



CALIFORNIA CHEMICAL


555 S Anton Blvd Suite 150
Costa Mesa, CA 92626
424-292-3424

Safety Data Sheet for Isopropyl alcohol Page 1 of 8

I. Identification

Chemicals name: Isopropyl Alcohol 70%
Other Information: -
Relevant identified uses of the substance or mixture and uses advised against: Manufacture of acetone and derivatives; manufacture of glycerol and isopropyl acetate; solvents for aromatic and other oils, alkaloids, glue, resins; possible solvents for cellulose derivatives; coating solvent; anti-freeze agent for liquid fuel; enamel; extract processing; dehydrating agent; preservatives; lotion; denaturant.

II. Hazard Identification:

Hazard Category: Class 2 flammable liquids, Class 5 acute toxic substance (ingestion), Class 3 Specific target organ systematic toxicity - Single exposure, Class 2A severe injury/eye irritation substance
Labeled Contents:  Symbols: Flames, Warning Warning: Danger Hazard Warning Information: Highly flammable liquid and vapor Harmful if ingested Causes slight skin irritation Causes eye irritation Hazard Prevention Measures: Place container in a well-ventilated area Keep away from inflammables. – Smoking prohibited Prevent eye contact Do not dispose in drainage canals Prevent static electricity
Other Hazards: -

III. COMPOSITION OF MIXTURE

Single:



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English Name: Isopropyl Alcohol
Synonyms: isopropanol, 2-Propanol, Dimethylcarbinol, Sec-Propyl Alcohol, Isopropanol
Chemical Abstracts Number (CAS No.): 67-63-0
Percentage for Chemical Ingredient (%): >70%

VI. First Aid Measures:

Emergency and First Aid Procedures:

Inhalation:

1. Remove the pollution source or the patient to a place with fresh air.
2. If breathing has stopped, apply artificial respiration immediately.
3. Seek medical attention immediately.

Skin Contact:

1. Use gentle, running warm water to rinse the injured area for more than 15 minutes as soon as possible.
2. Remove contaminated clothes and shoes when flushing with water. Contaminated clothes must be washed thoroughly before disposal.
3. If irritation persists, seek medical attention immediately.

Eye Contact:

1. Immediately lift upper and lower eyelids, wash with warm water for 20 minutes.
2. Seek medical attention immediately.

Ingestion:

1. Let the patient drink large amount of water to induce vomiting unless the patient is unconscious or has a cramp.
2. Seek medical attention immediately.

Major Disease and Harm Effect: Irritation. Exposure to large amounts can cause unconsciousness and death.

First-Aid Personal Protection: Must wear Class C protective gear and perform emergency rescue in safe area.

Prompt to Doctor: -

V. Fire Fighting Measure:

Suitable Extinguishing Media: carbon dioxide, chemical powder, and alcoholic foam

Special Exposure Hazards:

1. Vapors and liquids are flammable. Liquid will accumulate electric charges. Vapor is heavier than air and may float to places far away, and may flashback from ignition sources.
2. High heat will cause this material to decompose and produce toxic gas. The containers in a fire site may rupture and explode. ,

Special Extinguishing Procedure:

1. Retreat and extinguish the fire from a safe distance or a protected area.
2. Stay upwind to keep away from hazardous vapor and toxic decomposition.
3. Any leakage should be stopped before extinguishing the fire. If the leakage cannot be stopped and there is no immediate danger in the surrounding area, allow it to burn away. If the leakage is not stopped before extinguishing the fire, the vapor and the air will form an explosive mixture and ignite afterwards.
4. Separate materials that are not on fire and protect the personnel.
5. Move the container



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away from the fire field under safe conditions. 6. Use water mist to cool the tanks or containers in exposed the fire field. 7. Using water fog to extinguish fire may be ineffective without trained fire-fighting personnel. 8. If the leakage is not ignited, spray water mist to disperse vapor and protect the personnel who try to stop the leakage. 9. A water spout is ineffective for extinguishing the fire. 10. For a big fire in a large area, use the unmanned water mist stand or the automatic water fire monitor. 11. Retreat from the fire field and allow the fire to burn out. 12. Stay away from the tanks. 13. When the safety valve alarm of the tank sounds or the color changes due to fire, retreat immediately. 14. Personnel without special protective equipment should not enter the fire field.

