# Topper's Answers

## C.B.S.E. 2019

### Science

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## Class–X Delhi/Outside Delhi Set

Time: 3 Hours Max. Marks: 80

#### **General Instructions:**

- (i) The question paper comprises five Sections, A, B, C, D and E. You are to attempt All the sections.
- (ii) All questions are Compulsory.
- (iii) Internal choice is given in Sections B, C, D and E.
- (iv) Questions number 1 and 2 in Section A are one-mark questions. They are to be answered in one word or in one sentence.
- (v) Questions number 3 to 5 in Section B are two-marks questions. These are to be answered in about 30 words each.
- (vi) Questions number 6 to 15 in Section C are three-marks questions. These are to be answered in about 50 words each.
- (vii) Questions number 16 to 21 in Section D are five-marks questions. These are to be answered in about 70 words each.
- (viii) Questions number 22 to 27 in Section E are based on practical skills. Each question is a two-marks question. These are to be answered in brief.

#### SECTION - A

1. State the S.I. unit of potential difference and name the device used to measure it.

Ans. The S.I. unit of potential difference is worth in honour of Alessandero Valta.

Voltmeter is a device that is always connected in parallel in a circuit to measure the voltage.

2. List two characteristics of a good fuel.

Ans. 2. A good fuel is the one that 
1) does a large amount of work per unit volume or mass

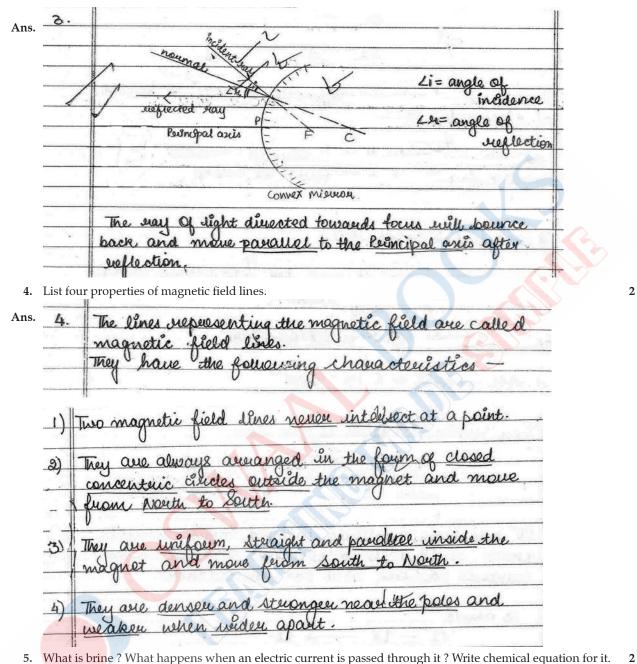
i.e. it has high calouific volue.

3) has a moderate ignition temperature.

3) is easy to store and transport.

#### SECTION – B

3. Draw a labelled ray diagram to show the path of the reflected ray corresponding to the ray which is directed towards the principal focus of a convex mirror. Mark the angle of incidence and angle of reflection on it.



OR

List the changes that are observed when dil. HCl is added to a small amount of copper oxide in a beaker. Write

List the changes that are observed when dil. HCl is added to a small amount of copper oxide in a beaker. Write balanced chemical equation for the reaction.

s. <b>B.</b>	buine is the cold and concentrated solution of looking
	chlouide.
_3)_	when electricity is passed through it, the Nace
	breaks to give lans in the solution.  Nac electricity, Nat + a
	Sodium Sodium Chlouide Chlouide ions

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	Then water also splits to give ions.
	H20 electricity H++OH-
	water hudespen Hyderanide
	ions. ions
9	Ch ions being negatively charged moves towards
	and and Hobeing positively charged moves
	towards cathode and are collected there.
	At anode
	$\alpha^ 1\bar{e} \longrightarrow \alpha$
	$a + a \longrightarrow a_2$
	culouine gas
	At carnoole
	$H^+ + 1\bar{e} \longrightarrow H$
	$H + H \longrightarrow H_2$
	Hydrogen gas
	The Nat & OH ions are left in solution
	which combine to give NaOH (Sookin Hydroxide)
_	NaCl +2HOO electricity > 2NaOH + H2^ + Cl2 1
2 noti	
Juda	houide water Sodium regarden chroning
	Buine

