “n” for natural or “a” for artificial:
- _____ the force of earth pulling on the moon
 - _____ the force of a spring pushing a pogo stick upward
 - _____ the force of gravity pulling the pogo stick downward
 - _____ the force of your feet kicking a trampoline
 - _____ the force of one electron repelling another
 - _____ the force holding two protons together in the nucleus of an atom

Please provide the requested information.

17. Elements with large, unstable nuclei that give off energy are described as being

_____.

18. In a steam engine, steam is used to turn a _____, which turns a shaft, which operates a generator.

19. A substance which reduces surface tension is called a _____.

Red Quiz 7

Lessons 25, 26, 27, 28, Labs 13, 14

Circle the best response(s) to complete the sentence(s).

1. A (photometer, refractometer) is a device that is used to measure how much light passes through a sample. It can be used to measure how much of a chemical is present in that sample.
2. A (photometer, refractometer) is a device that is used to measure the amount of bending of light that occurs as it passes through a sample. It can be used to identify an unknown substance by how much it bends light.
3. The more energetic electromagnetic radiation is, the (greater, lesser) is its degree of refraction (bending) when it passes from one substance into another.
4. Our eyes perceive the wavelengths of light that are (absorbed into, reflected from) an object we are observing.
5. A mirror is a device that (absorbs, reflects) light with near perfection.
6. A lens having a (shorter, longer) focal length will magnify any object to a greater extent than a lens having a (shorter, longer) one.

True or false?

7. Light may be accurately described as self-perpetuating waves of electromagnetism.
8. The wavelength of an electromagnetic ray is a proper expression of its energy.
9. A long tubular light bulb is an incandescent bulb.
10. The sun gives off both visible and invisible electromagnetic rays.
11. Light travels at a rate of almost 200,000 miles per second.
12. Electrons can absorb light causing them to increase in kinetic energy.
13. The angle at which light leaves a mirror is unrelated to the angle at which it hits the mirror.

Number the following forms of electromagnetic radiation in their correct order from highest to lowest energy (smallest to greatest wavelength):

14. _____ ultraviolet rays
15. _____ radio waves
16. _____ visible light
17. _____ gamma rays
18. _____ microwaves
19. _____ X rays
20. _____ infrared

Circle the letter corresponding to the best answer.

21. Bending of light—a change in its direction that often occurs when it moves from one substance into another—is called:
 - a. absorption
 - b. refraction
 - c. fraction
 - d. transmission

22. If an object absorbs all of the wavelengths of light except yellow and red ones, our eyes will perceive that object as being:
 - a. a mixture of colors that does not include yellow and red, probably brown
 - b. sometimes yellow and sometimes red
 - c. red, because it is dominant
 - d. orange

23. The magnification of an object can be calculated as follows:
 - a. height of the image/height of the object
 - b. height of the object/height of the image
 - c. focal length/height of the object
 - d. focal length/height of the image

Please provide the requested information.

24. Our eyes detect only “ _____ ” light.

25. Name three things that can happen to light when it strikes an object (use the proper scientific terms):

26. What color do we perceive when our eyes detect no light?

27. When we detect a wide variety of wavelengths at the same time?

See Lab Workbook

Red Section Review

Yellow Quiz 1

Lessons 1, 2, 3, 4, Labs 1, 2

True or false?

1. Evaporation is the process that converts water vapor to liquid water.
2. The opposite of evaporation is constipation.
3. Mass is a measure of an amount of matter.
4. Atomic masses are measured in AMB (atomic mass bundles).
5. Grams are a measure of liquid volume.
6. All liquids freeze at the same temperature.
7. One milliliter (ml) is the same as one cubic centimeter (cm³)
8. Aluminum is more dense than iron.

Match the following words with their descriptions:

- | | |
|--------------------|--|
| 9. _____ atom | a. pure substance combining two or more atoms of different kinds |
| 10. _____ proton | b. pure substance combining two or more bound atoms |
| 11. _____ electron | c. pure substance consisting of only one kind of atom |
| 12. _____ element | d. particle in an atom carrying a positive electric charge |
| 13. _____ molecule | e. smallest stable form of matter, all matter is made up of them |
| 14. _____ compound | f. particle in an atom carrying a negative electric charge |

Match the following words with their descriptions:

- | | |
|----------------------|---|
| 15. _____ solid | a. neither as stiff as a solid nor as fluid as a liquid |
| 16. _____ liquid | b. distant molecules, free motion, little attraction among molecules |
| 17. _____ gas | c. tight molecules, little motion, strong attractions among molecules |
| 18. _____ semi-solid | d. close molecules, fluid motion, neither strong nor weak attractions |

Circle the letter corresponding to the best answer.

19. The value, 6.02×10^{23} , is:

- a. Avogadro's number
- b. the number of atoms of any element in one gram-atomic-mass of that element
- c. both of the above
- d. neither of the above

20. The density of any substance is correctly calculated as follows:

- a. mass/distance
- b. number of atoms/volume
- c. mass/volume
- d. volume/mass

21. The usual units of density are:

- a. g/m
- b. #/ml or #/cm³
- c. g/ml or g/cm³
- d. ml/g

22. Archimedes principle states that:

- a. a floating object displaces its mass in liquid
- b. a floating object displaces its volume in liquid
- c. a sinking object displaces its mass in liquid
- d. a sinking object displaces its volume in liquid
- e. a and c
- f. a and d
- g. b and c
- h. b and d
- i. all of a through d

Provide the requested information.

23. What is the normal boiling point of water in °C?

24. In °F?

25. What is the freezing point of water in °C?

26. What is the freezing point of water in °F?

27. Hot gas that results from boiling water is called _____.

28. Changes of a substance from solid to liquid, liquid to gas or any of the changes in reverse are called changes of _____.

Yellow Quiz 2

Lessons 5, 6, 7, 8, Labs 3, 4

True or false?

1. When chemicals interact with one another they can be said to have *reacted*.
2. When shared electrons are shared equally so that they spend about the same amount of time with each atom, the resulting atom is likely to be polar.
3. A double bond is stronger than a triple bond.
4. A diatomic molecule is one having two of the same kind of atom bound together.
5. Halogens often form diatomic molecules.
6. The lanthanoids (or lanthanides) are also called the rare earth metals.
7. None of the actinoid (or actinide) series elements are radioactive.
8. Polar molecules tend to dissolve readily in water.
9. Atoms consist of a few particles moving about in mostly empty space.

