

SAFETY DATA SHEET

1. PRODUCT AND COMPANY IDENTIFICATION

Product name Lead shot

Synonyms Reload Lead, Magnum Shot, Hard shot, Chill Shot, Copper plated

shot, Buck shot, Free flow shot (dust), Ballast shot, Radiation shot

Recommended uses Reload shells (ammunition), ballast, radiation shielding

Uses advised against Jewelry, toys

2. HAZARDS IDENTIFICATION

Classification

This product is considered hazardous by the 2012 OSHA Hazard Communication Standard (29 CFR

1910.1200)

Carcinogenicity Category 1B
Reproductive toxicity Category 1A
Specific target organ toxicity Category 1

(repeated exposure)

Label elements

Danger

Hazard statements May cause cancer

May damage fertility or the unborn child May cause harm to breast-fed children

Cause damage to central nervous system, blood formation and kidneys and cardiovascular system through prolonged or repeated

exposure





Appearance Gray with bluish or silvery cast depending on physical alloy

State Solid Odor Odorless

Precautionary Statements –

Obtain special instructions before use

Prevention Do not handle until all safety precautions have been read and

understood

Use personal protective equipment as required

Wash face, hands and any exposed skin thoroughly after handling

Do not eat, drink or smoke when using this product Use only outdoors or in a well-ventilated area Do not breathe dust/fume/gas/mist/vapors/spray

Precautionary Statements -

Response

If exposed or concerned Get medical advice/attention

If Inhaled Remove victim to fresh air and keep at rest in a position

comfortable for breathing

If swallowed Call a POISON CENTER or doctor/physician if you feel unwell rinse

mouth

Precautionary Statements –

Storage

Store locked up

Precautionary Statements –

Disposal

Dispose of contents/container to an approved waste disposal plant

Other information Very toxic to aquatic life with long lasting effects

Very toxic to aquatic life

3. COMPOSITION/INFORMATION ON INGREDIENTS

Material	% by Wt.	CAS#	OSHA EXPOSURE
			LIMIT
Lead	91 – 99.99	7439-92-1	0.05 mg/m ³
Antimony	0.5 – 6.5	7440-36-0	0.50 mg/m ³
Arsenic	0.1 – 2.0	7440-38-2	0.01 mg/m ³
Copper	0.1 – 1.0	7440-50-8	0.10 mg/m ³

4. FIRST AID MEASURES

First aid measures

Eye contact In case of eye contact, immediately flush eyes with fresh water for at

least 15 minutes while holding the eyelids open. Remove contact lenses if worn. Get medical attention if irritation persists. Do not rub affected

area.

Skin contact Wash off immediately with soap and plenty of water. If skin irritation persists

call a Physician.

Inhalation Remove to fresh air. If breathing has stopped, give artificial respiration.

Get medical Attention immediately. If conscious, have victim clear nasal

passages.

Ingestion Seek immediate medical attention. Rinse mouth. Drink plenty of water.

Induce Vomiting, but only if victim is fully conscious.

Most important symptoms and effects, both acute and delayed

Symptoms

Acute (short term) exposure Lead is a potent, systemic poison; taken in large enough doses, lead

can kill in a matter of days. Acute encephalopathy may arise which develops quickly to seizures, coma and death from cardiorespiratory

arrest.

Chronic (long term) exposure
Chronic overexposure to lead may result in severe damage to blood

forming. Nervous, urinary and reproductive systems. Some common symptoms of chronic overexposure include loss of appetite, metallic taste in mouth, anxiety, constipation, nausea, pallor, excessive tiredness, weakness, insomnia, headache, nervous irritability, muscle and joint pain, fine tremors, numbness, dizziness, hyperactivity, colic.

Indication of any immediate medical attention and special treatment needed

5. FIRE – FIGHTING MEASURES

Suitable extinguishing media Dry chemical, foam or CO2

Specific hazards arising from

the chemical

May give off toxic fumes in a fire, including lead and antimony

fumes

Explosion data:

Sensitivity to mechanical

impact

None known.

