# Considerations on the risk of precipitation of calcium phosphate salts following the use of Vismed<sup>®</sup> (0.18% sodium hyaluronate) formulation

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D. Aleo <sup>1</sup>, <u>L. Bossy</u> <sup>2</sup>, <sup>1</sup> Medivis R&D Laboratories, Catania, Italy, <sup>2</sup> TRB Chemedica International SA, Geneva, Switzerland, Correspondence: Leila Bossy TRB Chemedica International SA 12 rue Michel Servet 1211 Geneva, Switzerland Tel: +41 22 703 49 00 Email: leila.bossy@trbchemedica.com

## Introduction

- Recently, Bernauer<sup>1,2</sup> et al. showed that exaggerated use (up to 100 times per day) of artificial tears containing high concentration (>50 mM) of phosphate buffer (i.e. Hylo-Comod<sup>®</sup>) in patients with severe epithelial defects may cause corneal calcification due to precipitation of calcium phosphate.
- Vismed<sup>®</sup> is a unique patented formulation that contains sodium hyaluronate (SH) together with essential ions (calcium, magnesium, potassium, sodium and chloride) that are found in natural tears. It also contains sodium citrate to maintain these essential ions in solution and avoid their precipitation<sup>3</sup>.
- \* The presence of citrate in Vismed<sup>®</sup> may also help avoid the precipitation of calcium phosphate.
- The concentration of phosphate buffer in the Vismed<sup>®</sup> formulation is low and close to that found in natural tears, thus reducing the risk of precipitation of calcium phosphate (Table 1).

#### Results

# Precipitation of calcium phosphate in Vismed<sup>®</sup> in the presence of increasing amounts of phosphate

Increasing concentration of phosphate up to 100 mM (+1100%) did not lead to any precipitation (Table 4).

Table 4: Precipitation of calcium phosphate in Vismed<sup>®</sup> in the presence of increasing amounts of phosphate

lon	Vismed (mM)	Increase of phosphate (mM)							
Phosphate	8.99	17.98	26.97	35.96	44.95	53.94	71.92	98.89	
Calcium	0.61	Constant							
% increase	-	200	300	400	500	600	800	1100	
Precipitate	-	NO	NO	NO	NO	NO	NO	NO	

 Table 1: Concentrations of ions in natural tears and Vismed<sup>®</sup>

lon	Concentration (mM)					
	Tears	Vismed®				
Phosphate	8.0	8.99				
Calcium	0.3 - 2.4	0.61				
Magnesium	0.3 - 1.1	0.45				
Potassium	15 - 35	13.9				
Citrate	_	1.00				
Sodium	108 - 175	65.7 (hypotonic)				
Chloride	102 - 168	63.6 (hypotonic)				

**Study objectives** 

# **Precipitation of calcium phosphate in Vismed<sup>®</sup> in the presence of increasing amounts of calcium**

The addition of calcium up to 2.44mM (+400%) to the Vismed<sup>®</sup> formulation did not lead to any precipitation (Table 5). A precipitate was visible at 3.05 mM (+500%) of calcium and above.

Table 5: Precipitation of calcium phosphate in Vismed<sup>®</sup> in the presence of increasing amounts of calcium

lon	Vismed (mM)	Increase of calcium (mM)								
Phosphate	8.99		Constant							
Calcium	0.61	1.22	1.83	2.44	3.05	3.66	4.27	4.88		
% increase	-	200	300	400	500	600	700	800		
Precipitate	-	NO	NO	NO	YES	YES	YES	YES		

The aim of this study was to evaluate *in vitro* the precipitation of calcium phosphate salts in the Vismed<sup>®</sup> formulation in presence of increasing amount of phosphate and/or calcium.

### **Methods**

To evaluate the risk of precipitation of calcium phosphate in Vismed<sup>®</sup>, increasing amounts of phosphate  $(Na_2HPO_4)$  were added to the orignal Vismed<sup>®</sup> formulation. The concentrations of phosphate added are shown in Table 2. The presence of calcium phosphate salts precipitate was assessed visually using a microscope.

#### Table 2: Increasing amount of phosphate added to Vismed<sup>®</sup>

lon	Vismed (mM)	Increase phosphate (mM)							
Phosphate	8.99	17.98	26.97	35.96	44.95	53.94	71.92	98.89	
% increase	-	200	300	400	500	600	800	1100	

A subsequent trial was carried out by adding increasing amounts of calcium (CaCl<sub>2</sub>) to the original Vismed<sup>®</sup> formulation, as shown in Table 3. The concentration of phosphate was kept constant (8.99 mM). The presence of calcium phosphate precipitate was assessed visually using a microscope.

#### Table 3: Increasing amount of calcium added to Vismed®

lon	Vismed (mM)	Increase calcium (mM)
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# Precipitation of calcium phosphate in Vismed<sup>®</sup> in the presence of increasing amounts of phosphate and constant amount of calcium

At a concentration of 54 mM (+600%) of phosphate and above, a precipitate was visible (Table 6).

Table 6: Precipitation of calcium phosphate in Vismed<sup>®</sup> in the presence of increasing amounts of phosphate and constant amount of calcium

lon	Vismed (mM)		Increase calcium and phosphate (mM)						
Calcium	0.61	1.22	1.22	1.22	1.22	1.22	1.22	1.22	
Calcium % increase	-	200	200 Constant						
Phosphate	8.99	8.99	17.98	26.97	35.96	44.95	53.94	62.93	
Phosphate % increase	-	-	200	300	400	500	600	700	
Precipitate	-	NO	NO	NO	NO	NO	YES	YES	

### **Discussion and conclusion**

In conclusion, it is unlikely that the repeated use of Vismed<sup>®</sup> will cause corneal calcification *in vivo*, since:

✤ Increased amount of phosphate up to 100 mM (+1100%) and calcium up to 2.44 mM (+400%) did not

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Calcium	0.61	1.22	1.83	2.44	3.05	3.66	4.27	4.88
% increase	-	200	300	400	500	600	700	800

A final experiment was performed by both, adding twice (*i.e.* 1.22 mM) the initial concentration of calcium in Vismed<sup>®</sup> and increasing the amount of phosphate up to 63 mM (+700%). The presence of calcium phosphate precipitate was assessed visually using a microscope.

- lead to any precipitation in Vismed<sup>®</sup>.
- Vismed<sup>®</sup> contains 5 times less phosphate (8.99 mM) than commercially available products like Hylo-Comod<sup>®</sup>.
- The results of several clinical trials in a total of more than 300 patients in patients with moderate dry eye, together with post marketing surveillance show a very positive result. Vismed<sup>®</sup> has never been implicated in such side effects to date. This shows that it is safe to use over time.

These results may be attributed to the presence of sodium citrate, that maintains the ions phosphate and calcium in solution. Further experiments are ongoing to confirm this hypothesis.

### References

- 1. Bernauer W, Thiel MA, Kurrer M et al. Corneal calcification following intensified treatment with sodium hyaluronate artificial tears. Br J Ophthalmol 2006; 90:285-8.
- 2. Bernauer W, Thiel MA, Langenauer UM, Rentsch KM. Phosphate concentration in artificial tears. Graefes Arch Clin Exp Ophthalmol 2006; 1-5.

3. Berg C, Tiselius HG. The effects of citrate on hydroxyapatite induced calcium oxalate crystallization and on the formation of calcium phosphate crystals. Urol Res 1989; 167-172.