Material Safety Data Sheet

Product Bella (Mid Brown)

1. Information on the chemical product and company

A. Product Bella (Mid Brown)

B. Recommended use and limitations of the use of the product

Recommended use of the product Pigment / Colorant

Limitations of the use of the product Skin contact / Eye contact / Ingestion / Inhalation C. Supplier information (For imported goods, state the information of domestic supplier to contact in case of emergency)

Company name Blossom Cloud Inc

Address 7,8F 45, Seodaegu-ro, Daegu, Republic of Korea

Emergency contact number +82-053-291-1111

2. Hazards identification

A. Hazard/Danger classification Skin corrosion/irritation: Class 2

Severe eye damage/eye irritation: Class 2 (2A/2B)

Specific target organ toxicity (single exposure): Class 3 (respiratory irritation) Specific target organ toxicity (single exposure): Class 3 (respiratory irritation)

Specific target organ toxicity (repeated exposures): Class 1

B. Warning label items, including precautionary statements

Pictograms



Signal word Danger

H315 Causes skin irritation H319 Causes serious eye irritation H335 May cause respiratory irritation

Hazard/danger statements
Hazard/danger statements
Hazard/danger statements
Hazard/danger statements
Hazard/danger statements

known to be affected). (Specify exposure paths that cause toxicity to the specific target organ (repeated exposure). However, it is limited to cases where there is conclusive evidence that toxicity to the specific target organ (repeated exposure) is not caused.)

Precautionary statements

 $P260\ Do\ not\ breathe\ dust/fume/gas/mist/vapours/spray.$

P261 Avoid breathing dust/fume/gas/mist/vapours/spray.

P264 Wash ... thoroughly after handling.

Preventive P270 Do not eat, drink or smoke when using this product.

P271 Use only outdoors or in a well-ventilated area.

P280 Wear protective gloves/protective clothing/eye protection/face protection.

P202 P252 FE ON SWIPL WE I THE PROPERTY OF THE

P302+P352 IF ON SKIN: Wash it with a large amount of water/....

P304+P340 IF INHALED: Move the victim to an area with fresh air and have him or

her take relaxed in a posture that allows for easy breathing.

P305+P351+P338 IF IN EYES: Rinse continuously with water for several minutes.

Remove contact lenses if possible. Keep washing.

Response P312 Consult with a medical institute/doctor/... if feeling uncomfortable.

P314 Get medical advice/attention if you feel unwell.

P321 Specific treatment.

P332+P313 If skin irritation occurs: Get medical advice/attention. P337+P313 If eye irritation persists: Get medical advice/attention.

Response P362+P364 Take off contaminated clothing and wash it before reuse.

P403+P233 Store the material in a well-ventilated place. Keep the container tightly

closed.

Storage P405 Store locked up.

Disposal P501 Dispose of contents/containers according to waste-related laws and regulations.

3. Component nan	nes and amount			
	Material	Other name (Common name)	CAS No.	Content (%
Iron oxide			1309-37-1	15
Glyceryl stearate		Octadecanoic acid ester 1.2.3-propagetrial contained	11000 07 2	_

Glyceryl stearate Octadecanoic acid, ester, 1,2,3-propanetriol contained 11099-07-3 5

C.I. pigment black PIGMENT BLACK 11) Iron oxide black; 12227-89-3 25

Iron oxide, yellow C.I. pigment yellow 42 (C.I. PIGMENT YELLOW 42): 51274-00-1 20

Water Dihydrogen oxide; 7732-18-5 15

4. First-aid measures

C. Inhaled

D. Ingested

A. Substance in eyes Get emergency medical treatment.

If contact with the material, wash the skin and eyes with running water for at least 20 minutes. Rinse eyes carefully with water for a few minutes if the eyes made contact with the substance.

Remove contact lenses if possible. Keep washing.

Seek medical treatment or advice if the eye irritation continues.

B. Contact with skin In the case of hot material, immerse or flush the affected area with a large amount of cold water

to dissipate heat.

Get emergency medical treatment.

Remove and separate the contaminated clothes and shoes and isolate the contaminated area. If contact with the material, wash the skin and eyes with running water for at least 20 minutes.

Prevent the spread of the contaminated area in case of slight skin contact.

Consult with a medical institute (doctor) if exposed to the material or feeling uncomfortable.

Consult with a medical institute (doctor) if feeling uncomfortable.

Seek medical treatment or advice if the skin is irritated.

Take off the contaminated clothing and wash it before using it again.

If the person does not breathe, perform artificial respiration.

Supply oxygen if the person has difficulty of breathing.

If a person is exposed to excessive dust or fumes of the product, remove the material with clean

air and seek medical attention if the victim coughs or shows any other symptoms.

Move the person to an area with fresh air.

Get emergency medical treatment.

If the person does not breathe, perform artificial respiration.

If the substance is ingested or inhaled, do not perform mouth-to-mouth artificial respiration but

use appropriate respiratory medical equipment.

Supply oxygen if the person has difficulty of breathing.

Keep the victim warm and stable.

Consult with a medical institute (doctor) if exposed to the material or feeling uncomfortable.

If the victim is unconscious, do not give anything by mouth.

Get emergency medical treatment.