Sensitivity to static discharge N

None known.

Protective equipment and precautions for firefighters

As in any fire, wear self-contained breathing apparatus pressuredemand, MSHA/NIOSH (approved or equivalent) and full protective gear. Lead is not considered to be a fire hazard. Powder/dust is

flammable when heated or exposed to flame.

6. ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

Personal precautions Evaluate personnel to safe areas. Avoid contact with skin, eyes

and inhalation of dusts. Use personal protection recommended in

Section 8.

For emergency responders Wear respiratory protection. Wear proper personal protective

equipment (gloves and goggles). Wear appropriate outer garment

to protect clothing

Environmental precautions

Environmental precautions Prevent entry into waterways, sewers, surface drainage systems

and poorly ventilated areas.

Methods and material for containment and cleaning up

Methods for containment Avoid creating dust. Safely stop source of spill. Restrict non-

essential personnel from area. All personnel involved in spill

cleanup should avoid skin and eye contact by wearing appropriate

personal protection equipment. Do not breathe dust.

Methods for cleaning up Avoid dust formation. Clean up dusts with high efficiency

particulate air (HEPA) filtered vacuum equipment or by wet

cleaning.

Prevention of secondary

hazards

Clean contaminated objects and area thoroughly observing

environmental regulations.

7. HANDLING AND STORAGE

Precautions for safe handling

Advice on safe handling Use personal protection recommended in Section 8. Avoid

generation of dust. Be familiar with the requirements set forth in the

OSHA Lead Standard, 29 CGR 1910.1025.

Conditions for safe storage, including any incompatibilities

Storage conditions Keep containers tightly closed in a dry, cool and well-ventilated

place.

Incompatible materials Strong oxidizing agents.

8. EXPOSURE CONTROLS / PERSONAL PROTECTION

Control parameters

Exposure guidelines

Chemical Name	ACGIH TLV	OSHA PEL	NIOSH IDLH
Lead Shot (As Lead)	TWA: 0.15 mg/m ³ Pb	TWA: 0.05 mg/m ³ Pb	IDLH: 100mg/m³Pb
7439-92- 1			TWA: 0.050 g/m ³ Pb
Antimony 7440-36-0	TWA: 0.5 mg/m ³ Sb	TWA: 0.5 mg/m ³ Sb	IDLH: 50mg/m ³ Sb
			TWA: 0.5mg/m ³ Sb
Arsenic 7440-38- 2	TWA: 0.01 mg/m ³ ,A1	TWA: 0.01 mg/m ³ As	IDLH: 5mg/m³As
			TWA: 0.002mg/m ³
			As 15minute ceiling
Copper 7440-50-8	TWA: 1.0 mg/m³Cu	TWA: 1.0 mg/m³Cu	IDLH: 2000 mg/m³Cu
			TWA: 1.0 mg/m ³ Cu

Appropriate engineering controls

engineering controls to maintain aerosols below the exposure limit. If

user operations generate dust, fume or mist use ventilation to keep exposure to airborne contaminates below the exposure limit.

Individual protection measures, such as personal protective equipment

Skin and body protection Protective clothing is required if exposure exceeds the PEL or TLV or

where possibility of skin or eye irritation exists. Full body cotton or disposable coveralls and disposable gloves should be worn during use and handling. Clothing should be left at work site and be properly disposed of or laundered after use. The wash water should be disposed of in accordance with local, state and federal regulations.

Personal clothing should be protected from contamination.

Respiratory protection If engineering controls cannot maintain airborne concentrations

below exposure limits, use appropriate, approved respiratory

protection (a 42 CFR 84 class N, R, or P-100 particulate filter cartridge). When exposure levels are unknown, a self-contained breathing apparatus which supplies a positive air pressure within a full face-piece mask should be worn. Utilization of respiratory equipment should be in accordance with 29 CFR 1910.1025 and 29

CFR 1910.134.

