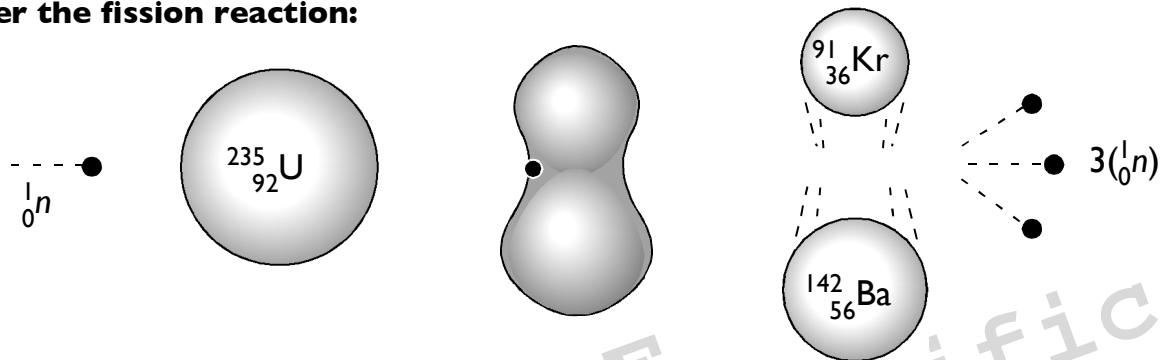


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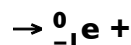
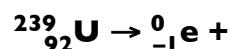
# Fission and Fusion

In this lecture, Paul Hewitt discusses the mechanics of nuclear fission and fusion. And he explains why each of the seemingly opposite processes is capable of releasing energy. Read the following questions before the presentation begins. Answer them while the presentation is in progress. [40 minutes]

Consider the fission reaction:

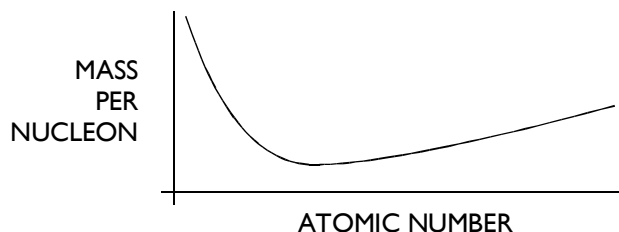
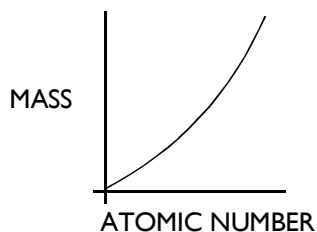


- The neutron that triggers the reaction
  - is a low-speed neutron.
  - is a high-speed neutron.
  - may be a low-speed or a high-speed neutron.
- A fast-moving golf ball would be slowed down the best if it
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  - entered an arena of Ping Pong balls.
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- What is the energy output of each?
  - A drop of water over Niagara Falls: \_\_\_\_\_ eV
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- Fill in the missing terms of the nuclear processes described in the lesson.



5. Approximately how much U-235 was left in the US arsenal after the bomb was dropped that devastated Hiroshima?

Consider the graphs.



6. The mass of a nucleon (proton or neutron)

A. is universally constant.      B. depends on the nucleus its in.

7. Consider the particles before and after fission. Which set has more mass?

A.  ${}^1_0n + {}^{235}_{92}\text{U}$       B.  ${}^{91}_{36}\text{Kr} + {}^{142}_{56}\text{Ba} + 3({}^1_0n)$       C. same for both

8. Energy released in nuclear fission can be used to boil water, turning it to steam. The steam can be used to spin a paddle wheel. The paddle wheel can be hooked to a \_\_\_\_\_ which—when spinning—will produce \_\_\_\_\_.

9. If you split a light nucleus such as magnesium,

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10. Helium has

A. less mass than hydrogen.      B. less mass per nucleon than hydrogen.  
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11. If 200 MeV is released when a uranium nucleus undergoes fission, how much energy is needed to fuse the fission by-products back together?