Match the following groups of atoms with their group names:

- | | |
|--------------------|-----------------------------|
| 10. _____ 1A | a. transition metals |
| 11. _____ 2A | b. post-transition elements |
| 12. _____ B groups | c. noble gases |
| 13. _____ 3A to 6A | d. halogens |
| 14. _____ 7A | e. alkali metals |
| 15. _____ 8A | f. alkaline earth metals |

Match the following symbols with their elements:

- | | |
|--------------|--------------|
| 16. _____ Mg | a. iron |
| 17. _____ Mn | b. sodium |
| 18. _____ O | c. oxygen |
| 19. _____ Cu | d. gold |
| 20. _____ Fe | e. potassium |
| 21. _____ Au | f. lead |
| 22. _____ Ag | g. magnesium |
| 23. _____ Pb | h. silver |
| 24. _____ K | i. manganese |
| 25. _____ Na | j. copper |

Match the following names of molecules with their molecular formulas:

- | | |
|---|-------------------|
| 26. _____ CH ₄ | a. carbon dioxide |
| 27. _____ H ₂ O | b. methane |
| 28. _____ CO ₂ | c. oxygen |
| 29. _____ O ₂ | d. ethylene |
| 30. _____ C ₂ H ₂ | e. water |

Circle the letter corresponding to the best answer.

31. A substance that can be readily shaped is said to be:

- a. duck-like
- b. malleable
- c. malevolent
- d. malleable

32. Semiconductors are found among the:

- a. metals
- b. metalloids
- c. Alloys
- d. non-metals

Provide the requested information.

33. Atoms on the far right-hand side of the periodic table that are completely satisfied with their number and arrangement of electrons. They are called the _____

_____.

34. An ion is an atom that bears a charge. A ion carrying a positive charge is called a(n) _____, while an ion carrying a negative charge is called a(n)

_____.

35. Name two types of chemical bonding: _____, _____.

36. Which of these two types is the stronger?

37. In writing a chemical equation, the _____ are shown on the left-hand side of the arrow and the _____ are shown on the right-hand side.

38. List four properties of metals:

Yellow Quiz 3

Lessons 9, 10, 11, 12, Labs 5, 6

True or false?

1. The combined mass of the products is always exactly the same as the combined mass of the reactants.
2. The amount of energy contained in the reactant molecules is the same as the amount of energy contained in the products of a chemical reaction.
3. A reaction equation can have different numbers of atoms on the left and right sides of the equation and still be correct.

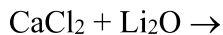
Match the atoms on the left with corresponding atoms on the right with which they will combine to form a compound:

- | | |
|---|-----------------------------------|
| 4. _____ one atom of a group 1A element | a. one atom of a group 4A element |
| 5. _____ two atoms of a group 1A element | b. one atom of a group 5A element |
| 6. _____ three atom of a group 1A element | c. one atom of a group 6A element |
| 7. _____ one atom of a group 2A element | d. one atom of a group 7A element |
| 8. _____ two atoms of a group 2A element | |
| 9. _____ one atom of a group 3A element | |
| 10. _____ one atom of a group 4A element | |

Circle the letter corresponding to the best answer.

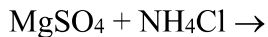
11. One water molecule (H_2O) consists of:
 - a. one hydrogen atom and two oxygen atoms
 - b. two hydrogen atoms and one oxygen atom
 - c. not enough information to give an answer
12. If the atomic weight of hydrogen is 1 AMU and the atomic weight of carbon is 12 AMU, then the molecular weight of methane (CH_4) is:
 - a. 13 AMU
 - b. 14 AMU
 - c. 16 AMU
 - d. 16 AMU plus a toothpick
13. Which of the following is a proper, balanced reaction equation:
 - a. $\text{NaOH} + \text{HCl} \rightarrow \text{NaCl} + \text{H}^2\text{O}$
 - b. $2 \text{NaOH} + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\text{O}$
 - c. $\text{NaOH} + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2$
 - d. $\text{NaOH} + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\text{O}$

14. Which is the proper way to complete the following reaction:



- a. $\text{CaCl}_2 + \text{Li}_2\text{O} \rightarrow \text{CaO} + 2 \text{LiCl}$
- b. $\text{CaCl}_2 + \text{Li}_2\text{O} \rightarrow 2 \text{CaO} + \text{LiCl}$
- c. $\text{CaCl}_2 + \text{Li}_2\text{O} \rightarrow \text{CaLi}_2 + \text{OCl}_2$

15. Sulfate (SO_4^{2-}) is a polyatomic ion with a 2 – charge. Ammonium (NH_4^+) is a polyatomic ion with a 1 + charge. Which is the proper way to complete the following reaction:



- a. $\text{MgSO}_4 + \text{NH}_4\text{Cl} \rightarrow \text{MgCl}_2 + \text{NH}_4\text{SO}_4$
- b. $\text{MgSO}_4 + \text{NH}_4\text{Cl} \rightarrow \text{Mg}_2\text{Cl}_2 + (\text{NH}_4)_2\text{SO}_4$
- c. $\text{MgSO}_4 + 2 \text{NH}_4\text{Cl} \rightarrow \text{MgCl}_2 + (\text{NH}_4)_2\text{SO}_4$

Provide the requested information.

16. Atoms that stay together in charged (ionic) groups that act as though they were a single element are called _____.

Yellow Quiz 4

Lessons 13, 14, 15, 16, Labs 7, 8

True or false?

1. Children should lose weight by fasting.
2. Fasting causes your body to accumulate fat during times when food is available.
3. Bodies use food molecules for generating energy for activities and for keeping itself warm.
4. A person should always eat anytime he/she feels hungry.
5. There are bacteria that can use petroleum for fuel (food).
6. The best way to lose weight is to eat less and exercise.
7. To lose weight we will probably have to be hungry sometimes.

Match the following carbon compounds with their descriptions:

- | | |
|------------------------------|---|
| 8. _____ alkane | a. a straight carbon chain with at least one side chain |
| 9. _____ alkene | b. a carbon chain consisting of only single bonds |
| 10. _____ alkyne | c. a carbon chain in the form of a ring |
| 11. _____ branched aliphatic | d. a carbon chain having at least one triple bond |
| 12. _____ cycloaliphatic | e. a carbon chain having at least one double bond |

Place the following petroleum fractions in the correct order in which they boil off from crude oil (the last one doesn't boil off at all):

13. _____ motor oil
14. _____ kerosene
15. _____ gasoline
16. _____ diesel fuel
17. _____ asphalt

Circle the letter(s) corresponding to the best answer(s).

18. Natural gas is gas that comes out of wells at a petroleum field that is used in homes for fuel.

It is composed of mostly:

- a. gasoline
- b. butane
- c. nitrogen
- d. methane

19. Which of the following come from petroleum?

- a. petrolatum
- b. beeswax
- c. paraffin
- d. potato chips

20. Which of the following equations are balanced?

- a. $\text{HCl} + \text{NaOH} \rightarrow \text{NaCl} + \text{H}_2\text{O}$
- b. $2 \text{NH}_3 + \text{O}_2 \rightarrow 2 \text{NO} + 3 \text{H}_2\text{O}$
- c. $2 \text{Mg} + \text{CO}_2 \rightarrow 2 \text{MgO} + \text{C}$
- d. $\text{CH}_4 + \text{H}_2\text{O} \rightarrow \text{CO} + 3 \text{H}_2$

Provide the requested information.

