Thinkwell's Homeschool Chemistry | Course Lesson Plan: 37 weeks

Welcome to Thinkwell's Homeschool Chemistry! We're thrilled that you've decided to make us part of your homeschool curriculum. This lesson plan is meant to be a guide for you and your homeschool student. Each day, you'll tackle a different topic and all the materials associated with that topic, such as video lectures, exercises, and interactivities. If you follow our day-by-day schedule, you'll complete the full curriculum for the course in 37 weeks. Feel free to modify and amend the plan as it best works for you. And, as always, please Let us know what we can do to help get you up and running with Thinkwell's Chemistry!

Schedule Overview:

Weeks 1 – 2	Chapter 1: An Introduction to Matter and Measurement
Weeks 3 – 4	Chapter 2: Atoms, Molecules, and Ions
Weeks 4 – 5	Chapter 3: Stoichiometry
Weeks 6 – 7	Chapter 4: Reactions in Aqueous Solutions
Weeks 7 – 9	Chapter 5: Gases
Weeks 9 – 10	Chapter 6: Thermochemistry
Weeks 11 – 12	Chapter 7: Modern Atomic Theory
Week 13	Chapter 8: Electron Configurations and Periodicity
Week 14	Chapter 9: Chemical Bonding: Fundamental Concepts
Weeks 15 – 16	Chapter 10: Molecular Geometry and Bonding Theory
Weeks 16 – 17	Chapter 11: Oxidation-Reduction Reactions
Weeks 17 – 18	Chapter 12: Condensed Phases: Liquids and Solids
Weeks 19 – 20	Chapter 13: Physical Properties of Solutions
Week 20	Midterm
Weeks 21 – 22	Chapter 14: Chemical Kinetics
Weeks 22 – 23	Chapter 15: Chemical Equilibrium
Weeks 24 – 25	Chapter 16: Acids and Bases
Weeks 25 – 26	Chapter 17: Equilibrium in Aqueous Solution
Weeks 27 – 28	Chapter 18: Introduction to Organic Reactions
Weeks 28 – 29	Chapter 19: Thermodynamics
Weeks 29 – 30	Chapter 20: Electrochemistry
Week 31	Chapter 21: Nuclear Chemistry
Week 32	Chapter 22: Chemistry of Metals
Week 33	Chapter 23: Transition Metals
Weeks 34 – 35	Chapter 24: Nonmetals
Weeks 35 – 36	Chapter 25: Organic Chemistry
Weeks 36 – 37	Chapter 26: Biochemistry
Week 37	Final Exam

Week 1	
Chapter 1: An Introduction to Matter and Measurement	
Assignments	Notes
Week 1, Day 1	
☐ 1.1.1 An Introduction to Chemistry	
☐ 1.1.2 The Scientific Method	
Week 1, Day 2	
☐ 1.2.1 States of Matter	
Week 1, Day 3	
☐ 1.2.2 A Word About Laboratory Safety	
☐ 1.2.3 CIA Demonstration: Differences in Density Due to	
Temperature	
Week 1, Day 4	
☐ 1.2.4 Properties of Matter	
☐ 1.3.1 The Measurement of Matter	
Week 1, Day 5	
☐ 1.3.2 Precision and Accuracy	
☐ 1.3.3 CIA Demonstration: Precision and Accuracy with Glassware	

Week 2	
Chapter 1: An Introduction to Matter and Measurement	
Chapter 1 Test	
Assignments	Notes
Week 2, Day 1	
☐ 1.3.4 Significant Figures	
☐ 1.3.5 Dimensional Analysis	
Week 2, Day 2	
☐ 1.4.1 Scientific (Exponential) Notation	
Week 2, Day 3	
☐ 1.4.2 Common Mathematical Functions	
Week 2, Day 4	
☐ Chapter 1 Practice Test	
Week 2, Day 5	Chapter 1 Test
☐ Chapter 1 Test	Score:

Week 3	
Chapter 2: Atoms, Molecules, and Ions	
Assignments	Notes
Week 3, Day 1	
2.1.1 Early Discoveries and the Atom	
☐ 2.1.2 Understanding Electrons	
Week 3, Day 2	
☐ 2.1.3 Understanding the Nucleus	
☐ 2.2.1 Mass Spectrometry: Determining Atomic Masses	
Week 3, Day 3	
☐ 2.2.2 Examining Atomic Structure	
☐ 2.2.3 CIA Demonstration: Flame Colors	
Week 3, Day 4	
2.3.1 Creating the Periodic Table	
☐ 2.4.1 Describing Chemical Formulas	
Week 3, Day 5	
☐ 2.4.2 Naming Chemical Compounds	
☐ 2.4.3 Organic Nomenclature	
Week 4	
Chapter 2 Test	
Chapter 3: Stoichiometry	
Assignments	Notes
Week 4, Day 1	
☐ Chapter 2 Practice Test	
Week 4, Day 2	Chapter 2 Test
☐ Chapter 2 Test	Score:
Week 4, Day 3	
☐ 3.1.1 An Introduction to Chemical Reactions and Equations	
☐ 3.1.2 CIA Demonstration: Magnesium and Dry Ice	
Week 4, Day 4	
☐ 3.1.3 Balancing Chemical Equations	
Week 4, Day 5	
☐ 3.2.1 The Mole and Avogadro's Number	
☐ 3.2.2 Introducing Conversions of Masses, Moles, and Number of	
Particles	