General hygiene considerations Do not eat, drink or smoke when using this product. Contaminated

work clothing should not be allowed out of the workplace. Wear disposable gloves and eye/face protection. Wash face, hands and any

exposed skin thoroughly after handling.

9. PHYSICAL AND CHEMICAL PROPERTIES

Information on basic physical and chemical properties

Physical state Solid

Appearance Gray with bluish or silvery cast depending on alloy

Odor Odorless

Property Values Remarks *Method

pH Not available
Melting point/freezing point >600°C
Boiling point/boiling range >600°C

Flash point Not applicable (high-melting point solid)
Evaporation rate Not applicable (high-melting point solid)

Flammability (solid, gas) Not combustible

Flammability limit in air

Upper flammability limit

Lower flammability limit

Not combustible

Not combustible

Vapor pressure Negligible

Vapor density Not applicable (high-melting point solid)

Specific gravity 9.96

Water solubility 70.2 mg/L at 20°C

Solubility in other solvents Lead compounds, soluble in 0.07 M hydrochloric acid

Partition coefficient Not applicable (inorganic)

Auto ignition temperature Not combustible

Decomposition temperature >600°C

Kinematic viscosity

Dynamic viscosity

Not applicable (solid)

Not applicable (solid)

Explosive properties Not considered to be explosive Oxidizing properties Not considered to be oxidizing

Other information

Softening point Not available Molecular weight Not available VOC content (%) Not available Bulk density Not available

10. STABILITY AND REACTIVITY

Reactivity Stable under normal conditions.
Chemical stability Stable under normal conditions.
Possibility of hazardous reactions None under normal processing.

Hazardous polymerization does not occur.

Conditions to avoid Avoid excessive exposure to heat.

Incompatible materials Strong oxidizing agents. Hazardous decomposition products Lead oxide fumes.

11. TOXICOLOGICAL INFORMATION

Information on likely routes of exposure

Hazardous exposure to lead compounds can occur only when product is heated, oxidized or otherwise processed or damaged to create dust, vapor or fume.

Inhalation Inhalation of lead dust or fumes may cause irritation of upper

respiratory tract and lungs

Eye contact Lead compounds may cause eye irritation

Skin contact

Lead compounds are poorly absorbed through the skin

Acute ingestion of lead compounds may cause abdominal

pain pausea vomiting diarrhea and severe cramping. This

pain, nausea, vomiting, diarrhea and severe cramping. This may lead to rapidly systemic toxicity and must be treated by

a physician.

Component information Lead is slowly absorbed by ingestion and inhalation and

poorly absorbed through the skin. If absorbed, lead will accumulate in the body with low rates of excretion, leading to long-term build up. Part of risk management is to take blood samples from workers for analysis to ensure that

exposure levels are acceptable.

Chemical Name	Oral LD50	Dermal LD50	Inhalation LC50
Lead 7439-92-1	450mg Pb/kg	Not available	100mg Pb/m³
Antimony 7440-36-0	7g Sb/kg Rat	Not available	Not available
Arsenic 7440-38- 2	763mg As/m³ Rat	Not available	5.2 mg As/m³ Cat

Copper 7440-50-8	1000 mg Cu/kg Rat	Not available	>2000	mg	Cu/m³
			Mamma	I	

Information on toxicological effects symptoms

Not available.

Delayed and immediate effects as well as chronic effects from short and long-term exposure

Skin corrosion/irritation Lead metal granules or dust: May cause skin irritation by

Mechanical action. Lead metal foil, shot or sheets: Not likely

to cause skin irritation.

Serious eye damage/eye irritation Lead metal granules or dust: Can irritate eyes by mechanical

action. Lead metal foil, shot or sheets: No hazard. Will not

cause eye irritation.