21. Automobiles have a device called a _____ that improves the efficiency of fuel burning.

22. The main dangerous gas that is removed by the device in the previous question is called _____.

23. There are always two products from an efficient fueling reaction. What are they?

24. Name two kinds of molecules people use for fuel.

Yellow Quiz 5

Lessons 17, 18, 19, 20, 21, Labs 9, 10

Note: this test includes a fifth lesson (Lesson 21) since it makes more sense to keep all this biochemistry together.

True or false?

1. Two things that are needed by bodies are fuel molecules and carbon dioxide.
2. Bodies produce carbon dioxide, energy, body mass and waste.
3. The energy produced by our bodies is used for our activities and to generate heat.
4. If we wish to lose weight we should be more physically active and eat more to fuel that extra activity.
5. If we wish to gain weight we should try to stop producing waste.
6. Two types of proteins are structural proteins and enzymes.
7. DNA and RNA store information that is used in guiding cellular processes.
8. Genetic engineering is a scientific discipline in which scientists manipulate polysaccharides.
9. Vitamin C is an oxidizing agent.
10. Rust is the result of chemical oxidation of iron.
11. Meat tenderizer contains enzymes that break down meat proteins.
12. Monosodium glutamate is a form of an amino acid used to flavor foods.
13. Dextrose is an example of a fatty acid.
14. Soap is made from nucleic acids.

Match the following with one of the terms on the right:

- | | |
|--|--------------------|
| 15. _____ made up of “base pairs” | a. nucleic acids |
| 16. _____ starch, for example | |
| 17. _____ made up of amino acids | b. proteins |
| 18. _____ glycogen, for example | |
| 19. _____ make up the genetic code | |
| 20. _____ made up of sugars | c. polysaccharides |
| 21. _____ made of adenine, guanine, thymine, cytosine | |
| 22. _____ made up of fatty acids | d. lipids |
| 23. _____ cellulose, for example | |
| 24. _____ DNA and RNA | |
| 25. _____ enzymes, for example | |
| 26. _____ deoxyribose and ribose sugars linked by PO_4^{2-} | |
| 27. _____ collagen, for example | |
| 28. _____ polymers of glucose, for example | |

Circle the letter(s) corresponding to the best answer(s).

29. Which of the following types of molecules have energy storage as a major function?
- a. nucleic acids
 - b. proteins
 - c. polysaccharides
 - d. lipids

Yellow Quiz 6

Lessons 22, 23, 24, Labs 11, 12

True or false?

1. Chemical reactions take place readily in solutions because molecules of two different substances can come into close contact with one another in solution.
2. Polar solutes dissolve readily in non-polar solvents.
3. True solutions are cloudy in appearance.
4. More viscous liquids create more stable suspensions.
5. Lakes have a considerable amount of suspended matter in them.
6. Vegetable oil becomes less viscous when warmed.
7. Sodium chloride is soluble in vegetable oil.

Provide the requested information from the word list at the end.

8. Fill in the blanks to complete this sentence regarding solutions:

A _____ results when a _____ is
_____ in a _____.

9. "Likes dissolve _____."
10. Oil mixed into vinegar is an example of a _____.
11. The chemists word for a thicker fluid is "_____."

solution, solvent, solute, dissolved, likes, dislikes, unlikes, solution, suspension, gluey, viscous, tacky, gooey

Yellow Quiz 7

Lessons 25, 26, 27, 28, Labs 13, 14

Circle the best response(s) to complete the sentence(s).

1. When water ionizes, it creates H^+ ions which are (acidic, basic) and OH^- ions which are (acidic, basic).
2. Evaporation of a liquid has a (warming, cooling) effect on the surface from which it evaporates.
3. A solution having a pH of 6 is (acidic, basic, neutral).

True or false?

4. The unique nature of water keeps the temperature of the earth's surface livable.
5. The unique nature of water keeps lakes from freezing solid in the winter months.
6. Blood and lymph are made of mostly protein.
7. Water serves as a medium in which food and nutrient substances dissolve to feed living cells.
8. Acid/base reactions produce water and a salt.
9. Soft drinks contain acid.
10. Milk of magnesia works by neutralizing stomach acid.

Circle the letter(s) corresponding to the best answer(s).

11. Which of the following reactions are methods for separating suspensions:
 - a. filtration
 - b. settling
 - c. centrifugation
 - d. mixing
12. Which of the following reactions shows the correct, complete ionization of sodium hydroxide:
 - a. $NaOH \rightarrow Na^+ + O^{2-} + H^+$
 - b. $NaOH \rightarrow Na^+ + OH^-$
 - c. $NaOH \rightarrow NaO^- + H^+$
 - d. none of the above
13. Which of the following reactions shows the correct way that hydrochloric acid normally reacts with sodium hydroxide:
 - a. $HCl + NaOH \rightarrow NaCl + H_2O$
 - b. $HCl + NaOH \rightarrow Na^+ + Cl^- + H^+ + OH^-$
 - c. $HCl + NaOH \rightarrow NaOCl + H_2$
 - d. none of the above

Yellow Section Review

See Lab Workbook

Blue Quiz 1

Lessons 1, 2, 3, 4, Labs 1, 2

Circle the best response(s) to complete the sentence(s).

1. Reproduction that depends on the participation of male and female organisms of the same kind is called (asexual, sexual) reproduction.

True or false?

2. A theory is a mental model for understanding our observations.
3. The general theory of evolution suggests that all life evolved from a single-celled ancestor.
4. Most scientists classify humans as animals.
5. Living things are different than non-living things because they do not obey the laws of physics.
6. Our scientific system for naming plants and animals is a “binomial” system.
7. The genome copies itself before a cell divides by a process called “duplicity.”
8. Sodium hypochlorite is added to city water to prevent the growth of harmful microbes.
9. The gelatin that is often used to grow bacteria is called agar.
10. Agar is layered on the inside of a special dish called a Putrid dish.
11. A common way diseases are transmitted is by shaking hands.
12. Chemical buffers are added to media to control the temperature.

Place the following levels of organization of living things in order beginning with the least organized and ending with the most highly organized:

13. _____ molecules
14. _____ subatomic particles (protons, neutrons, electrons)
15. _____ organ systems
16. _____ organelles
17. _____ atoms
18. _____ tissues
19. _____ organs
20. _____ large biomolecules
21. _____ organism
22. _____ cells

Place the following taxonomic classifications in order beginning with the broadest (kingdom) and ending with the narrowest:

- 23. _____ order
- 24. _____ genus
- 25. _____ class
- 26. _____ phylum
- 27. _____ kingdom
- 28. _____ species
- 29. _____ family

Circle the letter corresponding to the best answer.

