NAMING COMPOUNDS & ACIDS

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ANSITION METALS Compound Name to Chemical Formula







Transition Metal (Roman Numeral) Nonmetal

ending in ide



lead (IV) oxide



Step 1: Identify the symbol and charge of the metal and nonmetal.

$$Pb^{4+}$$
 O^{2-}

Step 2: Make sure charges are balanced/the same. In this case, they are not. So we have to add a subscript so the charges balance out.





Since we want the same charges, the common charge is 4, then O needs a 2 subscript.



By putting that 2 subscript, you're really just multiplying your charge by 2.

$$Pb^{4+} O^{2-}$$

Since charges have opposite signs and are the same number, they can cancel out. Final answer never shows charges.

Chemical Formula: PbO