Inhalation In an industrial setting, exposure to lead mainly occurs from

inhalation of dust or fumes. Lead dust or fumes: Can irritate the upper respiratory tract (nose, throat) as well as the bronchi and lungs by mechanical action. Lead dust can be absorbed through the respiratory system. However, inhaled lead does not accumulate in the lungs. All of an inhaled dose is eventually absorbed or transferred to the gastrointestinal

tract. Inhalation effects of exposure to fumes or dust or inorganic lead may not develop quickly. Symptoms may include metallic taste, chest pain, decreased physical fitness, fatigue, sleep disturbance, headache, and irritability, reduces

memory, mood and personality changes, aching bones and muscles, constipation, abdominal pains, decreasing appetite. Inhalation of large amounts may lead to ataxia, delirium,

convulsions/seizures, coma, and death. Lead metal foil, shot, or sheets: Not an inhalation hazard unless metal is heated. If metal is heated, fumes will be released. Inhalation of these

fumes may cause "fume metal fever", which is characterized by flulike symptoms. Symptoms may include metallic taste,

fever, nausea, vomiting, chills, cough, weakness, chest pain, generalized muscle pain/aches, and increased white blood

cell count.

Ingestion Lead metal granules or dust: The Symptoms of lead poisoning

include abdominal pain or cramps (lead colic), spasms,

nausea, vomiting, headache, muscle weakness,

hallucinations, distorted perceptions, "lead line" on the gums, metallic taste, loss of appetite, insomnia, dizziness and other symptoms similar to that of inhalation. Acute poisoning may result in high lead levels in the blood and urine, shock, coma and death in extreme cases. Lead metal foil, shot or sheets: Not an ingestion hazard for usual industrial handling.

Epidemiology studies or workers exposed to inorganic lead compounds have found a limited association with stomach cancer. This has led to the classification by IARC that inorganic lead compounds are probably carcinogenic to

humans.

Carcinogenic effects

Chemical Name	ACGIH	IARC	NTP	OSHA
Lead 7439-92-1	A3	2B	Reasonably, Anticipated	Category 1B

Antimony 7440-36-0	A2	2B	Not Listed	Category 2
Arsenic 7440-38-2	A1	1	K	Category 1B
Copper 7440-50-8	Not Listed	Not Listed	Not Listed	Not Listed

Reproductive toxicity Exposure to high levels of lead may cause adverse effects on

male and female, including adverse effects on sperm quality.

Prenatal exposure to lead and its compounds is also associated with adverse effects on fetal development. Lead has been found to be of relatively low acute toxicity by

ingestion, in contact with skin, and by inhalation, with no

evidence of any local or systemic toxicity from such

exposures.

STOT – repeated exposure Lead is a cumulative poison and may be absorbed into the

body through ingestion or inhalation. Inorganic lead compounds have been documented in observational human studies to produce toxicity in multiple organ systems and body function including the hematopoietic (blood) system, kidney function, reproductive function and the central nervous system. Postnatal exposure to lead compounds is

associated with impacts on neurobehavioral development in

children.

Chronic toxicity Lead is a cumulative poison. Increasing amounts of lead can

build up in the body and may reach a point where symptoms and disabilities occur. Continuous exposure may result in decreased fertility. Lead is a teratogen. Overexposure of lead by either parent before pregnancy may increase the chances of miscarriage or birth defects. May cause cancer. Contains a known or suspected reproductive toxin. May cause adverse

kidney effects.

Target organ effects Lead is a cumulative poison and may be absorbed into the

body through ingestion or inhalation. Inorganic lead compounds have been documented in observational human studies to produce toxicity in multiple organ systems and

body function including the hematopoietic (blood) system, kidney function, reproductive function and the central nervous system. Postnatal exposure to lead compounds is associated with impacts on neurobehavioral development in

children.

Aspiration hazard Not available.

The following values are calculated based on chapter 3.1 of the GHS document.

Inhalation LC50 None available

12. ECOLOGICAL INFORMATION

STOT – single exposure

Environmental fate Lead is very persistent in soil and sediments. No data on

environmental degradation. Mobility of metallic lead between ecological compartments is slow. Bioaccumulation of lead occurs in aquatic and terrestrial animals and plants, but little bioaccumulation occurs through the food chain. Most studies include lead compounds and not elemental

lead.

