# Material Safety Data Sheet

Product	Plat Brown (True)
1 Information on the chemical product and company	
A. Broduct	Diet Drown (True)
B. Recommended use and limitations of the use of the product	That Blown (Thue)
Recommended use of the product	Pigment / Colorant
Limitations of the use of the product	Skin contact / Eve contact / Ingestion / Inhalation
C. Supplier information (For imported goods, state the informat	ion of domestic supplier to contact in case of emergency)
Company name	Blossom Cloud Inc
Address	7,8F 45, Seodaegu-ro, Daegu, Republic of Korea
Emergency contact number	+82-053-291-1111
2 Hazards identification	
A Hazard/Danger classification	Skin corrosion/irritation: Class 2
A. Huzard Danger elassification	Severe eve damage/eve irritation: Class 2 (2A/2B)
	Carcinogenicity: Class 2
	Specific target organ toxicity (single exposure): Class 3 (respiratory irritation)
	Specific target organ toxicity (single exposure): Class 3 (respiratory irritation)
	Specific target organ toxicity (repeated exposures): Class 2
B. Warning label items, including precautionary statements Pictograms	
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	V
Signal word	Warning
	H315 Causes skin irritation
	H319 Causes serious eye irritation
	H335 May cause respiratory irritation
Hazard/danger statements	H351 Suspected of causing cancer (state route of exposure if it is conclusively proven that
	no other routes of exposure cause the hazard)
	H373 Causes damage to organs through prolonged or repeated exposure (list all organs
	known to be affected). (Specify exposure paths that cause toxicity to the specific target
	organ (repeated exposure). However, it is limited to cases where there is conclusive
Decountion are statements	evidence that toxicity to the specific target organ (repeated exposure) is not caused.)
r recautionary statements	P201 Obtain special instructions before use
	P202 Read and understand all safety precautions before handling
	P260 Do not breathe dust/fume/gas/mist/vanours/spray
Preventive	P261 Avoid breathing dust/fume/gas/mist/vapours/spray.
	P264 Wash thoroughly after handling.
	P271 Use only outdoors or in a well-ventilated area.
	P280 Wear protective gloves/protective clothing/eye protection/face protection.
	P302+P352 IF ON SKIN: Wash it with a large amount of water/
	P304+P340 IF INHALED: Move the victim to an area with fresh air and have him or her
	take relaxed in a posture that allows for easy breathing.
Response	P305+P351+P338 IF IN EYES: Rinse continuously with water for several minutes.
	Remove contact lenses if present and easy to do. Continue rinsing.
	P308+ P308 If exposed or concerned: Get medical advice/attention.
	P312 Consult with a medical institute/doctor/ if feeling uncomfortable.

Storage Disposal	<ul> <li>P314 Get medical advice/attention if you feel unwell</li> <li>P321 Specific treatment</li> <li>P332+P313 If skin irritation occurs: Get medical adv</li> <li>P337+P313 If eye irritation persists: Get medical adv</li> <li>P362+P364 Take off contaminated clothing and wast</li> <li>P403+P233 Store the material in a well-ventilated pl</li> <li>P405 Store locked up.</li> <li>P501 Dispose of contents/containers according to wast</li> </ul>	vice/attention. vice/attention. h it before reuse. ace. Keep the contain uste-related laws and	her tightly closed. regulations.
3. Component names and amount			
Material	Other name (Common name)	CAS No.	Content (%)
Titanium dioxide		13463-67-7	9
Iron oxide		1309-37-1	9
Glyceryl stearate	Octadecanoic acid, ester, 1,2,3-propanetriol contained	11099-07-3	10
C.I. pigment black 11	Iron oxide black;	12227-89-3	28
Iron oxide, yellow	C.I. pigment yellow 42	51274-00-1	24
Water	Dihydrogen oxide;	7732-18-5	20
4 First-aid measures			
A. Substance in eves	Get emergency medical treatment		
	If contact with the material, wash the skin and eyes w minutes. Rinse eyes carefully with water for a few minutes if substance. Remove contact lenses if possible. Keep w Seek medical treatment or advice if the eye irritation	vith running water fo the eyes made contac washing. continues.	r at least 20 ct with the
B. Contact with skin	In the case of not material, immerse of flush the affect water to dissipate heat. Get emergency medical treatment. Remove and separate the contaminated clothes and s If contact with the material, wash the skin and eyes w minutes. Prevent the spread of the contaminated area in case of Consult with a medical institute (doctor) if exposed t uncomfortable. Consult with a medical institute (doctor) if feeling un Seek medical treatment or advice if the skin is irritat Take off the contaminated clothing and wash it befor	hoes and isolate the o with running water fo of slight skin contact. o the material or feel ncomfortable. ed. e using it again.	contaminated area. or at least 20
C. mnaied	Supply oxygen if the person has difficulty of breathing If a person is exposed to excessive dust or fumes of the clean air and seek medical attention if the victim cour Move the person to an area with fresh air. Get emergency medical treatment. If the person does not breathe, perform artificial resp If the substance is ingested or inhaled, do not perform but use appropriate respiratory medical equipment. Supply oxygen if the person has difficulty of breathing Keep the victim warm and stable. Seek medical treatment or advice if exposed or conce Consult with a medical institute (doctor) if exposed the uncomfortable.	ng. he product, remove t ghs or shows any oth iration. n mouth-to-mouth ar ng. erned about being exp o the material or feel	he material with her symptoms. tificial respiration posed. ling
D. Ingested	If the victim is unconscious, do not give anything by Get emergency medical treatment.	mouth.	

D. Ingested	If the substance is ingested or inhaled, do not perform mouth-to-mouth artificial respiration but use appropriate respiratory medical equipment. Consult with a medical institute (doctor) if feeling uncomfortable when swallowing. Seek medical treatment or advice if exposed or concerned about being exposed. Consult with a medical institute (doctor) if exposed to the material or feeling uncomfortable. Rinse the mouth.
E. Other precautions for medical doctors	Let the medical staff be aware of the material before taking protective measures. If exposed to the substance, contact the medical staff and take special emergency measures such as tracking surveys. Inform the medical staff of the material before taking protective measures.
5. Explosion and firefighting measures	
A. Appropriate (and inappropriate) fire extinguishing agent	
Appropriate (and inappropriate) fire extinguishing agent	Use alcohol foam, carbon dioxide, or water spray to extinguish a fire involving this material.
B Specific hazards arising from the chemical	Ose dry said of son for extinguishing through sufficiently.
Specific hazards arising from the chemical	The container may explode when heated.
	The material can be decomposed at a high temperature and generate toxic gas.
	Irritating and very toxic gases may be produced by pyrolysis or combustion. The container may explode when heated.
	Leaked material has a risk of fire or explosion.
	May be reignited even after extinguishing.
	May ignite when in contact with moisture.
	Flammable/combustible material
	Some substances can flash and burn quickly.
	Some may build out are not easily ignitable.
	Non-flammable. Although the material itself does not burn, it may decompose to generate
	corrosive/toxic fumes when heated.
	Inhalation of decomposed products may lead to severe injury or death.
	Contact with the skin and eyes can cause severe burns.
	This material can generate irritating, corrosive, and toxic gas during a fire.
C. Protective gear to wear to suppress the fire and for preventive measures	
Titanium dioxide	Rescuers should wear appropriate protective gear.
	Maintain a safe distance when extinguishing the fire.
	Note that the material may be methed when transported.
	Move containers from the fire area if it is not dangerous.
	Extinguish the fire from a maximum distance or use an automatic fire extinguishing
	system during a tank fire.
	Cool the tank with a large amount of water even after the tank fire is extinguished.
	Immediately withdraw from the area if there is a high hissing sound from the pressure
	relief device, or the tank discolors during a tank fire.
	Stay away from the tank in flames during a tank fire.
	available, stav away and let it burn.
Iron oxide	Maintain a safe distance when extinguishing the fire.
	Move containers from the fire area if it is not dangerous.
	Extinguish the fire from a maximum distance or use an automatic fire extinguishing
	system during a tank fire.
	Do not allow water to get inside the container.
	Immediately withdraw from the area if there is a high hissing sound from the pressure
	relief device, or the tank discolors during a tank fire.
	Stay away from the tank in flames during a tank fire.