If the substance is ingested or inhaled, do not perform mouth-to-mouth artificial respiration but

use appropriate respiratory medical equipment.

Consult with a medical institute (doctor) if feeling uncomfortable when swallowing.

Consult with a medical institute (doctor) if exposed to the material or feeling uncomfortable.

Rinse the mouth.

E. Other precautions for medical doctors

Let the medical staff be aware of the material before taking protective measures.

Inform the medical staff of the material before taking protective measures.

5. Explosion and firefighting measures

A. Appropriate (and inappropriate) fire extinguishing agent Appropriate (and inappropriate) fire extinguishing agent

B. Specific hazards arising from the chemical Specific hazards arising from the chemical Use alcohol foam, carbon dioxide, or water spray to extinguish a fire involving this material.

Use dry sand or soil for extinguishing through suffocation.

The container may explode when heated.

The material can be decomposed at a high temperature and generate toxic gas.

Irritating and very toxic gases may be produced by pyrolysis or combustion.

The container may explode when heated. Leaked material has a risk of fire or explosion.

May be reignited even after extinguishing. May ignite when in contact with moisture.

Flammable/combustible material

Some substances can flash and burn quickly. Some may burn but are not easily ignitable.

Some may explosively decompose when heated or under fire.

Non-flammable. Although the material itself does not burn, it may decompose to generate corrosive/toxic fumes when heated.

Inhalation of decomposed products may lead to severe injury or death.

Contact with the skin and eyes can cause severe burns.

This material can generate irritating, corrosive, and toxic gas during a fire.

C. Protective gear to wear to suppress the fire and for preventive measures

Iron oxide Maintain a safe distance when extinguishing the fire.

Move containers from the fire area if it is not dangerous.

Extinguish the fire from a maximum distance or use an automatic fire extinguishing system

during a tank fire.

Do not allow water to get inside the container.

Cool the tank with a large amount of water even after the tank fire is extinguished.

Immediately withdraw from the area if there is a high hissing sound from the pressure relief

device, or the tank discolors during a tank fire.

Stay away from the tank in flames during a tank fire. Maintain a safe distance when extinguishing the fire.

Note that the material may be melted when transported.

Dig a ditch to dispose of extinguishing water and keep material from dispersing.

Move containers from the fire area if it is not dangerous.

Extinguish the fire from a maximum distance or use an automatic fire extinguishing system

during a tank fire.

Cool the tank with a large amount of water even after the tank fire is extinguished.

Immediately withdraw from the area if there is a high hissing sound from the pressure relief

device, or the tank discolors during a tank fire.

Stay away from the tank in flames during a tank fire.

In the case of a large tank fire, use unattained fire extinguishing equipment. If it is not available,

stay away and let it burn.

Rescuers should wear appropriate protective gear.

Maintain a safe distance when extinguishing the fire Note that the material may be melted when transported.

Dig a ditch to dispose of extinguishing water and keep material from dispersing.

Move containers from the fire area if it is not dangerous.

Extinguish the fire from a maximum distance or use an automatic fire extinguishing system during a tank fire.

Cool the tank with a large amount of water even after the tank fire is extinguished.

Immediately withdraw from the area if there is a high hissing sound from the pressure relief device, or the tank discolors during a tank fire.

Stay away from the tank in flames during a tank fire. In the case of a large tank fire, use unattained fire extinguishing equipment. If it is not available,

stay away and let it burn.

Glyceryl stearate

C.I. pigment black 11

Iron oxide, yellow Maintain a safe distance when extinguishing the fire

Note that the material may be melted when transported.

Dig a ditch to dispose of extinguishing water and keep material from dispersing.

Move containers from the fire area if it is not dangerous.

Extinguish the fire from a maximum distance or use an automatic fire extinguishing system during a tank fire.

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Immediately withdraw from the area if there is a high hissing sound from the pressure relief

device, or the tank discolors during a tank fire. Stay away from the tank in flames during a tank fire.

In the case of a large tank fire, use unattained fire extinguishing equipment. If it is not available,

stay away and let it burn.

Cool the tank with a large amount of water even after the tank fire is extinguished.

Immediately withdraw from the area if there is a high hissing sound from the pressure relief

device, or the tank discolors during a tank fire. Stay away from the tank in flames during a tank fire.

Water splashed from the container heated and exploded may cause burns to the skin and eyes.

6. Accidental release measures

Water

A. Personal precautions, protective equipment, and emergency procedures

Stop leakage if it is not dangerous.

Be aware of the materials and conditions to avoid.

Wipe up spills immediately, and follow preventive measures in the protective equipment section.

Do not touch or walk on exposed substances.

Remove all ignition sources.

Stop leakage if it is not dangerous.

Do not touch damaged containers or spills without wearing appropriate protective clothing.

For leaks without fire, wear full-face vapor protective clothing. Cover the material with plastic sheets to prevent diffusion.

Be aware of the materials and conditions to avoid. Do not inhale dust, fume, gas, mist, vapors, or spray. Avoid inhaling dust, fume, gas, mist, vapor, or spray.

Leaked material can cause contamination.

Prevent inflow into waterways, sewages, basements, or sealed areas.

C. Purification or removal Absorb the spill with inert material (e.g., dry sand or soil) and place chemical waste in a

container.