12. What could you do to release energy from iron nuclei?

A. split them (fission)      B. join them (fusion)      C. both      D. neither

13. How much matter is converted to energy each second by the Sun?

14. What makes fusion-based power production difficult on Earth?

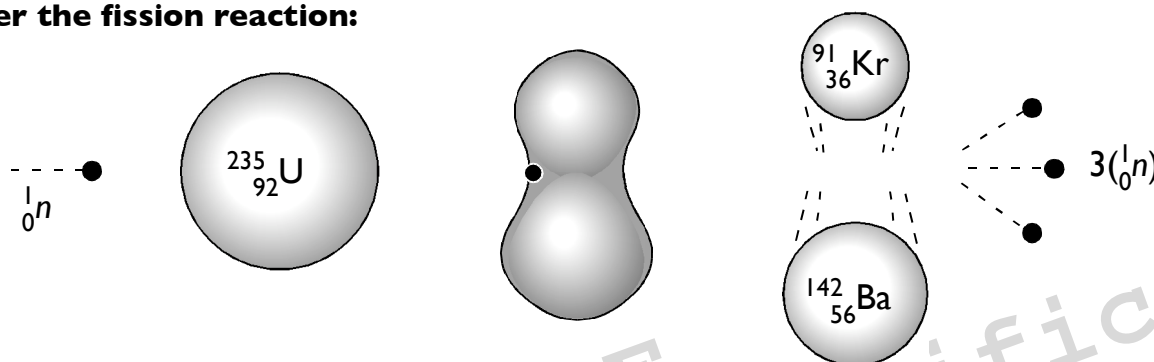
A. neutrons are hard to come by  
B. the nuclear strong force is hard to overcome  
C. hydrogen nuclei electrostatically repel  
D. fusion cannot occur at temperatures lower than that of stars

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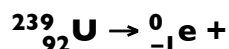
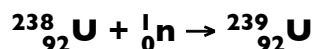
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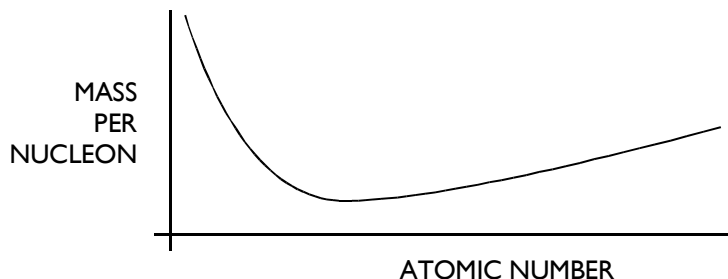
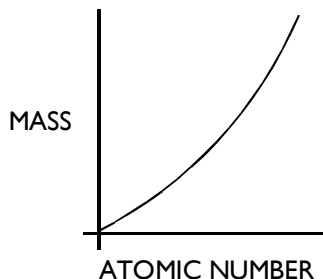
4. Fill in the missing terms of the nuclear processes described in the lesson.



5. The bomb that devastated Hiroshima was made with   ?; the bomb that devastated Nagasaki was made with   ?.

- A. uranium; uranium                      B. uranium; neptunium  
C. neptunium; plutonium                D. uranium; plutonium  
E. neptunium; uranium                    F. plutonium; plutonium

Consider the graphs.