Glyceryl stearate	Maintain a safe distance when extinguishing the fire. Note that the material may be melted when transported.
	Dig a ditch to dispose of extinguishing water and keep material from dispersing.
	Move containers from the fire area if it is not dangerous.
	during a tank fire.
	Cool the tank with a large amount of water even after the tank fire is extinguished.
	Immediately withdraw from the area if there is a high hissing sound from the pressure
	relief device, or the tank discolors during a tank fire.
	Stay away from the tank in flames during a tank fire.
	In the case of a large tank fire, use unattained fire extinguishing equipment. If it is not
C.L. nigment black 11	available, stay away and let it burn. Rescuers should wear appropriate protective gear
Chi pignicit olicit Ti	Maintain a safe distance when extinguishing the fire.
	Note that the material may be melted when transported.
	Dig a ditch to dispose of extinguishing water and keep material from dispersing.
	Move containers from the fire area if it is not dangerous.
	Extinguish the fire from a maximum distance or use an automatic fire extinguishing system
	during a tank fire.
	Immediately withdraw from the area if there is a high hissing sound from the pressure
	relief device, or the tank discolors during a tank fire.
	Stay away from the tank in flames during a tank fire.
	In the case of a large tank fire, use unattained fire extinguishing equipment. If it is not
· · · ·	available, stay away and let it burn.
Iron oxide, yellow	Maintain a safe distance when extinguishing the fire.
	Dig a ditch to dispose of extinguishing water and keep material from dispersing
	Move containers from the fire area if it is not dangerous.
	Extinguish the fire from a maximum distance or use an automatic fire extinguishing system
	during a tank fire.
	Cool the tank with a large amount of water even after the tank fire is extinguished.
	Immediately withdraw from the area if there is a high hissing sound from the pressure
	Stay away from the tank in flames during a tank fire.
	In the case of a large tank fire, use unattained fire extinguishing equipment. If it is not
	available, stay away and let it burn.
Water	Cool the tank with a large amount of water even after the tank fire is extinguished.
	Immediately withdraw from the area if there is a high hissing sound from the pressure
	relief device, or the tank discolors during a tank fire.
	Stay away from the tank in flames during a tank fire.
	eves.
6. Accidental release measures	
A. Personal precautions, protective equipment, and	Stop leakage if it is not dangerous.
emergency procedures	
	Be aware of the materials and conditions to avoid. Wine up spills immediately, and follow preventive measures in the protective equipment
	section.
	Do not touch or walk on exposed substances.
	Remove all ignition sources.
	Stop leakage if it is not dangerous.
	Do not touch damaged containers or spills without wearing appropriate protective clothing.
	Cover the material with plastic sheets to prevent diffusion
	Be aware of the materials and conditions to avoid.
	Do not inhale dust, fume, gas, mist, vapors, or spray.
	Avoid inhaling dust, fume, gas, mist, vapor, or spray.

<ul><li>B. Environmental precautions</li><li>C. Purification or removal</li></ul>	Leaked material can cause contamination. Prevent inflow into waterways, sewages, basements, or sealed areas. Absorb the spill with inert material (e.g., dry sand or soil) and place chemical waste in a container. Absorb the liquid and wash the contaminated area with detergent and water. Cover the material with dry sand/soil or other inert substance and then with plastic sheets to prevent diffusion and exposure to rain. Use a clean explosion-proof tool to collect spills and place them in a loosely covered plastic container.
7. Handling and storage	
A. Safe handling	<ul> <li>Be aware of the materials and conditions to avoid.</li> <li>Refer to engineering controls and personal protective gear when working.</li> <li>Be careful of the high temperature.</li> <li>Follow all MSDS/label precautions since product debris may remain after emptying the container.</li> <li>Carefully handle and store the material</li> <li>Carefully open the cap before opening it.</li> <li>Avoid long-term or continuous contact with the skin.</li> <li>Do not breathe vapors generated from heated material.</li> <li>Do not access the storage area if not ventilated appropriately.</li> <li>Be aware of the materials and conditions to avoid.</li> <li>Be aware of the materials and conditions to avoid.</li> <li>Refer to engineering controls and personal protective gear when working.</li> <li>Read and understand all safety precautions before handling.</li> <li>Avoid inhaling dust, fume, gas, mist, vapor, or spray.</li> <li>Wash the handling part thoroughly after handling the product.</li> <li>Do not eat, drink, or smoke when using the product.</li> <li>Use the material only outdoors or in wall wartilated areas</li> </ul>
B. Safe storage method	Be aware of the materials and conditions to avoid. Fully drain empty drums and properly close them to immediately return to the drum regulator or place them adequately. Keep away from food and beverages. Be aware of the materials and conditions to avoid. Thoroughly seal the container and store it in a well-ventilated area. Store the product in an area with a locking system.

#### 8. Exposure controls and personal protection

A. Chemical material exposure criteria, biological exposure criteria, etc. Domestic regulation

Titanium dioxide Iron oxide Glyceryl stearate C.I. pigment black 11 Iron oxide, yellow Water ACGIH regulation Titanium dioxide

Iron oxide Glyceryl stearate C.I. pigment black 11 Iron oxide, yellow Water TWA - 10 mg/m<sup>3</sup> carcinogenic class 2 TWA - 5 mg/m<sup>3</sup> No data TWA - 5 mg/m<sup>3</sup> iron oxide No data No data TWA 10 mg/m<sup>3</sup> TWA 5 mg/m<sup>3</sup>

No data

No data

No data

No data

Biological exposure criteria	
Titanium dioxide	No data
Iron oxide	No data
Glyceryl stearate	Not applicable
C.I. pigment black 11	No data
Iron oxide, yellow	No data
Other expedite criteria	Not applicable
Titanium dioxide	No data
Iron oxide	No data
Glyceryl stearate	No data
C.I. pigment black 11	No data
Iron oxide, yellow	No data
Water	No data
B. Adequate engineering control	Use process isolation, local ventilation, or other engineering control measures to lower the air level to the exposure criterion or below.
B. Adequate engineering control	If the operation generates dust, fume, or mist, ventilate the area to maintain the air pollution below the exposure limit.
B. Adequate engineering control	Install an eye-washing facility and safety shower in the facility using or storing this material.
C. Personal protective gear	
Protection of the respiratory system	
Titanium dioxide	Carcinogenic class 2
Titanium dioxide	Wear respiratory protective gear suitable for the exposed particulate material's physical and chemical characteristics that are certified by the Korea Occupational Safety and Health Agency or equivalent.
Titanium dioxide	If the exposure concentration is lower than 100 mg/m <sup>3</sup> , wear half-face respiratory protective
	gear equipped with an appropriate filter.
Titanium dioxide	If the exposure concentration is lower than 250 mg/m <sup>3</sup> , wear loose-fitting hood/helmet-type
	electric respiratory protective gear or continuous flow dust mask equipped with an appropriate filter.
Titanium dioxide	If the exposure concentration is lower than 500 mg/m <sup>2</sup> , wear full-face, electric half-face, or
	air-supplied continuous flow/pressure-demanding half-face respiratory protective gear equipped with an appropriate filter.
Titanium dioxide	If the exposure concentration is lower than 10000 mg/m <sup>3</sup> , wear a full-face or helmet/hood-
	type pressure-demanding air-supplied respirator equipped with an appropriate filter.
Titanium dioxide	If the exposure concentration is lower than 100000 mg/m³, wear SCBA (self-contained
	breathing apparatus) or pressure-demanding SCBA respiratory protective gear equipped with an appropriate filter.
Iron oxide	Wear respiratory protective gear suitable for the exposed particulate material's physical and chemical characteristics that are certified by the Korea Occupational Safety and Health Agency or equivalent.
Iron oxide	If the exposure concentration is lower than 50 mg/m <sup>3</sup> , wear half-face respiratory protective
	gear equipped with an appropriate filter.
Iron oxide	If the exposure concentration is lower than 125 mg/ $m^3$ , wear loose-fitting hood/helmet-type
	electric respiratory protective gear or continuous flow dust mask equipped with an appropriate filter.
Iron oxide	If the exposure concentration is lower than 250 mg/m', wear full-face, electric half-face, or
	air-supplied continuous flow/pressure-demanding half-face respiratory protective gear equipped with an appropriate filter.
Iron oxide	If the exposure concentration is lower than 5000 mg/m <sup>*</sup> , wear a full-face or helmet/hood-type
	pressure-demanding air-supplied respirator equipped with an appropriate filter.
Iron oxide	If the exposure concentration is lower than 50000 mg/m <sup>3</sup> , wear SCBA (self-contained
	breathing apparatus) or pressure-demanding SCBA respiratory protective gear equipped with an appropriate filter
Glyceryl stearate	Wear respiratory protective gear suitable for the exposed particulate material's physical and chemical characteristics that are certified by the Korea Occupational Safety and Health Agency or equivalent.
Glyceryl stearate	For particulate material, the following respiratory protective gear is recommended. - The face filter-type dust mask, air-filtering dust mask (high-efficiency particulate filter material), or dust mask with electric fan (filter material for dust mist and fume)
Glyceryl stearate	If short of oxygen (<19.6%), wear an air-supplied mask or SBCA (self-contained breathing apparatus) respiratory protective gear.