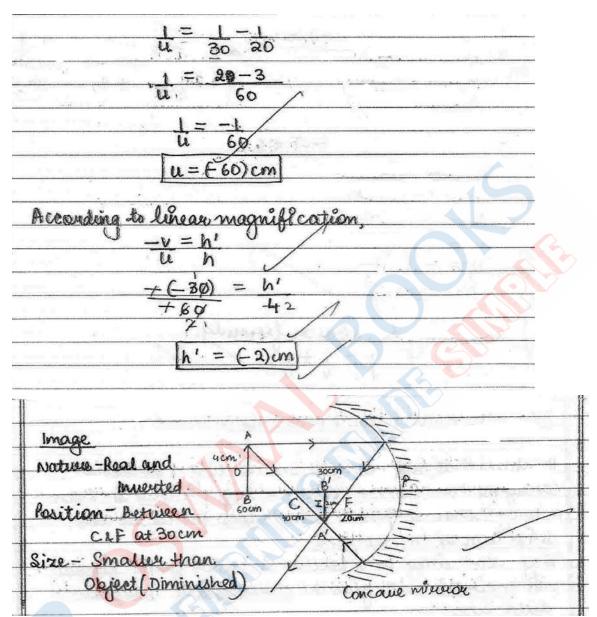
#### SECTION - C

6. A concave mirror has a focal length of 20 cm. At what distance from the mirror should a 4 cm tall object be placed so that it forms an image at a distance of 30 cm from the mirror? Also calculate the size of the image formed. 3

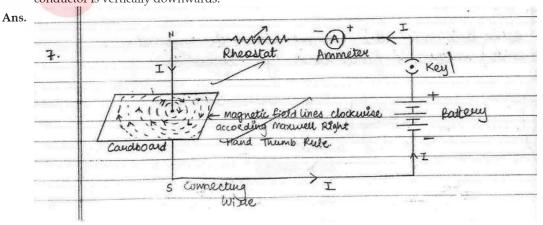
A real image 2/3rd of the size of an object is formed by a convex lens when the object is at a distance of 12 cm from it. Find the focal length of the lens.

н. гиа	the focal length of the lens.
6.	Concare marrie
ins.	f = -20  cm
	h = +4  cm
	v = -30  cm
	u = -?
	h'= -?
Fy-3	
- 1	According to mereon formula,
	1=1+1
1	fvu
	-1 = -1 + 1
	20 30 U
	-1 + 1 = 1
	20 30 U

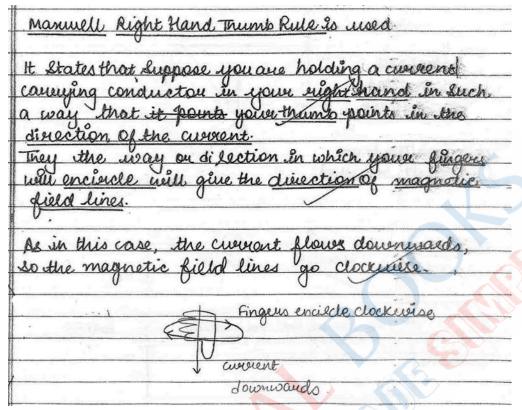
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7. Draw the pattern of the field lines of the magnetic field around a current carrying straight conductor passing through and held perpendicular to a horizontal cardboard. State right-hand thumb rule and explain how this rule is useful to determine the direction of the magnetic field in the above case, if the direction of current in the conductor is vertically downwards.



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8. An ore on treatment with dil. HCl gives the smell of rotten egg. Name the type of this ore. How can the metal be obtained from its concentrated ore?