Absorb the liquid and wash the contaminated area with detergent and water.

Cover the material with dry sand/soil or other inert substance and then with plastic sheets to

prevent diffusion and exposure to rain.

Use a clean explosion-proof tool to collect spills and place them in a loosely covered plastic container.

7. Handling and storage

B. Environmental precautions

A. Safe handling

Be aware of the materials and conditions to avoid.

Refer to engineering controls and personal protective gear when working.

Be careful of the high temperature.

Follow MSDS/label precautions since product debris may remain after emptying the container.

Carefully handle and store the material Carefully open the cap before opening it.

Avoid long-term or continuous contact with the skin.

Do not access the storage area if not ventilated appropriately.

Be aware of the materials and conditions to avoid. Be aware of the materials and conditions to avoid.

Refer to engineering controls and personal protective gear when working.

A. Safe handling Avoid inhaling dust, fume, gas, mist, vapor, or spray.

Wash the handling part thoroughly after handling the product.

Do not eat, drink, or smoke when using the product.
Use the material only outdoors or in well-ventilated areas.

B. Safe storage method Be aware of the materials and conditions to avoid.

Fully drain empty drums and properly close them to immediately return to the drum regulator

or place them adequately.

Keep away from food and beverages.

Be aware of the materials and conditions to avoid.

Thoroughly seal the container and store it in a well-ventilated area.

Store the product in an area with a locking system.

8. Exposure controls and personal protection

A. Chemical material exposure criteria, biological exposure criteria, etc.

Domestic regulation

Iron oxide TWA - 5 mg/m3

Glyceryl stearate No data

C.I. pigment black 11 TWA - 5 mg/m³ iron oxide

Iron oxide, yellow No data Water No data

ACGIH regulation

Iron oxide $TWA = 5 \text{ mg/m}^3$

Glyceryl stearate No data
C.I. pigment black 11 No data
Iron oxide, yellow No data
Water No data

Biological exposure criteria

Iron oxideNo dataGlyceryl stearateNot applicableC.I. pigment black 11No dataIron oxide, yellowNo dataWaterNot applicable

Other exposure criteria

Iron oxideNo dataGlyceryl stearateNo dataC.I. pigment black 11No dataIron oxide, yellowNo dataWaterNot applicable

B. Adequate engineering control

Use process isolation, local ventilation, or other engineering control measures to lower the air

level to the exposure criterion or below.

B. Adequate engineering control

If the operation generates dust, fume, or mist, ventilate the area to maintain the air pollution

below the exposure limit.

B. Adequate engineering control Install an eye-washing facility and safety shower in the facility using or storing this material.

C. Personal protective gear

Protection of the respiratory system

Wear respiratory protective gear suitable for the exposed particulate material's physical and Iron oxide chemical characteristics that are certified by the Korea Occupational Safety and Health Agency

or equivalent.

If the exposure concentration is lower than 50 mg/m³, wear half-face respiratory protective gear

equipped with an appropriate filter.

If the exposure concentration is lower than 125 mg/m³, wear loose-fitting hood/helmet-type Iron oxide electric respiratory protective gear or continuous flow dust mask equipped with an appropriate

filter.

If the exposure concentration is lower than 250 mg/m³, wear full-face, electric half-face, or air-Iron oxide supplied continuous flow/pressure-demanding half-face respiratory protective gear equipped

with an appropriate filter.

If the exposure concentration is lower than 5000 mg/m³, wear a full-face or helmet/hood-type

pressure-demanding air-supplied respirator equipped with an appropriate filter.

If the exposure concentration is lower than 50000 mg/m³, wear SCBA (self-contained breathing Iron oxide apparatus) or pressure-demanding SCBA respiratory protective gear equipped with an

appropriate filter.

Glyceryl stearate or equivalent. Glyceryl stearate Glyceryl stearate apparatus) respiratory protective gear. C.I. pigment black 11 Iron oxide C.I. pigment black 11 or equivalent. C.I. pigment black 11 equipped with an appropriate filter. C.I. pigment black 11 filter. C.I. pigment black 11 with an appropriate filter. C.I. pigment black 11 C.I. pigment black 11 appropriate filter. Iron oxide, yellow Iron oxide, yellow Iron oxide, yellow apparatus) respiratory protective gear. Water Water gas mask Water apparatus) respiratory protective gear. Protection of eyes goggles. Protection of eyes workers can easily access them. Protection of eyes cause eye irritation or other health hazards. Protection of eyes workers can easily access them. Protection of hands appropriate materials. Protection of body

Wear respiratory protective gear suitable for the exposed particulate material's physical and chemical characteristics that are certified by the Korea Occupational Safety and Health Agency

For particulate material, the following respiratory protective gear is recommended.

- The face filter-type dust mask, air-filtering dust mask (high-efficiency particulate filter material), or dust mask with electric fan (filter material for dust, mist, and fume)

If short of oxygen (<19.6%), wear an air-supplied mask or SBCA (self-contained breathing

Wear respiratory protective gear suitable for the exposed particulate material's physical and chemical characteristics that are certified by the Korea Occupational Safety and Health Agency

If the exposure concentration is lower than 50 mg/m³, wear half-face respiratory protective gear

If the exposure concentration is lower than 125 mg/m³, wear loose-fitting hood/helmet-type electric respiratory protective gear or continuous flow dust mask equipped with an appropriate

If the exposure concentration is lower than 250 mg/m³, wear full-face, electric half-face, or airsupplied continuous flow/pressure-demanding half-face respiratory protective gear equipped

If the exposure concentration is lower than 5000 mg/m³, wear a full-face or helmet/hood-type pressure-demanding air-supplied respirator equipped with an appropriate filter.