- 30. The fluid inside a cell is called:
 - a. cell juice
 - b. cytol
 - c. cytoplasm
 - d. plasma

- 31. The tiny functioning parts of a cell are called:
 - a. organelles
 - b. organs
 - c. protoplasm
 - d. cytoplasm

- 32. The entire genetic makeup of the cell is called the:
 - a. cytoplasm
 - b. gene
 - c. RNA
 - d. genome

- 33. The tiny device for assembling proteins is called:
 - a. enzyme
 - b. protein factory
 - c. ribosome
 - d. proteinosome

Provide the requested information.

34. Name five characteristics of living things.

35. Name five kingdoms of living things.

Choose the correct answers from the word list below

36. The kind of plate used for culturing bacteria is called a _____ plate.

37. The eyepiece of a microscope is also called the _____.

38. The _____ of the microscope contains the lower lens of the microscope.

39. The higher magnification provided by a microscope allows you to _____,
or separate out, fine details.

40. Organisms that require oxygen to live are called _____.

occipital, ocular, oculary, occult, objection, object, objective, obtrusive, resolution, resin,
reason, resolve, dissolve, aerobic, aesthetic, aerobic, anaerobic, pizza, pecuniary, Petra, Petri,
petrified, flat, fluted

Blue Quiz 2

Lessons 5, 6, 7, 8, Labs 3, 4

True or false?

1. Fungi, plants, archaea and bacteria cells all have some kind of rigid cell wall.
2. Animal cells are known for having chlorophyll.
3. Bacteria cells have a nucleus with a nuclear membrane.
4. Plants often have starch storage granules in their cells.
5. Fungi always have cells that are clearly separated from one another.
6. Another word for chemical evolution is abiogenesis.
7. A materialistic viewpoint is one based only on what can be directly observed.
8. Photosynthesis is the process by which animals make their own food.
9. Bacteria are all round in shape.

Match the following organelles with their descriptions:

- | | |
|---------------------------------|--|
| 10. _____ nucleus | a. folded, complex inner membrane |
| 11. _____ nuclear membrane | b. paired strands of DNA in higher organisms |
| 12. _____ nucleoplasm | c. rigid covering outside the cell membrane |
| 13. _____ chromosome | d. fluid inside the nucleus |
| 14. _____ cell wall | e. "powerhouse of the cell," site of energy generation |
| 15. _____ endoplasmic reticulum | f. thin membrane surrounding the nucleus |
| 16. _____ mitochondria | g. part of eukaryotic cells containing the genome |

Circle the letter corresponding to the best answer.

17. The big bang theory is an attempt to explain:
 - a. how matter was originally dispersed throughout the universe
 - b. where matter came from
 - c. where people came from
 - d. Darwinism
18. The theory of chemical evolution is an attempt to explain:
 - a. where chemicals came from
 - b. how life arose from non-living matter
 - c. where Darwin came from
 - d. where Elijah Wood came from

19. Darwinian evolution is an attempt to explain:
- where the first living thing came from
 - why apes are smarter than your siblings
 - how a single ancestor organism developed into all the diverse organisms present today
 - how a fish turned into a person
20. In which century did Charles Darwin live?
- 1400's
 - 1600's
 - 1800's
 - He's still alive
21. Which of the following support(s) the big bang theory:
- the universe is dark
 - the universe appears to be expanding
 - the universe appears to be contracting
 - the universe exists
22. Which of the following is(are) taken to support chemical evolution:
- bacteria are extremely complex
 - the universe appears to be expanding
 - large chemicals tend to be broken down into small ones
 - biochemicals can be manufactured in the laboratory under conditions that might be thought to be like early earth conditions
23. On agar plates, bacteria grow in clusters called:
- groups
 - hog piles
 - colonies
 - states

Blue Quiz 3

Lessons 9, 10, 11, 12, Labs 5, 6

Circle the best response(s) to complete the sentence(s).

1. For a living thing to increase in complexity over time is to go (against, with) the direction in which natural processes tend to go.
2. Organisms that are less complex are considered “less evolved” and are referred to as “(lower organisms, higher organisms).”
3. (Plants, Animals) are at the bottom of the food chain.
4. Animals are (producers, consumers).
5. Oxygen is a (reactant in, product of) photosynthesis.
6. (Producers, Consumers) break plant material down into food chemicals which they use to make new cells and energy.

True or false?

7. The word *evolution* means “change.”
8. Organisms do change over time.
9. By selective breeding, people can bring about desired changes in organisms.
10. Plants require carbon dioxide, water and sunlight in order to produce glucose.
11. Carbon dioxide and water are products of consumers.

Match the following microbes with their descriptions:

- | | |
|----------------------|--|
| 12. _____ virus | a. microscopic animals |
| 13. _____ archaea | b. microscopic plants |
| 14. _____ bacteria | c. eukaryotic microbes including yeasts and molds |
| 15. _____ microfungi | d. a non-living particle that reproduces inside living cells |
| 16. _____ microphyta | e. prokaryotic organisms occupying virtually every environment |
| 17. _____ microzoa | f. prokaryotic “ancient ones” that occupy harsh environments |

Circle the letter(s) corresponding to the best answer(s).

18. Three characteristics that all fungi have in common are:
 - a. eukaryotic
 - b. cell walls made of chitin
 - c. photosynthetic
 - d. feed on pre-formed organic matter
 - e. a squishy consistency

19. Three common types of fungi are:

- a. yeasts
- b. algae
- c. molds
- d. slime molds
- e. bacteria

20. Three structures that typically appear in mold mycelia are:

- a. seeds
- b. hyphae
- c. spores
- d. hairs
- e. fruiting bodies

Provide the requested information.

21. Small changes that take place in organisms from one generation to another are called _____.

22. If an organism can be seen only with the help of a microscope we say that that organism is _____.

_____ . The opposite of this term is _____.

23. A bacterium that can no longer be defeated by antibiotics is referred to as antibiotic _____.

_____.

24. An organism that loves high temperatures is called a _____.

25. Another name for blue-green algae is _____.

26. The process by which plants produce glucose from sunlight and carbon dioxide is called _____.

_____.

27. Plants capture sunlight using chemical _____.

28. _____ is the best known and most common of these chemicals.

microlutions, adaptations, macroevolution, really small, microscopic, a virus, a bacterium, redundant, reclusive, resistant, macroscopic, intermedioscopic, largoscopic, really large, a whopper, heatophile, tempophile, thermophile, cyanide, bacteria-cyanoverde, cyanobacteria, blue-green slime, gluconeogenesis, glycolysis, photophotosynthesis, phototrophism, stains, paints, greens, pigments, pygmies, Chlorophyte *a*, Chlorophyll *a*, Clorets *a*, Claustrophobi *a*, Clavulinic acid

Blue Quiz 4

Lessons 13, 14, 15, 16, Labs 7, 8

Circle the best response(s) to complete the sentence(s).