Smell of notten eggs is usually penduced by Sulphane.

i. It must be a sulphide one.

let us suppose that the one is Tinc blende (ZnS).

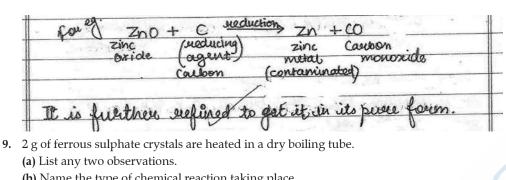
Concentration of one

First of all, the one is concentrated by the method of furth floatation.

Convenien into metal onide.

Then it is mostled convented into its metal onide via possting in supply of excess air.

Fourth The Concentration of the conc



- - (b) Name the type of chemical reaction taking place.
  - (c) Write balanced chemical equation for the reaction and name the products formed.

You might have noted that when copper powder is heated in a china dish, the reddish brown surface of copper powder becomes coated with a black substance.

(a) Why has this black substance formed?

Ans.

(b) What is this black substance?

(c) Write the chemical equation of the reaction that takes place.

026) Ans. Copper oxid flom black)

(d) How can the black coating on the surface be turned reddish brown?

This black coating can be

CH(5)+ H2O(4) water oppen

10. A white powder is added while baking cakes to make it soft and spongy. Name its main ingredients. Explain the function of each ingredient. Write the chemical reaction taking place when the powder is heated during baking. 3

The white powder added to cakes is baking Ans. Baking bourden is prepared with Tautacic acid is mainly for moducing effectivescence effect of sodium sicaresmate.

2NaHCO3 -> Na2CO3 + CO2+ + H2O(6)

Baking Soda Socium Carbon Water
in form of Carbonate dioxide (vapours)

Baking powder.

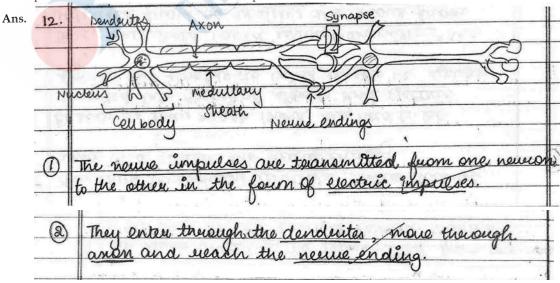
The carbon dioxide areleased gets trapped in the care layers and makes it soft and spongy.

11. List in tabular form three differences between blood and lymph.

	l >	Blood	Lymph
		It is a fluid connective	1. It is an entra cellular
	l to	tissue consisting of plasma	fluid consisting of
	_	Red blood cells, white	plasma, puoteins,
-		blood cells, platelets etc.	white blood cells etc.
	9.	It contains ned blood	a. It egopie coloure med
L .		cours and is eved in	blood celts and is
	19-1	colour due to	The state of the s
_		presence of harmoglosin	. therefore colourless.
3	1	Il contains many puntains	3. It has lessen puoteins.

12. Why does the flow of signals in a synapse from axonal end of one neuron to dendritic end of another neuron take place but not in the reverse direction? Explain.

It courses absorbed



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As soon as they reach there, they need to be converted into chemical signals and electric signals are unable to cross the gap i.e. symapse.
signals are unable to cross the gap i.e. synapse.
Thus, chemicals called newotransmitters are released from the resides that more from neure ending of one newson to the denotrite of the next one.
released from the resides that more from
neure ending of one never to the denduite of
the next one.
As soon as the Chemicals reach the dendritic top, they again get connected to electerial impulses and more on.
they again get connected to electerial impulsor
and more on.
THE REPORT OF THE PROPERTY OF THE PARTY OF T
They chemicals are released and more only in one direction and do not rieverse increder to
one direction and do not sieverse inorder to
reach their destination that can be spinal
(They are vieleased in one direction)
(

13. How does the creation of variations in a species promote survival? Explain with the help of an example.

Ans. 13. (1) Populations of organisms live in well defined places or niches.