If the exposure concentration is lower than 50000 mg/m³, wear SCBA (self-contained breathing apparatus) or pressure-demanding SCBA respiratory protective gear equipped with an

Wear respiratory protective gear suitable for the exposed particulate material's physical and chemical characteristics that are certified by the Korea Occupational Safety and Health Agency

For particulate material, the following respiratory protective gear is recommended.

- The face filter-type dust mask, air-filtering dust mask (high-efficiency particulate filter material), or dust mask with electric fan (filter material for dust, mist, and fume)

If short of oxygen (<19.6%), wear an air-supplied mask or SBCA (self-contained breathing

Wear respiratory protective gear suitable for the exposed gas/liquid's physical and chemical characteristics that are certified by the Korea Occupational Safety and Health Agency or

For gas/liquid material, the following respiratory protective gear is recommended.

- Separate-type full-face gas mask (organic compound use (acid gas use in case of acid gas)), separate-type half-face gas mask (organic compound use (acid gas use in case of acid gas)), direct connection-type full-face gas mask (organic compound use (acid gas use in case of acid gas)), half-face gas mask (organic compound use (acid gas use in case of acid gas)), or electric

If short of oxygen (<19.5%), wear an air-supplied mask or SBCA (self-contained breathing

For particulate material that may cause eye irritation or other health hazards, wear breathable

Install emergency washing facilities (shower type) and eye-washing facilities at places where

Wear protective or breathable goggles to protect your eyes from organic compound vapors that

Install emergency washing facilities (shower type) and eye-washing facilities at places where

Considering the chemical's physical and chemical properties, wear protective gloves made of

Considering the chemical's physical and chemical properties, wear protective clothing made of

appropriate materials.

Wear protective clothing for high-temperature or high-pressure splash protection as necessary.

9. Physicochemical characteristics

Protection of body

A. Physical state

No data Appearance Color No data B. Odor No data C. Odor threshold No data D. pH No data E. Melting point/freezing point No data F. Initial boiling point and boiling point range No data G. Flash point No data H. Evaporation rate No data I. Flammability (Solid and vapor) No data J. Upper/lower limits of the ignition or explosion range No data No data K. Vapor pressure L. Solubility No data M. Vapor density No data No data N. Relative density O. n-octanol-water partition coefficient (Kow) No data P. Spontaneous ignition temperature No data Q. Decomposition temperature No data R. Viscosity No data S. Molecular weight No data

Iron oxide

A. Physical state
Appearance

Solid (crystal, powder) Color Red to black B. Odor No data C. Odor threshold No data D. pH No data E. Melting point/freezing point 1565°C F. Initial boiling point and boiling point range No data G. Flash point No data H. Evaporation rate No data I. Flammability (Solid and vapor) Non-flammable

J. Upper/lower limits of the ignition or explosion range -/-K. Vapor pressure No data L. Solubility (Insoluble) M. Vapor density No data N. Relative density 5.24 O. n-octanol-water partition coefficient (Kow) No data No data P. Spontaneous ignition temperature Q. Decomposition temperature No data R. Viscosity No data S. Molecular weight 159.70

Glyceryl stearate

A. Physical state

Appearance Solid
Color No data
B. Odor No data
C. Odor threshold No data
D. pH No data
E. Melting point/freezing point No data

F. Initial boiling point and boiling point range No data G. Flash point No data H. Evaporation rate No data I. Flammability (Solid and vapor) No data J. Upper/lower limits of the ignition or explosion range No data K. Vapor pressure No data L. Solubility No data M. Vapor density 13 N. Relative density No data O. n-octanol-water partition coefficient (Kow) No data No data P. Spontaneous ignition temperature Q. Decomposition temperature No data R. Viscosity No data S. Molecular weight 376.574

C.I. pigment black PIGMENT BLACK 11)

A. Physical state

Appearance Solid (fine black powder)

Color Black B. Odor Odorless C. Odor threshold No data D. pH No data E. Melting point/freezing point 1527°C F. Initial boiling point and boiling point range No data G. Flash point No data H. Evaporation rate No data I. Flammability (Solid and vapor) No data J. Upper/lower limits of the ignition or explosion range -/-K. Vapor pressure No data L. Solubility No data M. Vapor density No data N. Relative density 5.18 O. n-octanol-water partition coefficient (Kow) No data P. Spontaneous ignition temperature No data Q. Decomposition temperature No data R. Viscosity No data 231.54 S. Molecular weight

Iron oxide, yellow

A. Physical state

Appearance Solid (nano material)

Color No data
B. Odor Scentless
C. Odor threshold No data

D. pH 3.5 to 7.5 (50 g/l, sediment)
E. Melting point/freezing point > 160°C (approximately 1013 hPa)

F. Initial boiling point and boiling point range No data
G. Flash point No data
H. Evaporation rate No data
I. Flammability (Solid and vapor) Not flammable
J. Upper/lower limits of the ignition or explosion range No data
K. Vapor pressure No data

