

Fluoro H₂O₂TM

Hydrogen Peroxide/Peroxidase Detection Kit

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Notes

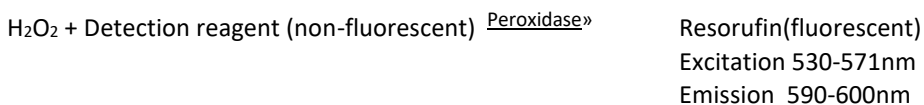
Revised protocol 5/06
Update 1/07
Revised protocol 06/05/2023

I. Assay Principle:

The Fluoro H₂O₂ detection kit utilizes a non-fluorescent detection reagent to measure H₂O₂. H₂O₂ oxidizes the detection reagent in a 1:1 stoichiometry to produce a fluorescent product resorufin. This oxidation is catalyzed by Peroxidase in a homogeneous no wash assay system.

The detection reagent can be utilized to measure H₂O₂ release from cells or enzyme coupled reactions (1-7)

Reaction:



II. Storage:

1. Short term (several weeks): at 2-8°C and away from light.
2. Long term: see individual components.
3. Once a vial of the Detection reagent is opened, it should be used promptly since it is subject to oxidation by air.

III. Warnings and Precautions:

1. For Research use only. Not for use in diagnostic procedures.
2. Practice safe laboratory procedures by wearing protective clothing and eyewear.
3. The fluorescent product of the detection reagent is not stable in the presence of thiols (DTT or 2-mercaptoethanol). Keep these reactants below 10uM. If you are using your own buffer, keep the reaction between pH 7-8 (optimal pH 7.4).
4. NADH and glutathione (reduced form: GSH) may interfere with the assay. See Technical note #5.

IV. Part # 5016. Kit contents (for 500 assays):

1. **Part # 3011. 5X Reaction Buffer:** 20 ml pH 7.4.
2. **Part # 4007. Detection reagent:** One vial for 500 assays.
3. **Part # 3012. Hydrogen Peroxide:** 200µL of a stabilized 3% solution.
4. **Part # 6004. Horseradish Peroxidase:** 10µL, enzyme unit varies batch to batch

Materials required but not supplied:

1. Dimethyl sulfoxide (DMSO)
2. Black 96-well plates
3. Fluorescence plate reader
4. Deionized water

V. Preparation of reagent working solutions:

1. **20 ml of 1X Reaction buffer Part# 3011:** 4ml of 5X Reaction buffer is added to 16ml of deionized water to make 1X reaction buffer. This should be sufficient for performing 100 assays of 100µL each and for preparing stock solutions of the enzyme and H₂O₂.

Note: It is important to equilibrate the buffer to room temperature before use as crystals may form on storage. This can be done by warming in a 37°C water bath or incubator for a few minutes.

2. **20mM Hydrogen Peroxide Part # 3012:** To 977 μ L of 1X Reaction buffer, add 22.7 μ L of the 3% H₂O₂ (0.88M) to make a 20mM solution. Once diluted, the H₂O₂ should be used promptly as it degrades rapidly.
3. **10U/ml Horseradish Peroxidase Part# 6004:** Quickly spin down the contents of the vial before opening. Based on the provided enzyme activity, calculate the amount of 1X reaction buffer needed to bring HRP down to 10units/ mL. Use entire content and dilute the vial. Once diluted, the unused HRP should be stored at -20°C as single use aliquots.
Note: due to small volume of HRP, if the entire vial is not diluted, over the time HRP can dry out.
See Technical note #4.
4. **10mM Detection Reagent Part# 4007:** Dissolve the contents of the vial in 500 μ L of DMSO. Once opened, should be used promptly and any remaining reagent can be aliquoted and frozen at -70°C. **Avoid repeated freeze thaw cycles.**

VI. Assay Protocol: Detection of Hydrogen Peroxide in Supernatants

1. Prepare 5ml **reaction cocktail** (for 100 assays) as follows:
100 μ L of 10mM Detection Reagent (50 μ M final)
200 μ L of 10U/ml HRP (0.1 U/ml)
4.7ml of 1X Reaction buffer
2. To prepare an H₂O₂ standard curve, dilute the appropriate amount of 20mM H₂O₂ in 1X Reaction buffer to make concentrations ranging from 0 to 10 μ M, each in a volume of 100 μ L (for duplicate runs).
Note: See Technical note #4.
3. Be sure to include a positive and negative (no H₂O₂) control in the assay if a standard curve is not used.
4. Pipette 50 μ L of the H₂O₂ dilutions, controls and samples into the wells of a 96- well plate.
5. Add 50 μ L of the reaction cocktail to each well to begin the reaction.
6. Incubate for 10 minutes, at RT, away from light.
7. Measure fluorescence at excitation: 530-570nm (570nm is the optimal excitation) and emission 590-600nm in a fluorescent plate reader.
8. Subtract background fluorescence (mean RFU value without H₂O₂) from each reading.

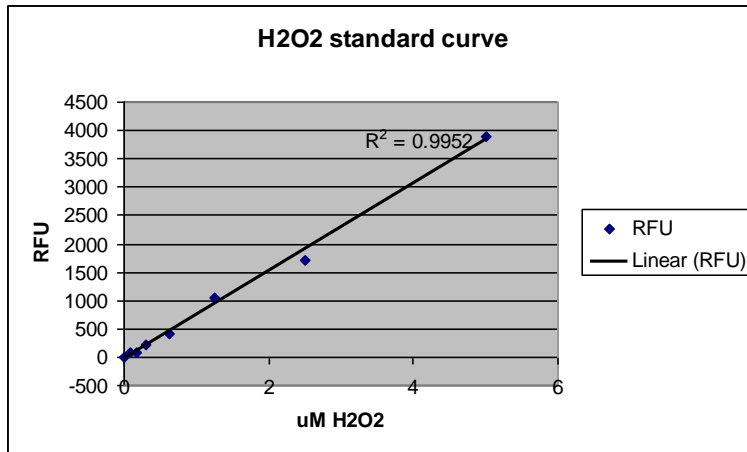


Figure 1: Example of standard curve in 1X reaction Buffer.

VII. Assay Protocol: Detection of Hydrogen Peroxide from Cells.

The Fluoro H₂O₂TM detection kit can be used to measure the release of H₂O₂ from cells. The following is a suggested protocol and can be modified to suit your particular research needs.

1. The reaction cocktail should be prepared in media that is used in your cell culture system. Serum should be reduced or avoided, as it will interfere with the reaction cocktail and may contain catalase activity, consequently compromising sensitivity. Suggested media to use:
 - A. Hanks Balanced Salt Solution
 - B. Krebs Ringers Phosphate Buffer
 - C. Serum Free Media (use with caution).
2. Just prior to use, prepare 5ml **reaction cocktail** (for 50-100 assays) as follows:
 - 100μL of 10mM Detection Reagent (50 μM final)
 - 200μL of 10U/ml HRP (0.1 U/ml)
 - 4.7ml of Media (media as suggested above or used in your experiment).
3. Harvest cells and wash in fresh media (see media step V 1). Plate cells in a 96 well plate at a concentration of 10,000 to 50,000 cells per well in a volume of 50 to 100 μL. Include the appropriate negative controls to measure background fluorescence.
 1. Media alone.
 2. Non-activated cells.

Construct a standard H₂O₂ curve (see above step IV 2) in the same media in which the cells are plated in. Keep the volumes in the cell sample and standard H₂O₂ curve constant.

Note: Each investigator should optimize the cell concentration and volume for their particular protocol. See Technical note 4.

5. Activate your cells according to your experimental protocol.
6. After activation add 50-100 μl of reaction cocktail to your cells and standard curve.
7. Incubate for 10 minutes, at RT, away from light.
8. Measure fluorescence at excitation: 530-570nm and emission 590-600nm in a fluorescent plate reader.

VIII. Assay Protocol: Detection of Peroxidase Activity:

The Fluoro H₂O₂™ kit can also be used to assay peroxidase activity.

1. Prepare 5ml of **reaction cocktail** (for 100 assays) as follows:
500 µL of 20mM H₂O₂ solution (2 mM final concentration)
50 µL of 10mM Detection Reagent (100µM final concentration)
4.45 ml of 1X Reaction buffer
2. To prepare a Peroxidase standard curve, make dilutions of the supplied Horse radish peroxidase in 1X Reaction buffer to make concentrations ranging from 0 to 25 mU/ml, each in a volume of 100 µL for duplicate runs.
Note: High levels of HRP (100mU/ml, final concentration) will produce lower fluorescence than 1mU/ml, because the excess HRP oxidizes the fluorescent reaction product, resorufin to non-fluorescent resazurin.
3. Be sure to include a positive and negative (no H₂O₂) control in the assay if a standard curve is not used.
4. Pipette 50µL of the H₂O₂ dilutions, controls, and samples into the wells of a 96- well plate.
5. Add 50µL of the reaction cocktail to each well to begin the reaction.
6. Incubate, away from light, if necessary.
Note: This step may need to be optimized as incubation times could vary from a few minutes to several minutes.
7. Measure the fluorescence at excitation: 530-570nm and emission at 590-600nm in a fluorescent plate reader.
8. Subtract the background fluorescence (mean RFU value without H₂O₂) from each reading.