They are notive to that area. The same

(3) Reprioduction Involves DNA replication which can generate everous that is the main source of variations.

(4) This Consistency of DNA is responsible for maintainence of body design.

(8) fou e.g. Bacterias are one of the oldest surving organisms.

Due to variations in their body designs, they ap adapt most of the inhospitable climates like deserts, ice etc.

They would further expunduce and survive in nature

14. What is a food chain? Why is the flow of energy in an ecosystem unidirectional? Explain briefly.

OR

(a) Why should National Parks be allowed to remain in their pristine form?

**(b)** Why is reuse of materials better than recycling?

s. 14. ([	one consumes the other to transfer energy.
	Cura
	foreg.
-	Plant Greass 10% House 10% Lion
	(levoduces) 90% (neubinoue) 90% (carninoue).
100	(Issuance).
<u> </u>	The flow of energy in an ecosystem is always -
	unidirectional as they cannot vever back
	the energy consumed our lost in emissonment.
~	
(3)	Pari enample, plants cannot evenent back the chemical
	everigiz anto solar energy.
-	
(4)	Since they more purguesserety from one trophe
	leve to the other, the energy content goes on
	decreasing according to 10%. law.
6	
0	They do not have that much energy to evereuse the
	genelvenen if they want to
	aueg
	out eg
	Plant 10.1. Cow 10% Tiger
	(Rusdutou) & (herbinore) Jao 7. (couri none)
	1000 ] (000 ]
	Top Countraise
	Caeninoues Flow of energy
	Herbinores Flow of energy Peroduceus for ecosystem
	Peroduceus
The state of	Sunlight

15. Mention the environmental consequences of the increasing demand for energy. List four steps you would suggest to reduce the consumption of energy.

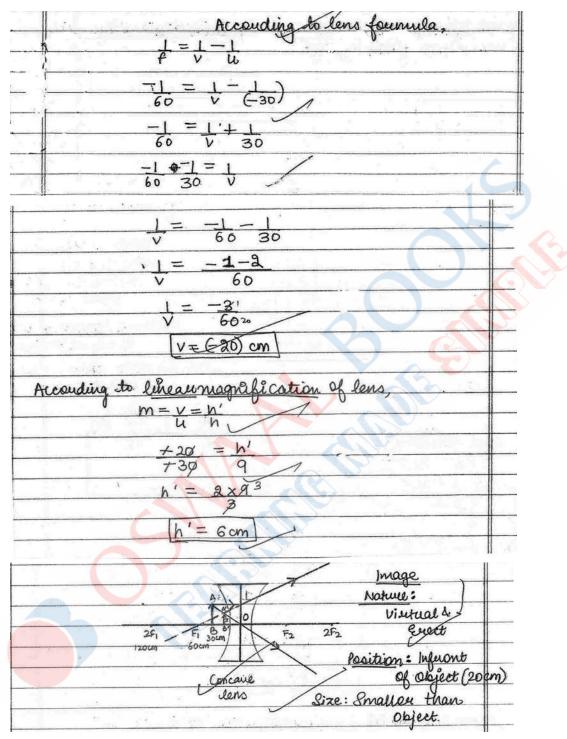
10.	& since the	ce industrial	ulustution,	you the devel	opineia
	of living	ve undustrial Standauds has also inc	of people.	the demand	1 fore
	Burney	Jan alan Sua	and and	10	0

<u></u>	This energy is mainly met by fossilfucts like coal and petroleum.
(3)	They have the following effects on environment:
i)	Incue ased pollution of ain, water, soil due to puesence of oxides of caebon, nitrogén, sulphur.
ii°)	Coal contains caupon which leads to emmission of caupon dioxide incheasing green house effect.
3/1)	This leads to global warning.
_tv)_	They also couse acid rain that convodes marble on our heuitage, and affects plants lije.
	ō Reduce their consumption:
1) 4	ve must use attennatives like Gron which are cleaner
a) u	le must develop healthy habits like using public ranspout instead of planate.
3) u	le must havness non-conventional sources of energy.
4) u	nol use them judiclously.