Week 5	
Chapter 3: Stoichiometry	
Chapter 3 Test	
Assignments	Notes
Week 5, Day 1	
☐ 3.3.1 Finding Empirical and Molecular Formulas	
☐ 3.3.2 Stoichiometry and Chemical Equations	
Week 5, Day 2	
☐ 3.3.3 Finding Limiting Reagents	
☐ 3.3.4 CIA Demonstration: Self-Inflating Hydrogen Balloons	
Week 5, Day 3	
☐ 3.3.5 Theoretical Yield and Percent Yield	
☐ 3.3.6 A Problem Using the Combined Concepts of Stoichiometry	
Week 5, Day 4	
☐ Chapter 3 Practice Test	
Week 5, Day 5	Chapter 3 Test
☐ Chapter 3 Test	Score:
Week 6	
Chapter 4: Reactions in Aqueous Solutions	
Assignments	Notes
Week 6, Day 1	
4.1.1 Properties of Solutions	
☐ 4.1.2 CIA Demonstration: The Electric Pickle	
Week 6, Day 2	
4.1.3 Concentrations of Solutions	
☐ 4.1.4 Factors Determining Solubility	
Week 6, Day 3	
☐ 4.2.1 Precipitation Reactions	
☐ 4.2.2 Acid-Base Reactions	
Week 6, Day 4	
☐ 4.2.3 Oxidation-Reduction Reactions	
☐ 4.3.1 Acid-Base Titrations	
Week 6, Day 5	
4.3.2 Solving Titration Problems	
☐ 4.3.3 Gravimetric Analysis	

Week 7	
Chapter 4 Test	
Chapter 5: Gases	
Assignments	Notes
Week 7, Day 1	
☐ Chapter 4 Practice Test	
Week 7, Day 2	Chapter 4 Test
☐ Chapter 4 Test	Score:
Week 7, Day 3	
☐ 5.1.1 Properties of Gases	
☐ 5.1.2 Boyle's Law	
Week 7, Day 4	
☐ 5.1.3 Charles's Law	
☐ 5.1.4 The Combined Gas Law	
Week 7, Day 5	
☐ 5.1.5 Avogadro's Law	
☐ 5.1.6 CIA Demonstration: The Potato Cannon	
Week 8	
Chapter 5: Gases	
Assignments	Notes
Week 8, Day 1	
5.2.1 The Ideal Gas Law	
☐ 5.2.2 Partial Pressure and Dalton's Law	
Week 8, Day 2	
5.2.3 Applications of the Gas Laws	
Week 8, Day 3	
5.2.4 The Kinetic-Molecular Theory of Gases	
☐ 5.2.5 CIA Demonstration: The Ammonia Fountain	
Week 8, Day 4	
☐ 5.3.1 Molecular Speeds	
☐ 5.3.2 Effusion and Diffusion	
Week 8, Day 5	
☐ 5.4.1 Comparing Real and Ideal Gases	

Week 9	
Chapter 5 Test	
Chapter 6: Thermochemistry	
Assignments	Notes
Week 9, Day 1	
☐ Chapter 5 Practice Test	
Week 9, Day 2	Chapter 5 Test
☐ Chapter 5 Test	Score:
Week 9, Day 3	
☐ 6.1.1 The Nature of Energy	
☐ 6.1.2 Energy, Calories, and Nutrition	
Week 9, Day 4	
☐ 6.1.3 The First Law of Thermodynamics	
☐ 6.1.4 Work	
Week 9, Day 5	
☐ 6.1.5 Heat	
☐ 6.1.6 CIA Demonstration: Cool Fire	
	T
Week 10	
Chapter 6: Thermochemistry	
Chapter 6 Test	
Assignments	Notes
Week 10, Day 1	
☐ 6.2.1 Heats of Reaction: Enthalpy	
☐ 6.2.2 CIA Demonstration: The Thermite Reaction	
Week 10, Day 2	
☐ 6.3.1 Constant Pressure Calorimetry	
☐ 6.3.2 Bomb Calorimetry (Constant Volume)	
Week 10, Day 3	
☐ 6.4.1 Hess's Law	
☐ 6.4.2 Enthalpies of Formation	
Week 10, Day 4	
☐ Chapter 6 Practice Test	
Week 10, Day 5	Chapter 6 Test
☐ Chapter 6 Test	Score:

Week 11	
Chapter 7: Modern Atomic Theory	
Assignments	Notes
Week 11, Day 1	
☐ 7.1.1 The Wave Nature of Light	
☐ 7.1.2 Absorption and Emission	
Week 11, Day 2	
☐ 7.1.3 CIA Demonstration: Luminol	
☐ 7.1.4 The Ultraviolet Catastrophe	
Week 11, Day 3	
☐ 7.1.5 The Photoelectric Effect	
☐ 7.1.6 The Bohr Model	
Week 11, Day 4	
7.1.7 The Heisenberg Uncertainty Principle	
☐ 7.2.1 The Wave Nature of Matter	
Week 11, Day 5	
7.2.2 Radial Solutions to the Schrödinger Equation	
☐ 7.2.3 Angular Solutions to the Schrödinger Equation	
Week 12	
Chapter 7: Modern Atomic Theory	
Chapter 7 Test	
Assignments	Notes
Week 12, Day 1	
7.3.1 Atomic Orbital Size	
☐ 7.3.2 Atomic Orbital Shapes and Quantum Numbers	
Week 12, Day 2	
7.3.3 Atomic Orbital Energy	
Week 12, Day 3	
☐ Chapter 7 Practice Test	
Week 12, Day 4	Chapter 7 Test
☐ Chapter 7 Test	Score:
Week 12, Day 5	
☐ 8.1.1 Understanding Electron Spin	
□ 8.1.2 Electron Shielding	