1. The part of the life cycle of the plant in which sexual reproduction takes place is called the (sporophyte, gametophyte) generation.
2. The seed is a product of (sexual, asexual) reproduction, while the spore is the product of (sexual, asexual) reproduction.
3. Plants that produce pollen and seeds in cones are called (gymnosperms, angiosperms) while those that produce pollen and seeds in flowers are called (gymnosperms, angiosperms).
4. Flowers having only male parts are called (staminate, pistillate), while those having only female parts are called (staminate, pistillate).

True or false?

5. The word *phytoplankton* means “suspended plants.”
6. The word *microphyta* means “large plants.”
7. If something is too large to be microscopic it is called *macrohagic*.
8. Mosses and worts are likely to be found in the dessert.
9. Vascular plants are plants having vessels to carry water and food within the plant.
10. The evolutionist might describe a vascular plant as a “higher plant” and an alga as a “lower plant.”
11. A spore is a dormant, resistant form of plant life that germinates when the conditions are right.
12. The fruit of a plant may be referred to as vegetable matter.
13. Germination can refer to the emergence of a plant from either a seed or a spore.

Match the following phyla of phytoplankton with their representative organisms:

- | | |
|------------------------|-------------------------------|
| 14. _____ Pyrrophyta | a. <i>Volvox</i> , spirogyra |
| 15. _____ Chrysophyta | b. <i>Euglena</i> |
| 16. _____ Euglenophyta | c. dinoflagellates |
| 17. _____ Chlorophyta | d. coccolithophorids, diatoms |

Match the following phyla of plants with their meanings:

- | | |
|------------------------|--------------------------|
| 18. _____ Pyrrophyta | a. “red plants” |
| 19. _____ Chrysophyta | b. “gold plants” |
| 20. _____ Euglenophyta | c. “dusky plants” |
| 21. _____ Chlorophyta | d. “green plants” |
| 22. _____ Rhodophyta | e. “good-eyeball plants” |
| 23. _____ Phaeophyta | f. “fire plants” |

Match the following phyla of plants with the types of plants represented:

- | | |
|------------------------|----------------------------|
| 24. _____ Chlorophyta | a. brown algae (kelp) |
| 25. _____ Rhodophyta | b. mosses and worts |
| 26. _____ Phaeophyta | c. red algae |
| 27. _____ Bryophyta | d. vascular plants |
| 28. _____ Tracheophyta | e. green filamentous algae |

Match the following flower parts to their descriptions:

- | | |
|--------------------|---|
| 29. _____ sepal | a. entire male organ of a flower including the anther and filament |
| 30. _____ petal | b. top surface of the carpel that receives pollen for fertilization |
| 31. _____ calyx | c. hollow organ that contains the ovules |
| 32. _____ anther | d. stalk connecting the stigma to the ovary |
| 33. _____ filament | e. long slender stalk to which an anther is attached |
| 34. _____ stamen | f. entire female organ of a flower including the stigma and style |
| 35. _____ carpel | g. brightly colored landing pad for insects |
| 36. _____ stigma | h. contains a single egg which may be fertilized to form a seed |
| 37. _____ ovary | i. male part on which pollen forms |
| 38. _____ style | j. green, leaf-like structures making up the calyx |
| 39. _____ ovule | k. entire cup-like base of a flower |

Circle the letter corresponding to the best answer.

40. Water and oxygen from photosynthesis are released from the leaves of vascular plants through special pores called
- stomata
 - tomato
 - stromboli
 - stigma

41. Which of the following plants is a tracheophyte (vascular plant)?
- liverwort
 - wart
 - oak tree
 - green alga
42. Which of the following lines the pores in leaves to control the escape of water from the plant?
- centurion cells
 - soldier cells
 - guard cells
 - dehydration protection thingies
43. Inside a seed you will find a(n):
- embryo
 - fetus
 - bugaboo
 - egg
44. Division of plants into monocotyledons and dicotyledons is based on the number of :
- flowers
 - flower petals
 - centimeters in a meter
 - seed leaves
45. A sugary fluid that insects remove from flowers is called:
- neck tar
 - nectar
 - honey
 - beeswax
46. A seed-bearing or spore-bearing part of a plant or the mature ovary are called:
- fruiting bodies
 - juicy fruit
 - bodies
 - vegetables

Blue Quiz 5

Lessons 17, 18, 19, 20

True or false?

1. The classification *Mesozoa* consists mainly of parasitic worms.
2. Eucoelomates are animals that have legs.
3. “Convergent evolution” is the process by which two very different organisms are supposed to have evolved very similar features.
4. In an evolutionary sense, arthropods are a successful group of organisms.
5. The word *diversity* refers to how many of a single exact kind of organism is alive at one time.
6. A trilobite is an extinct arthropod.
7. Spiders are in the class Merostomata.
8. Spiders and horseshoe crabs are arthropods.
9. Spiders and horseshoe crabs are Trilobites.
10. Spiders, ticks, mites and scorpions are all arachnids.

Match the following phyla of metazoans with their representative animals:

- | | |
|---------------------------|--|
| 11. _____ Porifera | a. flatworm |
| 12. _____ Coelenterata | b. proboscis, nemertine and ribbon worm |
| 13. _____ Ctenophora | c. roundworm |
| 14. _____ Platyhelminthes | d. arrow worm |
| 15. _____ Rhynchocoela | e. spiny-headed worm |
| 16. _____ Nematoda | f. sponge |
| 17. _____ Acanthocephala | g. acorn worm |
| 18. _____ Chaetognatha | h. comb jelly |
| 19. _____ Nematomorpha | i. horsehair worm |
| 20. _____ Brachiopoda | j. jellyfish, anemone, sea fan, coral |
| 21. _____ Tardigrada | k. chiton, snail, abalone, whelk, slug, periwinkle, cowrie |
| 22. _____ Mollusca | l. “water bear” |
| 23. _____ Hemichordata | m. lamp shells |

Match the following classes of mollusks with their representative animals:

- | | |
|-----------------------|------------------------------|
| 24. _____ Amphineura | a. snail, whelk, slug |
| 25. _____ Pelecypoda | b. tooth shell or tusk shell |
| 26. _____ Scaphopoda | c. octopus, squid, nautilus |
| 27. _____ Gastropoda | d. bivalve |
| 28. _____ Cephalopoda | e. chiton |

Match the following classes of annelid worms with their representative animals:

- | | |
|-------------------------|---------------------------------------|
| 29. _____ Archiannelida | a. leeches |
| 30. _____ Polychaeta | b. bristled worms with few appendages |
| 31. _____ Oligochaeta | c. a group of simple marine worms |
| 32. _____ Hirudinea | d. worms with many appendages |

Circle the letter corresponding to the best answer(s).