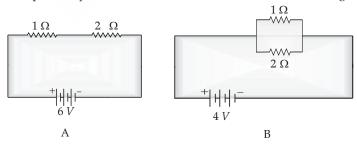
#### SECTION – D

16. A concave lens of focal length 60 cm is used to form an image of an object of length 9 cm kept at a distance of 30 cm from it. Use lens formula to determine the nature, position and length of the image formed. Also draw labelled ray diagram to show the image formation in the above case.

Ans.	16.	Concave lens
		$f = -60  \mathrm{cm}$
		h' = 9cm
		f = - 60 cm
		h'=+: ?
		h = + 9 cm
		$u = -30  \mathrm{cm}$
		V = - ?



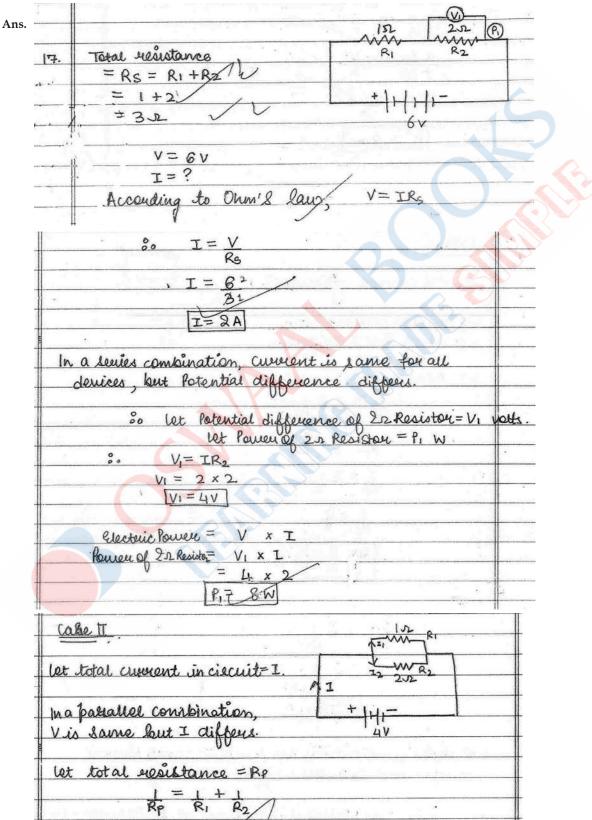
17. Compare the power used in  $2\Omega$  resistor in each of the following circuits :



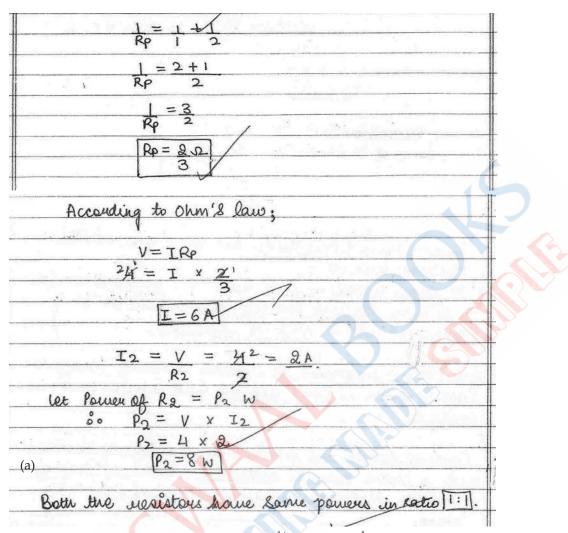
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#### OR

A bulb is rated 40 W; 220 V. Find the current drawn by it, when it is connected to a 220 V supply. Also find its resistance. If the given bulb is replaced by a bulb of rating 25 W; 220 V, will there be any change in the value of current and resistance? Justify your answer and determine the change.



