# **Air Cleaner Test Report**

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Application Number : KJ2005002 - 09

Report Number : REPAP20064401

Report Issue Date : 19 Jun 2020

Total Page : 8 Pages (including this page)

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## 1. Sample Description

Product : Air Cleaner

Brand Name : b-MOLA

Model No. : BM100

No. of Sample Received : 1

Test Date : 17 Jun 2020 – 17 Jun 2020

Test Item(s) : Pollutants Removal Efficiency

Test Requested : Benzene

Test Reference(s) : In-house method SOP200 (for VOC removal rate)

Test Equipment : Honeywell instrument ppbRAE 3000

Equipment no. : E002 - 001

Test Result : See the attached sheets

Remark : Client claimed that model BM100 same as IA30, IA1019 and

IA1019S. Only difference are the selling platforms.

## 2. Detail Description of the sample



b-MOLA/BM100



NCCO Reactor (NA213020300) and Activated Carbon HEPA

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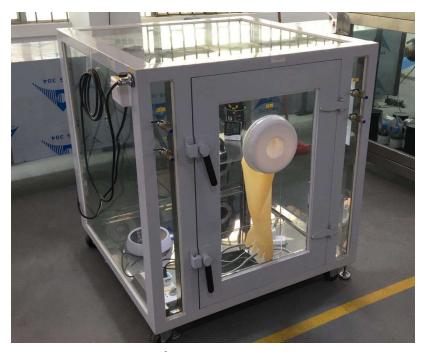
# 3. Testing Environment

Temperature : 26.3°C

Relative Humidity : 42%

Testing Chamber : 1m<sup>3</sup> Testing Chamber

Size (W × H × D) mm :  $1000 \times 1000 \times 1000$ 



1m<sup>3</sup> Testing Chamber

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4. Testing Method of Removal Efficiency

In a  $1\text{m}^3$  chamber, chemical was injected into the chamber by a syringe and evaporated by a hot plate. Internal circulation was turned on throughout the test to ensure the uniformity of chemical concentration inside the chamber. Initial concentration ( $C_0$ ) of the chemical was recorded before switching on the air cleaner with a range of  $100 \ (\pm 10) \ \text{mg/m}^3$ . Then, the air cleaner is switched on for  $60 \ \text{minutes}$  and the chemical concentration was recorded as  $C_{60}$ , the final

concentration of chemical.

The test was repeated without the air cleaner to determine the natural decay of the chemical at the test chamber. Chemical was injected into the chamber by a syringe and evaporated by a hot plate with an initial concentration ( $C_{N0}$ ). The final concentration ( $C_{N60}$ ) was determined 60 minutes later

New filters and HEPA have been used for each chemical test.

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## 5. Results of Removal Efficiency

Brand/ Model No.	Operation Mode	Test Chemical	Volume of use
			(mL)
b-MOLA/ BM100	SS	Benzene	0.41

Initial Concentration	Natural Decay, kn	Total Decay, ke	Removal Efficiency	
$(mg/m^3)$	(min <sup>-1</sup> )	(min <sup>-1</sup> )	(%)	
106.80	0.002	0.110	>99.9	

Remark: Initial concentration is set within 100±10mg/m<sup>3</sup>.

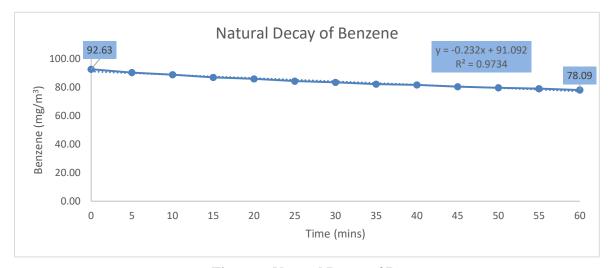


Figure a. Natural Decay of Benzene

IAQ Contractor, IAQ Control Facilities Supplier, IAQ Consultant Subsidiary company of the Hong Kong University of Science and Technology under the Entrepreneurship Program

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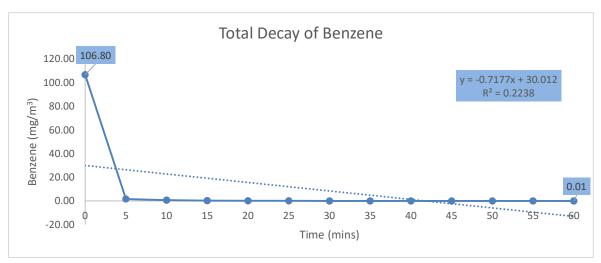


Figure b. Total Decay of Benzene

#### Calculation:

$$A_{1} = \cfrac{C_{0} - C_{60}}{C_{0}}$$
 
$$A_{2} = \cfrac{C_{N0} - C_{N60}}{C_{N0}}$$
 
$$C_{0}(1 - A_{2}) - C_{60}$$
 Removal Efficiency = 
$$\cfrac{C_{0}(1 - A_{2})}{C_{0}(1 - A_{2})}$$

A<sub>1</sub>: Removal rate (%)

A<sub>2</sub>: Natural decay rate (%)

C: Concentration of testing subject (mg/m<sup>3</sup>)

\*\*\* End of Report \*\*\*

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