Week 13	
Chapter 8: Electron Configurations and Periodicity	
Chapter 8 Test	
Assignments	Notes
Week 13, Day 1	
☐ 8.1.3 Electron Configurations through Neon	
☐ 8.1.4 Electron Configurations beyond Neon	
☐ 8.1.5 Periodic Relationships	
Week 13, Day 2	
☐ 8.2.1 Periods and Atomic Size	
☐ 8.2.2 Ionization Energy	
☐ 8.2.3 Electron Affinity	
Week 13, Day 3	
☐ 8.2.4 An Introduction to Electronegativity	
☐ 8.3.1 Hydrogen, Alkali Metals and Alkaline Earth Metals	
☐ 8.3.2 Transition Metals and Nonmetals	
Week 13, Day 4	
☐ Chapter 8 Practice Test	
Week 13, Day 5	Chapter 8 Test
☐ Chapter 8 Test	Score:

Week 14	
Chapter 9: Chemical Bonding: Fundamental Concepts	
Chapter 9 Test	
Assignments	Notes
Week 14, Day 1	
9.1.1 Valence Electrons and Chemical Bonding	
☐ 9.1.2 Ionic Bonds	
☐ 9.1.3 CIA Demonstration: Conductivity Apparatus-Ionic versus	
Covalent Bonds	
Week 14, Day 2	
☐ 9.2.1 Lewis Dot Structures for Covalent Bonds	
☐ 9.2.2 Predicting Lewis Dot Structures	
☐ 9.3.1 Resonance Structures	
☐ 9.3.2 Formal Charge	
Week 14, Day 3	
9.3.3 Electronegativity, Formal Charge, and Resonance	
☐ 9.4.1 Bond Properties	
☐ 9.4.2 Using Bond Dissociation Energies	
Week 14, Day 4	
☐ Chapter 9 Practice Test	
Week 14, Day 5	Chapter 9 Test
☐ Chapter 9 Test	Score:

Week 15	
Chapter 10: Molecular Geometry and Bonding Theory	
Assignments	Notes
Week 15, Day 1	
☐ 10.1.1 Valence-Shell Electron-Pair Repulsion Theory	
☐ 10.1.2 Molecular Shapes for Steric Numbers 2-4	
Week 15, Day 2	
☐ 10.1.3 Molecular Shapes for Steric Numbers 5 & 6	
☐ 10.1.4 Predicting Molecular Characteristics Using VSEPR Theory	
Week 15, Day 3	
☐ 10.2.1 Valence Bond Theory	
☐ 10.2.2 An Introduction to Hybrid Orbitals	
Week 15, Day 4	
☐ 10.2.3 Pi Bonds	
☐ 10.2.4 Molecular Orbital Theory	
☐ 10.2.5 Applications of the Molecular Orbital Theory	
Week 15, Day 5	
☐ 10.2.6 Beyond Homonuclear Diatomics	
☐ 10.2.7 CIA Demonstration: The Paramagnetism of Oxygen	
Week 16	
Chapter 10 Test	
Chapter 11: Oxidation-Reduction Reactions	
Assignments	Notes
Week 16, Day 1	
☐ Chapter 10 Practice Test	
Week 16, Day 2	Chapter 10 Test
☐ Chapter 10 Test	Score:
Week 16, Day 3	
☐ 11.1.1 Oxidation Numbers	
Week 16, Day 4	
☐ 11.1.2 Balancing Redox Reactions by the Oxidation Number	
Method	
☐ 11.1.3 Balancing Redox Reactions Using the Half-Reaction	
Method	
Week 16, Day 5	

☐ 11.1.4 The Activity Series of the Elements

 \square 11.1.5 CIA Demonstration: The Reaction between Al and Br₂

Week 17	
Chapter 11 Test	
Chapter 12: Condensed Phases: Liquids and Solids	
Assignments	Notes
Week 17, Day 1	
☐ Chapter 11 Practice Test	
	Chapter 11 Test
☐ Chapter 11 Test	Score:
Week 17, Day 3	
☐ 12.1.1 An Introduction to Intermolecular Forces and States of	
Matter	
☐ 12.1.2 Intermolecular Forces	
☐ 12.2.1 Properties of Liquids	
Week 17, Day 4	
☐ 12.2.2 CIA Demonstration: Boiling Water at Reduced Pressure	
☐ 12.2.3 Vapor Pressure and Boiling Point	
☐ 12.2.4 Molecular Structure and Boiling Point	
Week 17, Day 5	
☐ 12.2.5 Phase Diagrams	
☐ 12.2.6 CIA Demonstration: Boiling Water in a Paper Cup	
☐ 12.3.1 Types of Solids	

Week 18	
Chapter 12: Condensed Phases: Liquids and Solids	
Chapter 12 Test	
Assignments	Notes
Week 18, Day 1	
☐ 12.3.2 CIA Demonstration: The Conductivity of Molten Salts	
☐ 12.3.3 Crystal Structure	
Week 18, Day 2	
☐ 12.3.4 Calculating Atomic Mass and Radius from a Unit Cell	
☐ 12.3.5 Crystal Packing	
Week 18, Day 3	
☐ 12.4.1 Ceramics and Glass	
☐ 12.4.2 CIA Demonstration: Superconductivity	
Week 18, Day 4	
☐ Chapter 12 Practice Test	
Week 18, Day 5	Chapter 12 Test
☐ Chapter 12 Test	Score:

Week 19	
Chapter 13: Physical Properties of Solutions	
Assignments	Notes
Week 19, Day 1	
☐ 13.1.1 Types of Solutions	
☐ 13.1.2 Molarity and the Mole Fraction	
☐ 13.1.3 Molality	
Week 19, Day 2	
☐ 13.1.4 Energy and the Solution Process	
☐ 13.2.1 Temperature Change and Solubility	
☐ 13.2.2 Extractions	
Week 19, Day 3	
☐ 13.2.3 Pressure Change and Solubility	
☐ 13.3.1 Vapor Pressure Lowering	
☐ 13.3.2 Boiling Point Elevation and Freezing Point Depression	
Week 19, Day 4	
☐ 13.3.4 Osmosis	
☐ 13.3.5 Colligative Properties of Ionic Solutions	
Week 19, Day 5	
☐ 13.4.1 Colloid Formation and Flocculation	
☐ 13.4.2 CIA Demonstration: The Tyndall Effect	