33. An arthropod has an exoskeleton made of:
- a. protein
 - b. mostly calcium
 - c. chitin
 - d. Jell-O

Provide the requested information from the list at the end.

34. The study of animals is called _____.
35. Single-celled and colonial animals are classified as _____.
36. The _____ are multi-celled animals having only a few different cell types.
37. Multi-celled animals having a large variety of cell structures and functions are classified as _____.

botany, zoology, protozoology, animology, zoography, protoplasm, protein, probiotic, protozoa, mesozoa, mesoderm, mesobiotic, metabiotic, metazoa, metaplasm

Labs 9, 10

1. The _____ family of plants includes marigolds, dandelions, chrysanthemums, asters and dahlias.
2. Circle one. This previous family is the most (diverse, similar) group of plants on earth.
3. This family of plants has special _____ flowers that are “strap-like.”
4. These flowers are made up of a strap-like _____, a female part, like a pistil, called a(n) _____; a male part, like a stamen, called a _____ . It also has a fuzzy, frilly _____, and a(n) _____ which contains the seed when it is mature.

dandelion, composite, deposit, expository, repository, suppository, ligand, leg-like, leprous, ligulate, pedal, petal, gymnasium, gynoecium, gynecologist, androgen, androecium, philanthropist, papilla, pupa, pappus, puppy, oval, Orville Redenbacher, ovary, ornery, obituary

Blue Section Review I

Lessons 21, 22, 23, 24 (and cumulative from Lesson 1)

Labs 11, 12 (and cumulative from Lab 1)

See Lab Workbook

Blue Quiz 7

Lessons 25, 26, 27, 28 (Future labs will be quizzed only in the section reviews in the Lab Workbook)

Circle the best response(s) to complete the sentence(s).

1. The mammalian subclass *Prototheria* includes the (monotremes, marsupials).
2. An animal that is active mostly by day is (diurnal, nocturnal) while one that is active mostly by night is (diurnal, nocturnal).

True or false?

3. Birds are in the class Aves.
4. Birds are vertebrates.
5. Birds are chordates.
6. The mammalian subclass *Eutheria* includes the kangaroo.
7. Opossums are marsupials.
8. Bats are in the class Aves.
9. Bats are marsupials.

Match the following orders of mammals with their representative animals:

- | | |
|--------------------------|---|
| 10. _____ Insectivora | a. pangolin |
| 11. _____ Dermoptera | b. gliding lemur |
| 12. _____ Chiroptera | c. bat |
| 13. _____ Edentalia | d. squirrel, beaver, paca |
| 14. _____ Pholidota | e. swine, camel, llama, deer, giraffe, antelope |
| 15. _____ Lagomorpha | f. rabbit, hare |
| 16. _____ Rodentia | g. whale, dolphin, porpoise |
| 17. _____ Hyracoidea | h. hyrax |
| 18. _____ Artiodactyla | i. sea lion, seal, walrus |
| 19. _____ Perissodactyla | j. horse, tapir, rhinoceros |
| 20. _____ Cetacea | k. shrew, mole, hedgehog |
| 21. _____ Pinnipedia | l. anteater, sloth, armadillo |

Blue Quiz 8

Lessons 29, 30, 31, 32

True or false?

1. Giant panda are classified as carnivores, even though they eat no meat.
2. An endangered animal is one that is in danger of becoming too popular with wildlife enthusiasts.
3. The word *anatomy* is the breaking apart of “an atom.”

Match the following orders of mammals with their representative animals:

- | | |
|----------------------|------------------------------------|
| 4. _____ Carnivora | a. elephant |
| 5. _____ Proboscidea | b. monkey, ape, man, lemur, baboon |
| 6. _____ Sirenia | c. manatee, dugong, sea cow |
| 7. _____ Primates | d. cat, dog, bear, weasel |

Circle the letter(s) corresponding to the best answer(s).

8. The following are characteristics of primates:
 - a. opposable thumbs
 - b. opposable big toes
 - c. ball and socket joints
 - d. large brain
 - e. flat nails
 - f. tails
 - g. all of the above
9. The upper appendages include which of the following:
 - a. upper extremities
 - b. pelvic girdle
 - c. collar bones (clavicles)
 - d. shoulder blades (scapulas)

10. Which of the following are types of bone joints:

- a. ball and socket
- b. dovetail
- c. hinge
- d. saddle
- e. pivot
- f. gliding
- g. flying
- h. spreading
- i. ball and bat
- j. twist

11. The tips of our noses and outer ears are made of:

- a. rubber
- b. chiton
- c. cartilage
- d. Carthage
- e. agar

12. Name three types of connective tissue:

Blue Quiz 9

Lessons 33, 34, 35, 36

Circle the best response(s) to complete the sentence(s).

1. The word (gamete, zygote) describes either the egg cell or the sperm cell alone. The word (gamete, zygote) describes the new individual formed when the egg and the sperm come together.

True or false?

2. Bronchitis is a condition which describes an inflamed glottis.
3. Ovulation is the production of a new egg for possible fertilization.
4. Womb is another name for the ovary.
5. Sperm cells are produced in the testicles and stored in the epididymus.
6. The prostate is a female organ.

Match the following organs with their descriptions or functions:

- | | |
|---------------------------|---|
| 7. _____ liver | a. site where food chemicals are absorbed into the blood |
| 8. _____ gall bladder | b. place where undigested waste is packaged and processed |
| 9. _____ salivary glands | c. produces an emulsifier to aid digestion |
| 10. _____ small intestine | d. the site of storage of some digested food chemicals |
| 11. _____ large intestine | e. temporary storage for solid waste |
| 12. _____ rectum | f. located about the mouth |

Match the following organs with their descriptions or functions:

- | | |
|---------------------------|--|
| 13. _____ kidney | a. temporary storage for liquid waste |
| 14. _____ ureter | b. organ that removes excess water and waste chemicals from blood |
| 15. _____ urinary bladder | c. tube that carries urine from the kidneys to the urinary bladder |
| 16. _____ urethra | d. tube that removes urine from the urinary bladder |

Place the following organs in the correct order in which food passes through them, beginning with the mouth as number 1:

17. _____ anus
18. _____ esophagus
19. _____ rectum
20. _____ small intestine
21. _____ large intestine
22. _____ mouth
23. _____ stomach

Place the following organs in the correct order in which air passes into them from the outside, beginning with the mouth:

24. _____ larynx
25. _____ mouth
26. _____ trachea
27. _____ bronchioles
28. _____ glottis
29. _____ bronchi
30. _____ alveoli

Circle the letter corresponding to the best answer.