Chemical Name	Algae/aquatic plants	Fish	Toxicity to	Crustacean
			microorganisms	
Lead 7439-92- 1	0.072-0.388: 72h	0.298: 96h		0.074-0.656: 48h
	Pseudokirchneriella	Pimephales		Daphnia magna,
	subcapitatia,	promelas mg/L		Ceriodaphnia
	Chlorella kessierii	LC50 static 0.041-		dubia mg/L LC50
	mg/L ErC50 (pH 5.5-	1.810: 96h		(pH 5.5- 6.5)
	6.5) 0.026-0.080:72h	Pimephales		0.029-1.18: 48h
	Pseudokirchneriella	promelas,		Daphnia magna,
	subcapitatia,	Oncorhynchus		Ceriodaphnia
	Chlorella kessierii	mykiss mg/L LC50		dubia mg/L LC50
	mg/L ErC50 (pH	(pH 5.56.5)0.052-		(pH >6.5-7.5)
	>6.5-7.5) 0.021-	3.60: 96h		0.026-3.12: 48h
	0.050: 72h	Pimephales		
		promelas,		
		Oncorhynchus		
		mykiss mg/L LC50		
		(pH >6.5- 7.5)		
		0.114-3.25: 96h		
	Pseudokirchneriella	Pimephales		Daphnia magna,
	subcapitatia,	promelas,		Ceriodaphnia
	Chlorella kessierii	Oncorhynchus		dubia mg/L
	mg/L ErC50 (pH 7.5-	mykiss mg/L		LC50(pH >7.5-8.5)
	8.5)	LC50(pH >7.5-8.5)		
		56000: 96h		
		Gambusia affinis		
		mg/L LC50 static		
Antimony 7440-	Not Listed	Cyprinodont	Not Listed	Not Listed
36-0		variegates: LC50 =		
		6.2-8.3 mg/L/96h		
Arsenic 7440- 38-	Not Listed	Pimephales	Daphnia magna	Not Listed
2		promelas	(water flea)	
		(flathead	EC50= 3.8mg/l –	
		minnow) LC50=	48h	
		9.9mg/l – 96.0h		
Copper 7440- 50-	Pseudokirchneriella	Pimephales	None listed	Daphnia magna:
8	subcapitata: EC50 =	promelas: LC50 =		EC50 = .03
	.04260535	.00680156		mg/L/48h
	mg/L/72h	mg/L/96h.		
		Pimephales		
		promelas (static):		
		LC50 =		
		.3mg/L/96h		

Bioaccumulation

While lead metal and its compounds are generally insoluble, its processing or extended exposure in aquatic and terrestrial environments may lead to the release of lead in bioavailable forms. Lead compounds are not particularly mobile in the

aquatic environments, but can be toxic for organisms, especially fish, at low concentrations. Water hardness, pH and dissolved organic carbon content are factors which regulate the degree of toxicity. In soil, lead compounds are

generally not very bioavailable.

Mobility Lead and lead compounds will partially settle out due to their

fairly low solubility and partially dissolve. In soil, lead and

lead compounds are generally not very mobile or bioavailable, as they can be strongly absorbed on soil

particles, increasingly over time. It also forms complexes with organic matter and clay minerals that limit its mobility. When released into the soil, this material is not expected to leach

into groundwater.

Other adverse effects Not available.

13. DISPOSAL CONSIDERATIONS

Waste treatment methods

Disposal of wastes Disposal should be in accordance with applicable regional,

national and local laws and regulations.

Contaminated packaging Disposal should be in accordance with applicable regional,

national and local laws and regulations.

14. TRANSPORT INFORMATION

Note: This product is not regulated for domestic transport by land, air or rail.