Week 20	
Chapter 13 Test	
Midterm Exam	
Assignments	Notes
Week 20, Day 1	
☐ Chapter 13 Practice Test	
Week 20, Day 2	Chapter 13 Test
☐ Chapter 13 Test	Score:
Week 20, Day 3	
☐ Study for Midterm Exam	
Week 20, Day 4	
☐ Study for Midterm Exam	
Week 20, Day 5	Midterm Exam
☐ Midterm Exam	Score:

Week 21	
Chapter 14: Chemical Kinetics	
Assignments	Notes
Week 21, Day 1	
☐ 14.1.1 An Introduction to Reaction Rates	
☐ 14.1.2 Rate Laws: How the Reaction Rate Depends on	
Concentration	
☐ 14.1.3 Determining the Form of a Rate Law	
<u>Week 21, Day 2</u>	
14.2.1 First-Order Reactions	
☐ 14.2.2 Second-Order Reactions	
☐ 14.2.3 A Kinetics Problem	
<u>Week 21, Day 3</u>	
☐ 14.3.1 The Collision Model	
☐ 14.3.2 The Arrhenius Equation	
☐ 14.3.3 Using the Arrhenius Equation	
Week 21, Day 4	
☐ 14.4.1 Defining the Molecularity of a Reaction	
☐ 14.4.2 Determining the Rate Laws of Elementary Reactions	
☐ 14.4.3 Calculating the Rate Laws of Multistep Reactions	
Week 21, Day 5	
☐ 14.4.4 Steady State Kinetics	
☐ 14.5.1 Catalysts and Types of Catalysts	
☐ 14.5.2 A Word About Laboratory Safety	
	1
Week 22	
Chapter 14: Chemical Kinetics	
Chapter 14 Test	
Chapter 15: Chemical Equilibrium	
Assignments	Notes
Week 22, Day 1	
☐ 14.5.3 CIA Demonstration: Elephant Snot	
☐ 14.5.4 CIA Demonstration: The Cobalt(II)-Catalyzed Reaction of	
Potassium Sodium Tartrate	
☐ 14.5.5 CIA Demonstration: The Copper-Catalyzed	
Decomposition of Acetone	
Week 22, Day 2	
☐ Chapter 14 Practice Test	
Week 22, Day 3	Chapter 14 Test
☐ Chapter 14 Test	Score:
Week 22, Day 4	
☐ 15.1.1 The Concept of Equilibrium	
☐ 15.1.2 The Law of Mass Action and Types of Equilibrium	
☐ 15.1.3 Converting Between K ₂ and K ₂	