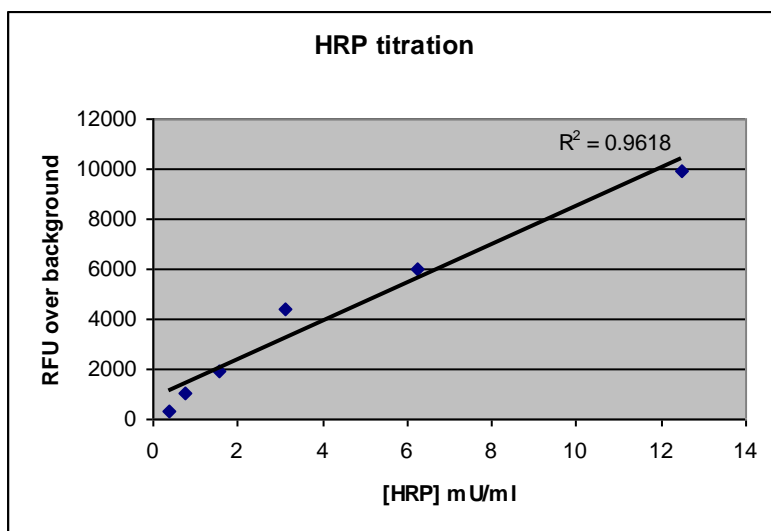


Figure 2: Standard curve of Peroxidase in 1X Reaction buffer

IX. Technical Notes:

1. The final concentrations of H₂O₂ will be 2X lower than used since the final reaction volume is 100μL. (0 to 5μM). The final concentration of the detection reagent is 100μM and that of HRP is 0.2U/ml in a 100μl reaction.
2. High concentrations of H₂O₂ (>100μM) will result in lower fluorescence due to oxidation of the fluorescent reaction product, resorufin to non-fluorescent resazurin.
3. The reaction cocktail once prepared, cannot be stored.
4. Preparation of H₂O₂ standard curve: It is important to prepare the standard curve in the same matrix or media as your sample. If your samples are in PBS, you can use our **1X Reaction Buffer** to construct your standard curve. If your samples are in media, prepare your standard curve in the same media. Serum based media tends to suppress the fluorescent signal.
5. At NADH levels above 10uM, detection reagent oxidation results from side chain reaction between NADH and HRP. This could result in aberrant readings. To minimize this interference, it is recommended to add superoxide dismutase (SOD) at 40U/mL to the reaction ⁸.
At glutathione (reduced form GSH) above 300uM, detection reagent oxidation results from side chain reaction between GSH and HRP. This could result in aberrant readings. To minimize this interference, it is recommended to add superoxide dismutase (SOD) at 40U/mL to the reaction ⁸.
6. The 5X reaction buffer should be equilibrated to room temperature before use as crystals may form upon storage.
7. Small, salinized tube might not be enough to add entire 1X buffer volume. Wash the salinized tube few times and transfer the content into appropriate size container. Bring the volume up to desired value with 1X reaction buffer. Be sure to thoroughly Pipette the HRP stock up and down several times and vortex the vial to be mixed properly prior to pipetting out.

References:

1. Mingjie Zhou, Zhenjun Diwu, Nataliya Panchuk-Voloshina and Richard P. Haugland. A Stable Nonfluorescent Derivative of Resorufin for the Fluorometric Determination of Trace Hydrogen Peroxide: Applications in Detecting the Activity of Phagocyte NADPH Oxidase and Other Oxidases. *Anal Biochem* 253, 162 (1997).
2. J. G. Mohanty, Jonathan S. Jaffe, Edward S. Schulman and Donald G. Raible. A highly sensitive fluorescent micro-assay of H₂O₂ release from activated human leukocytes using a dihydroxyphenoxazine derivative. *J. Immunol Methods* 202, 133 (1997).
3. Tatyana V. Votyakova and Ian J. Reynolds. Membrane Potential dependent and -independent production of reactive oxygen species by rat brain mitochondria. *J Neurochem* 79, 266 (2001).
4. Chun Song, Abu B. Al-Mehdi, and Aron B. Fisher. An immediate endothelial cell signaling response to lung ischemia. *Am J Physiol Lung Cell Mol Physiol* 281, L993 (2001).
5. Samantha C. Richer and W.C.L. Ford. A critical investigation of NADPH oxidase activity in human spermatozoa. *Mol Hum Reprod* 7, 237 (2001).
6. William G. Gutheil, Miglena E. Stefanova and Robert A. Nicholas. Fluorescent Coupled Enzyme Assays for Γ -Alanine: Application to Penicillin-Binding Protein and Vancomycin Activity Assays. *Anal Biochem* 287, 196 (2000).
7. Dominik Peus, Remus A. Vasa, Astrid Beyerle, Alexander Meves, Carsten Krautmacher and Mark R. Pittelkow. UVB Activates ERK1/2 and p38 Signaling Pathways via Reactive Oxygen Species in Cultured Keratinocytes. *J Invest Dermatol* 112, 751 (1999).
8. Tatyana V.Votyakova ,Ian J.Reynolds. Detection of hydrogen peroxide with Amplex Red:interference by NADH and reduced glutathione auto-oxidation. *Archives of Biochemistry and Biophysics*, 431: 138-144 (2004)."