C.I. pigment black 11 C.I. pigment black 11	Iron oxide Wear respiratory protective gear suitable for the exposed particulate material's physical and chemical characteristics that are certified by the Korea Occupational Safety and Health Agency or equivalent.
C.I. pigment black 11	If the exposure concentration is lower than 50 mg/m <sup>*</sup> , wear half-face respiratory protective
	gear equipped with an appropriate filter.
C.I. pigment black 11	If the exposure concentration is lower than 125 mg/m <sup>3</sup> , wear loose-fitting hood/helmet-type
	electric respiratory protective gear or continuous flow dust mask equipped with an appropriate filter.
C.I. pigment black 11	If the exposure concentration is lower than 250 mg/m <sup>3</sup> , wear full-face, electric half-face, or
	air-supplied continuous flow/pressure-demanding half-face respiratory protective gear equipped with an appropriate filter.
C.I. pigment black 11	If the exposure concentration is lower than 5000 mg/m <sup>3</sup> , wear a full-face or helmet/hood-
	type pressure-demanding air-supplied respirator equipped with an appropriate filter.
C.I. pigment black 11	If the exposure concentration is lower than 50000 mg/m <sup>3</sup> , wear SCBA (self-contained
	breathing apparatus) or pressure-demanding SCBA respiratory protective gear equipped with an appropriate filter.
Iron oxide, yellow	Wear respiratory protective gear suitable for the exposed particulate material's physical and chemical characteristics that are certified by the Korea Occupational Safety and Health Agency or equivalent.
Iron oxide, yellow	For particulate material, the following respiratory protective gear is recommended. - The face filter-type dust mask, air-filtering dust mask (high-efficiency particulate filter material) or dust mask with electric fan (filter material for dust mist, and fume)
Iron oxide, yellow	If short of oxygen (<19.6%), wear an air-supplied mask or SBCA (self-contained breathing apparatus) respiratory protective gear.
Water	Wear respiratory protective gear suitable for the exposed gas/liquid's physical and chemical characteristics that are certified by the Korea Occupational Safety and Health Agency or equivalent.
Water	For gas/liquid material, the following respiratory protective gear is recommended. - Separate-type full-face gas mask (organic compound use (acid gas use in case of acid gas)), separate-type half-face gas mask (organic compound use (acid gas use in case of acid gas)), direct connection-type full-face gas mask (organic compound use (acid gas use in case of acid gas)), half-face gas mask (organic compound use (acid gas use in case of acid gas)), or electric gas mask
Water	If short of oxygen (<19.5%), wear an air-supplied mask or SBCA (self-contained breathing apparatus) respiratory protective gear.
Protection of eyes	Wear protective or breathable goggles to protect your eyes from particulate material that may cause eye irritation or other health hazards.
Protection of eyes	Install emergency washing facilities (shower type) and eye-washing facilities at places where workers can easily access them.
Protection of eyes	Wear protective or breathable goggles to protect your eyes from organic compound vapors that cause eye irritation or other health hazards.
Protection of eyes	Install emergency washing facilities (shower type) and eye-washing facilities at places where workers can easily access them.
Protection of hands	Considering the chemical's physical and chemical properties, wear protective gloves made of appropriate materials.
Protection of body	Considering the chemical's physical and chemical properties, wear protective clothing made of appropriate materials.
Protection of body	Wear protective clothing for high-temperature or high-pressure splash protection as necessary.
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#### 9. Physicochemical characteristics A. Physical state

A. Physical state	
Appearance	No data
Color	No data
B. Odor	No data
C. Odor threshold	No data
D. pH	No data
E. Melting point/freezing point	No data
F. Initial boiling point and boiling point range	No data
G. Flash point	No data
H. Evaporation rate	No data
I. Flammability (Solid and vapor)	No data
J. Upper/lower limits of the ignition or explosion range	No data