Week 22, Day 5	
☐ 15.2.1 Approaching Chemical Equilibrium	
☐ 15.2.2 Predicting the Direction of a Reaction	
☐ 15.2.3 Strategies for Solving Equilibrium Problems	
<u> </u>	
Week 23	
Chapter 15: Chemical Equilibrium	
Chapter 15 Test	
Assignments	Notes
Week 23, Day 1	
☐ 15.2.4 Solving Problems Far from Equilibrium	
☐ 15.2.5 An Equilibrium Problem Using the Quadratic Equation	
Week 23, Day 2	
☐ 15.3.1 Le Châtelier's Principle	
☐ 15.3.2 The Effect of Changing Amounts on Equilibrium	
☐ 15.3.3 The Effect of Pressure and Volume on Equilibrium	
Week 23, Day 3	
☐ 15.3.4 The Effects of Temperature and Catalysts on Equilibrium	
☐ 15.3.5 CIA Demonstration: NO ₂ /N ₂ O ₄	
☐ 15.3.6 CIA Demonstration: Shifting the Equilibrium of FeSCN ²⁺	
Week 23, Day 4	
☐ Chapter 15 Practice Test	
1	Charles AF Table
Week 23, Day 5	Chapter 15 Test
Week 23, Day 5 ☐ Chapter 15 Test	Score:
	1
	1
☐ Chapter 15 Test	1
Chapter 15 Test Week 24 Chapter 16: Acids and Bases Assignments	1
Chapter 15 Test Week 24 Chapter 16: Acids and Bases Assignments Week 24, Day 1	Score:
Chapter 15 Test Week 24 Chapter 16: Acids and Bases Assignments Week 24, Day 1 □ 16.1.1 Arrhenius/Brønsted-Lowry Definitions of Acids and Bases	Score:
Chapter 15 Test Week 24 Chapter 16: Acids and Bases Assignments Week 24, Day 1 □ 16.1.1 Arrhenius/Brønsted-Lowry Definitions of Acids and Bases □ 16.1.2 Hydronium, Hydroxide, and the pH Scale	Score:
Chapter 15 Test Week 24 Chapter 16: Acids and Bases Assignments Week 24, Day 1 □ 16.1.1 Arrhenius/Brønsted-Lowry Definitions of Acids and Bases □ 16.1.2 Hydronium, Hydroxide, and the pH Scale Week 24, Day 2	Score:
Chapter 15 Test Week 24 Chapter 16: Acids and Bases Assignments Week 24, Day 1 □ 16.1.1 Arrhenius/Brønsted-Lowry Definitions of Acids and Bases □ 16.1.2 Hydronium, Hydroxide, and the pH Scale Week 24, Day 2 □ 16.2.1 Strong Acids and Bases	Score:
Chapter 15 Test Week 24 Chapter 16: Acids and Bases Assignments Week 24, Day 1 □ 16.1.1 Arrhenius/Brønsted-Lowry Definitions of Acids and Bases □ 16.1.2 Hydronium, Hydroxide, and the pH Scale Week 24, Day 2 □ 16.2.1 Strong Acids and Bases □ 16.2.2 CIA Demonstration: Natural Acid-Base Indicators	Score:
Chapter 15 Test Week 24 Chapter 16: Acids and Bases Assignments Week 24, Day 1 □ 16.1.1 Arrhenius/Brønsted-Lowry Definitions of Acids and Bases □ 16.1.2 Hydronium, Hydroxide, and the pH Scale Week 24, Day 2 □ 16.2.1 Strong Acids and Bases □ 16.2.2 CIA Demonstration: Natural Acid-Base Indicators □ 16.2.3 Weak Acids	Score:
Chapter 15 Test Week 24 Chapter 16: Acids and Bases Assignments Week 24, Day 1 □ 16.1.1 Arrhenius/Brønsted-Lowry Definitions of Acids and Bases □ 16.1.2 Hydronium, Hydroxide, and the pH Scale Week 24, Day 2 □ 16.2.1 Strong Acids and Bases □ 16.2.2 CIA Demonstration: Natural Acid-Base Indicators □ 16.2.3 Weak Acids Week 24, Day 3	Score:
Chapter 15 Test Week 24 Chapter 16: Acids and Bases Assignments Week 24, Day 1 □ 16.1.1 Arrhenius/Brønsted-Lowry Definitions of Acids and Bases □ 16.1.2 Hydronium, Hydroxide, and the pH Scale Week 24, Day 2 □ 16.2.1 Strong Acids and Bases □ 16.2.2 CIA Demonstration: Natural Acid-Base Indicators □ 16.2.3 Weak Acids Week 24, Day 3 □ 16.2.4 Weak Bases	Score:
Chapter 15 Test Week 24 Chapter 16: Acids and Bases Assignments Week 24, Day 1 □ 16.1.1 Arrhenius/Brønsted-Lowry Definitions of Acids and Bases □ 16.1.2 Hydronium, Hydroxide, and the pH Scale Week 24, Day 2 □ 16.2.1 Strong Acids and Bases □ 16.2.2 CIA Demonstration: Natural Acid-Base Indicators □ 16.2.3 Weak Acids Week 24, Day 3 □ 16.2.4 Weak Bases □ 16.2.5 Lewis Acids and Bases	Score:
Chapter 15 Test Week 24 Chapter 16: Acids and Bases Assignments Week 24, Day 1 □ 16.1.1 Arrhenius/Brønsted-Lowry Definitions of Acids and Bases □ 16.1.2 Hydronium, Hydroxide, and the pH Scale Week 24, Day 2 □ 16.2.1 Strong Acids and Bases □ 16.2.2 CIA Demonstration: Natural Acid-Base Indicators □ 16.2.3 Weak Acids Week 24, Day 3 □ 16.2.4 Weak Bases □ 16.2.5 Lewis Acids and Bases Strengths	Score:
Chapter 15 Test Week 24 Chapter 16: Acids and Bases Assignments Week 24, Day 1 16.1.1 Arrhenius/Brønsted-Lowry Definitions of Acids and Bases 16.1.2 Hydronium, Hydroxide, and the pH Scale Week 24, Day 2 16.2.1 Strong Acids and Bases 16.2.2 CIA Demonstration: Natural Acid-Base Indicators 16.2.3 Weak Acids Week 24, Day 3 16.2.4 Weak Bases 16.2.5 Lewis Acids and Bases 16.2.6 Trends in Acid and Base Strengths Week 24, Day 4	Score:
Chapter 15 Test Week 24 Chapter 16: Acids and Bases Assignments Week 24, Day 1 □ 16.1.1 Arrhenius/Brønsted-Lowry Definitions of Acids and Bases □ 16.1.2 Hydronium, Hydroxide, and the pH Scale Week 24, Day 2 □ 16.2.1 Strong Acids and Bases □ 16.2.2 CIA Demonstration: Natural Acid-Base Indicators □ 16.2.3 Weak Acids Week 24, Day 3 □ 16.2.4 Weak Bases □ 16.2.5 Lewis Acids and Bases □ 16.2.6 Trends in Acid and Base Strengths Week 24, Day 4 □ 16.3.1 Examining Polyprotic Acids	Score:
Week 24 Chapter 16: Acids and Bases Assignments Week 24, Day 1 □ 16.1.1 Arrhenius/Brønsted-Lowry Definitions of Acids and Bases □ 16.1.2 Hydronium, Hydroxide, and the pH Scale Week 24, Day 2 □ 16.2.1 Strong Acids and Bases □ 16.2.2 CIA Demonstration: Natural Acid-Base Indicators □ 16.2.3 Weak Acids Week 24, Day 3 □ 16.2.4 Weak Bases □ 16.2.5 Lewis Acids and Bases □ 16.2.6 Trends in Acid and Base Strengths Week 24, Day 4 □ 16.3.1 Examining Polyprotic Acids □ 16.4.1 Acid-Base Properties of Salt Solutions	Score:
Chapter 15 Test Week 24 Chapter 16: Acids and Bases Assignments Week 24, Day 1 □ 16.1.1 Arrhenius/Brønsted-Lowry Definitions of Acids and Bases □ 16.1.2 Hydronium, Hydroxide, and the pH Scale Week 24, Day 2 □ 16.2.1 Strong Acids and Bases □ 16.2.2 CIA Demonstration: Natural Acid-Base Indicators □ 16.2.3 Weak Acids Week 24, Day 3 □ 16.2.4 Weak Bases □ 16.2.5 Lewis Acids and Bases □ 16.2.6 Trends in Acid and Base Strengths Week 24, Day 4 □ 16.3.1 Examining Polyprotic Acids	Score:

Week 25	
Chapter 16 Test	
Chapter 17: Equilibrium in Aqueous Solution	
Assignments	Notes
Week 25, Day 1	Chapter 16 Test
☐ Chapter 16 Test	Score:
Week 25, Day 2	
☐ 17.1.1 Strong Acid-Strong Base and Weak Acid-Strong Base	
Reactions	
☐ 17.1.2 Strong Acid-Weak Base and Weak Acid-Weak Base	
Reactions	
☐ 17.1.3 The Common Ion Effect	
Week 25, Day 3	
☐ 17.2.1 An Introduction to Buffers	
☐ 17.2.2 CIA Demonstration: Buffers in Action	
☐ 17.2.3 Acidic Buffers	
Week 25, Day 4	
☐ 17.2.4 Basic Buffers	
☐ 17.2.5 The Henderson-Hasselbalch Equation	
Week 25, Day 5	
☐ 17.3.1 Strong Acid-Strong Base Titration	
☐ 17.3.2 CIA Demonstration: Barium Hydroxide-Sulfuric Acid	
Titration	
☐ 17.3.3 Weak Acid-Strong Base Titration	
[1
Week 26	
Chapter 17: Equilibrium in Aqueous Solution	
Assignments	Notes
Week 26, Day 1	
☐ 17.3.4 Polyprotic Acid-Strong Base Titration	
☐ 17.3.5 Weak Base-Strong Acid Titration	
☐ 17.3.6 Acid-Base Indicators	
Week 26, Day 2	
☐ 17.4.1 The Solubility Product Constant	
☐ 17.4.2 CIA Demonstration: Silver Chloride and Ammonia	
☐ 17.4.3 Solubility and the Common Ion Effect	
Week 26, Day 3	
☐ 17.4.4 Fractional Precipitation	
☐ 17.4.5 The Effects of pH on Solubility	
☐ 17.5.1 The Formation of Complex Ions	
☐ 17.5.2 Amphoteric Metal Hydroxides	
Week 26, Day 4	
☐ Chapter 17 Practice Test	Chantan 17 Tark
Week 26, Day 5	Chapter 17 Test Score:
☐ Chapter 17 Test	JC01 E

Week 27	
Chapter 18: Introduction to Organic Reactions	
Assignments	Notes
Week 27, Day 1	
☐ 18.1.1 An Introduction to Reactivity	
☐ 18.1.2 Bond Strengths	
☐ 18.1.3 Inductive Effects	
Week 27, Day 2	
☐ 18.1.4 Hybridization Effects	
☐ 18.1.5 Resonance Effects	
☐ 18.1.6 Solvent Effects: Acid Dissociation versus Proton Affinity	
Week 27, Day 3	
☐ 18.2.1 A Review of Relationship between Acids and Conjugate	
Bases	
☐ 18.2.2 Strengths of Organic Bases	
☐ 18.2.3 Solvent Effects on Organic Base Strength	
Week 27, Day 4	
☐ 18.3.1 Lewis Acids and the Formation of Acid-Base Adducts	
☐ 18.3.2 Oxides as Lewis Acids	
Week 27, Day 5	
☐ 18.4.1 Nucleophilic Substitution at sp³ Carbon	
☐ 18.4.2 Nucleophilic Substitution at sp ² Carbon	
Week 28	
Chapter 18: Introduction to Organic Reactions	
Chapter 18 Test	
Chapter 19: Thermodynamics	
Assignments	Notes
Week 28, Day 1	
☐ 18.4.3 Elimination Reactions	
☐ 18.4.4 CIA Demonstration: Slime	
Week 28, Day 2	
☐ Chapter 18 Practice Test	
Week 28, Day 3	Chapter 18
☐ Chapter 18 Test	Test Score:
Week 28, Day 4	
☐ 19.1.1 Spontaneous Processes	
☐ 19.2.1 Entropy and the Second Law of Thermodynamics	
☐ 19.2.2 Entropy and Temperature	
Week 28, Day 5	
☐ 19.3.1 Gibbs Free Energy	
☐ 19.3.2 Standard Free Energy Changes of Formation	