L. Solubility < 1 ug/L (20°C, pH: 8) 4.26 g/cm³ (density) M. Vapor density 4.26 (20°C) N. Relative density O. n-octanol-water partition coefficient (Kow) No data P. Spontaneous ignition temperature No data Q. Decomposition temperature No data R. Viscosity No data S. Molecular weight 159.69

A. Physical state

Appearance Liquid

Color Colorless (transparent)

B. Odor Odorless C. Odor threshold (Not applicable)

D. pH E. Melting point/freezing point 0°C F. Initial boiling point and boiling point range 100°C

G. Flash point (Not applicable) H. Evaporation rate No data I. Flammability (Solid and vapor) Not applicable J. Upper/lower limits of the ignition or explosion range -/- (not applicable) 23.8 mmHg (25°C) K. Vapor pressure L. Solubility 100 g/100 mlM. Vapor density No data N. Relative density 1 O. n-octanol-water partition coefficient (Kow) -1.38No data P. Spontaneous ignition temperature Q. Decomposition temperature No data R. Viscosity No data S. Molecular weight 18.02

10. Stability and reactivity

A. Chemical stability and the possibility of the harmful reaction

Iron oxide The container may explode when heated. Iron oxide Leaked material has a risk of fire or explosion. Iron oxide May be reignited even after extinguishing. Iron oxide May ignite when in contact with moisture. Iron oxide Flammable/combustible material

Iron oxide Some substances can flash and burn quickly.

Iron oxide Some react violently with water.

Iron oxide Some may explosively decompose when heated or under fire. Iron oxide Inhalation of decomposed products may lead to severe injury or death.

Iron oxide Contact with the skin and eyes can cause severe burns.

Iron oxide This material can generate irritating, corrosive, and toxic gas during a fire.

Glyceryl stearate The container may explode when heated. Glyceryl stearate Some may burn but are not easily ignitable.

Glyceryl stearate Non-flammable. Although the material itself does not burn, it may decompose to generate

corrosive/toxic fumes when heated.

Glyceryl stearate This material can generate irritating, corrosive, and toxic gas during a fire. C.I. pigment black 11 The container may explode when heated.
C.I. pigment black 11 Some may burn but are not easily ignitable.

C.I. pigment black 11 Non-flammable. Although the material itself does not burn, it may decompose to generate

corrosive/toxic fumes when heated.

C.I. pigment black 11 This material can generate irritating, corrosive, and toxic gas during a fire.

The material can be decomposed at a high temperature and generate toxic gas.

Iron oxide, yellow The container may explode when heated.

Iron oxide, yellow Some may burn but are not easily ignitable.

Non-flammable. Although the material itself does not burn, it may decompose to generate

corrosive/toxic fumes when heated.

Water Stable at room temperature and pressure condition

Water The container may explode when heated.

B. Conditions to avoid

Iron oxide, yellow

Iron oxide Humidity

Iron oxideIgnition sources such as heat, sparks, and flamesGlyceryl stearateIgnition sources such as heat, sparks, and flamesC.I. pigment black 11Ignition sources such as heat, sparks, and flamesIron oxide, yellowIgnition sources such as heat, sparks, and flames

Water Heat and pollution

C. Materials to avoid

Water

Iron oxide Water

Glyceryl stearate Combustible and reducing materials
C.I. pigment black 11 Combustible and reducing materials
Iron oxide, yellow Combustible and reducing materials

Water-reactive substances

D. Harmful materials generated during decomposition

Iron oxide Irritating, corrosive, and toxic gases

Glyceryl stearate Corrosive/toxic fume

Glyceryl stearate Irritating, corrosive, and toxic gases

C.I. pigment black 11 Irritating and very toxic gases may be produced by pyrolysis or combustion.

C.I. pigment black 11 Corrosive/toxic fume Iron oxide, yellow Corrosive/toxic fume

Iron oxide, yellow Irritating, corrosive, and toxic gases

Water No data

11. Toxicological information

A. Information on likely exposure paths

Iron oxide No data
Glyceryl stearate No data
C.I. pigment black 11 No data
Iron oxide, yellow No data
Water No data

B. Information on health hazards

Acute toxicity

Orally administered

Iron oxide LD50 > 10000 mg/kg rat (EU Method B.1; no disturbance observed)

Glyceryl stearate No data
C.I. pigment black 11 No data

Iron oxide, yellow LD50 > 10000 mg/kg rat

Iron oxide, yellow No data

Water LD50 90000 mg/kg rat (LD50 > 90 ml/kg (rat))

Transdermal

Iron oxideNo dataGlyceryl stearateNo dataC.I. pigment black 11No dataIron oxide, yellowNo dataWaterNo data

Inhaled

Iron oxide Dust LC50 5.05 mg/l 4 hr rat (OECD TG403, GLP)

Glyceryl stearate No data
C.I. pigment black 11 No data

Iron oxide, yellow Mist discriminating conc. 5.05 mg/l 4 hr rat

Iron oxide, yellow No data Water No data

Skin corrosion or irritation

Iron oxide By the result of the OECD G404 test, the skin irritation test using rabbits, no irritation is

observed. Erythema and edema irritation index $\boldsymbol{0}$

Glyceryl stearate No data

C.I. pigment black 11 Short-term contact causes skin irritation.