	No data
L. Solubility M. Vener density	No data No data
N. Relative density	No data
O. n-octanol-water partition coefficient (Kow)	No data
P. Spontaneous ignition temperature	No data
Q. Decomposition temperature	No data
R. Viscosity	No data
S. Molecular weight	No data
Titanium dioxide	
A. Physical state	<b>6 1</b> 1 (
Appearance	Solid (crystal)
B Odor	Odorless
C. Odor threshold	No data
D. pH	7
E. Melting point/freezing point	1843°C
F. Initial boiling point and boiling point range	3000°C (ca)
H Evanoration rate	No data
I. Flammability (Solid and vapor)	No data
J. Upper/lower limits of the ignition or explosion range	- / -
K. Vapor pressure	No data
L. Solubility	500.6 mg/l No. doto
M. vapor density	
N. Relative density	3.9 (g/m <sup>-</sup> )
O. n-octanol-water partition coefficient (Kow)	No data
$\Omega$ Decomposition temperature	No data
R. Viscosity	No data
S. Molecular weight	79.865
Iron oxide	
A. Physical state	Colid (amotal marridan)
Appearance	Red to black
B. Odor	No data
C. Odor threshold	No data
D. pH	No data
E. Melting point/freezing point	1565°C No dete
G. Flash point	No data
	1 to data
H. Evaporation rate	No data
H. Evaporation rate I. Flammability (Solid and vapor)	No data Non-flammable
H. Evaporation rate I. Flammability (Solid and vapor) J. Upper/lower limits of the ignition or explosion range	No data Non-flammable - / -
H. Evaporation rate I. Flammability (Solid and vapor) J. Upper/lower limits of the ignition or explosion range K. Vapor pressure	No data Non-flammable - / - No data (Incolublo)
<ul> <li>H. Evaporation rate</li> <li>I. Flammability (Solid and vapor)</li> <li>J. Upper/lower limits of the ignition or explosion range</li> <li>K. Vapor pressure</li> <li>L. Solubility</li> <li>M. Vapor density</li> </ul>	No data Non-flammable - / - No data (Insoluble) No data
<ul> <li>H. Evaporation rate</li> <li>I. Flammability (Solid and vapor)</li> <li>J. Upper/lower limits of the ignition or explosion range</li> <li>K. Vapor pressure</li> <li>L. Solubility</li> <li>M. Vapor density</li> <li>N. Relative density</li> </ul>	No data Non-flammable - / - No data (Insoluble) No data 5.24
<ul> <li>H. Evaporation rate</li> <li>I. Flammability (Solid and vapor)</li> <li>J. Upper/lower limits of the ignition or explosion range</li> <li>K. Vapor pressure</li> <li>L. Solubility</li> <li>M. Vapor density</li> <li>N. Relative density</li> <li>O. n-octanol-water partition coefficient (Kow)</li> </ul>	No data Non-flammable - / - No data (Insoluble) No data 5.24 No data
<ul> <li>H. Evaporation rate</li> <li>I. Flammability (Solid and vapor)</li> <li>J. Upper/lower limits of the ignition or explosion range</li> <li>K. Vapor pressure</li> <li>L. Solubility</li> <li>M. Vapor density</li> <li>N. Relative density</li> <li>O. n-octanol-water partition coefficient (Kow)</li> <li>P. Spontaneous ignition temperature</li> </ul>	No data Non-flammable - / - No data (Insoluble) No data 5.24 No data No data
<ul> <li>H. Evaporation rate</li> <li>I. Flammability (Solid and vapor)</li> <li>J. Upper/lower limits of the ignition or explosion range</li> <li>K. Vapor pressure</li> <li>L. Solubility</li> <li>M. Vapor density</li> <li>N. Relative density</li> <li>O. n-octanol-water partition coefficient (Kow)</li> <li>P. Spontaneous ignition temperature</li> <li>Q. Decomposition temperature</li> <li>B. Viscocity</li> </ul>	No data Non-flammable - / - No data (Insoluble) No data 5.24 No data No data No data No data
<ul> <li>H. Evaporation rate</li> <li>I. Flammability (Solid and vapor)</li> <li>J. Upper/lower limits of the ignition or explosion range</li> <li>K. Vapor pressure</li> <li>L. Solubility</li> <li>M. Vapor density</li> <li>N. Relative density</li> <li>O. n-octanol-water partition coefficient (Kow)</li> <li>P. Spontaneous ignition temperature</li> <li>Q. Decomposition temperature</li> <li>R. Viscosity</li> <li>S. Molecular weight</li> </ul>	No data Non-flammable - / - No data (Insoluble) No data 5.24 No data No data No data No data 159 70
<ul> <li>H. Evaporation rate</li> <li>I. Flammability (Solid and vapor)</li> <li>J. Upper/lower limits of the ignition or explosion range</li> <li>K. Vapor pressure</li> <li>L. Solubility</li> <li>M. Vapor density</li> <li>M. Relative density</li> <li>O. n-octanol-water partition coefficient (Kow)</li> <li>P. Spontaneous ignition temperature</li> <li>Q. Decomposition temperature</li> <li>R. Viscosity</li> <li>S. Molecular weight</li> <li>Glyceryl stearate</li> </ul>	No data Non-flammable - / - No data (Insoluble) No data 5.24 No data No data No data No data No data 159.70
<ul> <li>H. Evaporation rate</li> <li>I. Flammability (Solid and vapor)</li> <li>J. Upper/lower limits of the ignition or explosion range</li> <li>K. Vapor pressure</li> <li>L. Solubility</li> <li>M. Vapor density</li> <li>M. Vapor density</li> <li>N. Relative density</li> <li>O. n-octanol-water partition coefficient (Kow)</li> <li>P. Spontaneous ignition temperature</li> <li>Q. Decomposition temperature</li> <li>R. Viscosity</li> <li>S. Molecular weight</li> <li>Glyceryl stearate</li> <li>A. Physical state</li> </ul>	No data Non-flammable - / - No data (Insoluble) No data 5.24 No data No data No data No data No data 159.70
<ul> <li>H. Evaporation rate</li> <li>I. Flammability (Solid and vapor)</li> <li>J. Upper/lower limits of the ignition or explosion range</li> <li>K. Vapor pressure</li> <li>L. Solubility</li> <li>M. Vapor density</li> <li>M. Vapor density</li> <li>N. Relative density</li> <li>O. n-octanol-water partition coefficient (Kow)</li> <li>P. Spontaneous ignition temperature</li> <li>Q. Decomposition temperature</li> <li>R. Viscosity</li> <li>S. Molecular weight</li> <li>Glyceryl stearate</li> <li>A. Physical state</li> <li>Appearance</li> </ul>	No data Non-flammable - / - No data (Insoluble) No data 5.24 No data No data No data No data 159.70
<ul> <li>H. Evaporation rate</li> <li>I. Flammability (Solid and vapor)</li> <li>J. Upper/lower limits of the ignition or explosion range</li> <li>K. Vapor pressure</li> <li>L. Solubility</li> <li>M. Vapor density</li> <li>N. Relative density</li> <li>O. n-octanol-water partition coefficient (Kow)</li> <li>P. Spontaneous ignition temperature</li> <li>Q. Decomposition temperature</li> <li>R. Viscosity</li> <li>S. Molecular weight</li> <li>Glyceryl stearate</li> <li>A. Physical state</li> <li>Appearance</li> <li>Color</li> <li>P. Odor</li> </ul>	No data Non-flammable - / - No data (Insoluble) No data 5.24 No data No data No data No data 159.70 Solid No data
<ul> <li>H. Evaporation rate</li> <li>I. Flammability (Solid and vapor)</li> <li>J. Upper/lower limits of the ignition or explosion range</li> <li>K. Vapor pressure</li> <li>L. Solubility</li> <li>M. Vapor density</li> <li>M. Relative density</li> <li>O. n-octanol-water partition coefficient (Kow)</li> <li>P. Spontaneous ignition temperature</li> <li>Q. Decomposition temperature</li> <li>R. Viscosity</li> <li>S. Molecular weight</li> <li>Glyceryl stearate</li> <li>A. Physical state</li> <li>Appearance</li> <li>Color</li> <li>B. Odor</li> <li>C. Odor threshold</li> </ul>	No data Non-flammable - / - No data (Insoluble) No data 5.24 No data No data No data 159.70 Solid No data No data No data
<ul> <li>H. Evaporation rate</li> <li>I. Flammability (Solid and vapor)</li> <li>J. Upper/lower limits of the ignition or explosion range</li> <li>K. Vapor pressure</li> <li>L. Solubility</li> <li>M. Vapor density</li> <li>N. Relative density</li> <li>O. n-octanol-water partition coefficient (Kow)</li> <li>P. Spontaneous ignition temperature</li> <li>Q. Decomposition temperature</li> <li>R. Viscosity</li> <li>S. Molecular weight</li> <li>Glyceryl stearate</li> <li>A. Physical state</li> <li>Appearance</li> <li>Color</li> <li>B. Odor</li> <li>C. Odor threshold</li> <li>D. pH</li> </ul>	No data Non-flammable - / - No data (Insoluble) No data 5.24 No data No data No data 159.70 Solid No data No data No data No data No data No data No data
<ul> <li>H. Evaporation rate</li> <li>I. Flammability (Solid and vapor)</li> <li>J. Upper/lower limits of the ignition or explosion range</li> <li>K. Vapor pressure</li> <li>L. Solubility</li> <li>M. Vapor density</li> <li>N. Relative density</li> <li>O. n-octanol-water partition coefficient (Kow)</li> <li>P. Spontaneous ignition temperature</li> <li>Q. Decomposition temperature</li> <li>R. Viscosity</li> <li>S. Molecular weight</li> <li>Glyceryl stearate</li> <li>A. Physical state</li> <li>Appearance</li> <li>Color</li> <li>B. Odor</li> <li>C. Odor threshold</li> <li>D. pH</li> <li>E. Melting point/freezing point</li> </ul>	No data Non-flammable - / - No data (Insoluble) No data 5.24 No data No data No data No data 159.70 Solid No data No data No data No data No data No data No data No data No data
<ul> <li>H. Evaporation rate</li> <li>I. Flammability (Solid and vapor)</li> <li>J. Upper/lower limits of the ignition or explosion range</li> <li>K. Vapor pressure</li> <li>L. Solubility</li> <li>M. Vapor density</li> <li>N. Relative density</li> <li>O. n-octanol-water partition coefficient (Kow)</li> <li>P. Spontaneous ignition temperature</li> <li>Q. Decomposition temperature</li> <li>R. Viscosity</li> <li>S. Molecular weight</li> <li>Glyceryl stearate</li> <li>A. Physical state</li> <li>Appearance</li> <li>Color</li> <li>B. Odor</li> <li>C. Odor threshold</li> <li>D. pH</li> <li>E. Melting point/freezing point</li> <li>F. Initial boiling point and boiling point range</li> </ul>	No data Non-flammable - / - No data (Insoluble) No data 5.24 No data No data No data No data 159.70 Solid No data No data No data No data No data No data No data No data No data No data
<ul> <li>H. Evaporation rate</li> <li>I. Flammability (Solid and vapor)</li> <li>J. Upper/lower limits of the ignition or explosion range</li> <li>K. Vapor pressure</li> <li>L. Solubility</li> <li>M. Vapor density</li> <li>N. Relative density</li> <li>O. n-octanol-water partition coefficient (Kow)</li> <li>P. Spontaneous ignition temperature</li> <li>Q. Decomposition temperature</li> <li>R. Viscosity</li> <li>S. Molecular weight</li> <li>Glyceryl stearate</li> <li>A. Physical state</li> <li>Appearance</li> <li>Color</li> <li>B. Odor</li> <li>C. Odor threshold</li> <li>D. pH</li> <li>E. Melting point/freezing point</li> <li>F. Initial boiling point and boiling point range</li> <li>G. Flash point</li> </ul>	No data Non-flammable - / - No data (Insoluble) No data 5.24 No data No data No data No data 159.70 Solid No data No data No data No data No data No data No data No data No data
<ul> <li>H. Evaporation rate</li> <li>I. Flammability (Solid and vapor)</li> <li>J. Upper/lower limits of the ignition or explosion range</li> <li>K. Vapor pressure</li> <li>L. Solubility</li> <li>M. Vapor density</li> <li>N. Relative density</li> <li>O. n-octanol-water partition coefficient (Kow)</li> <li>P. Spontaneous ignition temperature</li> <li>Q. Decomposition temperature</li> <li>Q. Decomposition temperature</li> <li>R. Viscosity</li> <li>S. Molecular weight</li> <li>Glyceryl stearate</li> <li>A. Physical state</li> <li>Appearance</li> <li>Color</li> <li>B. Odor</li> <li>C. Odor threshold</li> <li>D. pH</li> <li>E. Melting point/freezing point</li> <li>F. Initial boiling point and boiling point range</li> <li>G. Flash point</li> <li>H. Evaporation rate</li> <li>L. Flammability (Solid and vapor)</li> </ul>	No data Non-flammable - / - No data (Insoluble) No data 5.24 No data No data
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<ul> <li>H. Evaporation rate</li> <li>I. Flammability (Solid and vapor)</li> <li>J. Upper/lower limits of the ignition or explosion range</li> <li>K. Vapor pressure</li> <li>L. Solubility</li> <li>M. Vapor density</li> <li>N. Relative density</li> <li>O. n-octanol-water partition coefficient (Kow)</li> <li>P. Spontaneous ignition temperature</li> <li>Q. Decomposition temperature</li> <li>Q. Decomposition temperature</li> <li>R. Viscosity</li> <li>S. Molecular weight</li> <li>Glyceryl stearate</li> <li>A. Physical state</li> <li>Appearance</li> <li>Color</li> <li>B. Odor</li> <li>C. Odor threshold</li> <li>D. pH</li> <li>E. Melting point/freezing point</li> <li>F. Initial boiling point and boiling point range</li> <li>G. Flash point</li> <li>H. Evaporation rate</li> <li>I. Flammability (Solid and vapor)</li> <li>J. Upper/lower limits of the ignition or explosion range</li> <li>K. Vapor pressure</li> <li>L. Solubility</li> <li>M. Vapor density</li> <li>N. Relative density</li> <li>O. n-octanol-water partition coefficient (Kow)</li> </ul>	No data Non-flammable - / - No data (Insoluble) No data 5.24 No data No data
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C L nigment black 11	
A Physical state	
Appearance	Solid (fine black powder)
Color	Black
B. Odor	Odorless
C. Odor threshold	No data
D. pH	No data
E. Melting point/freezing point	1527°C
F. Initial boiling point and boiling point range	No data
G. Flash point	No data
H. Evaporation rate	No data
I. Flammability (Solid and vapor)	No data
J. Upper/lower limits of the ignition or explosion range	- / -
K. Vapor pressure	No data
L. Solubility	No data
M. Vapor density	No data
N. Relative density	5.18
O. n-octanol-water partition coefficient (Kow)	No data
P. Spontaneous ignition temperature	No data
Q. Decomposition temperature	No data
R. Viscosity	No data
S. Molecular weight	231.54

