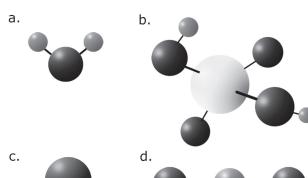
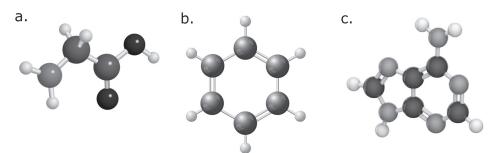
CLASS Chemistry: Matter and How It's Made, Introduction to Atoms

Your steadfast love, O Lord, extends to the heavens, your faithfulness to the clouds. Psalm 36:5

- 1. **True or False?** Almost all the familiar types of matter—air, plants, animals, food, water, automobiles, furniture, etc.—are made from small "pieces" of matter called molecules.
- 2. **True or False?** There are really only a few different types of molecule.
- 3. In the ball-and-stick diagrams, what do the "balls" represent?
- 4. Match the molecule below to its correct name.

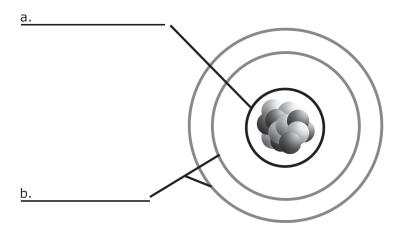


- 1. sulfuric acid
- 2. water
- 3. carbon dioxide
- 4. ammonia
- 5. How many atoms are the molecules in 4a, 4b, 4c and 4d made of? ______
- 6. How many atoms are the molecules below made from?



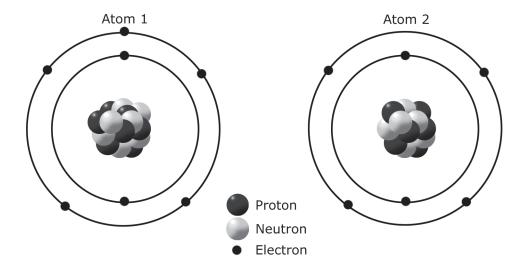
- 7. The abbreviation symbol for a type of atom is listed below. Provide its name.
 - a. N stands for _____
 - **b.** O stands for _____

- 8. What is the symbol for sulfur? _____
- 9. How many different kinds of atoms are there?
 - **a.** 68
 - **b.** 81
 - **c.** 181
 - **d.** 118
 - **e.** 128
- 10. True or False? Because an atom is so small, it is not made from anything smaller.
- 11. **True or False?** The fact that we can see an individual atom with a very powerful microscope clearly shows that even though something is very, very small, it is still matter because it can be seen and takes up space.
- 12. Choose the correct statements.
 - **a.** Large types of matter—like trees, people, animals and mountains—are not made from tiny types of matter called molecules.
 - **b.** Even though they are tiny, all atoms are made from even smaller pieces of matter.
 - c. It is how an atom is built that makes one type of atom different from another type.
 - **d.** By counting the "balls" in a ball and stick model, you can tell how many atoms a molecule contains.
- 13. Label the parts of an atom.



- 14. What are the three particles that atoms are built from?
- 15. Which particles make up the nucleus?
- 16. True or False? Electrons are in the nucleus.
- 17. **True or False?** The two atoms below are the same type of atom.

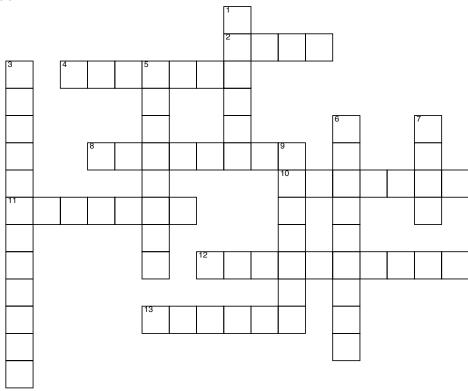
Why did you answer as you did?



CROSSWORD PUZZLE =

Finish the crossword puzzle below using the words provided.

Atom
Ball and Stick
Bond
Chemist
Chemistry
Electron
Element
Matter
Microscope
Molecule
Neutron
Nucleus
Proton



ACROSS

- 2. The smallest whole unit of matter is a/an ______
- 4. A scientist who studies how atoms and molecules interact.
- 8. The particle that whizzes around the outside of an atom's nucleus.
- 10. A kind of atom is called a/an _____.
- 11. Protons and neutrons "live" in the _____ of an atom.
- 12. An instrument that can be used to see pieces of matter that are too small to be seen with the naked eye.
- 13. One of the types of particles in an atom's nucleus.

DOWN

- 1. Anything that you can see or feel or that takes up space.
- 3. A _____ diagram is one way we can use to draw molecules.
- 5. Two or more atoms bond together to form a
- 6. The scientific study of how atoms and molecules interact.
- 7. The thing in a molecule that holds two atoms together.
- 9. One of the types of particles in an atom's nucleus.