Iron oxide, yellow Total irritation score: 0/4, no irritation

Water Not applicable

Severe eye damage or irritation

Iron oxide By the result of the OECD G405 GLP test, the eye corrosion/irritation test using rabbits, no

irritation is observed. Irritation index $\boldsymbol{0}$

Glyceryl stearate No data

C.I. pigment black 11 Short-term contact causes skin irritation.

Iron oxide, yellow No irritation. Rabbit Water Not applicable

Respiratory sensitization

Iron oxideNo dataGlyceryl stearateNo dataC.I. pigment black 11No dataIron oxide, yellowNo dataWaterNot applicable

Skin hypersensitivity

Iron oxide By the result of the hypersensitivity test using guinea pigs, it is not hypersensitive.

Glyceryl stearate No data
C.I. pigment black 11 No data

Iron oxide, yellow Not hypersensitive. Guinea pig Water Not applicable Carcinogenicity Occupational Safety and Health Act Iron oxide No data Glyceryl stearate No data No data C.I. pigment black 11 Iron oxide, yellow No data Water No data Public Notice by the Ministry of Employment and Labor Iron oxide No data No data Glyceryl stearate No data C.I. pigment black 11 Iron oxide, yellow No data Water No data IARC Iron oxide 3 No data Glyceryl stearate C.I. pigment black 11 No data Iron oxide, yellow No data Water No data **OSHA** Iron oxide No data Glyceryl stearate No data C.I. pigment black 11 No data Iron oxide, yellow No data Water No data ACGIH Iron oxide A4 Glyceryl stearate No data C.I. pigment black 11 No data Iron oxide, yellow No data Water No data NTP Iron oxide No data Glyceryl stearate No data C.I. pigment black 11 No data Iron oxide, yellow No data

No data

No data

No data

No data

No data

Water

Glyceryl stearate

Iron oxide, yellow

C.I. pigment black 11

EU CLP Iron oxide Water No data

Germ cell mutagenicity

Iron oxide The results of the reverse mutation assay using in vitro microorganisms and the OECD TG473 GLP

test, the chromosome aberration test using mammal (Chinese hamster) cells, are negative. The result

of the in vivo comet assay using rats is negative. Analog material CAS No. 1317-61-9

Glyceryl stearate No data
C.I. pigment black 11 No data

Iron oxide, yellow Reverse mutation test using in vitro bacteria: Negative (S. typhimurium TA1535, TA1537, TA98,

TA100 regardless of the metabolic activation system)

Water Not applicable

Reproductive toxicity

Iron oxideNo dataGlyceryl stearateNo dataC.I. pigment black 11No dataIron oxide, yellowNo dataWaterNot applicable

Toxicity to the specific target organ (One-time exposure)

Iron oxide Causes irritation to the respiratory system.

Glyceryl stearate No data

C.I. pigment black 11 Inhaling causes irritation to the respiratory system.

Iron oxide, yellow Inhaling: 5 male and 5 female rats (Wistars) were si

Inhaling: 5 male and 5 female rats (Wistars) were singly exposed to 5 mg/l CERAC- pigment (average particle size = 35 nm) for 4 hours. Their mortality, clinical signs, and weights were observed for 14 days. All animals deceased during the observation period and sacrificed at the end of test period were pathologically examined. A single snout was only exposed for inhalation to CERAC-pigment for 4 hours at an aerosol concentration of 5 mg/L; all animals could tolerate exposure. Consequently, it was

It has been reported that repeated inhalation exposure in humans may cause respiratory effects

considered that the MLC (median lethal concentration) exceeded 5 mg/L. (OECD TG 403) $\,$

Not applicable

Toxicity to the specific target organ (Repeated exposures)

Iron oxide

(atresia) and may cause metal fume fever.

Glyceryl stearate

No data

C.I. pigment black 11 No data

Iron oxide, yellow

Subchronic inhalation toxicity (90 days) was tested on rats (male/female). By the test result, an

increase in lung and lung-associated lymph node weights was observed, but no other effects were

observed. NOAEL 4.7 mg/L air (OECD TG 413, GLP).

Inhalation (subchronic): Rats were exposed to 3 different concentrations irregularly. The test was clearly consistent with poorly soluble particles and showed a typical result. No evidence of lung

toxicity (NOAEL = 4.7 mg/m³), rat, OECD TG 413, GLP

Water Not applicable

Inhalation hazard

Water

Iron oxideNo dataGlyceryl stearateNo dataC.I. pigment black 11No dataIron oxide, yellowNo dataWaterNot applicable

Other hazardous impacts

Iron oxide No data
Glyceryl stearate No data

C.I. pigment black 11 No data
Iron oxide, yellow No data
Water No data

12. Ecological information

A. Ecological toxicity

Fish

Iron oxide $LC0 \ge 50000 \text{ mg/l } 96 \text{ hr and (Danio rerio)}$

Glyceryl stearate LC50 47.228 mg/l 96 hr

C.I. pigment black 11 No data

Iron oxide, yellow $LC0 \ge 100000 \text{ mg} / 196 \text{ hr Danio rerio}$

Iron oxide, yellow (Running freshwater)