Turn and the sections	
A Dhysical state	
A. Physical state	Solid (nano motorial)
Color	No data
P. Odar	Scontless
C. Oder threshold	No data
	$25 \pm 75 (50 \text{ c/l} \text{ addiment})$
D. pri E. Malting point/fragging point	> 160% (or province to by 1012 h Da)
E. Initial hailing point and hailing point	> 100 C (approximately 1015 hPa)
F. Initial boining point and boining point range	No data
U. Frasil point	No data
I. Evaporation rate	No data Nat flowmahla
I. Frammaonity (Sond and vapor)	Not frammable
<i>K</i> Vanor processo	No data
K. vapor pressure	$< 1 u_{\alpha}/L (20^{\circ}C \text{ pH}; 8)$
E. Solubility	< 1 ug/L (20 C, pil. 8)
M. Vapor density	4.26 g/cm <sup>2</sup> (density)
N. Relative density	4.26 (20°C)
O. n-octanol-water partition coefficient (Kow)	No data
P. Spontaneous ignition temperature	No data
Q. Decomposition temperature	No data
R. Viscosity	No data
S. Molecular weight	159.69
Water	
A. Physical state	
Appearance	Liquid
Color	Colorless (transparent)
B. Odor	Odorless
C. Odor threshold	(Not applicable)
D. pH	7
E. Melting point/freezing point	0°C
F. Initial boiling point and boiling point range	100°C
G. Flash point	(Not applicable)
H. Evaporation rate	No data
I. Flammability (Solid and vapor)	Not applicable
J. Upper/lower limits of the ignition or explosion range	- / - (not applicable)
K. Vapor pressure	23.8 mmHg (25°C)
L. Solubility	100 g/100 ml
M. Vapor density	No data
N. Relative density	1
O. n-octanol-water partition coefficient (Kow)	-1.38

- P. Spontaneous ignition temperature Q. Decomposition temperature R. Viscosity
- S. Molecular weight

## 10. Stability and reactivity

A. Chemical stability and the possibility of the harmful

reaction

Titanium dioxide	The material can be decomposed at a high temperature and generate toxic gas.
Titanium dioxide	The container may explode when heated.
Titanium dioxide	Some may burn but are not easily ignitable.
Titanium dioxide	Non-flammable. Although the material itself does not burn, it may decompose to generate
	corrosive/toxic fumes when heated.