31. Which of the following are glands which secrete digestive juices into the stomach:

- a. small intestine
- b. gall bladder
- c. liver
- d. pancreas

32. The female menstrual cycle lasts:

- a. exactly 14 days
- b. exactly 28 days
- c. about 48 days
- d. about 28 days, but it varies from one individual to another

33. The common term for the menstrual cycle is the:

- a. comma
- b. period
- c. exclamation point
- d. fertilization

34. Fertilization usually takes place inside the:

- a. ovary
- b. oviduct
- c. uterus
- d. vagina

Provide the requested information.

35. Name four types of teeth in order from the front of the mouth to the back:

- incisor
- canine
- premolar
- molar

36. Explain why hemoglobin is usually thought of as part of the circulatory system, but could be considered part of the respiratory system as well.

Blue Quiz 10

Lessons 37, 38, 39, 40

Circle the best response(s) to complete the sentence(s).

1. Blood with the cells taken out of it is called (lymph, plasma, tissue fluid). The fluid, similar to plasma, that keeps the cells bathed in their required substances is called (lymph, plasma, tissue fluid). When this fluid collects in the lymphatic vessels, it is called (lymph, plasma, tissue fluid).
2. The largest single artery of the body is the (aorta, pulmonary artery). The large (pulmonary arteries, aortas) take blood to the lungs.
3. The large arteries that carry blood to the arms are the (femorals, subclavians) while those that carry blood to the legs are the (femorals, subclavians).
4. Athlete's foot and diaper rash are examples of two types of (bacterial, fungal, viral) infection. (Bacteria, Fungi, Viruses) are the only group of organisms that are treated with antibiotics. (Bacteria, Fungi, Viruses) on the other hand are non-living pathogens that cause a large variety of diseases including AIDS and the common cold.

True or false?

5. The blood flows to your head through two large arteries, one on each side of your neck called the carotid arteries.
6. A nerve cell is called a neurotic.
7. The sensation of the need to vomit is called malaise.
8. Olfactory receptors are special nerve endings that sense pain.
9. The brain sends signals to certain glands in the body causing them to release chemicals into the blood.
10. The brain and spinal cord together are called the Grand Central Station (GCS).
11. Genetics is the study of inheritance.
12. The father of Genetics was Gregor Mendel.
13. Offspring having specific characteristics can be directed by a process called selective eating.
14. Genetic engineering is based on chemical methods for changing DNA.
15. Bacterial cells and the cells of higher organisms reproduce in exactly the same way.
16. Trichinosis is a disease caused by worms in poorly cooked pork.

Place the following in the order in which they receive oxygenated blood:

17. _____ venules
18. _____ arteries
19. _____ veins
20. _____ arterioles
21. _____ capillaries

Circle the letter corresponding to the best answer(s).

22. Which are the six major portions of the brain?

- a. frontal lobe
- b. backal lobe
- c. auditory lobe
- d. temple lobe
- e. temporal lobe
- f. spatial lobe
- g. brain stem
- h. brain root
- i. cerebellum
- j. antebellum
- k. occidental lobe
- l. occipital lobe
- m. parietal lobe
- n. parental lobe

See Lab Workbook

Blue Section Review II

Rainbow Quiz 1

Lessons 1, 2, 3, 4, Labs 1, 2

Circle the best response(s) to complete the sentence(s).

1. An experimental (“control group,” “treatment group”) is changed in some important detail while the (“control group,” “treatment group”) is held constant.
2. Since feelings and beliefs are (objective, subjective), they do not serve as adequate bases for a scientific conclusion. The decision about an experiment’s outcome must be (objective, subjective).
3. (Plaster of Paris, Cement) is calcined calcium and aluminum silicates while (plaster of Paris, cement) is heat-treated gypsum.

True or false?

4. A theory might be described as a model for thinking.
5. A rock (according to the science of geology) is a small piece of crystalline material that might be held in the palm of your hand.
6. Settled materials are called sediments.
7. The Ring of Fire is the ring of activity around a volcano.
8. Ground water is water that lies on top of the ground.
9. A spring that arises from pressure from beneath the ground surface is called an artesian spring.
10. Water enters the atmosphere when it is warmed by the sun by a process we call exasperation.
11. Water reenters the liquid phase by a process we call consternation.
12. The study of fresh surface water is called limnology.
13. The study of ocean water is oceanology.
14. An “abiotic control” is a poisoned, negative-control sample.
15. Corundum is also called “alumina,” and is made of aluminum trioxide.

Circle the letter corresponding to the best answer.

16. Molten rock within the earth is called.

- a. magnetite
- b. lava
- c. magma
- d. mantle
- e. lavage

17. One method of detecting the growth of bacteria is monitoring for:

- a. morbidity
- b. turbidity
- c. light production
- d. condensation

18. What are the main components of concrete:

- a. sand, gravel, glue
- b. sand, lava rock, textile fibers
- c. duct tape, dinosaur bones, library paste, saliva
- d. sand, gravel, cement

19. The German scientist after whom the hardness scale is named is:

- a. Frederich Moz
- b. Frederich Mohs
- c. Moses
- d. Fred Moe

20. Which of the following may be products of a volcano.

- a. tectonic plates
- b. gases
- c. lava
- d. oceans
- e. mudslides
- f. ash

21. Which of the following are thought to be the result of tectonic activity:

- a. tidal waves
- b. tsunamis
- c. earthquakes
- d. Botswana

Provide the missing information.

22. The central part of the earth is called the _____. Which is probably made of magnetic solid and magma. Surrounding this inner part is the thick, dense, crystalline rock we call the _____. The outermost layer is a relatively thin layer of soil, rock and water that we call the _____.
23. A large crack in the crust at the bottom of the ocean is called a _____.
24. Soil is presumed to have formed by the _____ of rock.
25. A _____ is a place where water collects to form a lake or pond.
26. Artificial lakes for holding water, called _____, are formed by damming up a river.
27. An experiment is a test of a _____.
28. What is the hardest natural substance? _____.

mantle, hearth, heart, mantissa, cloak, core, crust, corps, corpus, corpse, filling, crust, crest, crud, fissure, fizzy, trench, wench, crevasse, explosion, erosion, evasion, invasion, incantation, bassoon, basin, sink, sinkhole, lavatory, compoundments, impoundments, hypothesis, theory, suspicion, student, corundum, emerite, diamond, science

Rainbow Quiz 2

Lessons 5, 6, 7, 8, Labs 3, 4

Circle the best response(s) to complete the sentence(s).