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- 18. (a) State the reason why carbon can neither form C<sup>4+</sup> cations nor C<sup>4-</sup> anions, but forms covalent bonds. Also state reasons to explain why covalent compounds5
  - (i) are bad conductors of electricity.
  - (ii) have low melting and boiling points.

Ans.

(auton has a tetravalency but can't gain on losse electrons because — .

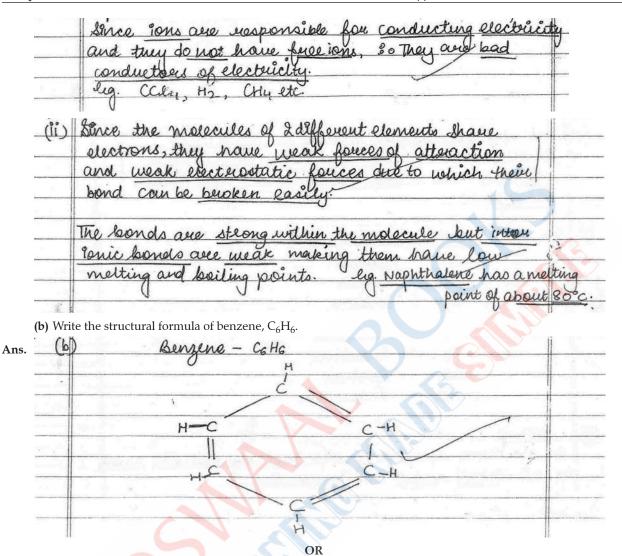
(b) If it gains 4 electrons, it with C<sup>-4</sup> negative charge.

It is very difficult for 4 protons to hold on to 8 electrons and it becomes unstable.

(a) If it loses 4 electrons, it evapores a lot of energy to lose it which it can't afford and again pecomes unotable.

(i) Covalent bonds are bounded by sharing of electrons and a chave them as molecules, so no no teransfer of electrons of electrons of involvement of ions takes place.

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- (a) Define the term 'isomer'.
- (b) Two compounds have same molecular formula  $C_3H_6O$ . Write the name of these compounds and their structural formula.
- (c) How would you bring the following conversions:
  - (i) Ethanol to ethene
  - (ii) Propanol to propanoic acid
- 19. (a) Define groups in the Modern Periodic Table. How do valency, atomic size and metallic character vary in a group?

Ans.	19.0	The vertical columns present in the modern	-
	a)	Pourodio table having elements with similar	-
		watervier and a chemical properties are called	-
		Gracups.	+
	6	Company and Company	
	(2)	For eg. Alkali metals Group 1 H - Huduogen	123
		Li cithuin	
		Na Sodium	-
		K Potassium	1

Ans.

	Rb Rubidium	81
	Cs Cesium	_
	Fu Francium	<del>(0 m</del>
6	Valency	
9	As me more down a group, natercy remains same.	_
	As me move down a quoup, naterry remains same.  eg. all alkali metals are monovalent.	-
(4)	Atomicsixe	
	Atomic size is the distance from the contre of nucleus	-
	to the last shell.	
	As we move down a guoup, the atomic size incurases	
	As we move down a guoup, the atomic size increases as number of shells increase.	
	leg. H-> &1	
	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	
(3)	mobility. Chance tar	
(5)	As we go down a guoup, the Chemical reactivity of	
	metals decreases increases as it is easier for a	
	bigger atom to lose electrons due to weaken	1
	electrostatic forces. lig. Francium is more metallic tha	h
	But in non-metals, it decueases as it is easier	-
	for a smaller atom on top to gain electrons due.	-
	Eg. Flouvine is more non-metallic fran Astanine.	
	de l'annue	

(b) The atomic number of an element is 14. Examine if this element will have metallic properties or not. Give reason to justify your answer.

5

Atomic mumber = 14

Electronic configuration = 2, 8, 4

Element = Silicon

This element has seni properties of both

netals and non-netals,

o It is a metalloid.