No data

No data

No data

18.02

Iron oxide Glyceryl stearate Glyceryl stearate Glyceryl stearate Glyceryl stearate C.I. pigment black 11 C.I. pigment black 11 C.I. pigment black 11 C.I. pigment black 11 Iron oxide, yellow Iron oxide, yellow Iron oxide, yellow Iron oxide, yellow Water Water B. Conditions to avoid Titanium dioxide Iron oxide Iron oxide Glyceryl stearate C.I. pigment black 11 Iron oxide, yellow Water C. Materials to avoid Titanium dioxide Iron oxide Glyceryl stearate C.I. pigment black 11 Iron oxide, yellow Water

The container may explode when heated. Leaked material has a risk of fire or explosion. May be reignited even after extinguishing. May ignite when in contact with moisture. Flammable/combustible material Some substances can flash and burn quickly. Some react violently with water. Some may explosively decompose when heated or under fire. Inhalation of decomposed products may lead to severe injury or death. Contact with the skin and eyes can cause severe burns. This material can generate irritating, corrosive, and toxic gas during a fire. The container may explode when heated. Some may burn but are not easily ignitable. Non-flammable. Although the material itself does not burn, it may decompose to generate corrosive/toxic fumes when heated. This material can generate irritating, corrosive, and toxic gas during a fire. The container may explode when heated. Some may burn but are not easily ignitable. Non-flammable. Although the material itself does not burn, it may decompose to generate corrosive/toxic fumes when heated. This material can generate irritating, corrosive, and toxic gas during a fire. The material can be decomposed at a high temperature and generate toxic gas. The container may explode when heated. Some may burn but are not easily ignitable. Non-flammable. Although the material itself does not burn, it may decompose to generate corrosive/toxic fumes when heated. Stable at room temperature and pressure condition The container may explode when heated. Ignition sources such as heat, sparks, and flames Humidity Ignition sources such as heat, sparks, and flames Ignition sources such as heat, sparks, and flames Ignition sources such as heat, sparks, and flames Ignition sources such as heat, sparks, and flames

Combustible and reducing materials Water Combustible and reducing materials

Heat and pollution

Combustible and reducing materials Combustible and reducing materials Water-reactive substances

D. Harmful materials generated during decomposition	
Titanium dioxide	Corrosive/toxic fume
Titanium dioxide	Irritating, corrosive, and toxic gases
Iron oxide	Irritating, corrosive, and toxic gases
Glyceryl stearate	Corrosive/toxic fume
Glyceryl stearate	Irritating, corrosive, and toxic gases
C.I. pigment black 11	Irritating and very toxic gases may be produced by pyrolysis or combustion.
C.I. pigment black 11	Corrosive/toxic fume
Iron oxide, yellow	Corrosive/toxic fume
Iron oxide, yellow	Irritating, corrosive, and toxic gases
Water	No data
11. Toxicological information	
A. Information on likely exposure paths	
Titanium dioxide	No data
Iron oxide	No data
Glyceryl stearate	No data
C.I. pigment black 11	No data
Iron oxide, yellow	No data
Water	No data
B. Information on health hazards	
Acute toxicity	
Orally administered	
Titanium dioxide	LD50 > 2000 mg/kg Mouse (OECD TG 420)
Iron oxide	LD50 > 10000 mg/kg rat (EU Method B.1; no disturbance observed)
Glyceryl stearate	No data
C.I. pigment black 11	No data
Iron oxide, yellow	LD50 > 10000 mg/kg Rat
Iron oxide, yellow	No data
Water	LD50 90000 mg/kg Rat (LD50 > 90 ml/kg (Rat))
Transdermal	
Titanium dioxide	No data
Iron oxide	No data
Glyceryl stearate	No data
C.I. pigment black 11	No data
Iron oxide, yellow	No data
Water	No data
Inhaled	
Titanium dioxide	Dust LC50> 6.82 mg/l rat (OECD TG 403, no mortality)
Iron oxide	Dust LC50 5.05 mg/l 4 hr rat (OECD TG403, GLP)
Glyceryl stearate	No data
C.I. pigment black 11	No data
Iron oxide, yellow	Mist discriminating conc. 5.05 mg/l 4 hr rat
Iron oxide, yellow	No data
Water	No data

Skin corrosion or irritation Titanium dioxide By the result of the skin corrosion/irritation test using rabbits, no irritation is observed. Erythema index = 0, OECD TG 404 Iron oxide By the result of the OECD G404 test, the skin irritation test using rabbits, no irritation is observed. Erythema and edema irritation index 0 Glyceryl stearate No data C.I. pigment black 11 Short-term contact causes skin irritation. Iron oxide, yellow Total irritation score: 0/4, no irritation Water Not applicable Severe eye damage or irritation Titanium dioxide By the result of the severe eye damage/irritation test using rabbits, no irritation is observed. Conjunctival redness index =1-2, OECD TG 405, GLP Iron oxide By the result of the OECD G405 GLP test, the eye corrosion/irritation test using rabbits, no irritation is observed. Irritation index 0 Glyceryl stearate No data C.I. pigment black 11 Short-term contact causes eye irritation. Iron oxide, yellow No irritation. Rabbit Not applicable Water Respiratory sensitization Titanium dioxide No data Iron oxide No data Glyceryl stearate No data C.I. pigment black 11 No data Iron oxide, yellow No data Water Not applicable Skin hypersensitivity By the result of the skin hypersensitivity test using guinea pigs, no skin hypersensitivity is Titanium dioxide observed. OECD TG 403 Iron oxide By the result of the hypersensitivity test using guinea pigs, it is not hypersensitive. Glyceryl stearate No data C.I. pigment black 11 No data Not hypersensitive. Guinea pig Iron oxide, yellow Water Not applicable Carcinogenicity Occupational Safety and Health Act Titanium dioxide No data Iron oxide No data Glyceryl stearate No data C.I. pigment black 11 No data Iron oxide, yellow No data No data Water Public Notice by the Ministry of Employment and Labor Titanium dioxide 2 Iron oxide No data Glyceryl stearate No data C.I. pigment black 11 No data Iron oxide, yellow No data Water No data

IARC	
Titanium dioxide	2B
Iron oxide	3
Glyceryl stearate	No data
C.I. pigment black 11	No data
Iron oxide, yellow	No data
Water	No data
OSHA	
Titanium dioxide	No data
Iron oxide	No data
Glyceryl stearate	No data
C.I. pigment black 11	No data
Iron oxide, yellow	No data
Water	No data
ACGIH	
Titanium dioxide	A4
Iron oxide	A4
Glyceryl stearate	No data
C.I. pigment black 11	No data
Iron oxide, yellow	No data
Water	No data
NTP	
Titanium dioxide	No data
Iron oxide	No data
Glyceryl stearate	No data
C.I. pigment black 11	No data
Iron oxide, yellow	No data
Water	No data
EU CLP	
Titanium dioxide	2 (only in the powder form containing 1% or more of particles with an aerodynamic diameter
Thailium dioxide	of 10 µm or less)
Iron oxide	No data
Glyceryl stearate	No data
C.I. pigment black 11	No data
Iron oxide, yellow	No data
Water	No data
Germ cell mutagenicity	
Titanium dioxide	The results of the reverse mutation assay using in vitro microorganisms (OECD TG 471), the mammalian cell gene mutation test (OECD TG 476), and the chromosomal abnormality test (OECD TG 473) are negative regardless of metabolic activation. Negative by the results of the in vivo chromosomal abnormality test and the decolorization test
Iron oxide	The results of the reverse mutation assay using in vitro microorganisms and the OECD TG473 GLP test, the chromosome aberration test using mammal (Chinese hamster) cells, are negative. The result of the in vivo comet assay using rats is negative. Analog material CAS No. 1317-61-9
Glyceryl stearate	No data
C.I. pigment black 11	No data
Iron oxide, yellow	Reverse mutation test using in vitro bacteria: Negative (S. typhimurium TA1535, TA1537, TA98, TA100 regardless of the metabolic activation system)