1. The dinosaurs are supposed to have first arisen during the (Jurassic, Triassic) period of the Mesozoic era.
2. Modern humans are supposed to have first arisen during the (Recent, Pleistocene) epoch of the Quaternary period of the Cenozoic era.
3. A representative of every vertebrate phylum appears in the (Precambrian era, Cambrian period).
4. Saltier water is (more, less) dense than less salty water.

True or false?

5. Carbon dioxide is a greenhouse gas.
6. Geology is the study of the solar system.
7. Beneath the earth's surface are layers of sediment called strata.
8. Fossil fuels are believed to come from aging of dead plants and animals.
9. Water that has more solid dissolved in it than it can continue to hold is referred to as "supercritical."

Match the following terms with their descriptions:

- | | |
|-----------------------------|--|
| 10. _____ continental shelf | a. 500-meter-thick layer of dead organic matter, mineral matter and shells of dead animals |
| 11. _____ continental rise | b. dives 600 meters to the ocean floor |
| 12. _____ continental slope | c. depth of rapidly-changing water temperature |
| 13. _____ continental drift | d. an underwater mountain range that occurs approximately halfway between continents |
| 14. _____ mid-ocean ridge | e. shallow-sloping border around continents having shallow water above it |
| 15. _____ sediment | f. dives 3500 meters from the continental shelf |
| 16. _____ ocean floor | g. the bottom of the ocean |
| 17. _____ thermocline | h. separation of the continents over time |

*Place the following terms eras of geochronology in their proper order **beginning with the most recent**:*

18. _____ Cenozoic
19. _____ Paleozoic
20. _____ Precambrian
21. _____ Mesozoic

Circle the letter corresponding to the best answer.

22. To remove the salt from water is to:

- a. decant
- b. desaltize
- c. desalinize
- d. disabuse

23. Which is not an example of the greenhouse effect:

- a. the heating of air inside a greenhouse
- b. the heating of air inside your car on a sunny day
- c. the heating of the atmosphere due to accumulation of gases that hold in heat
- d. the heating of air inside your house due to the use of an electric heater

24. Which of the following words is used to describe the study of life history on earth:

- a. geology
- b. mineralogy
- c. paleontology
- d. geochronology
- e. agronomy

25. Which of the following words is used to describe the study of the age of the earth:

- a. geology
- b. mineralogy
- c. paleontology
- d. geochronology
- e. agronomy

26. Which of the following words is used to describe the study of soil for agriculture:

- a. geology
- b. mineralogy
- c. paleontology
- d. geochronology
- e. agronomy

27. The opposite of catastrophism is:

- a. evolutionism
- b. uniformitarianism
- c. Darwinism
- d. geochronology

28. Sea water is approximately what percent salt-saturated? This is equal to what percent salt by weight?

- a. 0.125% saturated; 0.00361% salt
- b. 1.25% saturated; 0.0361% salt
- c. 12.5% saturated; 0.361% salt
- d. 125% saturated; 36.1% salt

29. Magnesium sulfate heptahydrate is the chemical name for:

- a. rock salt
- b. ice cream salt
- c. pickling salt
- d. Epsom salts

Rainbow Quiz 3

Lessons 9, 10, 11, 12, Labs 5, 6

True or false?

1. Ground water is water that fills the pore spaces of soil and rock beneath the ground surface.
2. Soil is always made up of particles that are about the same size.
3. Air that is holding its capacity of water is said to be *saturated*.
4. Commercial jetliners typically travel at the boundary between the thermosphere and exosphere.
5. The main factor affecting motion in the atmosphere is the heating of atmospheric gases by the sun.
6. Natural acids decrease the pH of groundwater, causing it to dissolve minerals.

Match the following weather phenomena with their descriptions:

- | | |
|--------------------|---|
| 7. _____ lightning | a. frozen precipitation that damages crops, houses and cars |
| 8. _____ flood | b. a condensation phenomenon |
| 9. _____ fog | c. generated from static charges |
| 10. _____ hail | d. frozen precipitation with driving wind |
| 11. _____ wildfire | e. usually results from unusually dry conditions |
| 12. _____ blizzard | f. generated from moisture on saturated ground |

Circle the letter corresponding to the best answer.

13. Which word refers to liquid, especially water, that is suitable for drinking:
 - a. potable
 - b. drinkable
 - c. palatable
 - d. portable
14. Which of the following is not a major function of the atmosphere:
 - a. keeps us warm by holding in solar heat
 - b. provides food for us
 - c. protects us from the sun's ultraviolet radiation
 - d. provides us with oxygen in the correct concentration to support life
 - e. brings about our climate and weather patterns

Provide the requested information.

15. Name the five major divisions of the atmosphere in order beginning at the earth's surface.

Rainbow Quiz 4

Lessons 13, 14, 15, 16, Labs 7, 8

Circle the best response(s) to complete the sentence(s).

1. While (climate, weather) has to do with long term conditions on earth, (climate, weather) has to do with day to day changes in those conditions.
2. A(n) (anticyclone, cyclone) is a(n) area of low atmospheric pressure, while a(n) (anticyclone, cyclone) is an area of high pressure.
3. In the northern hemisphere, an anticyclone rotates in a (clockwise, counterclockwise) direction while a cyclone rotates (clockwise, counterclockwise).
4. In North America, most of us live in a (temperate, tropical) climate.

True or false?

5. Prevailing winds are also called “trade winds.”
6. Ocean currents are brought about in part by the effects of prevailing winds.
7. Deforestation results in the loss of wildlife habitat.
8. Cyclones and anticyclones are collectively called climate systems.
9. Jet streams are tubes of rapidly moving air.
10. There are six permanent jet streams (three in each hemisphere)

Match the following climatic features with their descriptions:

- | | |
|--------------------------------|---|
| 11. _____ Hadley cells | a. largely caused by Hadley cells |
| 12. _____ prevailing winds | b. experiences constant low pressure from rising warm air |
| 13. _____ Coriolis effect | c. large masses of air circulating in the atmosphere |
| 14. _____ equator | d. receive moisture that is picked up by air at the equator then dropped north and south of the equator |
| 15. _____ deserts | e. from constant dry air pressing on the surface of the earth |
| 16. _____ tropical rainforests | f. effect of inertia on objects moving across a spinning earth |

Circle the letter corresponding to the best answer.

17. Hadley cells are:

- a. large areas of low pressure that circulate above the earth
- b. large areas of high pressure that circulate above the earth
- c. large areas of circulating air that result from warming at the equator and cooling as the air travels north and south toward the poles
- d. large air masses that loom over the equator

18. Hadley cells do not extend from the equator to the poles because:

- a. the earth rotates, breaking them up into several smaller cells
- b. surface features of the earth break them up
- c. radio waves from outer space cause interference
- d. high pressure battles with low pressure

19. Air pressure is measured using a:

- a. hygrometer
- b. hydrometer
- c. barometer
- d. thermometer

Rainbow Section Review

See Lab Workbook