Water No data

Shellfish

 $\begin{tabular}{ll} Iron oxide & EC50 > 100 mg/1 48 hr \\ Glyceryl stearate & LC50 318.38 mg/1 48 hr \\ \end{tabular}$

C.I. pigment black 11 No data

 $LC50 \ge 100 \text{ mg} / 148 \text{ hr Daphnia magna}$ Iron oxide, yellow (OECD TG 202, still freshwater, GLP)

Water No data

Birds

Iron oxide No data

Glyceryl stearate EC50 3.72 mg/l 96 hr

C.I. pigment black 11 No data

Iron oxide, yellow EC50 18 mg/l 72hr pseudokirchneriella subcapitata

Iron oxide, yellow (OECD TG 201)

Water No data

B. Persistence and degradability

Persistence

Iron oxide No data
Glyceryl stearate No data
C.I. pigment black 11 No data
Iron oxide, yellow No data
Water log Kow -1.38

Degradability

Iron oxideNo dataGlyceryl stearateNo dataC.I. pigment black 11No dataIron oxide, yellowNo dataWaterNo data

C. Bio-accumulation

Accumulation

Iron oxide No data
Glyceryl stearate No data

C.I. pigment black 11 No data
Iron oxide, yellow No data
Water No data

Biodegradability

Iron oxide No data
Glyceryl stearate No data
C.I. pigment black 11 No data
Iron oxide, yellow No data
Water No data

D. Soil mobility

Iron oxideNo dataGlyceryl stearateNo dataC.I. pigment black 11No dataIron oxide, yellowNo dataIron oxide, yellowNo dataWaterNo data

E. Other hazardous impacts

Iron oxide No data
Glyceryl stearate No data
C.I. pigment black 11 No data
Iron oxide, yellow No data
Water Not applicable

13. Disposal considerations

A. Method of disposal

Iron oxide 1) Treat neutralization, hydrolysis, oxidation, and reduction.

2) Incinerate or melt at a high temperature.

3) Treat solidification.

Glyceryl stearate If specified in the Wastes Control Act, dispose of contents and containers according to the

regulation.

C.I. pigment black 11 If specified in the Wastes Control Act, dispose of contents and containers according to the

egulation.

Iron oxide, yellow If specified in the Wastes Control Act, dispose of contents and containers according to the

egulation.

Water If specified in the Wastes Control Act, dispose of contents and containers according to the

regulation.

B. Disposal considerations

Iron oxide Dispose of contents and containers (according to the contents specified in the relevant laws).

Glyceryl stearate Dispose of contents and containers (according to the contents specified in the relevant laws).

C.I. pigment black 11 Dispose of contents and containers (according to the contents specified in the relevant laws).

Iron oxide, yellow Dispose of contents and containers (according to the contents specified in the relevant laws).

Water Consider precautions described in the regulation if specified in the Wastes Control Act.

14. Transport information

A. UN No.

Iron oxide 1376

Glyceryl stearate No information is available for its UN Classification of Transport Hazardous Substances.

C.I. pigment black 11 No information is available for its UN Classification of Transport Hazardous Substances.

Iron oxide, yellow No information is available for its UN Classification of Transport Hazardous Substances.

Water No information is available for its UN Classification of Transport Hazardous Substances.

B. Proper shipping name

Iron oxide Iron oxide, spent or iron oxide sponge, spent, obtained from coal gas purification

Glyceryl stearate Aluminum calcium oxide -

C.I. pigment black 11 Not applicable

Iron oxide, yellow 4-Methyl-N-[[(4-methylphenyl)amino]carbonyl]benzenesulfonamide

Water Not applicable

C. Transport hazard class

Iron oxide 4.2

Glyceryl stearate Not applicable
C.I. pigment black 11 Not applicable
Iron oxide, yellow Not applicable
Water Not applicable

D. Container classification

Iron oxide III

Glyceryl stearate Not applicable
C.I. pigment black 11 Not applicable
Iron oxide, yellow Not applicable
Water Not applicable

E. Marine pollutant

Iron oxideNot applicableGlyceryl stearateNo dataC.I. pigment black 11No dataIron oxide, yellowNo dataWaterNo data

F. Particular safety measures required, or the user should be aware of related to transport or transport means

Emergency measures during a fire

Iron oxide F-G

Glyceryl stearate Not applicable
C.I. pigment black 11 Not applicable
Iron oxide, yellow Not applicable
Water Not applicable

Emergency measures during leaking

Iron oxide S-P

Glyceryl stearate Not applicable
C.I. pigment black 11 Not applicable
Iron oxide, yellow Not applicable
Water Not applicable

15. Regulatory information

A. Regulation by the Occupational Safety and Health Act

Iron oxide Harmful material subject to control

Iron oxide Substance subject to work environment measurement (measurement cycle: 6 months)

Iron oxide Substance subject to special health examination (diagnosis cycle: 12 months)

Iron oxide Substance material to exposure criteria

Glyceryl stearate No data

C.I. pigment black 11 Harmful material subject to control (iron and its compounds)

C.I. pigment black 11 Substance subject to work environment measurement (measurement cycle: (iron oxide dust

and fumes))

