## **Air Cleaner Test Report**

Applicant : RHT Industries Limited

Address : Block B, 2/F, Goodwill Industrial Building, No. 36-44 Pak Tin Par

Street, Tsuen Wan, New Territories, Hong Kong

Application Number : KJ191002-16

Report Number : REPAP19113301

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## 1. Sample Description

Product : Air Cleaner

Brand Name : b-MOLA

Model No. : MOLA150

No. of Sample Received : 1

Test Date : 29 Nov 2019 – 29 Nov 2019

Test Item(s) : Pollutants Removal Efficiency

Test Requested : Toluene

Test Reference(s) : In-house method SOP200 (for VOC removal rate)

Test Equipment : Honeywell instrument ppbRAE 3000

Equipment no. : E002 - 002

Test Result : See the attached sheets

Remark : Client claimed that model MOLA150 same as IA50/BM150 in

terms of power, parts, components and structures. Only

different is the selling platform.

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## 2. Detail Description of the sample





b-MOLA/MOLA150

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NCCO Reactor (NA213020300) and Activated Carbon HEPA

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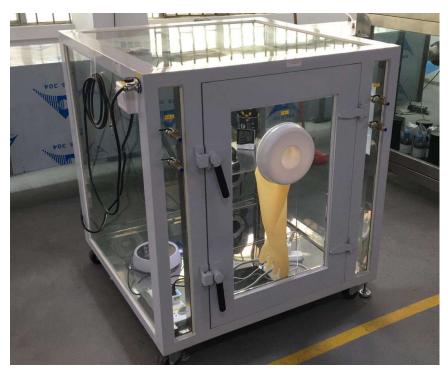
## **Testing Environment**

Temperature : 23.2 °C

Relative Humidity : 42 %

: 1m<sup>3</sup> Testing Chamber **Testing Chamber** 

Size  $(W \times H \times D)$  mm :  $1000 \times 1000 \times 1000$ 



1m<sup>3</sup> Testing Chamber

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**Testing Method of Removal Efficiency** 

In a 1m3 chamber, chemical was injected into the chamber by a syringe and evaporated by a

hot plate. Internal circulation was turned on throughout the test to ensure the uniformity of

chemical concentration inside the chamber. Initial concentration (C<sub>0</sub>) of the chemical was

recorded before switching on the air cleaner with a range of 100 (±10) mg/m<sup>3</sup>. Then, the air

cleaner is switched on for 60 minutes and the chemical concentration was recorded as C<sub>60</sub>, the

final concentration of chemical.

The test was repeated without the air cleaner to determine the natural decay of the chemical

at the test chamber. Chemical was injected into the chamber by a syringe and evaporated by a

hot plate with an initial concentration (C<sub>N0</sub>). The final concentration (C<sub>N60</sub>) was determined 60

minutes later

New filters and HEPA have been used for each chemical test.

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## 5. Results of Removal Efficiency

| Brand/ Model No. | <b>Operation Mode</b> | Test Chemical | Volume of use |
|------------------|-----------------------|---------------|---------------|
|                  |                       |               | (mL)          |
| b-MOLA/MOLA150   | SS                    | Toluene       | 0.185         |

| Initial Concentration | Natural Decay, kn    | Total Decay, ke      | Removal Efficiency |
|-----------------------|----------------------|----------------------|--------------------|
| $(mg/m^3)$            | (min <sup>-1</sup> ) | (min <sup>-1</sup> ) | (%)                |
| 107.30                | 0.003                | 0.056                | > 99               |

Remark: Initial concentration is set within 100±10mg/m<sup>3</sup>.

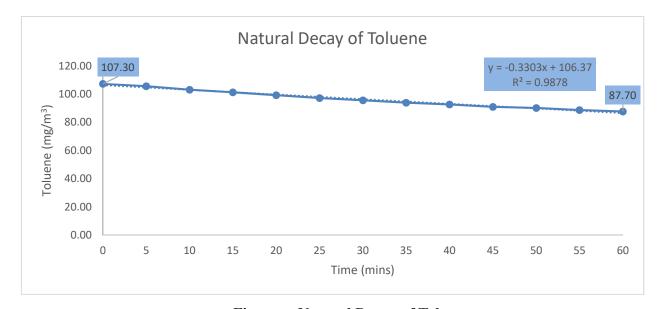


Figure a. Natural Decay of Toluene

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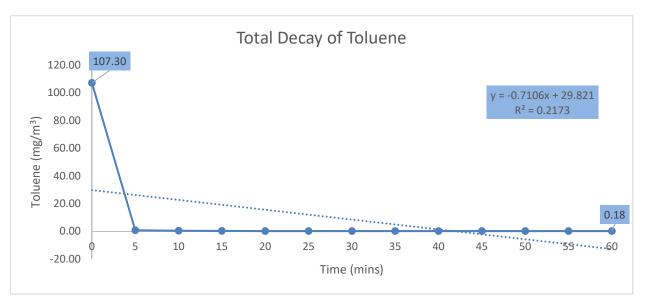


Figure b. Total Decay of Toluene

### Calculation:

$$A_{1} = \cfrac{C_{0} - C_{60}}{C_{0}}$$
 
$$A_{2} = \cfrac{C_{N0} - C_{N60}}{C_{N0}}$$
 
$$C_{0}(1 - A_{2}) - C_{60}$$
 Removal Efficiency = 
$$\cfrac{C_{0}(1 - A_{2})}{C_{0}(1 - A_{2})}$$

A<sub>1</sub>: Removal rate (%)

A<sub>2</sub>: Natural decay rate (%)

C: Concentration of testing subject (mg/m<sup>3</sup>)

\*\*\*End of Report\*\*\*

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