Air Cleaner Test Report

Applicant : RHT Industries Limited

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1. Sample Description

Product : Air Cleaner

Brand Name : b-MOLA

Model No. : MOLA300

No. of Sample Received : 1

Test Date : 12 Nov 2019 – 12 Nov 2019

Test Item(s) : Pollutants Removal Efficiency

Test Requested : Benzene

Test Reference(s) : In-house method SOP200 (for VOC removal rate)

Test Equipment : Honeywell instrument ppbRAE 3000

Equipment no. : E002 - 001

: See the attached sheets Test Result

Remark : Client claimed that model MOLA300 same as IA60/BM300 in

terms of power, parts, components and structures. Only

different is the selling platform.

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2. Detail Description of the sample





b-MOLA/MOLA300

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NCCO Reactor (NA213020300) and Activated Carbon HEPA

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3. Testing Environment

Temperature : 26.2 °C

Relative Humidity : 49 %

Testing Chamber : 3m³ Testing Chamber

Size $(W \times H \times D)$: $1.40m \times 1.40m \times 1.50m$



3m³ Testing Chamber

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4. Testing Method of Removal Efficiency

In a 3m^3 chamber, chemical was injected into the chamber by a syringe and evaporated by a hot plate. Internal circulation was turned on throughout the test to ensure the uniformity of chemical concentration inside the chamber. Initial concentration (C_0) of the chemical was recorded before switching on the air cleaner with a range of $100 \ (\pm 10) \ \text{mg/m}^3$. Then, the air cleaner is switched on for 60 minutes and the chemical concentration was recorded as C_{60} , the final concentration of chemical.

The test was repeated without the air cleaner to determine the natural decay of the chemical at the test chamber. Chemical was injected into the chamber by a syringe and evaporated by a hot plate with an initial concentration (C_{N0}). The final concentration (C_{N60}) was determined 60 minutes later

New filters and HEPA have been used for each chemical test.

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5. Results of Removal Efficiency

Brand/ Model No.	No. Operation Mode Test Chemical		Volume of use	
			(mL)	
B-MOLA/MOLA300	SS	Benzene	1.00	

Initial Concentration	Natural Decay, kn	Total Decay, ke	Removal Efficiency	
mg/m ³	(min ⁻¹)	(min ⁻¹)	(%)	
100.60	0.001	0.0547	>99	

Remark: Initial concentration is set within 100±10mg/m³.

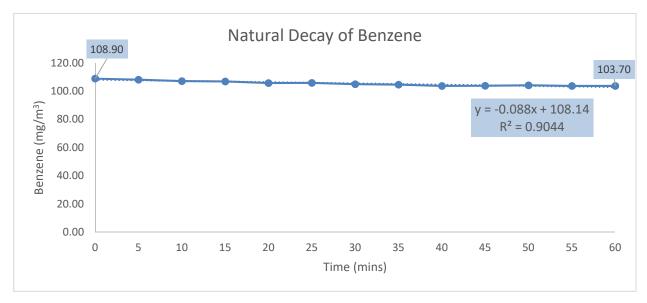


Figure a. Natural Decay of Benzene

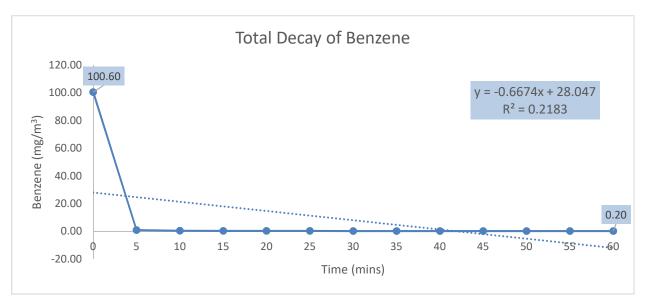


Figure b. Total Decay of Benzene

Calculation:

$$A_{1} = \cfrac{C_{0} - C_{60}}{C_{0}}$$

$$A_{2} = \cfrac{C_{N0} - C_{N60}}{C_{N0}}$$

$$C_{0}(1 - A_{2}) - C_{60}$$
 Removal Efficiency =
$$\cfrac{C_{0}(1 - A_{2})}{C_{0}(1 - A_{2})}$$

A₁: Removal rate (%)

A₂: Natural decay rate (%)

C: Concentration of testing subject (mg/m³)

End of Report

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