Special Protection Equipment: Fire fighters must wear air respirators, protective gloves, and fire fighting coats.

VI. Accidental Release Measures:

Personal Protection:

1. Before the polluted area is cleaned up completely, access to the area should be restricted.
2. Make sure the cleaning work is performed by trained personnel.
3. The personnel should wear appropriate personal protective equipment.

Environmental Protection:

1. The air in the area should be well ventilated.
2. All flammable sources should be extinguished or eliminated.
3. Report the accident to the safety, health, and environmental protection authorities of the government.

Methods for Cleaning Up:

1. Do not come in contact with the released chemical.
2. Avoid the released chemical from entering the sewers or sealed spaces.
3. Stop or reduce the leakage under safe conditions if possible.
4. Use soil, sand or similar inert non-combustible substances that will not react with the spill to surround the spill.
5. For small spills, absorb using an absorbent that will not react with the spill. The polluted absorbent becomes as harmful as the released chemical and should be placed in the appropriate container that is capped and labeled. Use water to clean up the leakage area.
6. For large spills, contact the fire department, emergency rescue units and supplier for assistance.

VII. Handling and Storage:

Handling:

1. This substance is a highly flammable and toxic liquid. Therefore, personal protective equipment should be operative and utilized during disposal. The staff should receive relevant trainings on the risk and safe handling of this substance.
2. All flammable sources should be removed and kept away from heat and incompatible substances.
3. The "Smoking Prohibited" sign should be present in the work area.
4. The liquid will accumulate an electric charge. Therefore, a design to increase the conductivity should be taken into consideration. For example, all tanks, transfer containers, and lines should be grounded. Any naked metal should be connected for grounding. During the operation, the flow rate should be reduced to increase the operation time, the duration of the liquid retained in the lines should be increased, and the operation should be performed under low temperature.
5. When the operation is not conducted in the sealed system, the connections between the



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operation container and the receiving transmission equipment should be at the equivalent electric potential. 6. The empty tanks, containers, and lines may contain harmful residues. Therefore, there should be no welding, cutting, drilling, or heating before they are cleaned. 7. The ventilation system or equipment that does not produce sparks used in the work area should be explosion-proof. 8. The formation of mist or vapor during the operation should be avoided. The operation should be conducted in a well-ventilated area in the smallest quantity possible. The operation area should be separated from the storage area. 9. Wear appropriate personal protective equipment when necessary to avoid contact with this chemical substance or the polluted equipment. 10. Do not use with incompatible chemicals (such as strong oxidants) in order to decrease the risks of fire and explosion.

Storage:

1. Make sure that the passageways and exits are clear and unobstructed. 2. Consider the installation of leakage and fire detection systems, auto fire extinguishing system or enough usable emergency handling equipment in the storage area and operation area for large amounts. 3. Use containers made of compatible materials. Be careful to avoid spills during repacking. 4. Do not try to send out the material by adding pressure using air or inert gases. 5. Do not mix the material in the storage area unless this area is isolated with fireproof structure. 6. Use approved container and mixing equipment for flammable liquids. 7. Do not pour contaminated liquid back to the original storage tank. 8. Containers must be labeled, tightly sealed and kept from damages when not in use. 9. The chemical should be stored in cool, dry, and well-ventilated area away from direct sunlight. Keep away from heat sources, flammable sources, and incompatibles. 10. Storage equipment should be constructed with fire-resistant materials. 11. The floor should be constructed with impermeable material to avoid absorption by the floor. 12. Set up slopes, doorsills or furrows at doorways so that leaking substances are discharged to a safe place. 13. The storage area should have clear signs and be free from impediments. Only designated or trained personnel are allowed to enter. 14. The storage area and the work area should be separated. The chemical should be stored away from lifts, buildings, room entrances, or major accesses. 15. Appropriate fire extinguishers and leak clean-up equipment should be available near the storage area. 16. Containers should be checked for damages or leakage regularly. 17. All new containers should be checked for the appropriate labels and any damage. 18. The quantity of chemical to be stored should be limited. 19. The containers made from compatible materials should be used to store the released chemical. 20. The storage tanks should be grounded and connected to other equipment at equivalent electric potential. 21. Tanks used for storing inflammable liquids must be installed with relief valve and vacuum relief valve. 22. The chemical should be stored at the temperature suggested by the chemical manufacturer or supplier. If necessary, the temperature detection alarm should be installed to alert if the temperature is too high or too low. 23. The storage of large quantities should be avoided. The chemical should be stored in the isolated fireproof building. 24. The exhaust pipes of tanks should be installed with flame arresters. 25. The storage tanks should be built on the ground level with the entire bottom sealed to prevent leakage. The liquid control dam able to hold the entire volume should be present in the surrounding area.