6. The mass of a nucleon (proton or neutron)

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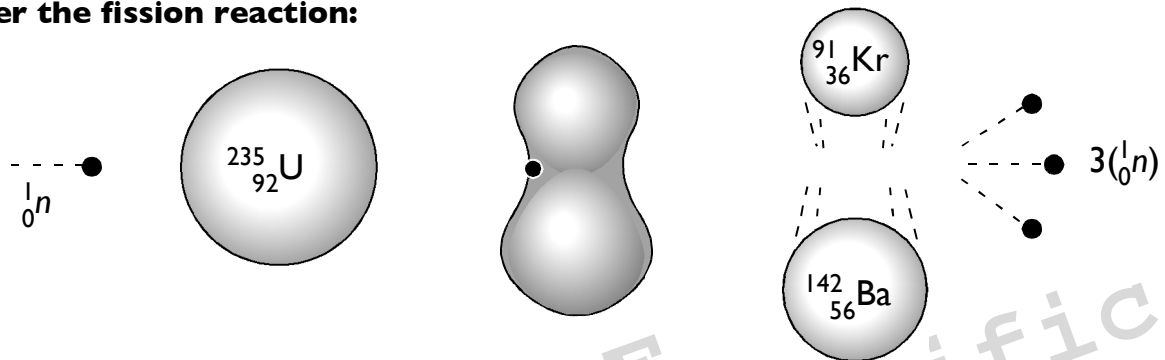
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B. undergo a series of chemical explosions.  
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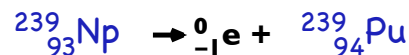
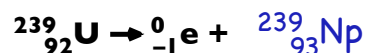
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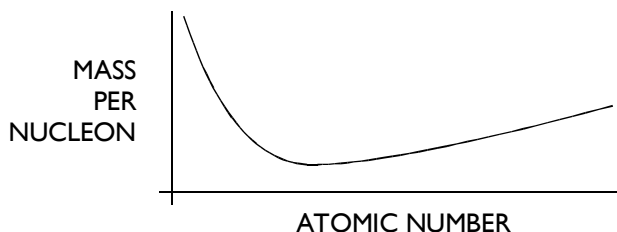
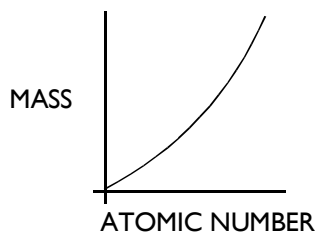
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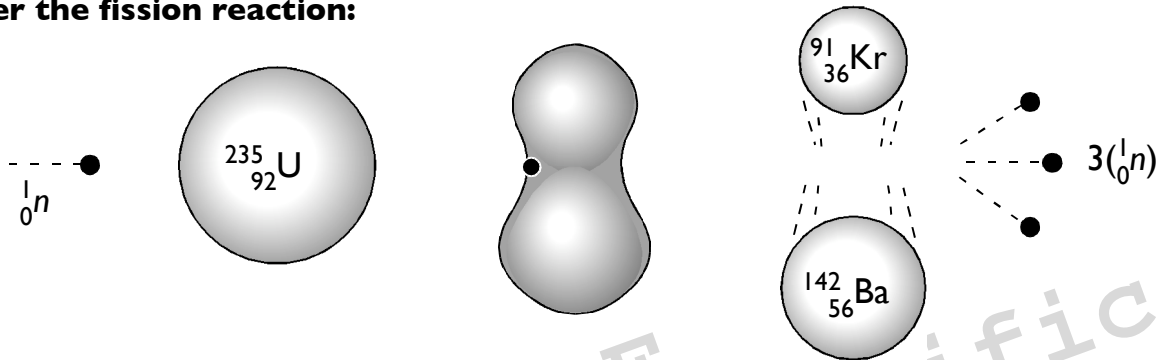
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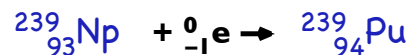
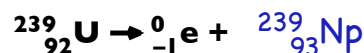
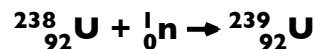
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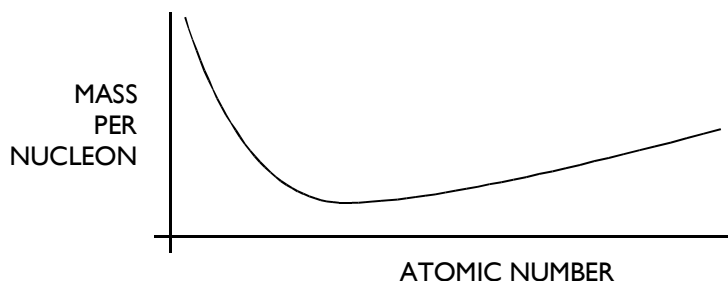
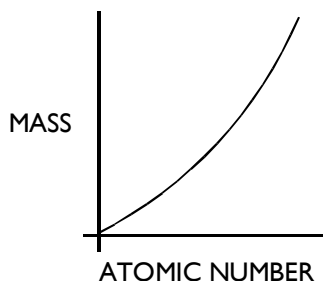


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