Water

Not applicable

Reproductive toxicity	
Titanium dioxide	By the result of the reproductive developmental toxicity test using rats, no clinical symptoms or weight changes are observed. NOAEL= 1000 mg/kg bw/day(OECD TG 210)
Iron oxide	No data
Glyceryl stearate	No data
C.I. pigment black 11	No data
Iron oxide, yellow	Reverse mutation test using in vitro bacteria: Negative (S. typhimurium TA1535, TA1537, TA98, TA100 regardless of the metabolic activation system)
Water	Not applicable
Toxicity to the specific target organ (One-time exposure)	
Titanium dioxide	By the result of the acute oral toxicity test using rats, no animals are deceased, and no weight changes or no significant lesions are observed at autopsy. OECD TG 425
Iron oxide	Causes irritation to the respiratory system.
Glyceryl stearate	No data
C.I. pigment black 11	Inhaling causes irritation to the respiratory system.
Iron oxide, yellow	Inhaling: 5 male and 5 female rats (Wistars) were singly exposed to 5 mg/l CERAC- pigment (average particle size = 35 nm) for 4 hours. Their mortality, clinical signs, and weights were observed for 14 days. All animals deceased during the observation period and sacrificed at the end of test period were pathologically examined. A single snout was only exposed for inhalation to CERAC-pigment for 4 hours at an aerosol concentration of 5 mg/L; all animals could tolerate exposure. Consequently, it was
<b>XX</b> 7 /	considered that the MLC (median lethal concentration) exceeded 5 mg/L. (OECD TG 403)
Water	Not applicable
Toxicity to the specific target organ (Repeated exposures)	The field of the transformed and the transformed
Titanium dioxide	By the result of the repeated oral toxicity test using rats, there was no death, and no significant effect was observed. NOAEL = $24,000 \text{ mg/kg}$ bw/davOECD TG 407
	By the result of the repeated inhalation toxicity test using mice and hamsters (0, 10, 50, or
	$250 \text{ mg/m}^3$ does 6 hours/dow 5 dows/weak for 12 weaks) multiple on a self
	250 mg/m dose, 6 hours/day, 5 days/week for 15 weeks), putnonary initialitation, cen
	toxicity, lung cell proliferation, and histopathological changes were observed. NOAEC = $10 \text{ mg/m3}$ . When testing animals such as rats, lung damage is observed if exposed to poorly soluble particles under overload conditions, which is considered species-specific. However, no pathological observations have been reported in similar tests on humans and other primates. In addition, no significant respiratory organ toxicity was found in epidemiologic studies in humans. Considering the above comprehensively, the data is insufficient to classify the toxicity to the specific target organ (repeated exposure).
Iron oxide	It has been reported that repeated inhalation exposure in humans may cause respiratory
~	effects (atresia) and may cause metal fume fever.
Glyceryl stearate	No data
C.I. pigment black 11	No data
Iron oxide, yellow	Subchronic inhalation toxicity (90 days) was tested on rats (male/temale). By the test result, an increase in lung and lung-associated lymph node weights was observed, but no other effects were observed. NOAEL 4.7 mg/L air (OECD TG413, GLP). Inhalation (subchronic): Rats were exposed to 3 different concentrations irregularly. The test was clearly consistent with poorly soluble particles and showed a typical result. No evidence
	of lung toxicity (NOAFI = $4.7 \text{ mg/m}^3$ ) rat OFCD TG 413 GLP
<b>XX</b> 7 /	
Water	Not applicable
Titaninu dianida	Nr. 1-4-
Intanium dioxide	No data
Iron oxide Chaogail stoorete	No data
C L nigment block 11	No data
U.I. pignient black II Iron oxide vellow	No data
Woter	Not applicable
water	not applicable

Other hazardous impacts Titanium dioxide Iron oxide Glyceryl stearate C.I. pigment black 11 Iron oxide, yellow Water

### 12. Ecological information

A. Ecological toxicity Fish Titanium dioxide Iron oxide Glyceryl stearate C.I. pigment black 11 Iron oxide, yellow Iron oxide, yellow Water Shellfish Titanium dioxide Iron oxide Glyceryl stearate C.I. pigment black 11 Iron oxide, yellow Iron oxide, yellow Water Birds Titanium dioxide Iron oxide Glyceryl stearate C.I. pigment black 11 Iron oxide, yellow Iron oxide, yellow Water B. Persistence and degradability Persistence Titanium dioxide Iron oxide Glyceryl stearate C.I. pigment black 11 Iron oxide, yellow Water Degradability Titanium dioxide Iron oxide Glyceryl stearate C.I. pigment black 11 Iron oxide, yellow Water

No data No data No data No data No data No data

 $\label{eq:loss} LC50 > 100 \mbox{ mg/l 96 hr Carassius auratus (OECD Guideline 203)} \\ LC0 \ge 50000 \mbox{ mg/l 96 hr Other (Danio rerio)} \\ LC50 \ 47.228 \mbox{ mg/l 96 hr} \\ No \ data \\ LC0 \ge 100000 \mbox{ mg/l 96 hr Danio rerio} \\ (Running \ freshwater) \\ No \ data \\ LC50 > 500 \mbox{ mg/l 48 hr Daphnia magna} \\ EC50 > 100 \mbox{ mg/l 48 hr} \\ \end{tabular}$ 

LC50 318.38 mg/l 48 hr No data LC50  $\geq$  100 mg/l 48 hr Daphnia magna (OECD TG 202, still freshwater, GLP) No data

EC50 > 50 mg/l 72 hr Selenastrum capricornutum No data EC50 3.72 mg/l 96 hr No data EC50 18 mg/l 72 hr pseudokirchneriella subcapitata (OECD TG 201 ) No data

No data No data No data No data No data log Kow -1.38

No data No data No data No data No data

C. Bio-accumulation	
Accumulation	
Titanium dioxide	No data
Iron oxide	No data
Glyceryl stearate	No data
C.I. pigment black 11	No data
Iron oxide, yellow	No data
Water	No data
Biodegradability	
Titanium dioxide	No data
Iron oxide	No data
Glyceryl stearate	No data
C.I. pigment black 11	No data
Iron oxide, yellow	No data
Water	No data
D. Soil mobility	
Titanium dioxide	No data
Iron oxide	No data
Glyceryl stearate	No data
C.I. pigment black 11	No data
Iron oxide, yellow	No data
Water	No data
E. Other hazardous impacts	
Titanium dioxide	No data
Iron oxide	No data
Glyceryl stearate	No data
C.I. pigment black 11	No data
Iron oxide, yellow	No data
Water	No data
13. Disposal considerations	
A. Method of disposal	
Titanium dioxide	No data
Iron oxide	1) Treat neutralization, hydrolysis, oxidation, and reduction.
	2) Incinerate or melt at a high temperature.
	3) Treat solidification.
Glyceryl stearate	If specified in the Wastes Control Act, dispose of contents and containers according to the
	regulation.
C.I. pigment black 11	If specified in the Wastes Control Act, dispose of contents and containers according to the
	regulation.
Iron oxide, yellow	If specified in the Wastes Control Act, dispose of contents and containers according to the
	regulation.
Water	If specified in the Wastes Control Act, dispose of contents and containers according to the
	regulation.

B. Disposal considerations	
Titanium dioxide	Dispose of contents and containers (according to the contents specified in the relevant
	laws).
Iron oxide	Dispose of contents and containers (according to the contents specified in the relevant
	laws).
Glyceryl stearate	Dispose of contents and containers (according to the contents specified in the relevant
	laws).
C.I. pigment black 11	Dispose of contents and containers (according to the contents specified in the relevant
	laws).
Iron oxide, yellow	Dispose of contents and containers (according to the contents specified in the relevant
	laws).
Water	Consider precautions described in the regulation if specified in the Wastes Control Act.

#### 14. Transport information

A. UN No. Titanium dioxide Iron oxide Glyceryl stearate C.I. pigment black 11 Iron oxide, yellow Water B. Proper shipping name Titanium dioxide Iron oxide Glyceryl stearate C.I. pigment black 11 Iron oxide, yellow Water C. Transport hazard class Titanium dioxide Iron oxide Glyceryl stearate C.I. pigment black 11 Iron oxide, yellow Water D. Container classification Titanium dioxide Iron oxide Glyceryl stearate C.I. pigment black 11 Iron oxide, yellow Water E. Marine pollutant Titanium dioxide Iron oxide Glyceryl stearate C.I. pigment black 11 Iron oxide, yellow

Water

No information is available for its UN Classification of Transport Hazardous Substances. 1376 No information is available for its UN Classification of Transport Hazardous Substances. No information is available for its UN Classification of Transport Hazardous Substances. No information is available for its UN Classification of Transport Hazardous Substances.