VIII. Exposure Control / Personal Protection:



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Engineering Control:			
<ol style="list-style-type: none"> 1. Complete gas exchange or partial exhaust devices. 2. Will not produce sparks if used alone and grounded to the ventilation system. 3. Exhaust opening is connected directly outside the window. 4. Provide enough fresh air to supplement the air exhausted by the exhaust system. 			
Control Factor:			
TWA	STEL	CEILING	BEIs
400 ppm	500 ppm	—	40mg/L (Acetone in urine)
Personal Protection Equipment:			
<p>Respiratory Protection:</p> <ol style="list-style-type: none"> 1. Below 2000ppm: Fixed amount air supplied type respirator, respiratory respirator with organic vapor filter cartridge powered for air purification or full chemical filter cartridge, respirator with organic vapor filter cartridge, full self-contained or air supplied respirator. 2. Unknown Concentration: positive-pressure self-contained respiratory apparatus, positive-pressure full air-supply respiratory apparatus with positive-pressure self-contained respiratory apparatus. 3. Escape: Gas mask with organic vapor filter cartridge, life escape self-contained breathing apparatus. <p>Hand Protection:</p> <ol style="list-style-type: none"> 1. Impermeable gloves made of butyl rubber, rubber-like, Viton, 4H, CPF 3, Responder, etc. <p>Eye Protection:</p> <ol style="list-style-type: none"> 1. Anti chemical splashing safety goggles, full face masks <p>Skin & Body Protection:</p> <ol style="list-style-type: none"> 1. Leak-proof gloves made of butyl rubber, rubber-like, Viton, 4H, etc. 			
Hygiene Procedures:			
<ol style="list-style-type: none"> 1. Polluted clothes should be removed as soon as the work is completed. The clothes should be worn or discarded only after being washed. The washing staff should be informed of the harmful effects of the pollution. 2. Eating, drinking, and smoking are strictly prohibited in the work area. 3. Wash hands thoroughly after handling the substance. 4. Keep the work area clean. 			

IX. Physical and Chemical Properties / Characteristics:

Appearance: Colorless liquid	Odor: Rubbery alcohol odor
Odor threshold: 3.3-610 ppm (detected), 7.6-49ppm (sensed)	Melting point: - 88.5
pH value:	Boiling Point / Boiling Range: 82.3°C
Flammability (solid, gas):	Flash Point: 12 °C
Decomposition Temperature: -	Test Method: Close Cup
Spontaneous Temperature: 399 °C	Exposure Limits: 2.0 %~12
Vapor Pressure: 33 mmHg @ 20°C	Vapor Density: 2.07
Specific Gravity : 0.785	Solubility in Water: Completely soluble



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Log Kow: 0.05	Percent volatile: 1.5 (n-Butyl Acetate = 1)
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X. Stability and Reactivity:

Stability: Under normal stable condition, may slowly form into peroxides.
Special Conditions of Hazardous Reaction: 1. Strong oxidants (such as nitrates, perchlorates and peroxides): increased risks of fire and explosion. 2. Phosgene: produces isopropyl chlorocarbonate and hydrochloric acid. 3. Ferric salt: causes explosive heat decomposition reaction. 4. Hydrogen – Palladium: may catch fire if mixed in the air. 5. Strong acid: May cause violent reaction. 6. Alkali metals or alkali earth metals: may release flammable toxic gases.
Conditions to Avoid: Heat, sparks, static electricity, ignition sources, light
Incompatibility: Ferric salt, hydrogen – palladium, strong oxidizer, phosgene
Hazardous Decomposition Products: -

XI. Toxicological Information

Exposure route: skin contact, inhalation, ingestion, eye contact
Symptoms: Irritation, dizziness, narcosis, nausea, vomiting, diarrhea
Acute Toxicity: Skin: 1. Short period of exposure will not irritate skin. Inhalation: 1. Concentration of below 400 ppm will cause light irritation of the upper respiratory tract. 2. High concentration will cause dizziness, loss of motor functions (loss of coordination), and deep coma. Ingestion: 1. May cause dizziness, stomachache, painful cramps, nausea, vomiting and diarrhea. 2. Exposure to large amount will cause unconsciousness and death. 3. Estimated fatal dosage is about 131g. Eye: 1. Concentration of below 400 ppm will cause light irritation. 2. Direct contact of liquid with the eyes will cause acute irritation. LD50 (Test animal, absorption route): 4710mg/kg (rat, ingestion) LC50 (Test animal, absorption route): 16000ppm/8H (rat, inhalation)
Chronic: 1. Skin: Prolonged or frequent skin contact may cause dryness and peeling. ° 2. Ingestion: After daily ingestion of 6.4mg/kg isopropyl alcohol for 6 weeks, the blood and urine show no special changes in the chemical or cellular compositions. 3500ppm/7H (Female rats pregnant for 1-19 days, ingestion) causes incomplete development of the embryo. IARC listed it as Group 3: Cannot be determined as carcinogenic in humans.

XII. Ecological Information:

Eco-toxicity: LC50 (Fish): -



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EC50 (aquatic invertebrates): -

Bio-concentration Factor (BCF):

Durability and Degradability:

1. Results from 4 experiments showed that after 5 days (20) in the sewage, isopropyl alcohol can decompose 58% of the BOD theoretical value. 2. When released into water, it is expected to evaporate (estimated half-life is 5.4 days) and can be biodegraded (although it decomposed quickly in the laboratory but there is no relevant data in natural water sources). 3. When released into the air, it is expected to undergo photolysis (half-life is 1 to several days). Since it is water-soluble, it may be washed down by the rain.

Half-life (air): 6.2~72 hr

Half-life (water surface): 24~168 hr

Half-life (underground water): 48~336 hr

Half-life (soil): 24~168 hr

Biological Accumulation: Will not accumulate inside the body.

Fluidity in the Soil: When released into the soil, its high vapor pressure, faced with low adsorption from the soil, will cause it to evaporate quickly and seep into the ground.

Other adverse effects: Highly toxic to aquatic or marine life.

XIII. Disposal Information:

Disposal Information:

1. Bury in specified landfills or incinerate in approved solvent incinerators. 2. If small amounts flow into drainages or gutters, wash with large amounts of water to prevent the accumulation of flammable gases. 3. For large amounts report to the environmental protection unit.

XIV. Transport Information:

The United Nations Number (Un-No): 1219

UN Transport Name: Isopropyl alcohol

Transport Hazard Classification: Class 3 Flammable Liquids

Packaging Category: II

Marine Pollutant (Yes/No): No

Special Transport Way and Note: -

XV. Regulation Information:

Apply Regulation:

1. Enforcement Rules of the Occupational Safety and Health Act
2. Regulations of Hazard Communication on Dangerous and Harmful Material
3. Ordinance on Prevention of Organic Solvent Poisoning
4. Standards of Tolerable Hazardous Substance Concentration in the Air of Labor Working Environment



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5. Traffic Safety Regulations

6. Standards for the Storage, Clearance, and Disposal of Industrial Waste

7. Public Hazardous Materials and Flammable Pressurized Gases Establishment Standards and Safety Control Regulations