

**Bio-industrial Evolution** 

Metabolix DP9002 for Denitrification Aquarium in Applications







#### Overview

- What are Polyhydroxyalkanoates (PHAs)
- Three Nitrogen Conversion Pathways
- Conventional versus New Denitrification
- Oceanic Corals Canada Study
- References Denitrification

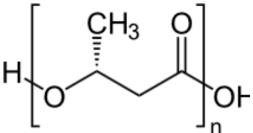


# Polyhydroxyalkanoates (PHAs)

- Polyhydroxyalkanoates (PHAs) are linear polyesters produced by bacterial fermentation of sugar or lipids
- PHAs produced by bacteria to store carbon and energy
- Polyhydroxybutyrates (PHBs) are polyhydroxyalkanoates (PHAs)
- The simplest and most commonly occurring form of PHA is poly-ß-hydroxybutyrate (P3HB) a PHB (produced by fermentative CH₃

production)

 P3HB consists of 1,000 to 30,000 hydroxy fatty acid monomers



Structure of poly-(R)-3-hydroxybutyrate (P3HB), a polyhydroxyalkanoate

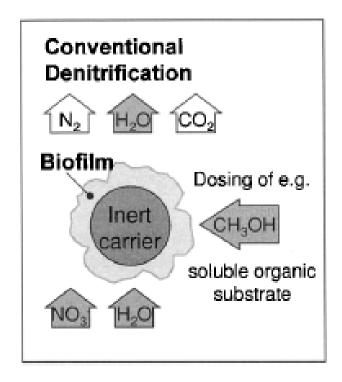


### Three Nitrogen Conversion Pathways

- Three nitrogen conversion pathways traditionally used for the removal of ammonia—nitrogen in water treatment, aquaculture and aquarium systems
  - 1) Photoautotrophic removal by algae
  - 2) Autotrophic bacterial conversion of ammonia–nitrogen to nitrate–nitrogen
  - 3) Heterotrophic bacterial conversion of ammonia—nitrogen directly to microbial biomass
- PHA/PHB is used as a carbon source for
  - 2) Autotrophic bacterial conversion pathway
  - 3) Heterotrophic bacterial conversion pathway



### Conventional versus New Denitrification



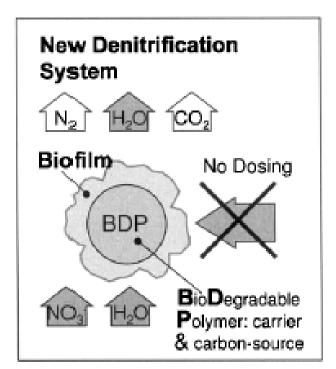


Fig. 2. Denitrification processes with different organic substrates.

Source: Boley et.al., Biodegradable polymers as solid substrate and biofilm carrier for denitrification in recirculated aquaculture systems, Aquacultural Engineering 22 (2000) 75-85



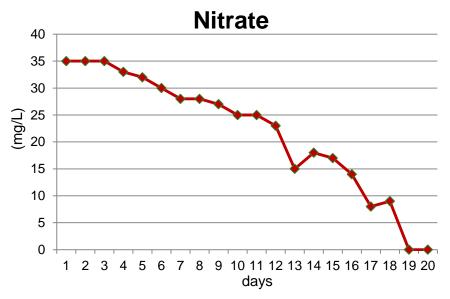
### DP9002 for Denitrification

 Small, white to light brown colored pellets developed for denitrification in aquariums



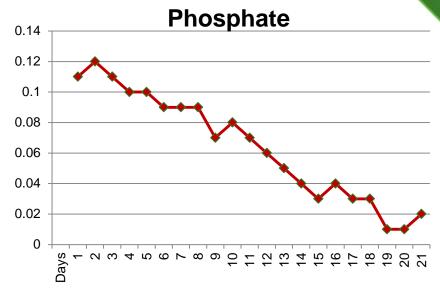


## Oceanic Corals Canada Study





- 175 gallons
- Computer controlled lighting cycles and mineral additives
- 22 fish, various corals and inverts
- Daily feeding regimen is identical controlled amounts
- Protein Skimmer runs 24/7
- No water changes were during time of study



#### Fluidized Reactor Specifications

- Vertex UF-20 with 1460 grams of PHA Pellets
- Water flow = 265 GPH

Source: Oceanic Corals Canada, 2010



### Data for PHA in Denitrification

- PHA is comparable with other carbon-based systems
  - Efficiency rate of approx. 2.4 7 g PHA / g of N<sub>2</sub>\*
- PHA is also active in phosphate removal
- PHA is certified Marine Biodegradable per ASTM 7081
- 100% Bio-based
- No fillers

<sup>\*</sup> Source: Gutierrez-Wing et.al., Polyhydroxyalkanoates as a carbon source for denitrification of water, World Environmental and Water Resources Congress 2007



### References – Denitrification

- 1. Boley A, Müller W-R (2001) Evaluation of different biodegradable polymers for nitrate removal in aquaria. Bulletin de l'Institut océanographique, Special 20
- 2. Boley A, Müller W-R, Haider G (2000) Biodegradable polymers as solid substrate and biofilm carrier for denitrification in recirculated aquaculture systems. Aquacultural Engineering 22: 75-85
- 3. Ebeling J M, Hightower P G (2010) Preliminary evaluation of three PHA formulations for passive self-regulating denitrification technology. Aquaculture Systems Technologies
- 4. Ebeling J M, Timmons M B, Bisogni JJ (2006) Engineering analysis of the stoichiometry of photoautotrophic, autotrophic, and heterotrophic removal of ammonia—nitrogen in aquaculture systems. Aquaculture 257: 346-358
- 5. Gutierrez-Wing M T, Rusch K A, Malone R F (2007) Polyhydroxyalkanoates as a carbon source for Denitrification of waters. World Environmental and Water Resources Congress 2007
- 6. Hiraishi A, Khan S T (2003) Application of polyhydroxyalkanoates for denitrification in water and wastewater treatment. Appl Microbiol Biotechnol 61:103-109
- 7. Qin L, Liu Y, Tay J-H (2005) Denitrification on poly-b-hydroxybutyrate in microbial granular sludge sequencing batch reactor. Water Research 39: 1503-1510
- 8. Van Rijn J, Tal Y, Schreier H J (2006) Denitrification in recirculating systems: Theory and applications. Aquacultural Engineering 34: 364-376



#### Disclaimer

NOTICE: Customer assumes all risk and liability for any use or handling of this product beyond Metabolix's direct control. Customer is responsible for obtaining any licenses or other rights necessary to make, use or sell products containing Mirel/Mevera compounds. Customer should consult its legal counsel to determine whether its labels for products made with Mirel/Mevera compounds are in compliance with applicable laws and regulations. Metabolix shall not be responsible for any consequential, special or incidental damages, and liability for breach of warranty, negligence or other claims is limited to the purchase price of material purchased. The information contained herein is believed to be reliable; however, Metabolix makes NO REPRESENTATIONS, GUARANTEES OR WARRANTIES, EXPRESS OR IMPLIED, INCLUDING, BUT NOT LIMITED TO, THE IMPLIED WARRANTIES OF MFRCHANTABILITY AND FITNESS FOR A PARTICULAR PURPOSE.

