Air Cleaner Test Report

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1. Sample Description

Product : Air Cleaner

Brand Name : b-MOLA

Model No. : BM20

No. of Sample Received : 1

Test Date : 26 May 2022 – 26 May 2022

Test Item(s) : Pollutants Removal Efficiency

Test Requested : Tetrachloroethylene

Test Reference(s) : In-house method SOP200 (for VOC removal rate)

Test Equipment : Honeywell instrument ppbRAE 3000

Equipment no. : E002 - 003

Test Result : See the attached sheets

Remark : N/A

2. Detailed Description of the sample





b-MOLA/BM20

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NCCO Filter and HEPA

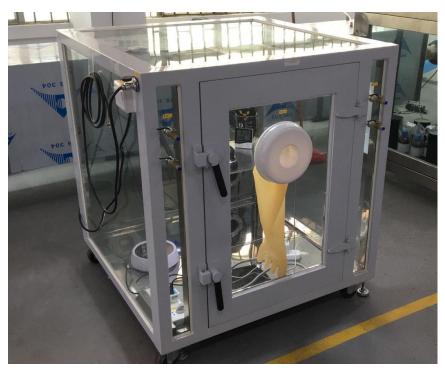
Testing Environment

: 26.4 °C Temperature

: 48 % **Relative Humidity**

: 1m³ Testing Chamber **Testing Chamber**

Size $(W \times H \times D)$ mm : $1000 \times 1000 \times 1000$



1m³ Testing Chamber

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Testing Method of Removal Efficiency

In a 1m³ chamber, chemical was injected into the chamber by a syringe and evaporated by a

hot plate. Internal circulation was turned on throughout the test to ensure the uniformity of

chemical concentration inside the chamber. Initial concentration (C₀) of the chemical was recorded

before switching on the air cleaner with a range of 100 (±10) mg/m³. Then, the air cleaner is

switched on for 60 minutes and the chemical concentration was recorded as C₆₀, the final

concentration of chemical.

The test was repeated without the air cleaner to determine the natural decay of the chemical

at the test chamber. Chemical was injected into the chamber by a syringe and evaporated by a hot

plate with an initial concentration (C_{N0}). The final concentration (C_{N60}) was determined 60 minutes

later.

New filters and HEPA have been used for each chemical test.

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under the Entrepreneurship Program

5. Results of Removal Efficiency

Brand/ Model No.	Operation Mode	Test Chemical	Volume of use
			(mL)
b-MOLA/ BM20	Н	Tetrachloroethylene	0.12

Initial Concentration	Natural Decay, kn	Total Decay, ke	Removal Efficiency
(mg/m^3)	(min ⁻¹)	(min ⁻¹)	(%)
105.8	0.0034	0.0391	91

Remark: Initial concentration is set within 100±10mg/m³.

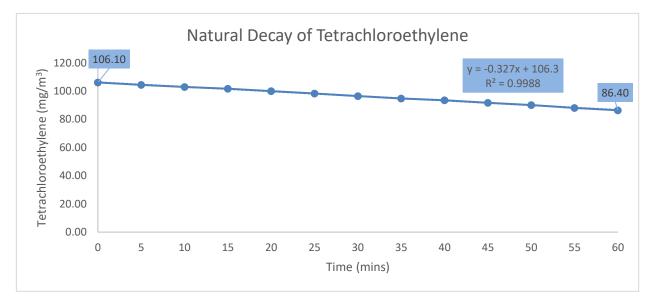


Figure a. Natural Decay of Tetrachloroethylene

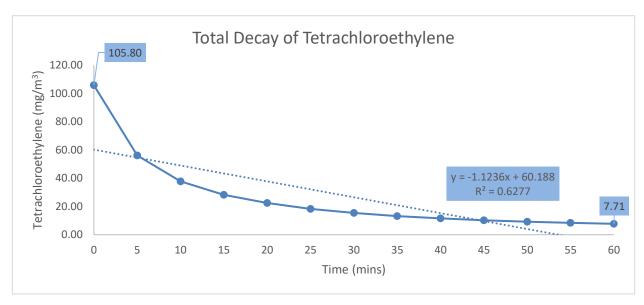


Figure b. Total Decay of Tetrachloroethylene

Calculation:

$$A_2 = \frac{C_{N0} - C_{N60}}{C_{N0}}$$

$$R = \frac{C_0(1 - A_2) - C_{60}}{C_0(1 - A_2)}$$

R: Removal efficiency (%)

A₂: Natural decay rate (%)

C: Concentration of testing subject (mg/m³)

*** End of Report ***

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