No information is available for its UN Classification of Transport Hazardous Substances.

Not applicable Iron oxide, spent or iron oxide sponge, spent, obtained from coal gas purification Aluminium calcium oxide -Not applicable 4-Methyl-N-[[(4-methylphenyl)amino]carbonyl]benzenesulfonamide Not applicable Not applicable 4.2 Not applicable Not applicable Not applicable Not applicable Not applicable

Not applicable III Not applicable Not applicable Not applicable Not applicable

No data Not applicable No data No data No data No data F. Particular safety measures required, or the user should be aware of related to transport or transport means Emergency measures during a fire

Emergency measures during a fire	
Titanium dioxide	Not applicable
Iron oxide	F-G
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable
Emergency measures during leaking	
Titanium dioxide	Not applicable
Iron oxide	S-P
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable

## 15. Regulatory information

A. Regulat	tion by the Occupational Safety and Health Act	
	Titanium dioxide	Harmful material subject to control
	Titanium dioxide	Substance subject to work environment measurement (measurement cycle: 6 months)
	Titanium dioxide	Substance material to exposure criteria
	Iron oxide	Harmful material subject to control
	Iron oxide	Substance subject to work environment measurement (measurement cycle: 6 months)
	Iron oxide	Substance subject to special health examination (diagnosis cycle: 12 months)
	Iron oxide	Substance material to exposure criteria
	Glyceryl stearate	No data
	C.I. pigment black 11	Harmful material subject to control (iron and its compounds)
	C.I. pigment black 11	Substance subject to work environment measurement (measurement cycle: (iron oxide dust and fumes))
	C.I. pigment black 11	Substance subject to special health examination (diagnosis cycle: iron oxide (dust and fumes only))
	C.I. pigment black 11	Substance material to exposure criteria
	Iron oxide, yellow	No data
	Water	No data
B. Regulat	tion by the Toxic Chemicals Control Act	
	Titanium dioxide	No data
	Iron oxide	No data
	Glyceryl stearate	No data
	C.I. pigment black 11	No data
	Iron oxide, yellow	No data
	Water	No data
C. Regulat	tion by the Act on the Safety Control of Hazardous	
Substances	S	
	Titanium dioxide	No data
	Iron oxide	No data
	Glyceryl stearate	No data
	C.I. pigment black 11	No data
	Iron oxide, yellow	No data
	Water	No data

D. Regulation by the Wastes Control Act	
Titanium dioxide	No data
Iron oxide	Designated waste
Glyceryl stearate	Designated waste
C.I. pigment black 11	No data
Iron oxide, yellow	No data
Water	No data
E. Other regulations by domestic and foreign laws	
Domestic regulation	
Titanium dioxide	
Iron oxide	
Glyceryl stearate	
C.I. pigment black 11	
Iron oxide, yellow	
Water	
Other domestic regulation	
Titanium dioxide	Not applicable
Iron oxide	Not applicable
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable
Foreign regulation	
US control information (OSHA regulation)	
Titanium dioxide	Not applicable
Iron oxide	Not applicable
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable
US control information (CERCLA regulation)	
Titanium dioxide	Not applicable
Iron oxide	Not applicable
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable
US control information (EPCRA 302 regulation)	
Titanium dioxide	Not applicable
Iron oxide	Not applicable
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable

US control information (EPCRA 304 regulation)	
Titanium dioxide	Not applicable
Iron oxide	Not applicable
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable
US control information (EPCRA 313 regulation)	
Titanium dioxide	Not applicable
Iron oxide	Not applicable
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable
US control information (Rotterdam Convention)	
Titanium dioxide	Not applicable
Iron oxide	Not applicable
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable
US control information (Stockholm Convention)	
Titanium dioxide	Not applicable
Iron oxide	Not applicable
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable
US control information (Montreal Protocol)	
Titanium dioxide	Not applicable
Iron oxide	Not applicable
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable
EU classification information (Finalized	
classification)	
Titanium dioxide	Not applicable
Iron oxide	Not applicable
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable
EU classification information (Hazard statement)	
Titanium dioxide	Not applicable
Iron oxide	Not applicable
Glyceryl stearate	Not applicable
C.I. pigment black 11	Not applicable
Iron oxide, yellow	Not applicable
Water	Not applicable

EU classification information (Safety statement) Titanium dioxide Iron oxide Glyceryl stearate C.I. pigment black 11 Iron oxide, yellow Water

Not applicable Not applicable Not applicable Not applicable Not applicable Not applicable

## 16. Other information

A. Source of information Titanium dioxide ECHA (Appearance) ECHA (Color) ECHA (B. Odor) ECHA (D. pH) ECHA (E. Melting point/freezing point) ECHA (F. Initial boiling point and boiling point range) ECHA (L. Solubility) ECHA (N. Relative density) ChemIDPlus (S. Molecular weight) ECHA (Orally administered) ECHA (Inhaled) OECD SIDS (Skin corrosion or irritation) ECHA (Severe eye damage or irritation) OECD SIDS (Skin hypersensitivity) OECD SIDS (Germ cell mutagenicity) OECD SIDS (Reproductive toxicity) OECD SIDS (Toxicity to the specific target organ (Onetime exposure)) OECD SIDS and ECHA (Toxicity to the specific target organ (Repeated exposures)) ECHA (Shellfish) ECHA (Birds) Iron oxide ICSC (Appearance) ICSC (Color) ICSC (E. Melting point/freezing point) HSDB (I. Flammability (Solid and vapor)) HSDB (L. Solubility) ICSC (N. Relative density) ECHA (Orally administered) ECHA (Inhaled) ECHA (Skin corrosion or irritation) ECHA (Severe eye damage or irritation) ECHA (Skin hypersensitivity) ECHA (Germ cell mutagenicity) (Reproductive toxicity)

ICSC (Toxicity to the specific target organ (One-time exposure)) nite, icsc, gestis (Toxicity to the specific target organ (Repeated exposures)) ECHA (Fish) ECHA (Shellfish) Glyceryl stearate Calculation by the molecular weight and the average molecular weight of air (M. Vapor density) ECOSAR (Fish) ECOSAR (Shellfish) ECOSAR (Birds) Lexemol T Inolex (Orally administered) | ECOSAR (Birds) | ECOSAR (Shellfish) | ECOSAR(Fish) | Calculation by the molecular weight and the average molecular weight of air (Vapor density) | ChemIDplus (Molecular weight) C.I. pigment black 11 14303 Chemicals (Japan) (Appearance) 14303 Chemicals (Japan) (N. Relative density) 14303 Chemicals (Japan) (S. Molecular weight) 14303 Chemicals (Japan) Iron oxide, yellow ECHA (Appearance) ECHA (B. Odor) GESTIS (D. pH) ECHA (E. Melting point/freezing point) ECHA (I. Flammability (Solid and vapor)) ECHA(L. Solubility) ECHA (M. Vapor density) ECHA (N. Relative density) ECHA (S. Molecular weight) ECHA (Orally administered) ECHA (Inhaled) ECHA (Skin corrosion or irritation) ECHA (Severe eye damage or irritation) ECHA (Skin hypersensitivity) ECHA (Germ cell mutagenicity) ECHA (Toxicity to the specific target organ (One-time exposure)) ECHA (Toxicity to the specific target organ (Repeated exposures)) ECHA (Fish) ECHA (Shellfish) ECHA (Birds) ECHA (D. Soil mobility) Water NLM B. Initial creation date 2022-09-21 C. # of revisions and the latest revision date # of revisions times The latest revision date 0 D. Other

• The prepared MSDS is the edited and partially corrected data based on the MSDS data provided by the KOSHA (Korea Occupational Safety and Health Agency).