It has 4 valence electrons due to which

it can neither gain nou lose electrons as it

would become unotable.

o It founs consent bonds.

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•		do the following provide evidences in favour of evolution in organisms? Explain with an example
	each	i. Homologous organs
		Analogous organs
		Fossils
ns.	20.	I Homologous organs
		The state of the s
	1.	They are the sugans that are similar in structure but
	1.	have been modified to perform different functions.
		7 38 170
	20: 9	
		Tourist of ratification, garage
	-	fring uses it to hop and as a shock absorber, humans use
		it to write, hold etc. and travels usent to cueep on wells.
		to the second se
	12.5	They emplois that maybe the reptiles, mammals and
		amphibians had common ancestors and evolved to be in
	-	the present four.
		Analogous Organs
	0	They made disperent store smeetures and appearance but
		heukoum similar function Ans.
		Egirds have feathlus and for the following body.
	(2)	for eg. usings of a bat and a bind have different structure;
	0	don't peupoum function of flying. [ Weat - Thin flaplike,
	(3)	They do not show common ancastory but tocce sublition due.
		to which they inherited the became capable of flying.
	(iii)	fossils
	0	They are the eremains of plants and animals found
		under earth that lived in exempte past.
		The state of the s
	0	The state of the s
	(8)	They toll us about the evolutionary relationships
		in the past.
	3	Four eg- Auchaeoptoup has Cranjum and wings
		similar to birids but claves and beak similar to
	and the same of	reptiles
	-	
	(4	The also thouse healthing between westiles and area
	- (9	0 -
	-	ou maybe ares evoluting from leptiles
	(b) Expl	ain two methods to determine the age of fossils.

(2)	as we die.
	The & pencentage of fossil to compared with the present
	percentage in living organisms to detoemine their age.
00	
(1)	Relative method
DI.	The earth is dug, the fassils found closer to Earth are
	recent whereas the ones found in deepen layers.
1	are the older ones.
1	Foureg. Dinasaurs are found in deeper layeus.
/ D:	tionaid between any willing Control and all well and the Marking the items of any deat of Carliffe
	stinguish between cross-pollination and self-pollination. Mention the site and product of fertilizativer.
	aw labelled diagram of a pistil showing the following parts :
ן טוני	
	Stigma, Style, Ovary, Female germ cell
) D	OR raw a diagram of human female reproductive system and label the parts :
	which produce an egg.
	) where fertilization takes place.
(11	
21.	Fallopian Tube (2) 5 Ovidact (2)
1	I fallopian Tube
a	
-	
	Ovary Por
	ovacy
1.	( Rusauces egg) 1) alla
1	( The state of the
	De de
	st two bacterial diseases which are transmitted sexually.
) Li	Granoughosa and Suphillies are bacterial diseases
) Li	Gronoughosa and suphilles are bacterial diseases
) Li	
) Li	transmitted semally
6)	
6)	transmitted semularity
6)	transmitted semually saturation at are contraceptive devices? Give two reasons for adopting contraceptive devices in humans.  The devices used to prevent see feetili lation in human
6)	transmitted semularity
6)	transmitted semually  at are contraceptive devices? Give two reasons for adopting contraceptive devices in humans.  The devices used to prevent so fentili Lation in human females are known as contraceptive devices.
6)	transmitted semually saturation at are contraceptive devices? Give two reasons for adopting contraceptive devices in humans.  The devices used to prevent see feetili lation in human

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2

#### **SECTION - E**

22. Draw labelled diagram to show the following parts in an embryo of a pea seed:

Cotyledon, Plumule, Radical

22. Hydra, neprvoduces by Budding

Small

Small

Small

Devotoping

Devotoping

Burnt Hydra

T

Tenchacles

Farent Hydra

T

Totales

Farent Hydra

T

Totales

T

Totales

T

Totales

T

Totales

T

TOTALES

OR

A student observed a permanent slide showing asexual reproduction in Hydra. Draw labelled diagram in proper sequence of the observations that must have been made by the student. Name the process of reproduction also.