C.I. pigment black 11 Substance subject to special health examination (diagnosis cycle: iron oxide (dust and fumes

only))

C.I. pigment black 11 Substance material to exposure criteria

Iron oxide, yellow No data Water No data

B. Regulation by the Toxic Chemicals Control Act

Iron oxideNo dataGlyceryl stearateNo dataC.I. pigment black 11No dataIron oxide, yellowNo dataWaterNo data

C. Regulation by the Act on the Safety Control of Hazardous Substances

Iron oxideNo dataGlyceryl stearateNo dataC.I. pigment black 11No dataIron oxide, yellowNo dataWaterNo data

D. Regulation by the Wastes Control Act

Iron oxide Designated waste Glyceryl stearate Designated waste

C.I. pigment black 11 No data
Iron oxide, yellow No data
Water No data

E. Other regulations by domestic and foreign laws

Domestic regulation
Iron oxide
Glyceryl stearate

C.I. pigment black 11 Iron oxide, yellow

Water

Other domestic regulation

Iron oxideNot applicableGlyceryl stearateNot applicableC.I. pigment black 11Not applicableIron oxide, yellowNot applicableWaterNot applicable

Foreign regulation

US control information (OSHA regulation)

Iron oxide Not applicable Glyceryl stearate Not applicable

C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable
US control information (CERCLA regulation)	
Iron oxide	Not applicable
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable
US control information (EPCRA 302 regulation)	
Iron oxide	Not applicable
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable
US control information (EPCRA 304 regulation)	
Iron oxide	Not applicable
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable
US control information (EPCRA 313 regulation)	
Iron oxide	Not applicable
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable
US control information (Rotterdam Convention)	11
Iron oxide	Not applicable
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable
US control information (Stockholm Convention)	11
Iron oxide	Not applicable
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable
US control information (Montreal Protocol)	T. T.F.
Iron oxide	Not applicable
Glyceryl stearate	Not applicable
CI : 11 1 11	Y

Not applicable

C.I. pigment black 11

Iron oxide, yellow Not applicable Water Not applicable EU classification information (Finalized classification) Iron oxide Not applicable Glyceryl stearate Not applicable C.I. pigment black 11 Not applicable Iron oxide, yellow Not applicable Water Not applicable EU classification information (Hazard statement) Iron oxide Not applicable Glyceryl stearate Not applicable C.I. pigment black 11 Not applicable Iron oxide, yellow Not applicable Water Not applicable EU classification information (Safety statement) Not applicable Iron oxide Glyceryl stearate Not applicable

Not applicable

Not applicable

Not applicable

16. Other information

Water

A. Source of information

Iron oxide

ICSC (Appearance)

C.I. pigment black 11

Iron oxide, yellow

ICSC (Color)

ICSC (E. Melting point/freezing point)

HSDB (I. Flammability (Solid and vapor))

HSDB (L. Solubility)

ICSC (N. Relative density)

ECHA (Orally administered)

ECHA (Inhaled)

ECHA (Skin corrosion or irritation)

ECHA (Severe eye damage or irritation)

ECHA (Skin hypersensitivity)

ECHA (Germ cell mutagenicity)

(Reproductive toxicity)

 $ICSC\ (Toxicity\ to\ the\ specific\ target\ organ\ (One-time\ exposure))$

nite, icsc, gestis (Toxicity to the specific target organ (Repeated exposures))

ECHA (Fish)

ECHA (Shellfish)

ECOSAR (Birds)

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molecular weight of air (Vapor density) | ChemIDplus (Molecular weight)
     C.I. pigment black PIGMENT BLACK 11)
        14303 Chemicals (Japan) (Appearance)
        14303 Chemicals (Japan) (N. Relative density)
        14303 Chemicals (Japan) (S. Molecular weight)
        14303 Chemicals (Japan)
     Iron oxide, yellow
        ECHA (Appearance)
        ECHA (B. Odor)
        GESTIS (D. pH)
        ECHA (E. Melting point/freezing point)
        ECHA (I. Flammability (Solid and vapor))
        ECHA (L. Solubility)
        ECHA (M. Vapor density)
        ECHA (N. Relative density)
        ECHA (S. Molecular weight)
        ECHA (Orally administered)
        ECHA (Inhaled)
        ECHA (Skin corrosion or irritation)
        ECHA (Severe eye damage or irritation)
        ECHA (Skin hypersensitivity)
        ECHA (Germ cell mutagenicity)
        ECHA (Toxicity to the specific target organ (One-time exposure))
        ECHA (Toxicity to the specific target organ (Repeated exposures))
        ECHA (Fish)
        ECHA (Shellfish)
        ECHA (Birds)
        ECHA (D. Soil mobility)
     Water
        NLM
B. Initial creation date
                                                                               September 23, 2022
C. # of revisions and the latest revision date
     # of revisions
                                                                                  times
                                                                               0
     The latest revision date
D. Other
```

Lexemol T Inolex (Orally administered) | ECOSAR (Birds) | ECOSAR (Shellfish) | ECOSAR(Fish) | Calculation by the molecular weight and the average

 The prepared MSDS is the edited and partially corrected data based on the MSDS data provided by the KOSHA (Korea Occupational Safety and Health Agency).