Week 29 Chapter 19 Test Chapter 20: Electrochemistry Assignments Notes Week 29, Day 1 19.4.1 Enthalpy and Entropy Contributions to K 19.4.2 The Temperature Dependence of K 19.4.3 Free Energy Away from Equilibrium Week 29, Day 2 Chapter 19 Practice Test Week 29, Day 3 Chapter 19 Test Chapter 19 Test Score: Week 29, Day 4 20.1.1 Reviewing Oxidation-Reduction Reactions 20.2.1 Electrochemical Cells Week 29, Day 5 20.2.2 Electromotive Force 20.2.3 Standard Reduction Potentials 20.2.4 Using Standard Reduction Potentials		
Chapter 19 Test Chapter 20: Electrochemistry Assignments Notes Week 29, Day 1 19.4.1 Enthalpy and Entropy Contributions to K 19.4.2 The Temperature Dependence of K 19.4.3 Free Energy Away from Equilibrium Week 29, Day 2 Chapter 19 Practice Test Week 29, Day 3 Chapter 19 Test Score: Week 29, Day 4 20.1.1 Reviewing Oxidation-Reduction Reactions 20.2.1 Electrochemical Cells Week 29, Day 5 20.2.2 Electromotive Force 20.2.3 Standard Reduction Potentials	Week 29	
Chapter 20: Electrochemistry Assignments Notes Week 29, Day 1 19.4.1 Enthalpy and Entropy Contributions to K 19.4.2 The Temperature Dependence of K 19.4.3 Free Energy Away from Equilibrium Week 29, Day 2 Chapter 19 Practice Test Week 29, Day 3 Chapter 19 Test Score: Week 29, Day 4 20.1.1 Reviewing Oxidation-Reduction Reactions 20.2.1 Electrochemical Cells Week 29, Day 5 20.2.2 Electromotive Force 20.2.3 Standard Reduction Potentials	Chapter 19: Thermodynamics	
Assignments Week 29, Day 1 19.4.1 Enthalpy and Entropy Contributions to K 19.4.2 The Temperature Dependence of K 19.4.3 Free Energy Away from Equilibrium Week 29, Day 2 Chapter 19 Practice Test Week 29, Day 3 Chapter 19 Test Score: Week 29, Day 4 20.1.1 Reviewing Oxidation-Reduction Reactions 20.2.1 Electrochemical Cells Week 29, Day 5 20.2.2 Electromotive Force 20.2.3 Standard Reduction Potentials	Chapter 19 Test	
Week 29, Day 1 □ 19.4.1 Enthalpy and Entropy Contributions to K □ 19.4.2 The Temperature Dependence of K □ 19.4.3 Free Energy Away from Equilibrium Week 29, Day 2 □ Chapter 19 Practice Test Week 29, Day 3 Chapter 19 Test □ Chapter 19 Test Score: Week 29, Day 4 Score: □ 20.1.1 Reviewing Oxidation-Reduction Reactions 20.2.1 Electrochemical Cells Week 29, Day 5 20.2.2 Electromotive Force □ 20.2.3 Standard Reduction Potentials	Chapter 20: Electrochemistry	
□ 19.4.1 Enthalpy and Entropy Contributions to K □ 19.4.2 The Temperature Dependence of K □ 19.4.3 Free Energy Away from Equilibrium Week 29, Day 2 □ Chapter 19 Practice Test Week 29, Day 3 □ Chapter 19 Test Chapter 19 Test Score: Week 29, Day 4 □ 20.1.1 Reviewing Oxidation-Reduction Reactions □ 20.2.1 Electrochemical Cells Week 29, Day 5 □ 20.2.2 Electromotive Force □ 20.2.3 Standard Reduction Potentials	Assignments	Notes
□ 19.4.2 The Temperature Dependence of K □ 19.4.3 Free Energy Away from Equilibrium Week 29, Day 2 □ Chapter 19 Practice Test Week 29, Day 3 Chapter 19 Test □ Chapter 19 Test Score: Week 29, Day 4 □ 20.1.1 Reviewing Oxidation-Reduction Reactions □ 20.2.1 Electrochemical Cells Week 29, Day 5 □ 20.2.2 Electromotive Force □ 20.2.3 Standard Reduction Potentials	Week 29, Day 1	
□ 19.4.3 Free Energy Away from Equilibrium Week 29, Day 2 □ Chapter 19 Practice Test Week 29, Day 3 □ Chapter 19 Test Score: □ Week 29, Day 4 □ 20.1.1 Reviewing Oxidation-Reduction Reactions □ 20.2.1 Electrochemical Cells Week 29, Day 5 □ 20.2.2 Electromotive Force □ 20.2.3 Standard Reduction Potentials	☐ 19.4.1 Enthalpy and Entropy Contributions to K	
Week 29, Day 2 ☐ Chapter 19 Practice Test Week 29, Day 3 ☐ Chapter 19 Test Score: Week 29, Day 4 ☐ 20.1.1 Reviewing Oxidation-Reduction Reactions ☐ 20.2.1 Electrochemical Cells Week 29, Day 5 ☐ 20.2.2 Electromotive Force ☐ 20.2.3 Standard Reduction Potentials	☐ 19.4.2 The Temperature Dependence of K	
☐ Chapter 19 Practice Test Week 29, Day 3 ☐ Chapter 19 Test Score: Week 29, Day 4 ☐ 20.1.1 Reviewing Oxidation-Reduction Reactions ☐ 20.2.1 Electrochemical Cells Week 29, Day 5 ☐ 20.2.2 Electromotive Force ☐ 20.2.3 Standard Reduction Potentials	☐ 19.4.3 Free Energy Away from Equilibrium	
Week 29, Day 3 Chapter 19 Test Chapter 19 Test Score:	Week 29, Day 2	
Chapter 19 Test Week 29, Day 4 20.1.1 Reviewing Oxidation-Reduction Reactions 20.2.1 Electrochemical Cells Week 29, Day 5 20.2.2 Electromotive Force 20.2.3 Standard Reduction Potentials	☐ Chapter 19 Practice Test	
Week 29, Day 4 □ 20.1.1 Reviewing Oxidation-Reduction Reactions □ 20.2.1 Electrochemical Cells Week 29, Day 5 □ 20.2.2 Electromotive Force □ 20.2.3 Standard Reduction Potentials	Week 29, Day 3	Chapter 19 Test
□ 20.1.1 Reviewing Oxidation-Reduction Reactions □ 20.2.1 Electrochemical Cells Week 29, Day 5 □ 20.2.2 Electromotive Force □ 20.2.3 Standard Reduction Potentials	☐ Chapter 19 Test	Score:
□ 20.2.1 Electrochemical Cells Week 29, Day 5 □ 20.2.2 Electromotive Force □ 20.2.3 Standard Reduction Potentials	Week 29, Day 4	
Week 29, Day 5 □ 20.2.2 Electromotive Force □ 20.2.3 Standard Reduction Potentials	☐ 20.1.1 Reviewing Oxidation-Reduction Reactions	
□ 20.2.2 Electromotive Force □ 20.2.3 Standard Reduction Potentials	☐ 20.2.1 Electrochemical Cells	
□ 20.2.3 Standard Reduction Potentials	Week 29, Day 5	
	☐ 20.2.2 Electromotive Force	
☐ 20.2.4 Using Standard Reduction Potentials	☐ 20.2.3 Standard Reduction Potentials	
	☐ 20.2.4 Using Standard Reduction Potentials	