23. In the experiment "To prepare a temporary mount of a leaf peel to show stomata", glycerine and safranin are used.When and why are these two liquids used? Explain.

Ans. 23: D Gilycoinne is used before placing the specimen on the clide.

(a) It helps it to keep most and powerents douging.

Saferanan is a synthetic due used to stain the cells of the loaf so that each part is unible peroperly.

(a) It is applied after the loaf pool is washed in water and kept in a watch glass.

24. How is the presence of an acid tested with a strip of red litmus paper?

OR

A student is performing an experiment to study the properties of acetic acid. Answer the following questions:

- (i) Name the substance he must add to acetic acid to produce carbon dioxide.
- (ii) Give the relevant chemical equation for the reaction.
- (iii) How would he test CO<sub>2</sub> gas in the laboratory?

Ans. (i) He must add sodium canbonate on Sodium hydrogen.

Carbonate to produce carbon dioxide.

(ii) 2CH2COOH + Naco2 >2BCH2COONa+CO2+H2O

Ethanoic Sodium Sodium Carbon water

chil	To test cog in a laboratory, bring a proving condle
	wear it It would exting usued as it is a non supporter
	of compustion.
	OR OP THE STATE OF
- 1	Pass it through freshly perpared linewater which tuens with
	(a (OH) 2 (ag) + CO2 > CaCO3 + H2O
	liene water Carbon Calcium water
1	(calcium dioxide careonate
	Myduorside) (cohite ppt.)

25. What would you observe on adding zinc granules to freshly prepared ferrous sulphate solution? Give reason for your answer.

Ans. Fe SO 4 (eq) + Zn (s) displacement In SO 4(eq) + Fe (s)

fermous Zinc Zinc Zinc Ferrows

Supporte Granules surphate minor

The gener coloure of the solution changes to greyish - white

an there is a displacement who seartion occurring.

Zine being more greature than bean displaced it

from its falt solution

Fe powder can be observed.

- **26.** A teacher gives a convex lens and a concave mirror of focal length of 20 cm each to his student and asks him to find their focal lengths by obtaining the image of a distant object. The student uses a distant tree as the object and obtains its sharp image, one by one, on a screen. The distances d<sub>1</sub> and d<sub>2</sub> between the lens/mirror and the screen in the two cases and the nature of their respective sharp images are likely to be
  - (a) (20 cm, 40 cm) and (erect and erect)
  - (b) (20 cm, 40 cm) and (inverted and inverted)
  - (c) (20 cm, 20 cm) and (inverted and inverted)
  - (d) (20 cm, 40 cm) and (erect and inverted)

Give reason for your answer.

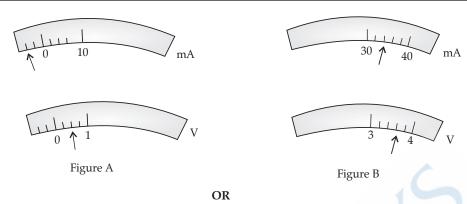
Ans. 26: (c) (20 m, 20 m) & (inverted and inverted) is correct

as forableogth remains the same.

The convex lens and concave mirror always four, real and inverted images encept in a few scare cases.

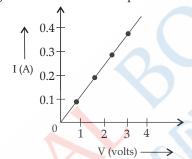
Since the object is a distant one, the smage obtained will be the focus and when calculated focal length remains the same is 20 cm.

27. The rest position of the needles in a milliammeter and voltmeter, not in use, are as shown in Figure A. When a student uses these instruments in his experiment, the readings of the needles are in the positions shown in Figure B. Determine the correct values of current and voltage the student should use in his calculations.2



In the experiment to study the dependence of current (I) on the potential difference (V) across a resistor, a student obtained a graph as shown.

- (i) What does the graph depict about the dependence of current on the potential difference?
- (ii) Find the current that flows through the resistor when the potential difference across it is 2.5 V.



Ans.