- Under 49 CFR 171.8, individual packages that contain lead metal (<100 micrometers) below the reportable quantity (RQ) are not regulated.
- Under 49 CFR 171.4, except when transporting aboard a vessel, the requirements of this subchapter specific to marine pollutants do not apply to non-bulk packaging transported by motor vehicles, rail cars and aircrafts.

DOT

Proper shipping name Not applicable
Hazard class Not applicable
Reportable quantity (RQ) Not applicable
Packing group Not applicable

Marine pollutant Soluble lead compounds are listed as a marine pollutant

according to DOT.

Emergency Response Guide Not applicable

15. REGULATORY INFORMATION

International Inventories

TSCA Complies
DSL/NDSL Complies

EINECS/ELINCS Complies
ENCS Complies
IECSC Complies
KECL Complies
PICCS Complies
AICS Complies

Legend:

TSCAUnited States Toxic Substances Control Act Section 8(b) Inventory **DSL/NDSL**Canadian Domestic Substances List/Non-Domestic Substances List

EINECS/ELINCS European Inventory of Existing Chemical Substances/European List of Notified

Chemical Substances

ENCSJapan Existing and New Chemical SubstancesIECSCChina Inventory of Existing Chemical SubstancesKECLKorean Existing and Evaluated Chemical SubstancesPICCSPhilippines Inventory of Chemicals and Chemical Substances

AICS Australian Inventory of Chemical Substances

US Federal Regulations SARA 313

Section 313 of Title III of the Superfund Amendments and Reauthorization Act of 1986 (SARA). This product contains a chemical or chemicals which are subject to the reporting requirements of the Act and Title 40 of the Code of Federal Regulations, Part 372.

Chemical Name	CAS No.	Weight - %	SARA 313 – Threshold Values %
Lead	7439-92-1	91-99.99	0.1
Antimony	7440-36-0	0.5-6.5	1.0
Arsenic	7440-38-2	0.1-2.0	0.1
Copper	7440-50-8	0.1 – 1.0	1.0

SARA 311/312 Hazard Categories

Acute health hazard Yes
Chronic health hazard Yes
Fire hazard No
Sudden release of pressure hazard No
Reactive hazard No

CWA (Clean Water Act)

This product contains the following substances which are regulated pollutants pursuant to the Clean Water Act (40 CFR 122.21 and 40 CFR 122.42)

Chemical Name	CWA – Reportable Quantities	CWA – Toxic Pollutants	CWA – Priority Pollutants	CWA – Hazardous Substances
Lead 7439-92-1	10 lb.	X	X	X
Antimony 7440-36-0	5000 lb.	Х	X	X
Arsenic 7440-38-2	1 lb.	Х	X	X
Copper 7440-50-8	1 lb.	-	Х	Х

CERCLA

This material, as supplied, contains one or more substances regulated as a hazardous substances under the Comprehensive Environmental Response Compensation and Liability Act (CERCLA) (40 CFR 302).

US State Regulations California Proposition 65

This product contains a chemical known to the state of California to cause birth defects or other reproductive harm.

Chemical Name	California Proposition 65	
Lead – 7439-92-1	Cancer	
Antimony – 7440-36-0	Cancer	
Arsenic – 7440-38-2	Developmental	
Copper – 7440-50-8	Not Listed	

US State Right-to-Know Regulations

Chemical Name	New Jersey	Massachusetts	Pennsylvania	Illinois	Rhode Island
Lead – 7439-92-1	х	х	х	-	х
Antimony – 7440-36-0	х	х	х	-	х
Arsenic – 7440-38-2	х	х	х	-	x
Copper – 7440-50-8	х	х	х	-	х

US EPA Label Information

EPA Pesticide Registration Number

Not available.

16. OTHER INFORMATION

Disclaimer

This information provided in this Safety Data Sheet is correct to the best of our knowledge, information and Belief at the date of its publication. The information given is designed only as guidance for safe handling, use, Processing, storage, transportation, disposal and release and is not to be considered a warranty or quality Specification. The information materials or in any process, unless specified in the text.

Version 2

Last revisión October 1st, 2015