Week 30	
Chapter 20: Electrochemistry	
Chapter 20 Test	
Assignments	Notes
Week 30, Day 1	
☐ 20.2.5 The Nernst Equation	
☐ 20.2.6 Electrochemical Determinants of Equilibria	
Week 30, Day 2	
☐ 20.3.1 Batteries	
☐ 20.3.2 CIA Demonstration: The Fruit-Powered Clock	
☐ 20.4.1 Corrosion and the Prevention of Corrosion	
Week 30, Day 3	
☐ 20.5.1 Electrolytic Cells	
☐ 20.5.2 The Stoichiometry of Electrolysis	
Week 30, Day 4	
☐ Chapter 20 Practice Test	
Week 30, Day 5	Chapter 20 Test
☐ Chapter 20 Test	Score:

Week 31	
Chapter 21: Nuclear Chemistry	
Chapter 21 Test	
Assignments	Notes
Week 31, Day 1	
☐ 21.1.1 The Nature of Radioactivity	
☐ 21.1.2 The Stability of Atomic Nuclei	
☐ 21.1.3 Binding Energy	
Week 31, Day 2	
☐ 21.2.1 Rates of Disintegration Reactions	
☐ 21.2.2 Radiochemical Dating	
Week 31, Day 3	
☐ 21.3.1 Nuclear Fission	
☐ 21.3.2 Nuclear Fusion	
☐ 21.3.3 Applications of Nuclear Chemistry	
Week 31, Day 4	
☐ Chapter 21 Practice Test	
Week 31, Day 5	Chapter 21 Test
☐ Chapter 21 Test	Score:

Week 32	
Chapter 22: Chemistry of Metals	
Chapter 22 Test	
Assignments	Notes
Week 32, Day 1	
☐ 22.1.1 Metallurgical Processes	
☐ 22.1.2 Band Theory of Conductivity	
☐ 22.1.3 Intrinsic Semiconductors	
Week 32, Day 2	
☐ 22.1.4 Doped Semiconductors	
☐ 22.2.1 The Alkali Metals	
Week 32, Day 3	
☐ 22.2.2 The Alkaline Earth Metals	
☐ 22.2.3 Aluminum	
☐ 22.2.4 CIA Demonstration: The Reaction between Al and Br ₂	
Week 32, Day 4	
☐ Chapter 22 Practice Test	
Week 32, Day 5	Chapter 22 Test
☐ Chapter 22 Test	Score:

Week 33	
Chapter 23: Transition Metals	
Chapter 23 Test	
Assignments	Notes
Week 33, Day 1	
☐ 23.1.1 Properties of Transition Metals	
☐ 23.1.2 CIA Demonstration: Copper One-Pot Reactions	
☐ 23.2.1 Complexes and Ligands	
Week 33, Day 2	
☐ 23.2.2 Naming Coordination Compounds	
☐ 23.2.3 Structures of Coordination Compounds and Isomers	
☐ 23.3.1 Color and Transition Metals	
Week 33, Day 3	
☐ 23.3.2 Crystal Field Theory	
☐ 23.3.3 Ligand Field Theory	
☐ 23.3.4 Magnetic Properties and Spin	
Week 33, Day 4	
☐ Chapter 23 Practice Test	
Week 33, Day 5	Chapter 23 Test
☐ Chapter 23 Test	Score:
Week 34	
Chapter 24: Nonmetals	
Assignments	Notes
Week 34, Day 1	
☐ 24.1.1 General Properties of Nonmetals	
☐ 24.1.2 Hydrogen	
Week 34, Day 2	
☐ 24.2.1 General Properties of Carbon	
☐ 24.2.2 Silicon	
Week 34, Day 3	
☐ 24.3.1 Nitrogen	
☐ 24.3.2 Phosphorus	
Week 34 Day 4	

☐ 24.4.1 Oxygen

☐ 24.4.3 Sulfur

☐ 24.5.1 Halogens

Week 34, Day 5

☐ 24.4.2 CIA Demonstration: Creating Acid Rain

□ 24.5.2 Aqueous Halogen Compounds□ 24.6.1 Properties of Noble Gases

Week 35	
Chapter 24 Test	
Chapter 25: Organic Chemistry	
Assignments	Notes
Week 35, Day 1	
☐ Chapter 24 Practice Test	
Week 35, Day 2	Chapter 24 Test
☐ Chapter 24 Test	Score:
Week 35, Day 3	
☐ 25.1.1 Alkanes	
☐ 25.1.2 Alkenes and Alkynes	
Week 35, Day 4	
☐ 25.1.3 Isomers	
☐ 25.1.4 Aromatic Hydrocarbons	
Week 35, Day 5	
☐ 25.2.1 Alcohols, Ethers, and Amines	
☐ 25.2.2 Carbonyl-Containing Functional Groups	

Week 36	
Chapter 25: Organic Chemistry	
Chapter 25 Test	
Chapter 26: Biochemistry	
Assignments	Notes
Week 36, Day 1	
☐ 25.3.1 Organic Polymers	
☐ 25.3.2 CIA Demonstration: The Synthesis of Nylon	
Week 36, Day 2	
☐ Chapter 25 Practice Test	
Week 36, Day 3	Chapter 25 Test
☐ Chapter 25 Test	Score:
Week 36, Day 4	
☐ 26.1.1 Proteins	
☐ 26.1.2 Nucleic Acids	
Week 36, Day 5	
☐ 26.1.3 Carbohydrates	
☐ 26.1.4 Lipids	

Week 37	
Chapter 26 Test	
Final Exam	
Assignments	Notes
Week 37, Day 1	
☐ Chapter 26 Practice Test	
Week 37, Day 2	Chapter 26 Test
☐ Chapter 26 Test	Score:
Week 37, Day 3	
☐ Study for Final Exam	
Week 37, Day 4	
☐ Study for Final Exam	
Week 37, Day 5	Final Exam
☐ Final Exam	Score: